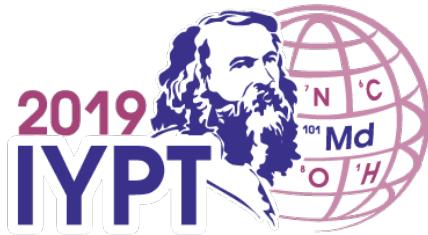


# Theoretical issues in physics of SHE : nuclear reaction perspectives



International Year  
of the Periodic Table  
of Chemical Elements

Kouichi Hagino  
Kyoto University

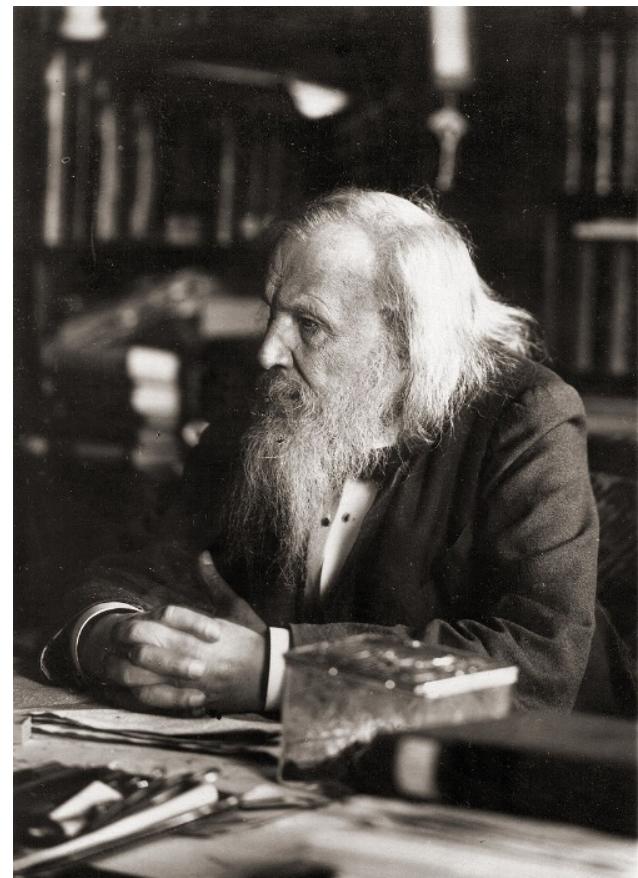


1. Introduction-1: periodic tables
2. Introduction-2: superheavy elements
3. Formation reactions
4. Theoretical issues
5. Physics of neutron-rich nuclei
6. Summary

# Periodic table of elemetns (1869)



International Year  
of the Periodic Table  
of Chemical Elements



Mendereev  
(1834-1907)

# periodic table of elements

noble  
gas

	Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
↓	Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1		1 H																		
2		3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	2 He	
3		11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	10 Ne	
4		19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	18 Ar	
5		37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6		55 Cs	56 Ba	57 La	*	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7		87 Fr	88 Ra	89 Ac	*	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
		*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
		*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				

# periodic table of elements

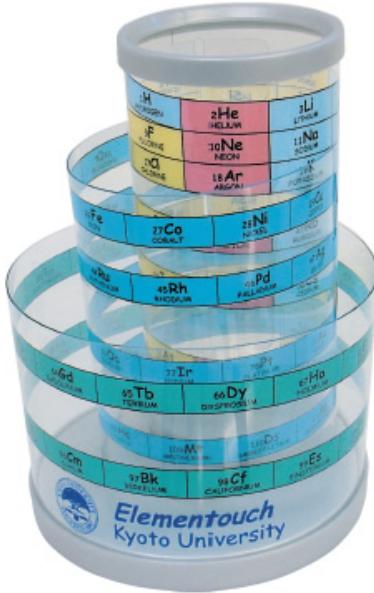
noble  
gas

	Group → 1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
↓ Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	1 H																	2 He	
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne	
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	57 La	*	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	89 Ac	*	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
	*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
	*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				



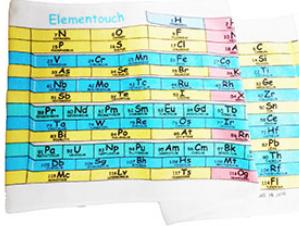
Prof. Yoshiteru Maeno (Kyoto U., cond. matt. expt.)

# periodic table of elements



mug cup

3D periodic table “elementouch”  
(Y. Maeno, 2001)



towel



T-shirt  
(Kyoto-U. coop)



Prof. Yoshiteru Maeno (Kyoto U., cond. matt. expt.)

# periodic table of elements-nuclei?

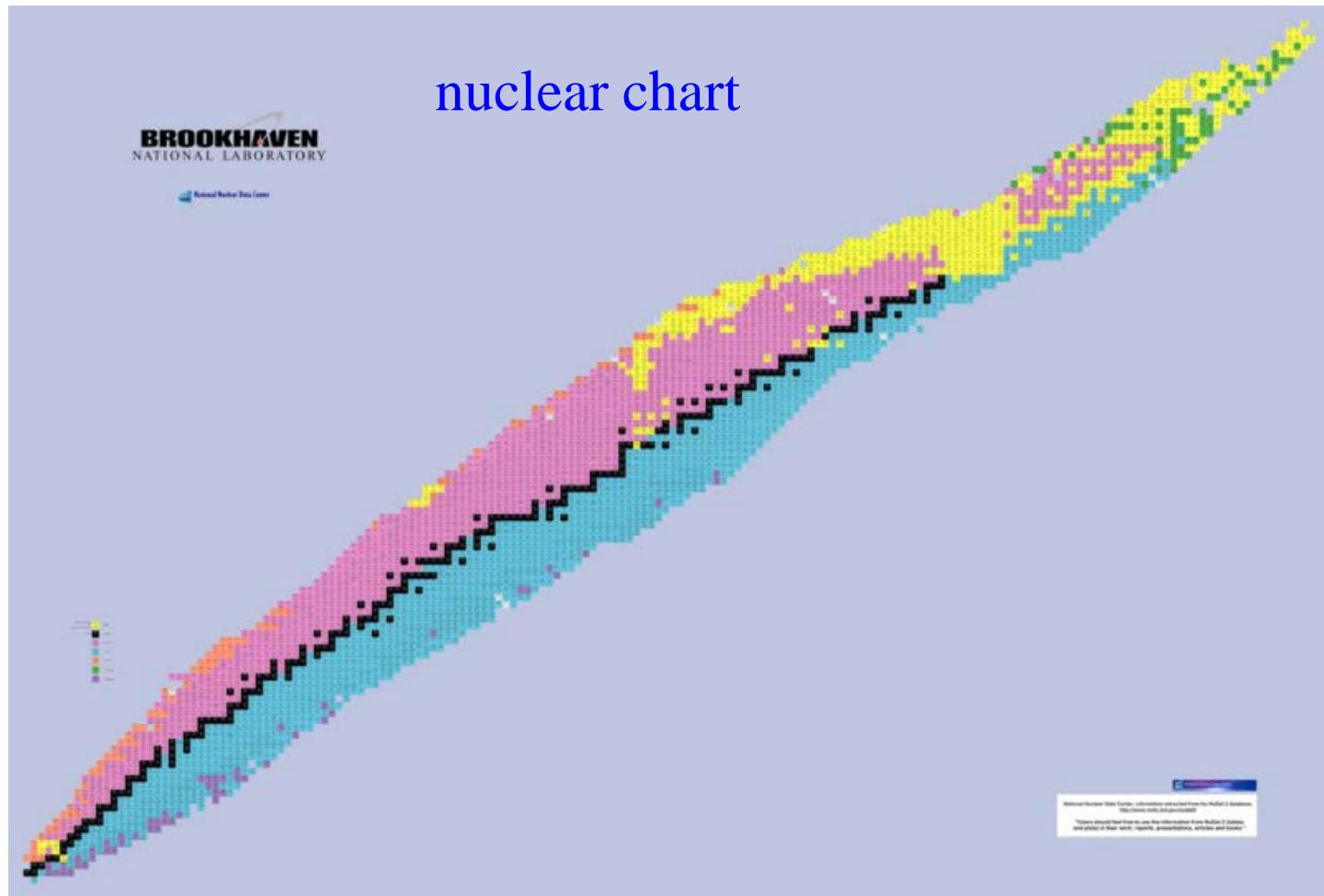
proton  
magic # ← noble  
gas

Group → 1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
↓ Period																		
1	1 H																2 He	
2	3 Li	4 Be										5 B	6 C	7 N	8 O	9 F	10 Ne	
3	11 Na	12 Mg										13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
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7	87 Fr	88 Ra	89 Ac	*	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts
	*	*	*	*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
	*	*	*	*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr



Prof. Yoshiteru Maeno (Kyoto U., cond. matt. expt.)

# Nuclear periodic table

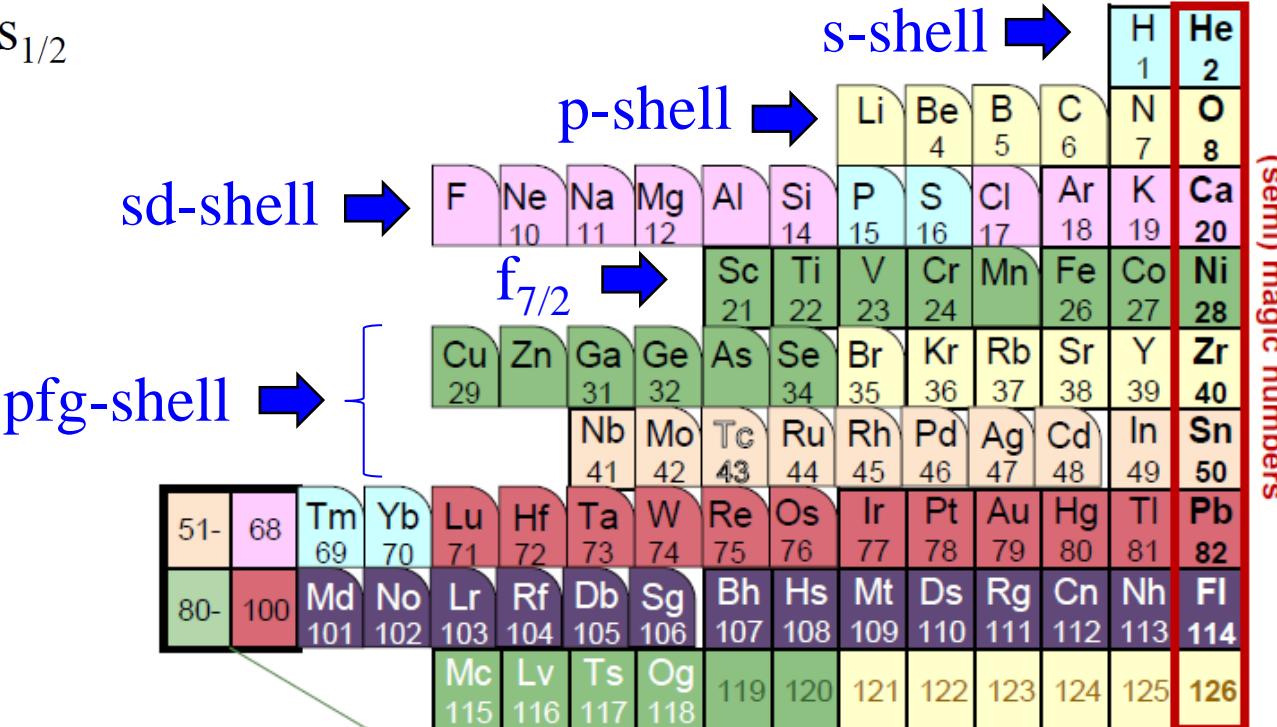
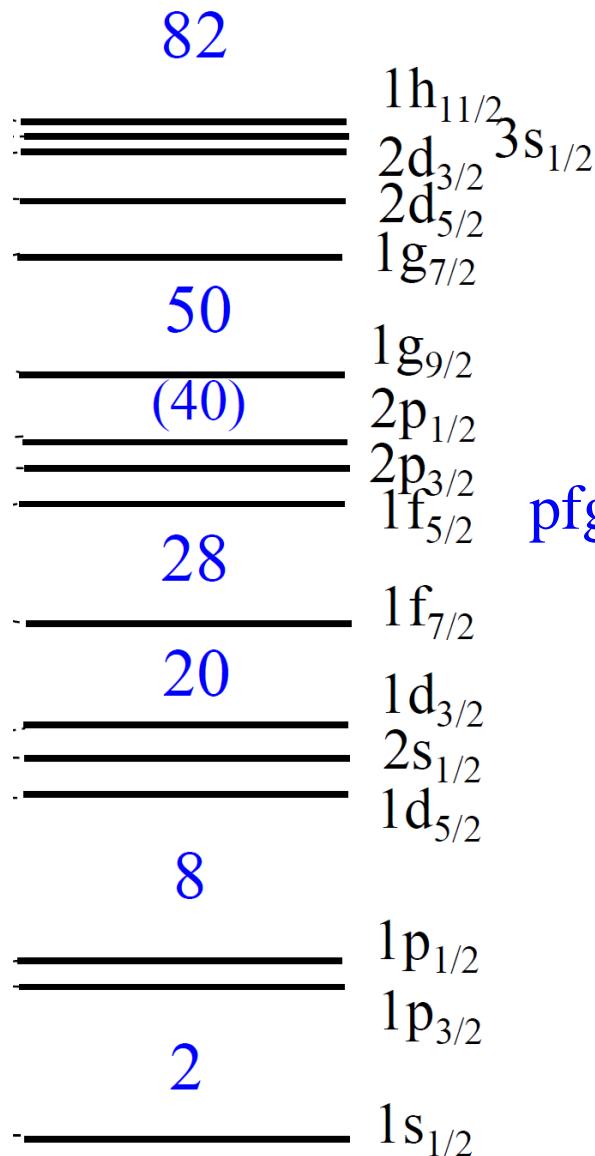


Yet, a pedagogical significance  
(to familiarize nuclear physics)

# Nuclear periodic table

K. Hagino and Y. Maeno,  
Found. of Chem. 22, 267 (2020).

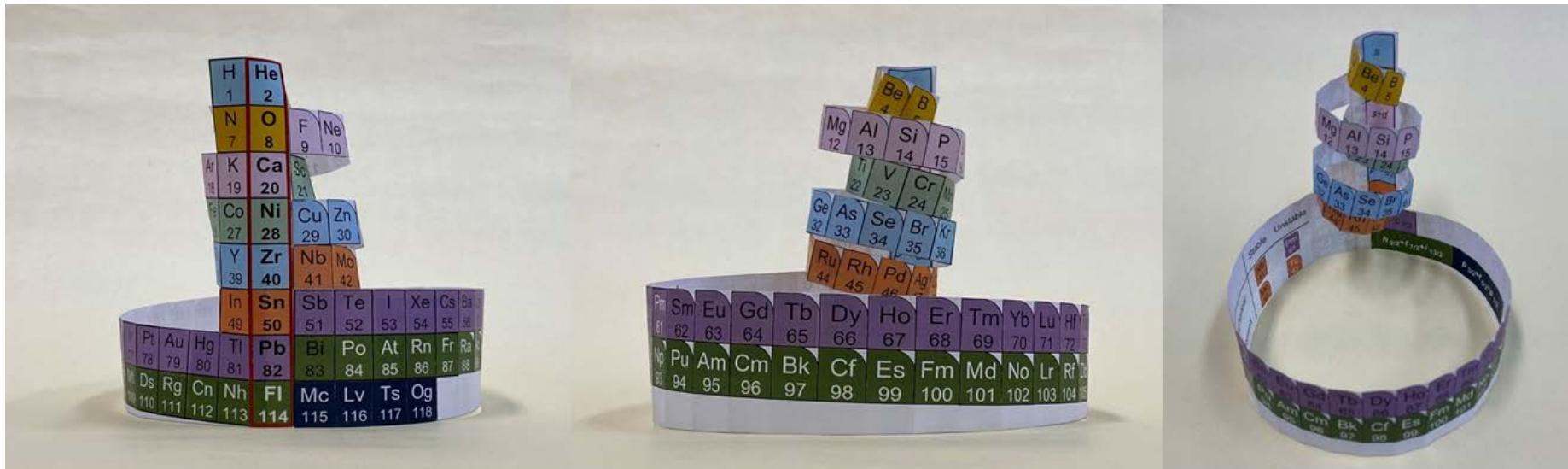
proton  
magic



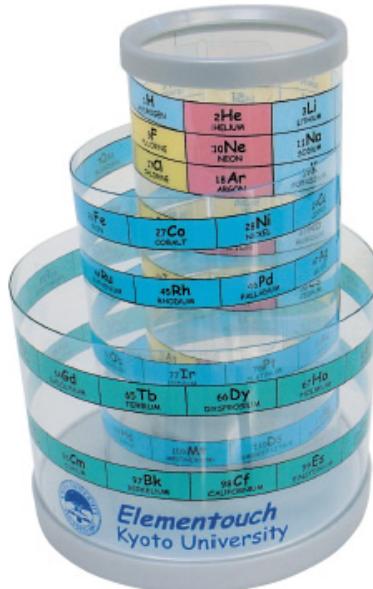
I	Xe	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er
53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68
At	Rn	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm
85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100

# Nuclear periodic table

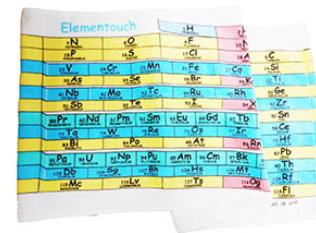
“nucletouch” (a 3D model)



cf. “elementouch”  
(Y. Maeno, 2001)



mug cup



towel



T-shirt  
(Kyoto-U. coop)

[家](#) > 科学研究 > 化学材料 > 掌握元素性质的新工具！京都大学创建原子核周期表

## 掌握元素性质的新工具！京都大学创建原子核周期表

2020年05月11日

化学材料

# 日本經濟新聞

朝刊



ストーリー



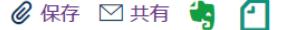
Myニュース 日

トップ 速報 マネー 経済・金融 政治 ビジネス マーケット テクノロジー

### 新しい周期表を考案 京大、原子核の性質を表現

2020/5/3付 | 日本経済新聞 朝刊

保存 共有



京都大学の前野悦輝教授と萩野浩一教授は、原子核の性質をわかりやすく示す新しい周期表を考案した。従来の周期表が元素の化学的な性質を知るのに役立つ一方で、新しい周期表は元素の原子核の性質を知るために使える。

NEWS RELEASE 27-MAY-2020

## A special elemental magic

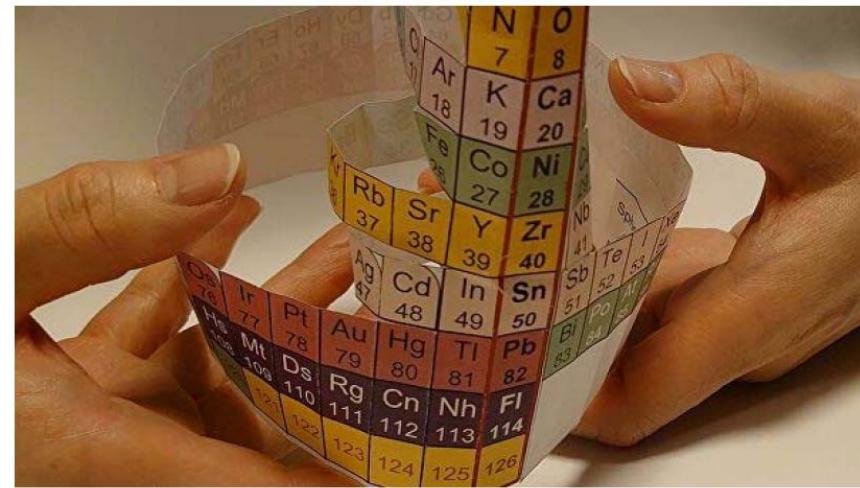
Kyoto scientists announce a 'nuclear' periodic table

KYOTO UNIVERSITY

### Японские физики представили новую периодическую таблицу элементов

19:34 27.05.2020

58122



© Фото : Kyoto University/Yoshiteru Maeno/Kouichi Hagino

МОСКВА, 27 мая — РИА Новости. Ученые из Кюотского университета представили периодическую таблицу элементов, которая в отличие от таблицы Менделеева, где за основу взяты электроны в атоме, основана на

# a magical coincidence

Y. Maeno, K. Hagino, and T. Ishiguro,  
Found. of Chem., in press.

periodic table of elements

5 B	6 C	7 N	8 O	9 F	10 Ne
13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As
46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb
78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi
110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc
64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm
96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md
				102 No	103 Lr

the nuclear magic numbers are there  
in the same column!  
(the same elements around them)

nuclear periodic table

2 He	1 H	2 He	Li	Be	B
	C	N 7	O 8	F	Ne 10
P 15	S 16	Cl 17	Ar 18	K 19	Ca 20
		Mn	Fe 26	Co 27	Ni 28
Br 35	Kr 36	Rb 37	Sr 38	Y 39	Zr 40
	Pd 49	Ag 50	Cd 51	In 52	Sn 53
I 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82
M 101	Ds 110	Rg 111	Cn 112	Nh 113	Fl 114
				Mc 115	Lv 116
				Ts 117	Og 118

a magical coincidence which  
can be recognized only after  
making a nuclear periodic table

# Superheavy elements and Neutron-rich Nuclei



International Year  
of the Periodic Table  
of Chemical Elements

1	1 H
2	3 Li      4 Be
3	11 Na    12 Mg
4	19 K     20 Ca    21 Sc    22 Ti    23 V    24 Cr    25 Mn    26 Fe    27 Co    28 Ni    29 Cu    30 Zn    31 Ga
5	37 Rb    38 Sr    39 Y    40 Zr    41 Nb    42 Mo    43 Tc    44 Ru    45 Rh    46 Pd    47 Ag    48 Cd    49 In
6	55 Cs    56 Ba    57 La    * 72 Hf    73 Ta    74 W    75 Re    76 Os    77 Ir    78 Pt    79 Au    80 Hg    81 Tl Ph II    Br II    Po II    At II    Bn II
7	87 Fr    88 Ra    89 Ac    104 Rf    105 Db    106 Sg    107 Bh    108 Hs    109 Mt    110 Ds    111 Rg    112 Cr    113 Nh    114 Fl    115 Mc    116 Lv    117 Ts    118 Og

\* 58 Ce   59 Pr   60 Nd   61 Pm   62 Sm   63 Eu   64 Gd   65 Tb   66 Dy   67 Ho   68 Er   69 Tm   70 Yb   71 Lu

\* 90 Th   91 Pa   92 U   93 Np   94 Pu   95 Am   96 Cm   97 Bk   98 Cf   99 Es   100 Fm   101 Md   102 No   103 Lr

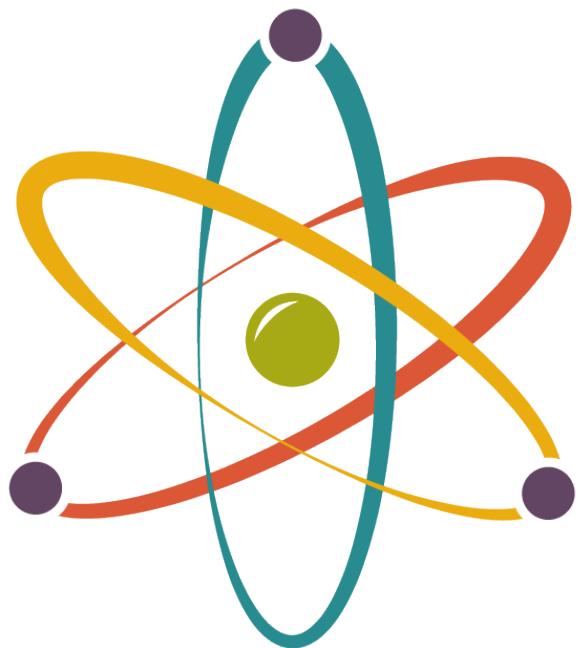
superheavy elements = trans-actinides  
(elements heavier than Z=104)

1	1 H																2 He	
2	3 Li	4 Be															10 Ne	
3	11 Na	12 Mg															18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	53 Te	54 Xe	
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7	87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
	*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
	*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			

## Interests in physics and chemistry of superheavy elements

- what is the heaviest element? → atomic property
- what is the double magic nucleus next to  $^{208}\text{Pb}$ ? → nuclear property
- should the periodic table be changed or not? → chemical property
- how do superheavy elements influence the r-process nucleosynthesis?  
→ astrophysics

what determines the limit of existence of elements?



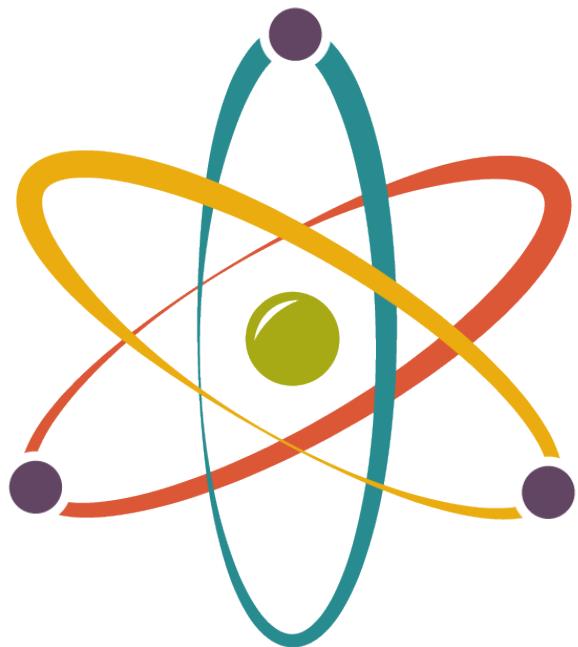
# INTERNATIONAL YEAR OF THE PERIODIC TABLE **2019**



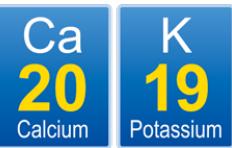
possibilities to be considered:

- ✓ electron orbitals in atom
- ✓ stability of nucleus in atom (← magic number)

what determines the limit of existence of elements?



# INTERNATIONAL YEAR OF THE PERIODIC TABLE **2019**



possibilities to be considered:

let's first discuss

- ✓ electron orbitals in atom
- ✓ stability of nucleus in atom (← magic number)

what determines the limit of existence of elements? (i) electron orbits

hydrogen-like atom

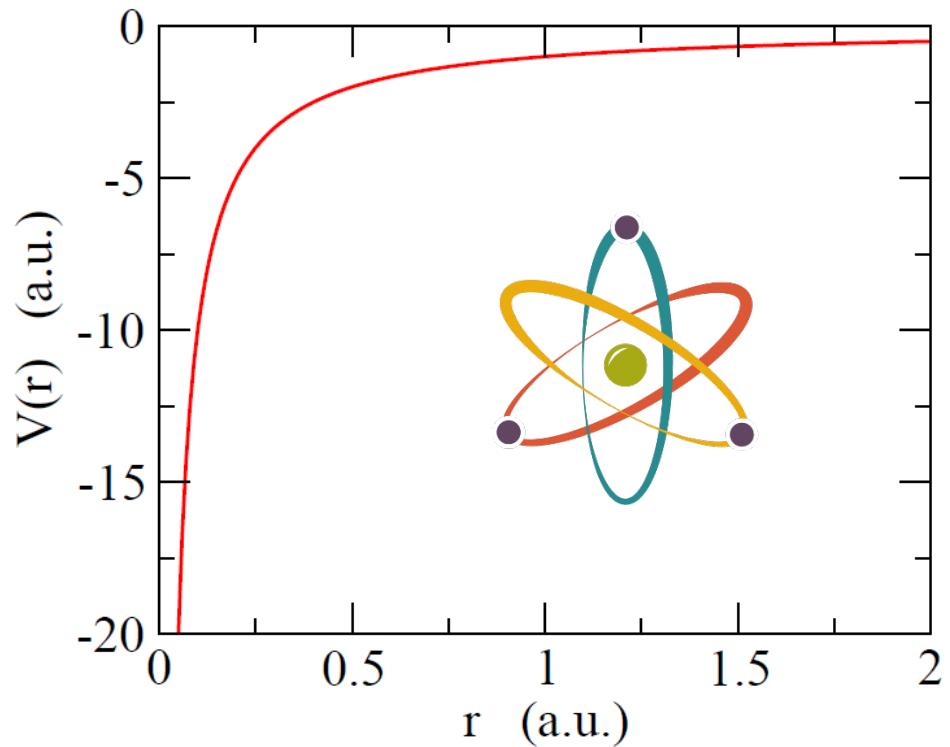
$$V(r) = -\frac{Ze^2}{r}$$

1S state

$$E_{1s} = -Z^2 \cdot \frac{me^4}{2\hbar^2}$$

$Z \rightarrow$ large,  $E_{1s} \rightarrow$ small

$\langle r \rangle \rightarrow$  small



## what determines the limit of existence of elements? (i) electron orbits

hydrogen-like atom

$$V(r) = -\frac{Ze^2}{r}$$

1S state

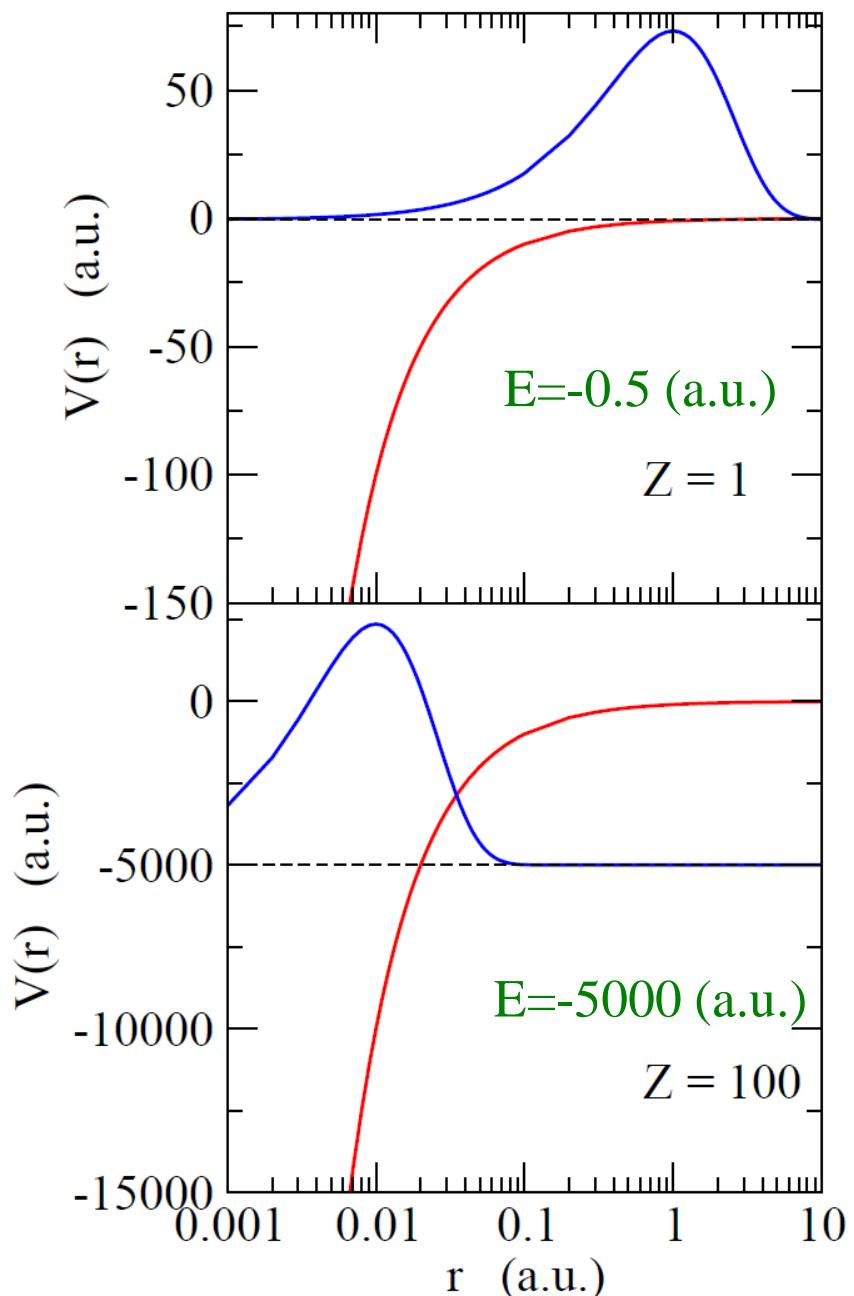
$$E_{1S} = -Z^2 \cdot \frac{me^4}{2\hbar^2}$$

$Z \rightarrow$ large,  $E_{1s} \rightarrow$ small

$\langle r \rangle \rightarrow$  small

$\langle p \rangle \rightarrow$  large (uncertainty)

→ relativistic effect



what determines the limit of existence of elements? (i) electron orbits

hydrogen-like atom

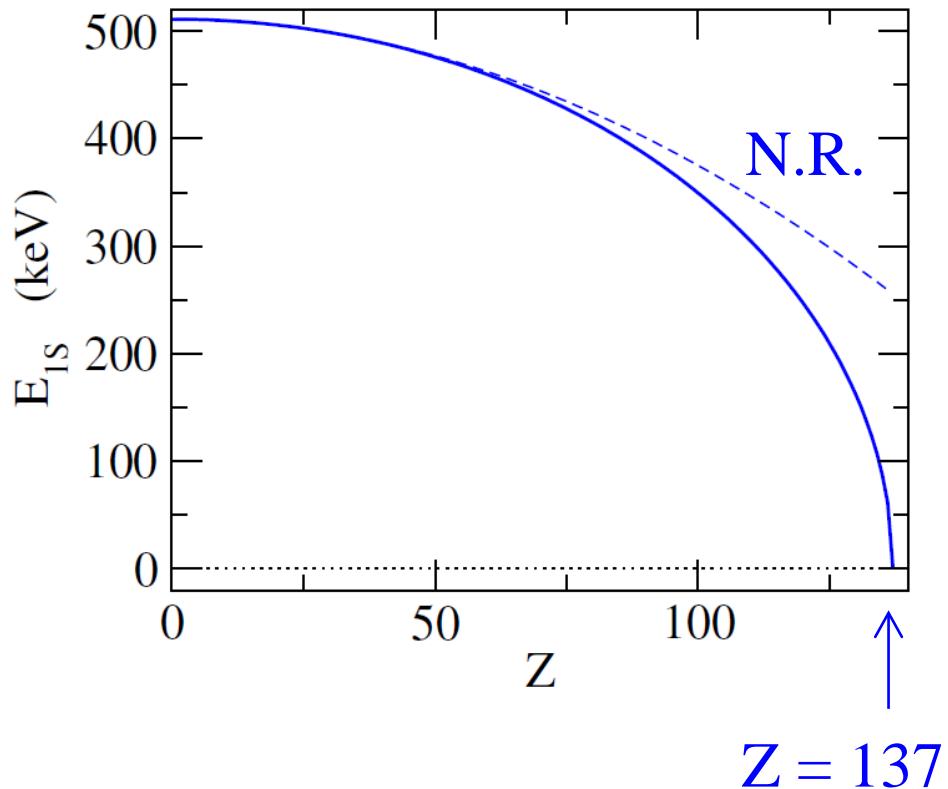
$$V(r) = -\frac{Ze^2}{r}$$

1S state (Dirac equation)

$$E_{1S} = mc^2 \sqrt{1 - (Z\alpha)^2}$$

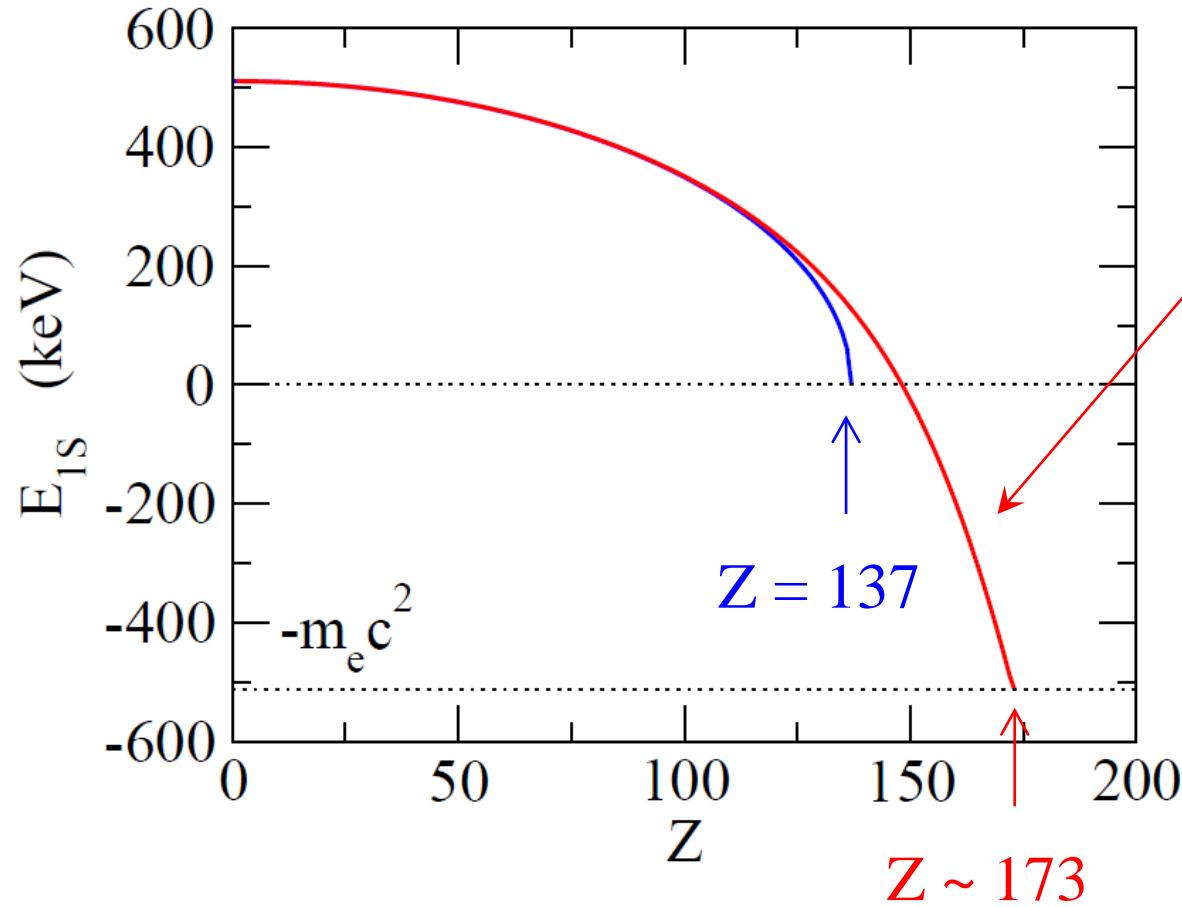
$$\alpha = \frac{e^2}{\hbar c} \sim \frac{1}{137}$$

$Z > 137 \rightarrow$  no solution



## what determines the limit of existence of elements? (i) electron orbits

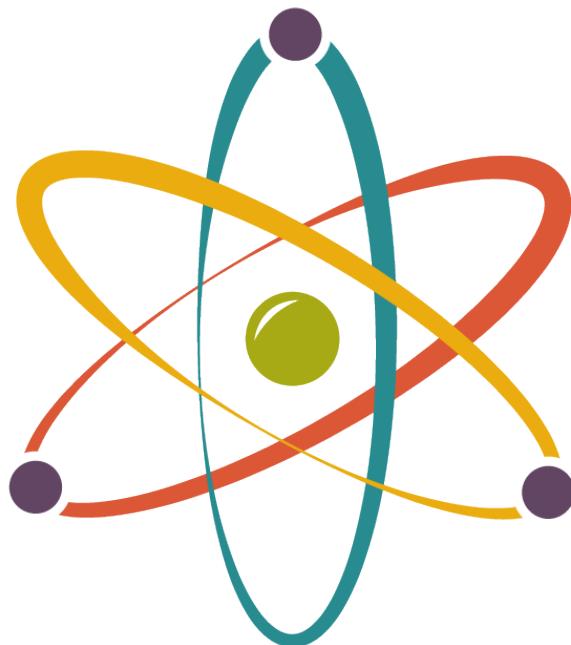
hydrogen-like atom



finite size of a  
nucleus  
(uniform charge)

$$R_N \sim 10^{-4} \text{ \AA}$$
$$\text{rms} \sim 10^{-3} \text{ \AA}$$
$$(A=173)$$

what determines the limit of existence of elements?



# INTERNATIONAL YEAR OF THE PERIODIC TABLE **2019**

I 53 Iodine	Y 39 Yttrium	Pt 78 Platinum	Ca 20 Calcium	K 19 Potassium
-------------------	--------------------	----------------------	---------------------	----------------------

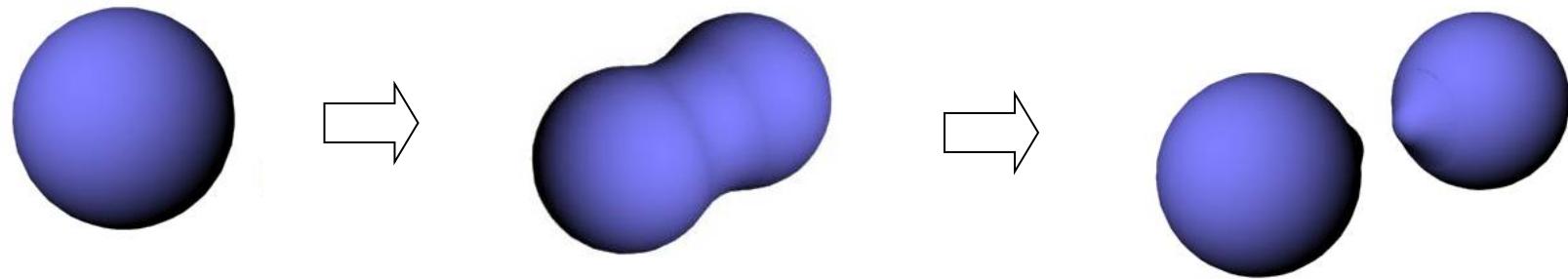
possibilities to be considered:

let's next discuss

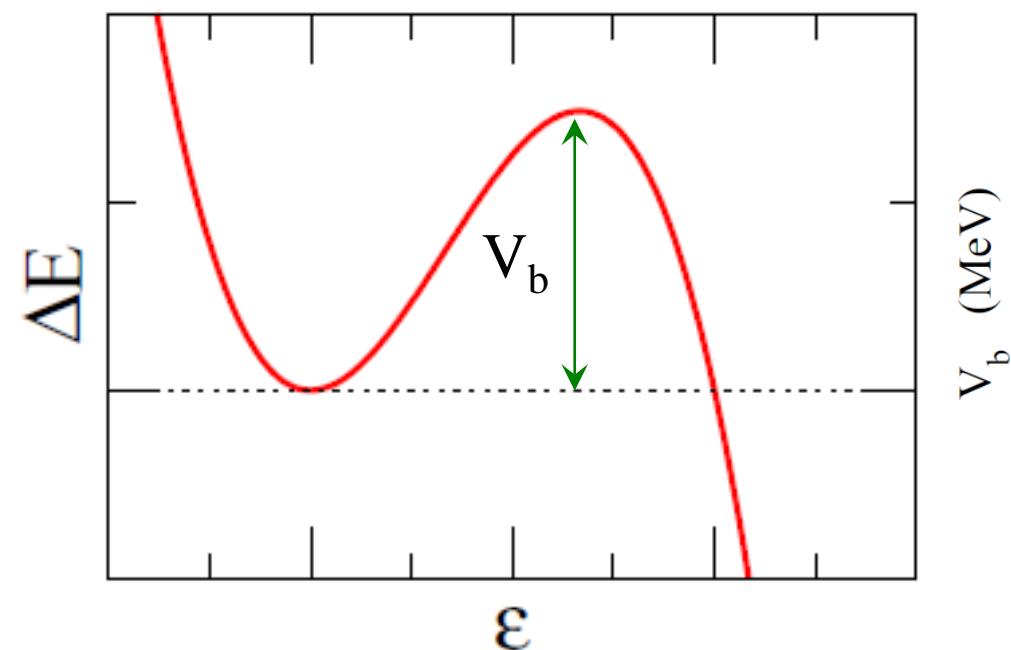
- ✓ electronic orbitals in atom
- ✓ stability of nucleus in atom

## what determines the limit of existence of elements? (ii) atomic nucleus

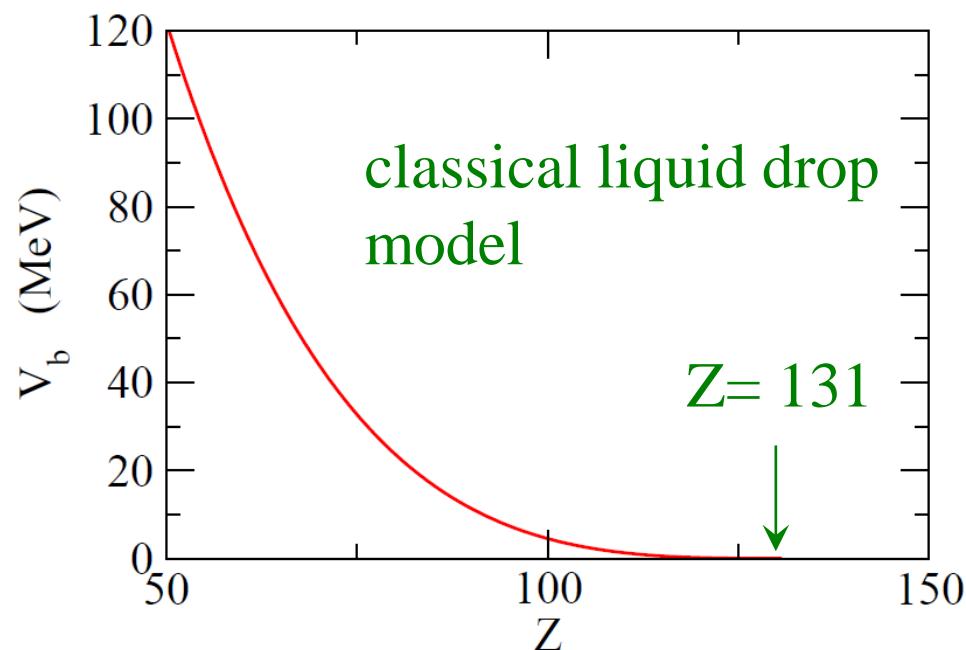
heavy nuclei → unstable against fission



fission barrier



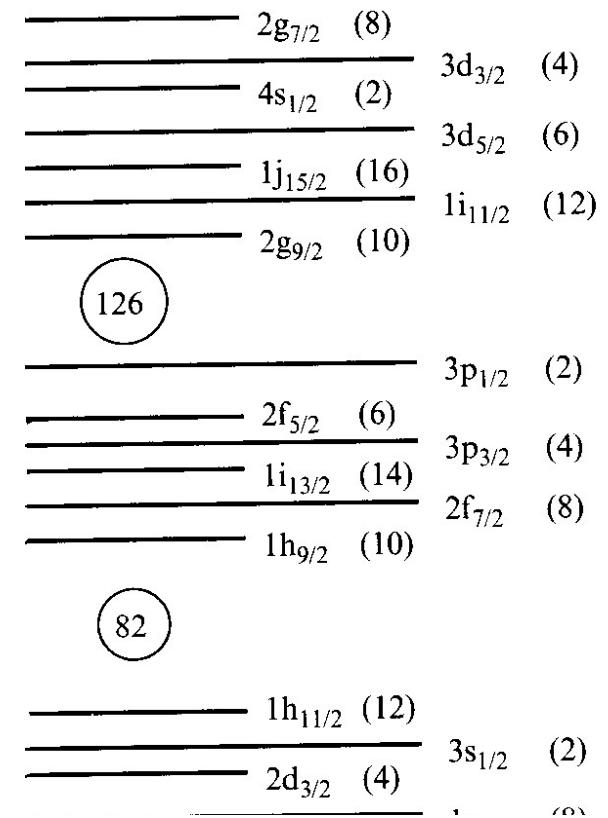
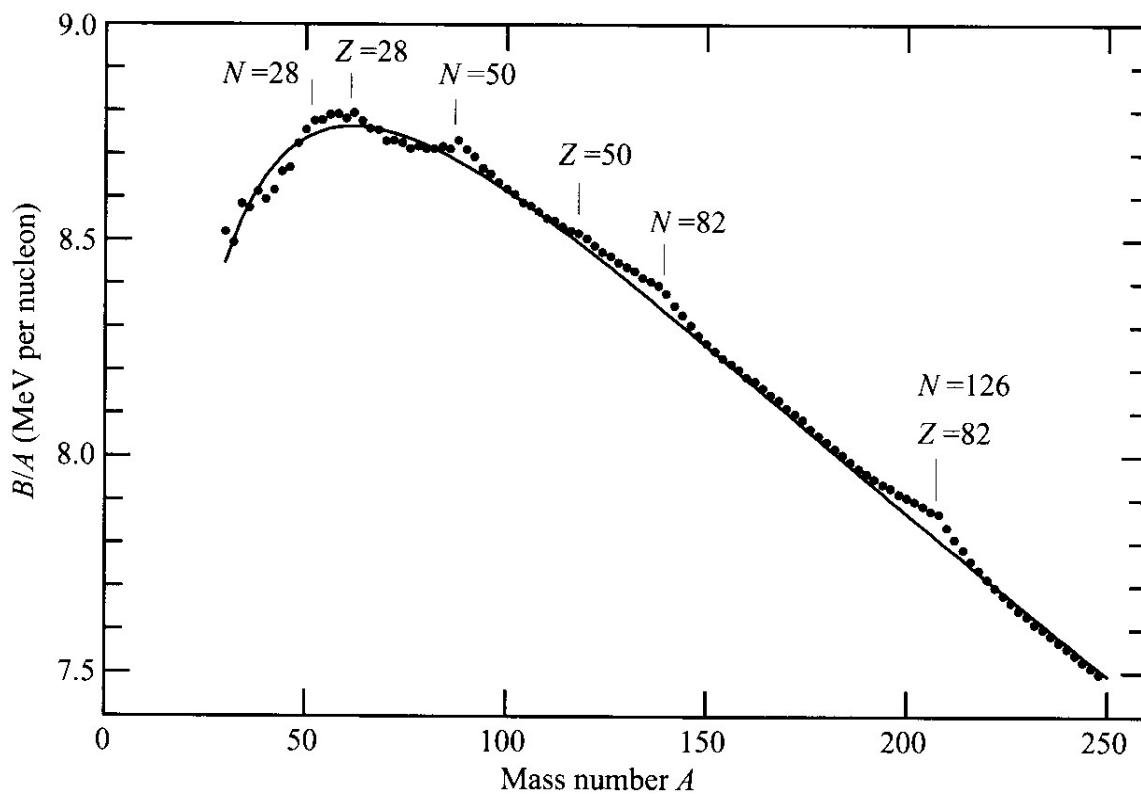
barrier height

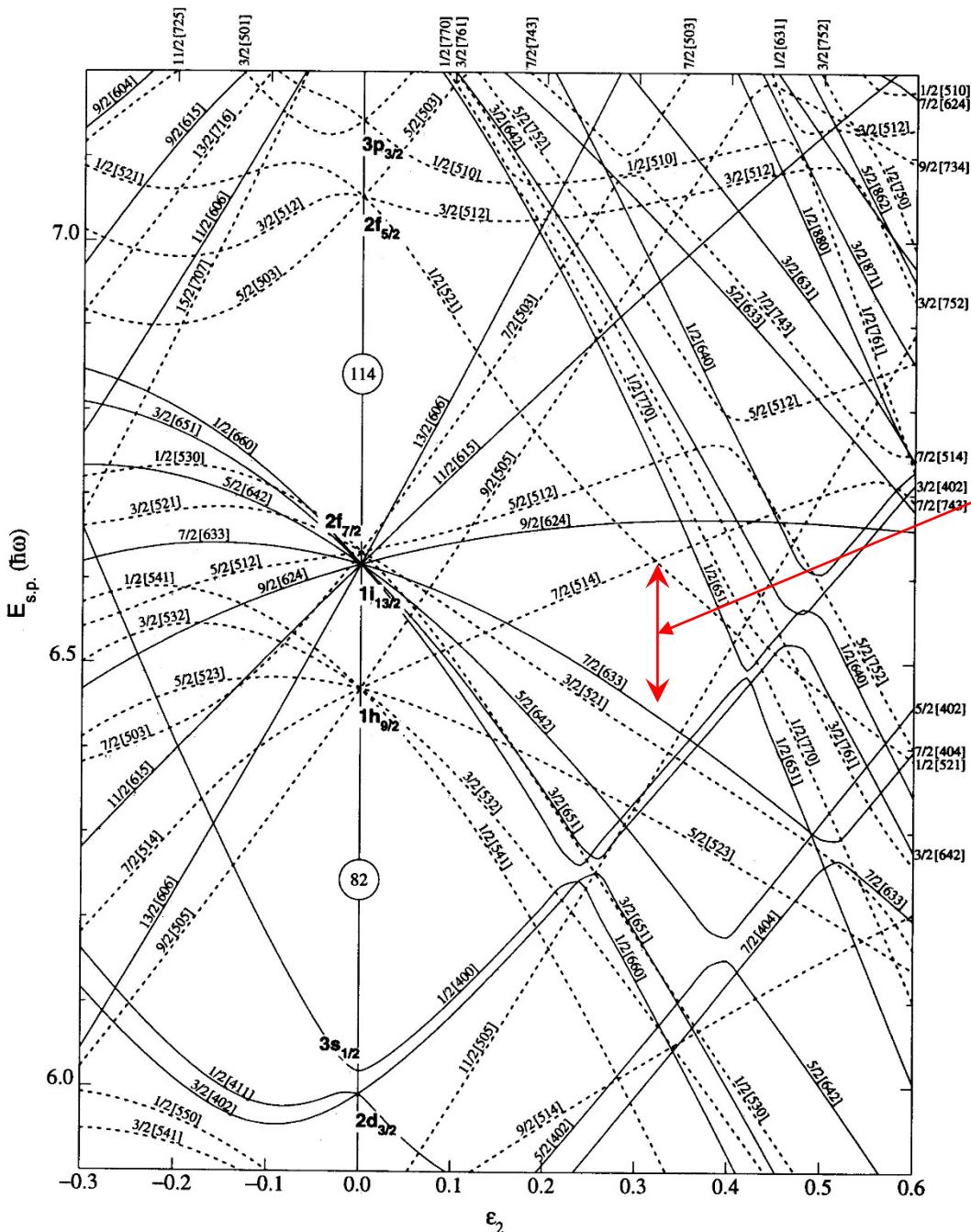


classical liquid drop model

## shell effect

$$B = B_{\text{LDM}} + B_{\text{shell}}$$



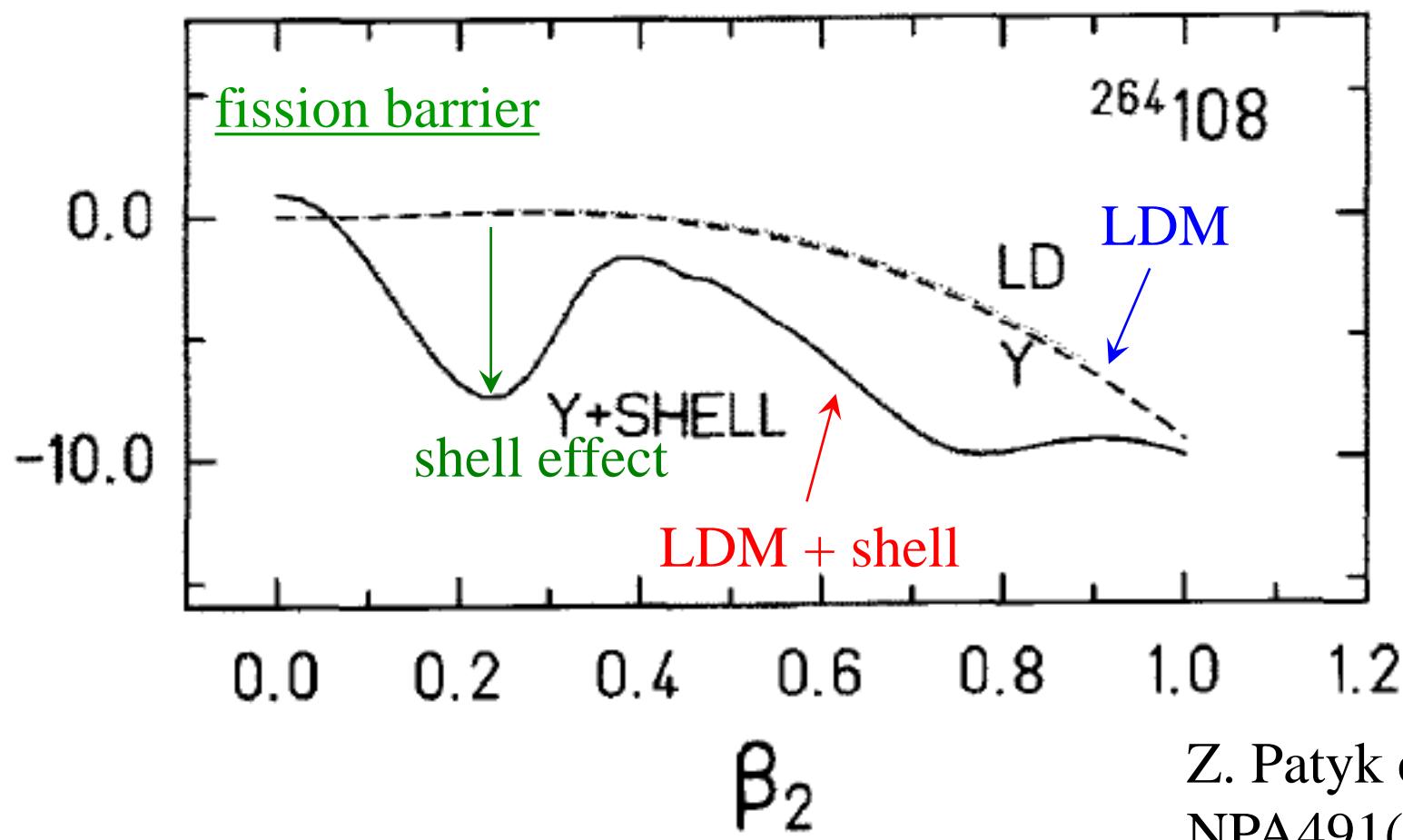


Nilsson diagram

energy gap if deformed

Figure 13. Nilsson diagram for protons,  $Z \geq 82$  ( $\epsilon_4 = \epsilon_2^2/6$ ).

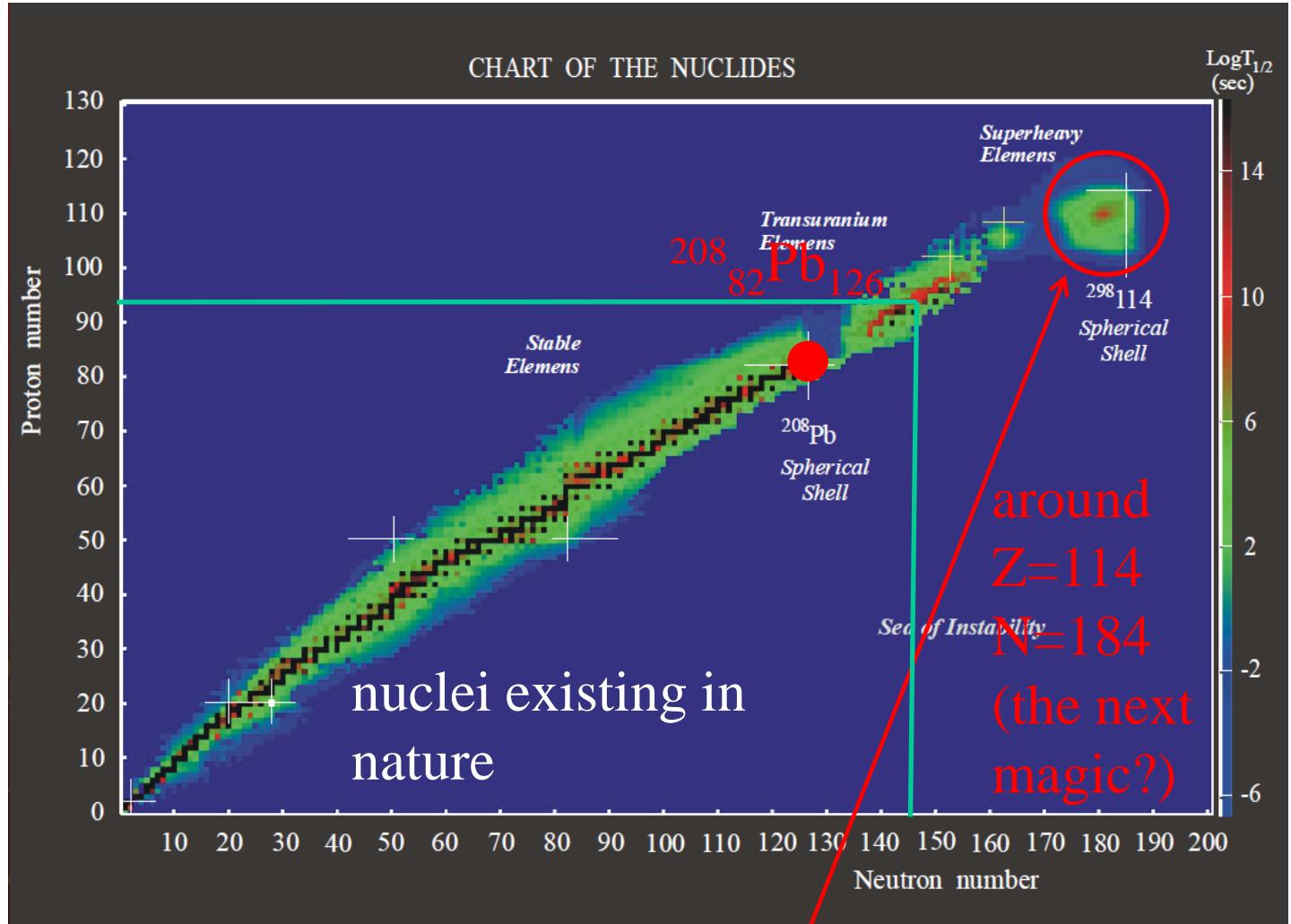
## what determines the limit of existence of elements? (ii) atomic nucleus



Z. Patyk et al.,  
NPA491('89) 267

QM shell effect (magic numbers) raises  $B_{\text{fiss}}$  and stabilizes a nucleus

# Superheavy elements (the island of stability)

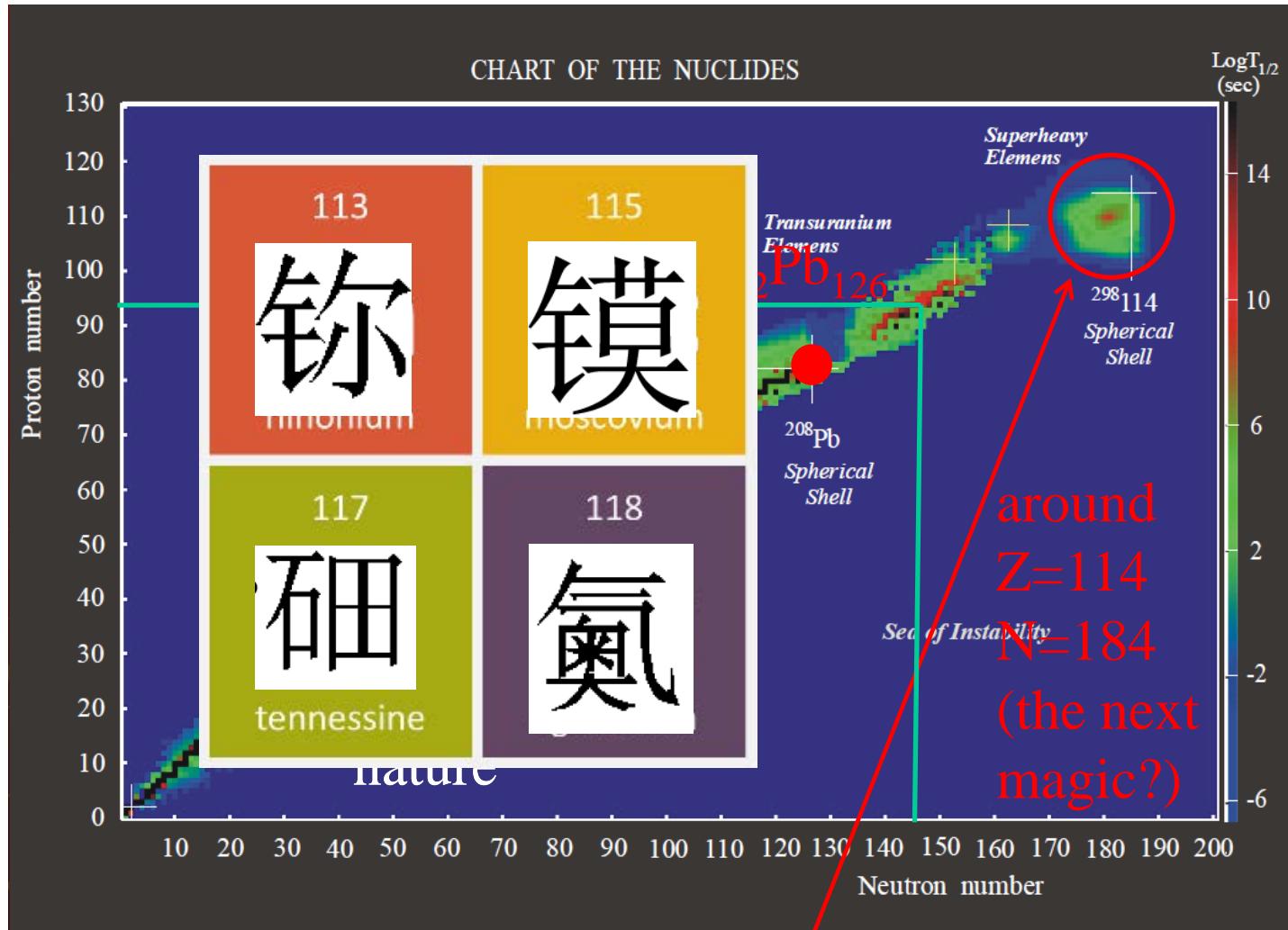


Yuri Oganessian

**long-lived with  $10^{3-5}$  years**

# Superheavy elements (the island of stability)

超重元素

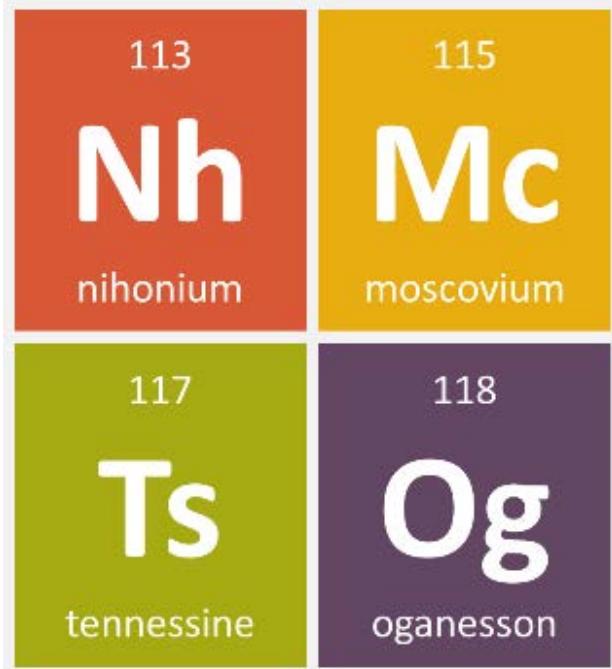


Yuri Oganessian

long-lived with  $10^{3-5}$  years

# Fusion reactions for SHE

the element 113: Nh

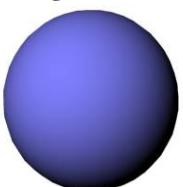
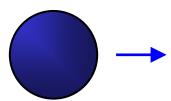


November, 2016



A detailed periodic table showing elements from hydrogen (H) to oganesson (Og). The element Nh (Nihonium) is highlighted with a red box in its group 13 position. Other elements are color-coded by group: groups 14-17 are green, groups 1-3 are blue, and groups 18-19 are red. Asterisks indicate synthetic elements.

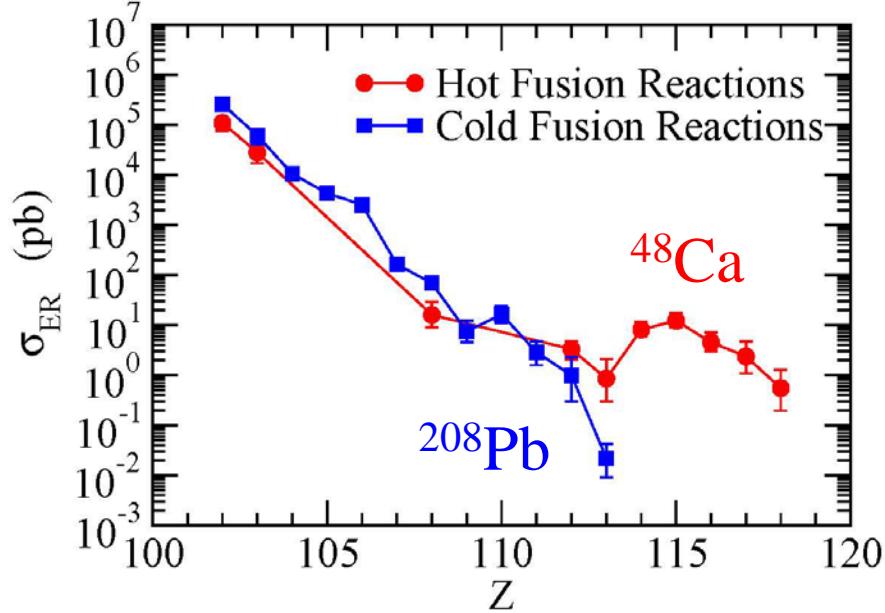
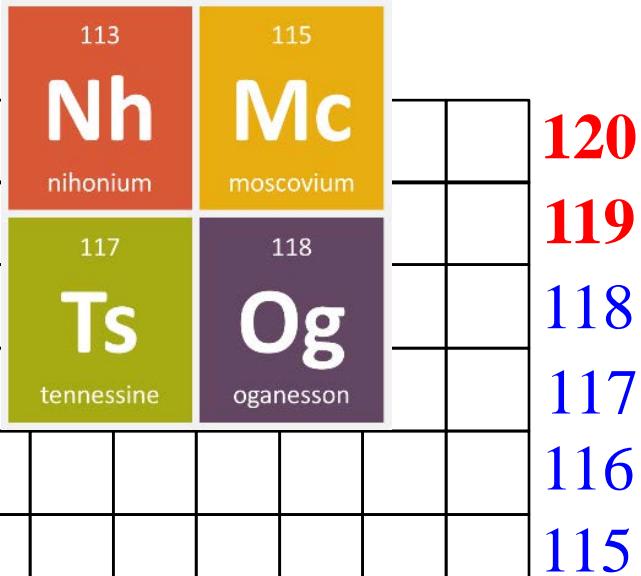
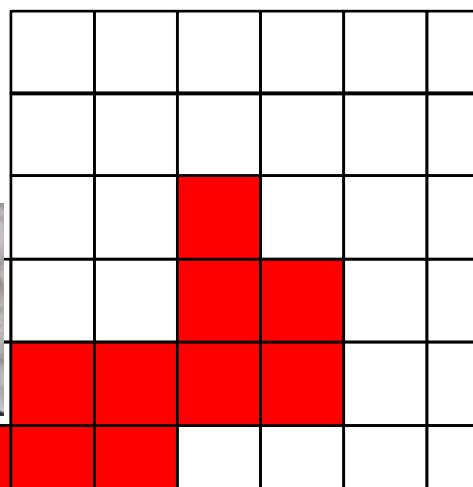
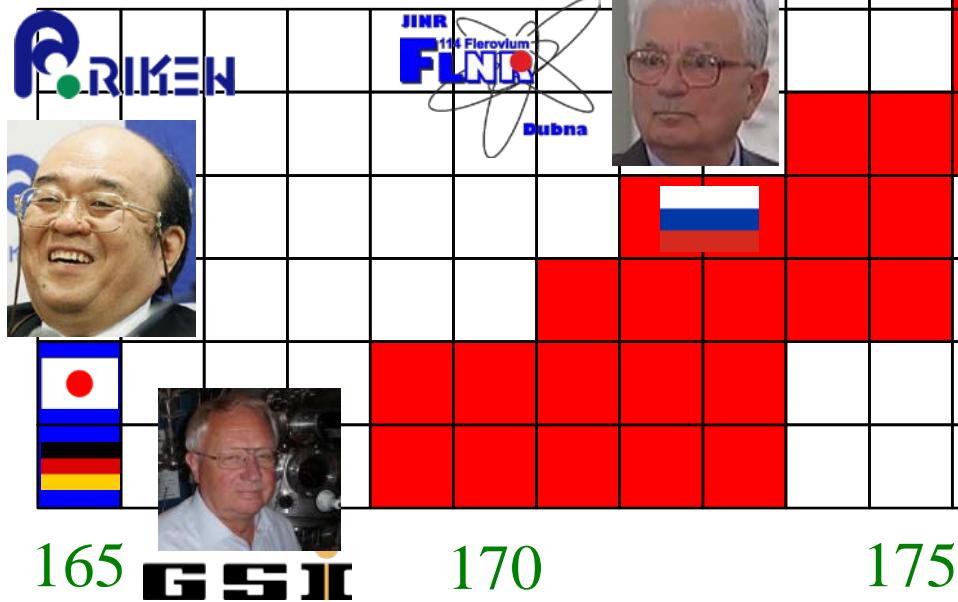
Wikipedia



Heavy-ion fusion reaction

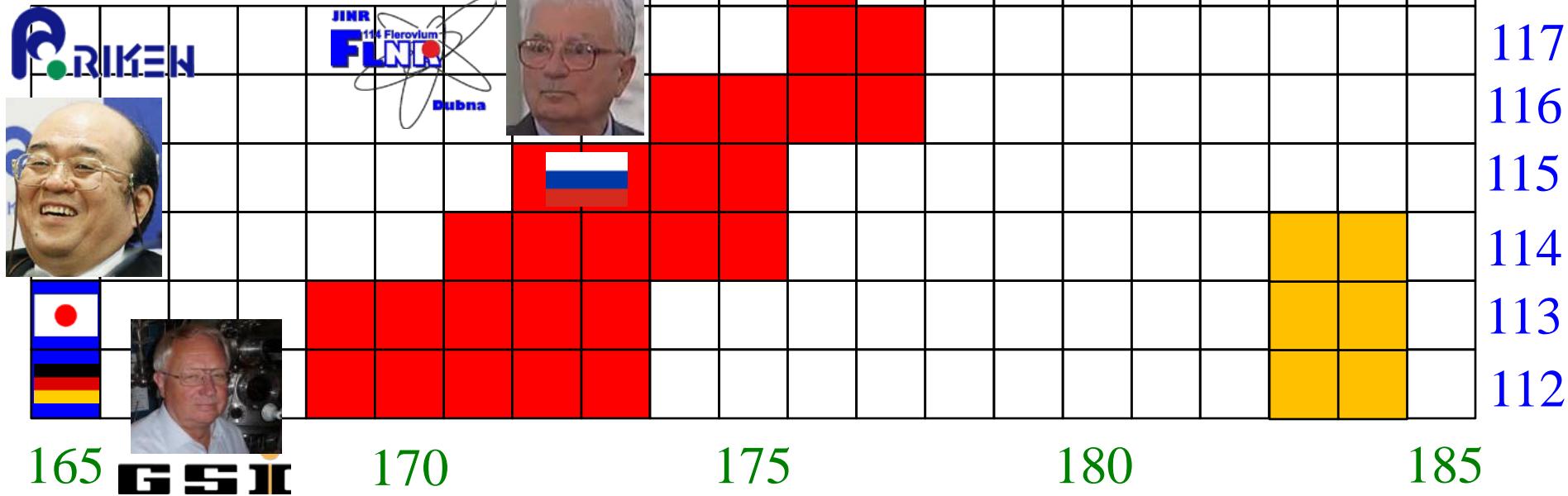
# Fusion for superheavy elements

Superheavy elements  
synthesized so far

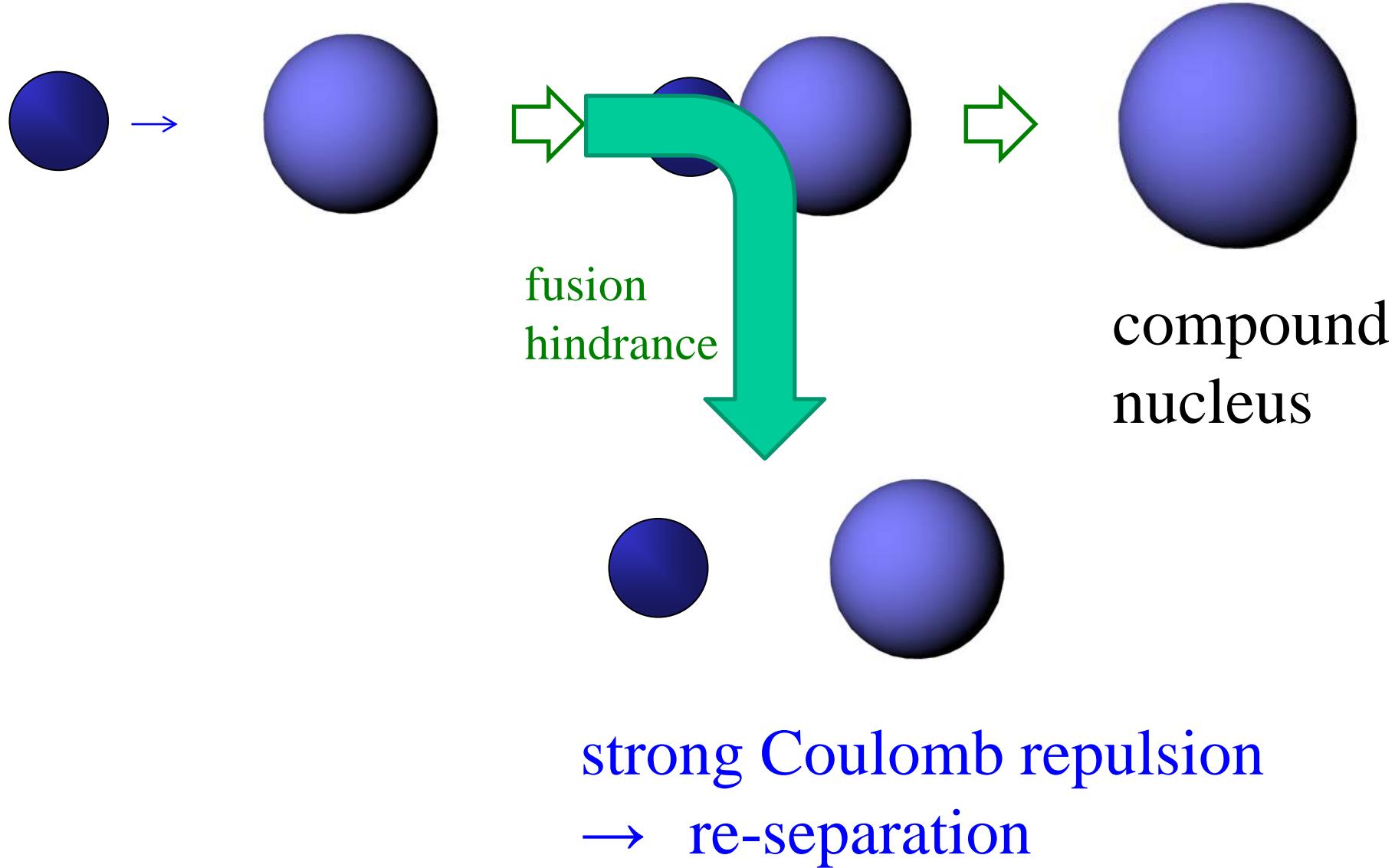


# Fusion for superheavy elements

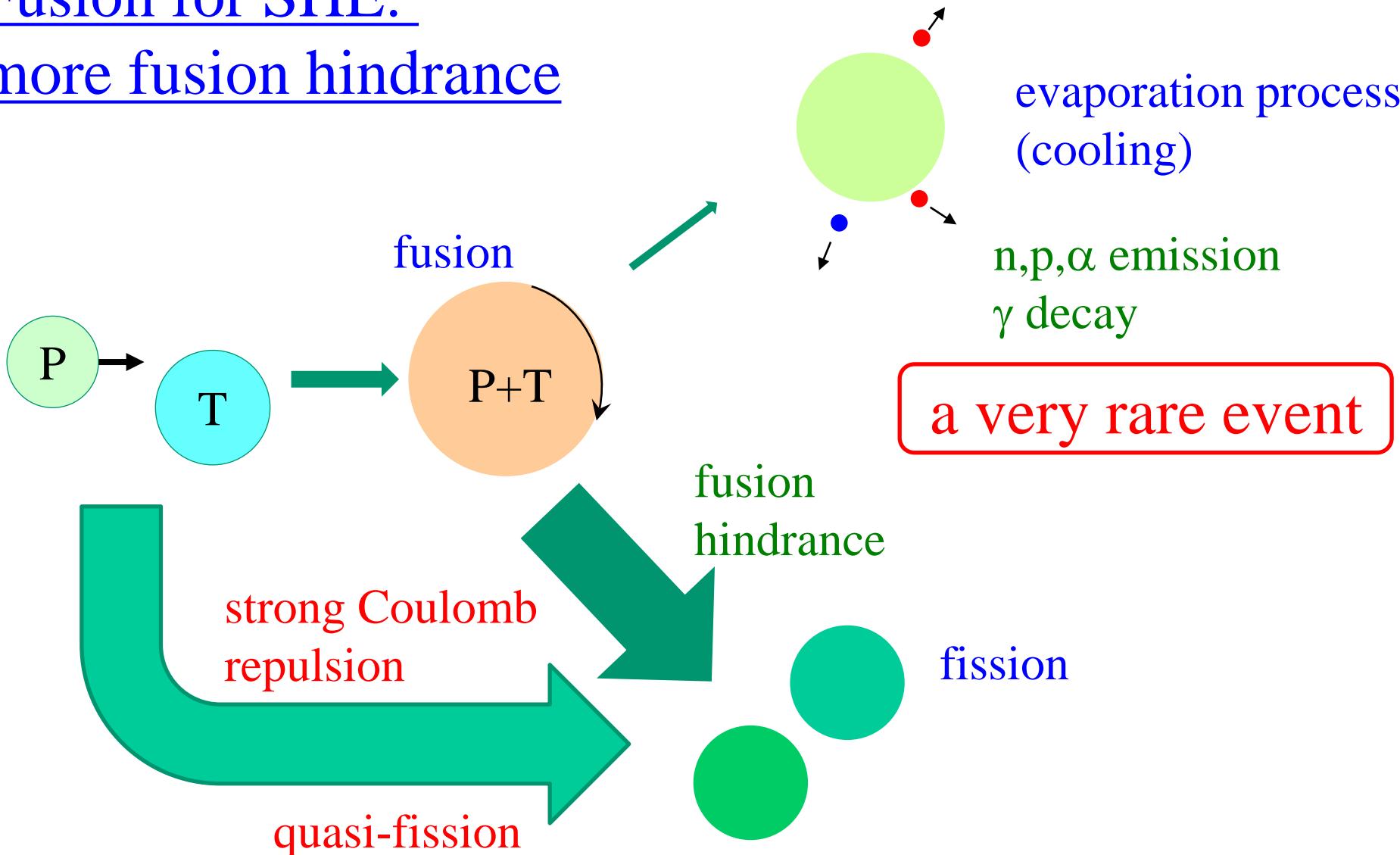
Superheavy elements  
synthesized so far



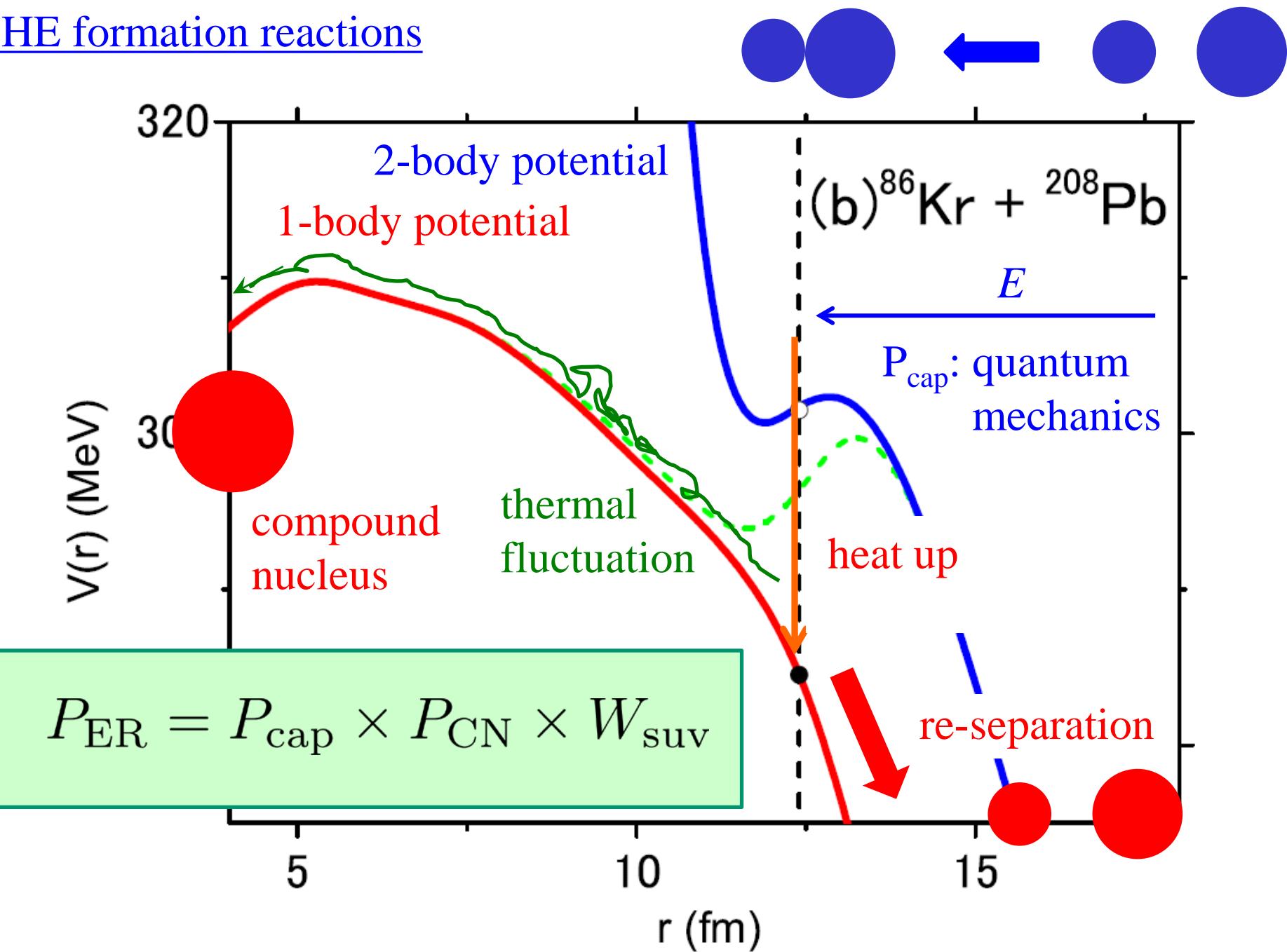
# Fusion for SHE: fusion hindrance



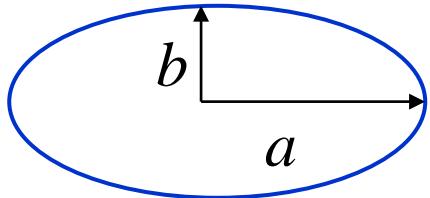
## Fusion for SHE: more fusion hindrance



## SHE formation reactions

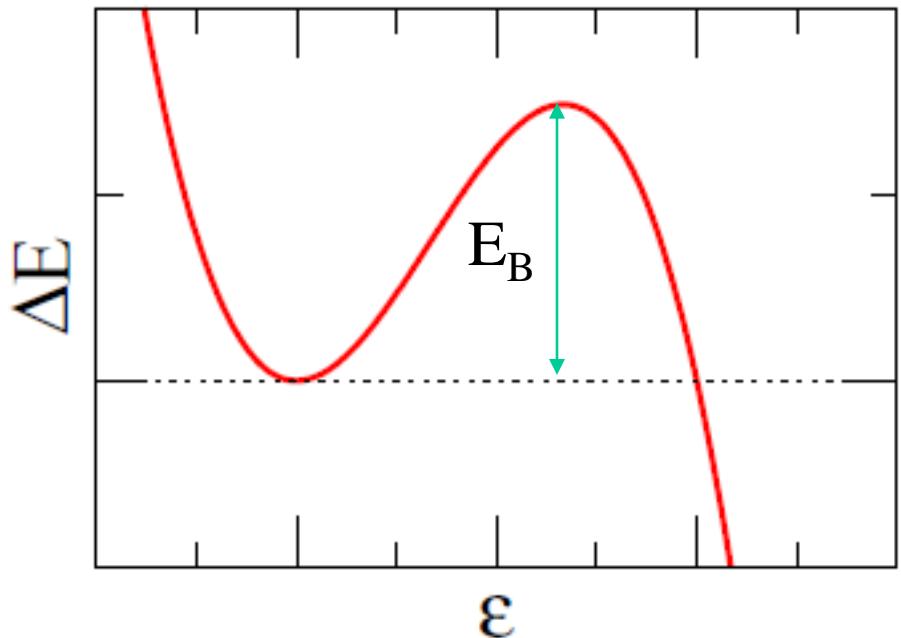


## (note) fission barrier in the liquid drop model



$$\begin{aligned} a &= R \cdot (1 + \epsilon) \\ b &= R \cdot (1 + \epsilon)^{-1/2} \\ ab^2 &= R^3 = \text{constant} \end{aligned}$$

$$\begin{aligned} \Delta E &= \Delta E_{\text{surf}} + \Delta E_{\text{coul}} \\ &= E_S^{(0)} \left\{ \frac{2}{5}(1-x)\epsilon^2 - \frac{4}{105}(1+2x)\epsilon^3 + \dots \right\} \end{aligned}$$

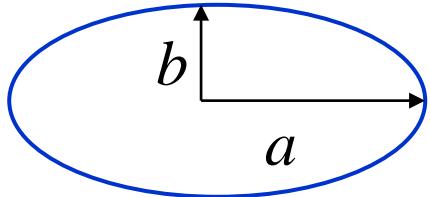


$$E_S^{(0)} = +a_S A^{2/3}$$

$$x \equiv \frac{E_C^{(0)}}{2E_S^{(0)}} = \frac{a_C}{2a_S} \cdot \frac{Z^2}{A} \sim \frac{1}{53.3} \cdot \frac{Z^2}{A}$$

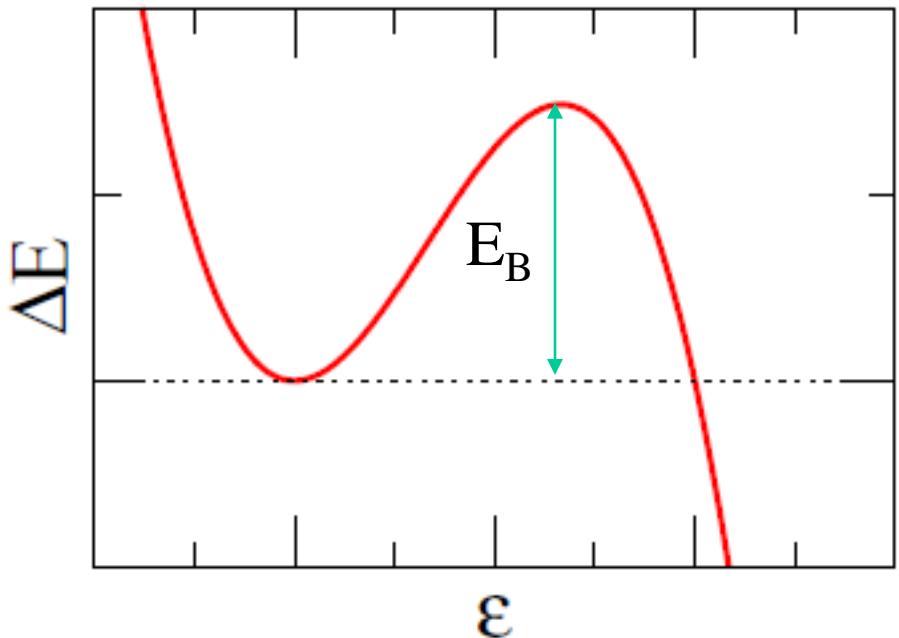
$$E_C^{(0)} = a_C Z(Z-1)/A^{1/3}$$

## (note) fission barrier in the liquid drop model



$$\begin{aligned} a &= R \cdot (1 + \epsilon) \\ b &= R \cdot (1 + \epsilon)^{-1/2} \\ ab^2 &= R^3 = \text{constant} \end{aligned}$$

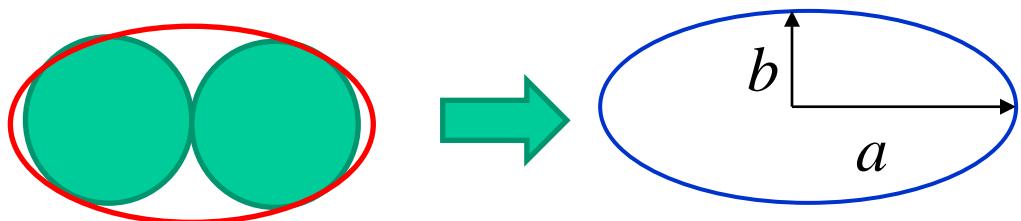
$$\begin{aligned} \Delta E &= \Delta E_{\text{surf}} + \Delta E_{\text{coul}} \\ &= E_S^{(0)} \left\{ \frac{2}{5}(1-x)\epsilon^2 - \frac{4}{105}(1+2x)\epsilon^3 + \dots \right\} \end{aligned}$$



fission barrier:

$$\begin{aligned} \epsilon_B &= \frac{21(1-x)}{3(1+2x)} \\ E_B &= \frac{98}{15} \cdot \frac{(1-x)^3}{(1+2x)^2} \cdot E_S^{(0)} \end{aligned}$$

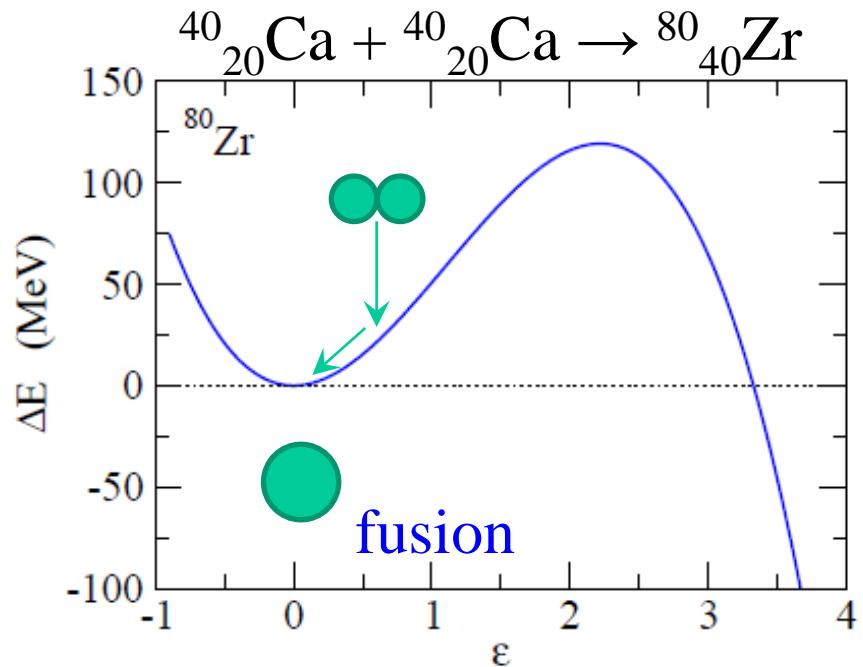
if two identical nuclei contact:



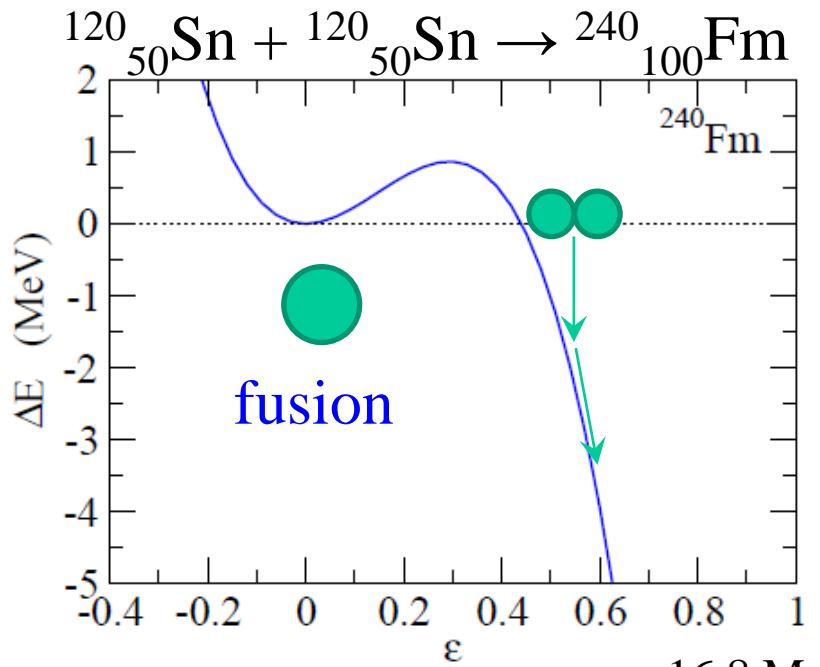
$$a = R_0 \cdot (1 + \epsilon)$$

$$b = R_0 \cdot (1 + \epsilon)^{-1/2}$$

$$\frac{a}{b} \sim \frac{2R}{R} = 2 \rightarrow \epsilon \sim 0.587$$

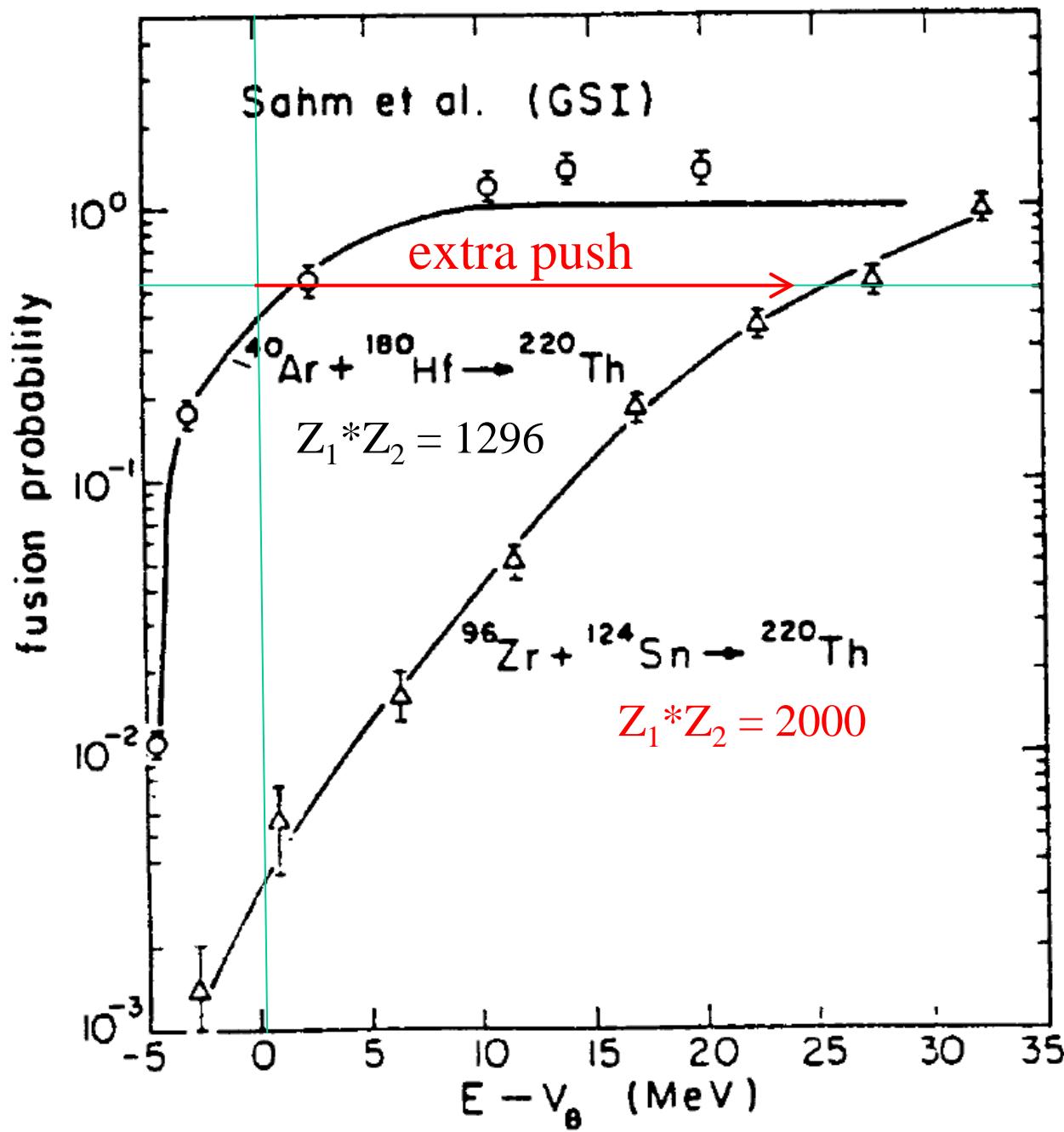


threshold:  $Z_1^* Z_2 = 1600 \sim 1800$

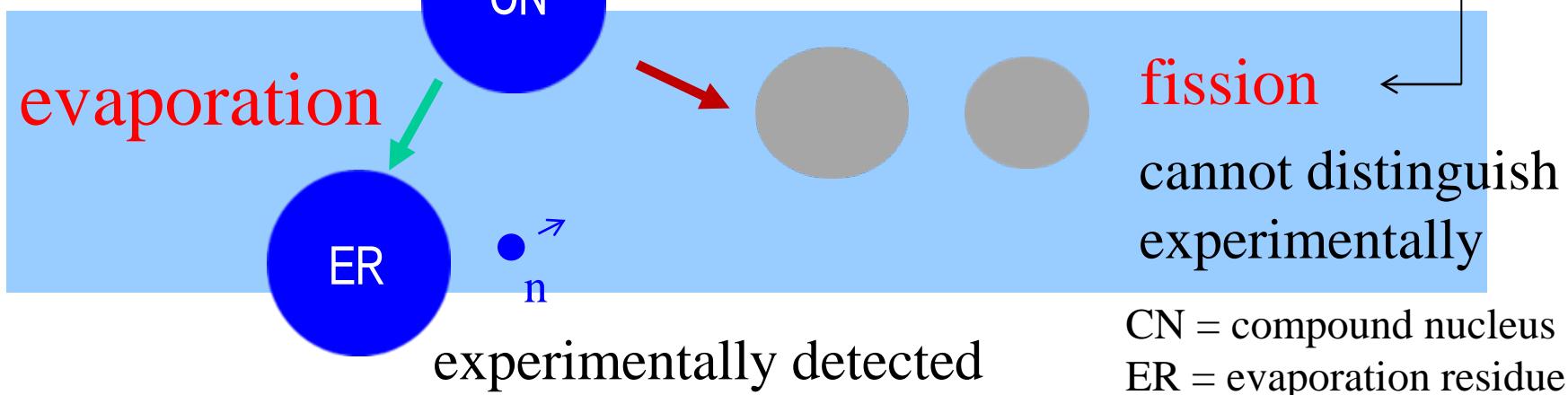
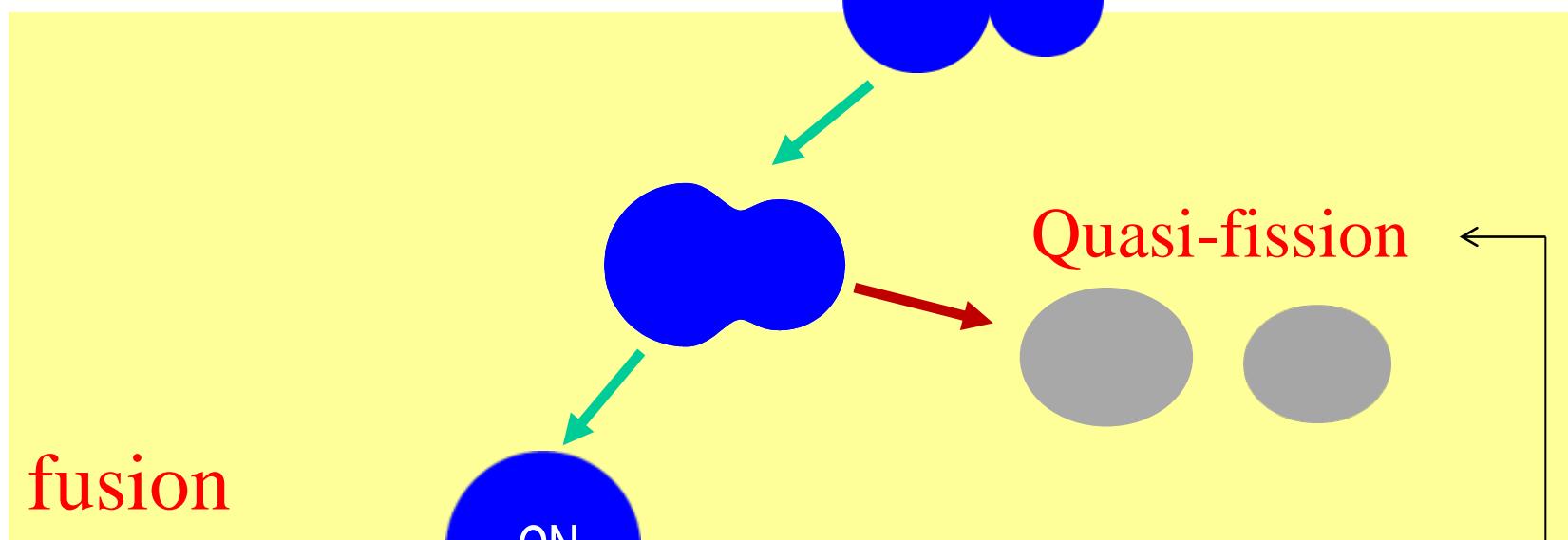


$$a_S = 16.8 \text{ MeV}$$

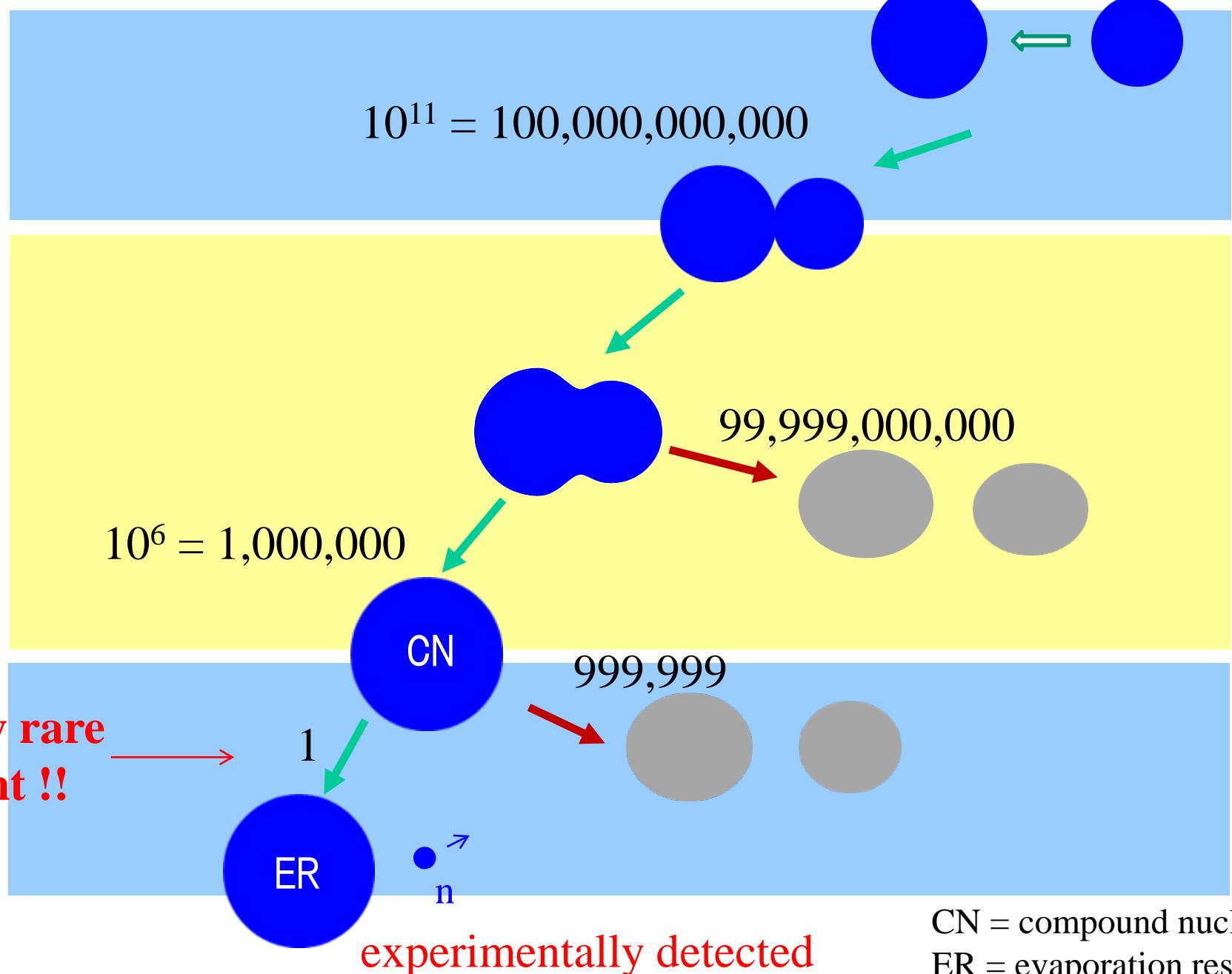
$$a_C = 0.72 \text{ MeV}$$



C.-C. Sahm et al.,  
Z. Phys. A319('84)113



# typical values for Ni + Pb reaction



# typical values for Ni + Pb reaction

$10^{11} = 100,000,000,000$

hot fusion  
:optimizes this process

$10^6 = 1,000,000$

very rare  
event !!

CN

$\bullet_n$

ER

experimentally detected

99,999,000,000

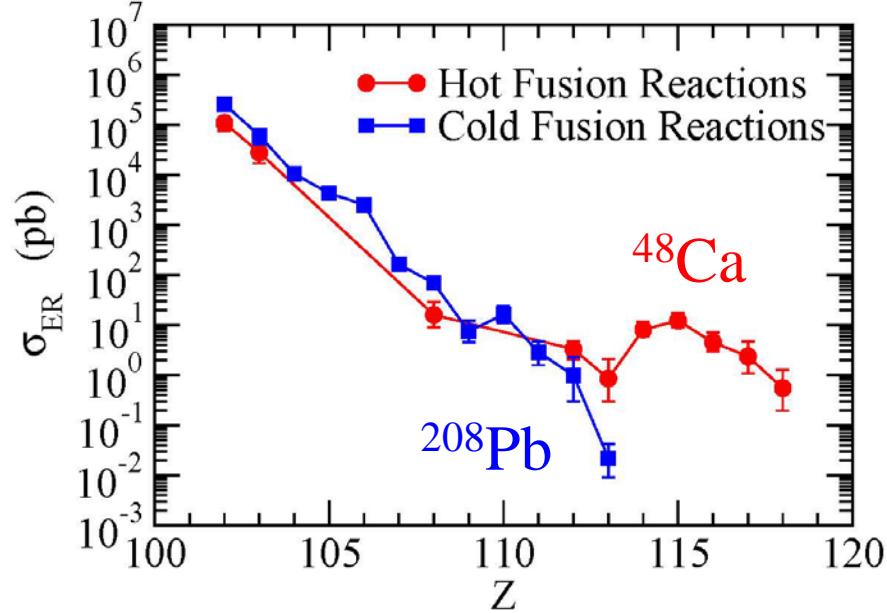
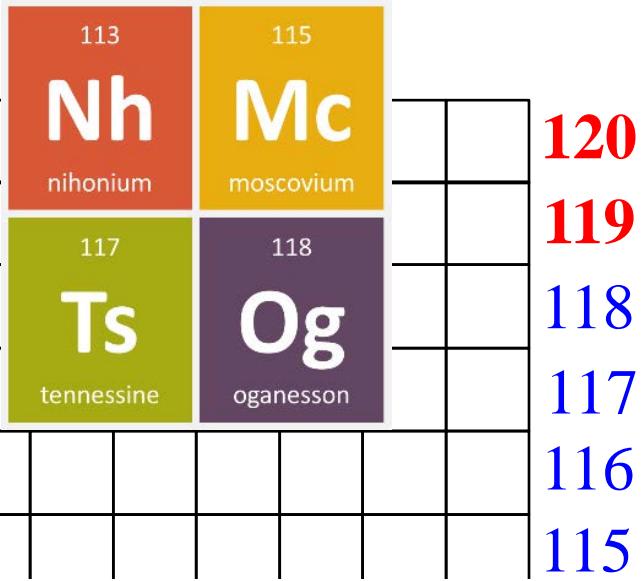
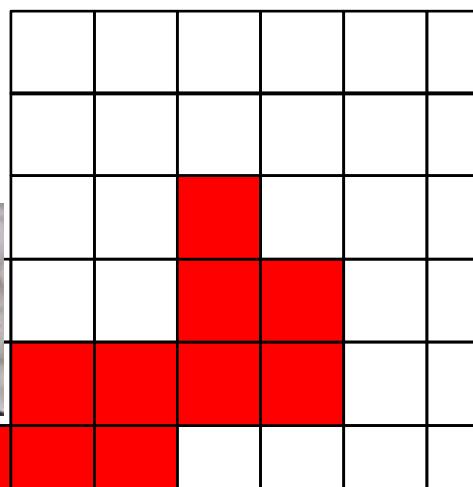
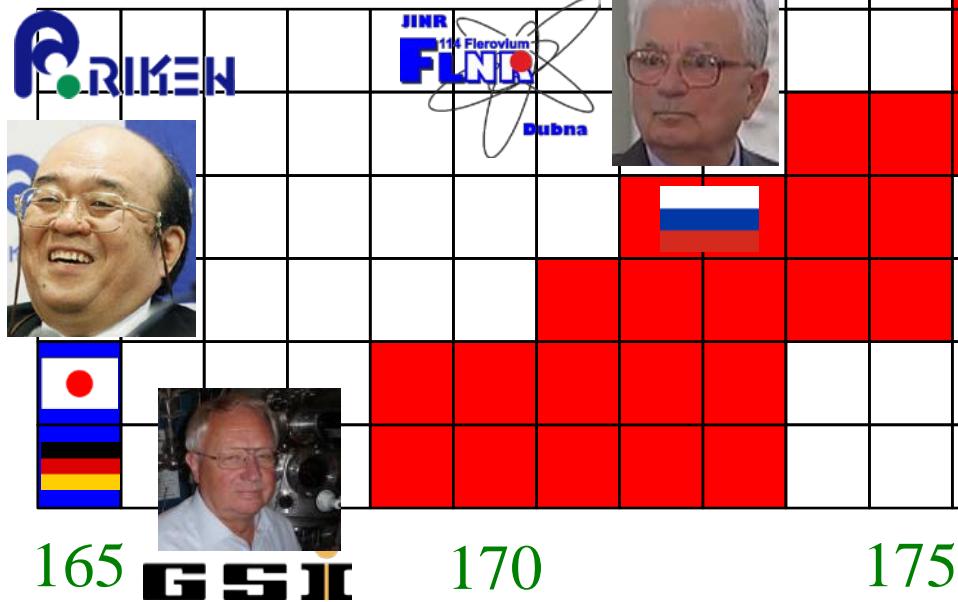
cold fusion  
:optimizes this process

CN = compound nucleus

ER = evaporation residue

# Fusion for superheavy elements

Superheavy elements  
synthesized so far



# Theoretical challenges

formation of SHE: very rare  
→ a large theoretical uncertainty

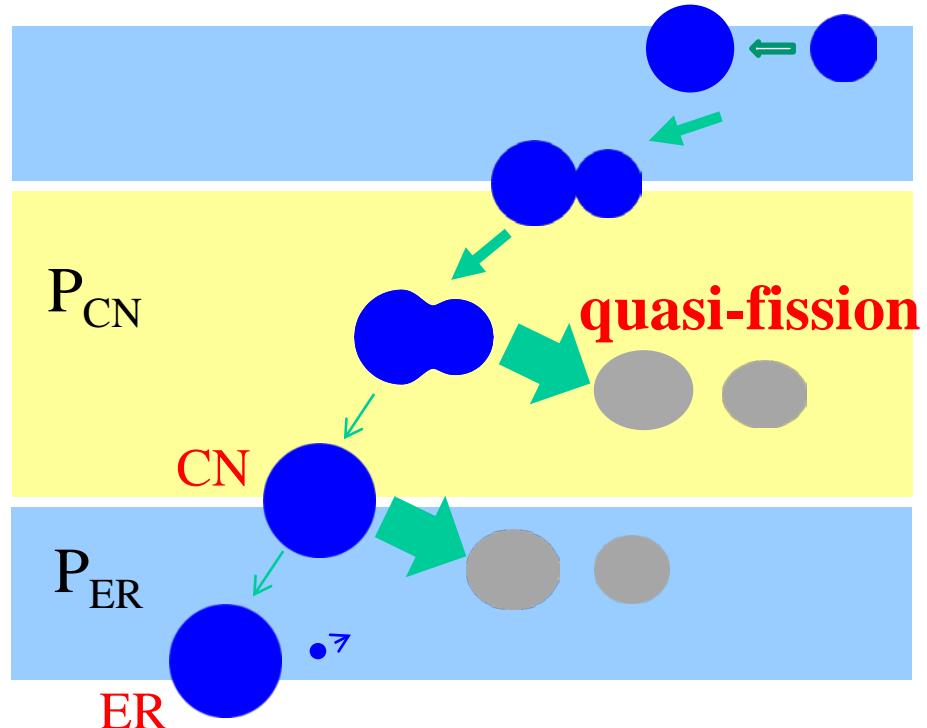
$$P_{\text{ER}} = P_{\text{cap}} \cdot P_{\text{CN}} \cdot W_{\text{suv}}$$

- ✓ no exp. data for  $P_{\text{CN}}$
- ✓ exp. data:  $P_{\text{ER}}$  only

CN=複合核、ER=蒸発残留核

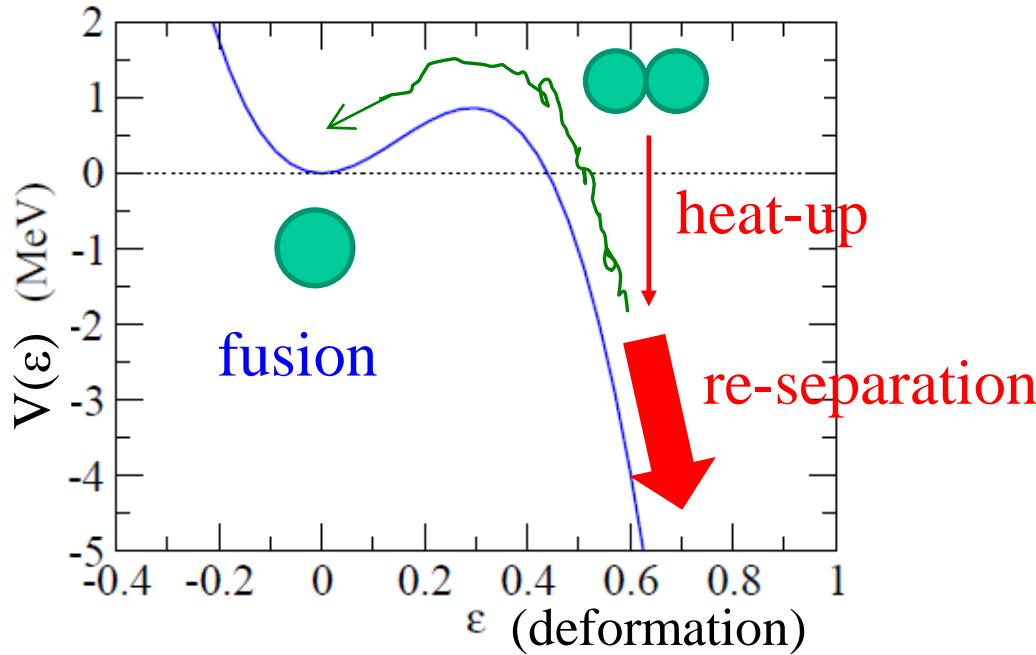
theoretical challenges:  
to reduce the uncertainties and  
make reliable predictions

fusion hindrance



Physics of open quantum systems  
量子開放系

## Langevin approach



**thermal fluctuation**

→ Langevin method  
(Brownian method)

## classical Langevin equation

$$m \frac{d^2 q}{dt^2} = -\frac{dV(q)}{dq} - \gamma \frac{dq}{dt} + R(t)$$

friction      random interaction  $\rightarrow \langle R(t) \rangle = 0$

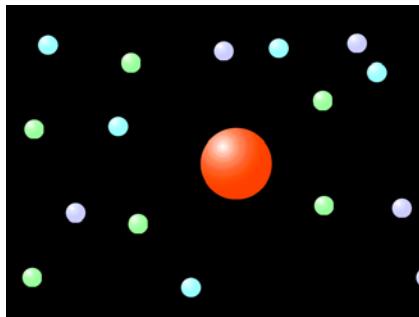
classical:

$$\langle R(t)R(t') \rangle = 2D \delta(t - t')$$

$D = \gamma T$  (Einstein relation)

(white noise; no memory)

Brownian motion



interaction of a Brownian  
particle with atoms

## classical Langevin equation

$$m \frac{d^2 q}{dt^2} = -\frac{dV(q)}{dq} - \gamma \frac{dq}{dt} + R(t)$$

friction

random interaction  $\rightarrow \langle R(t) \rangle = 0$

classical:

$$\langle R(t)R(t') \rangle = 2D \delta(t - t')$$

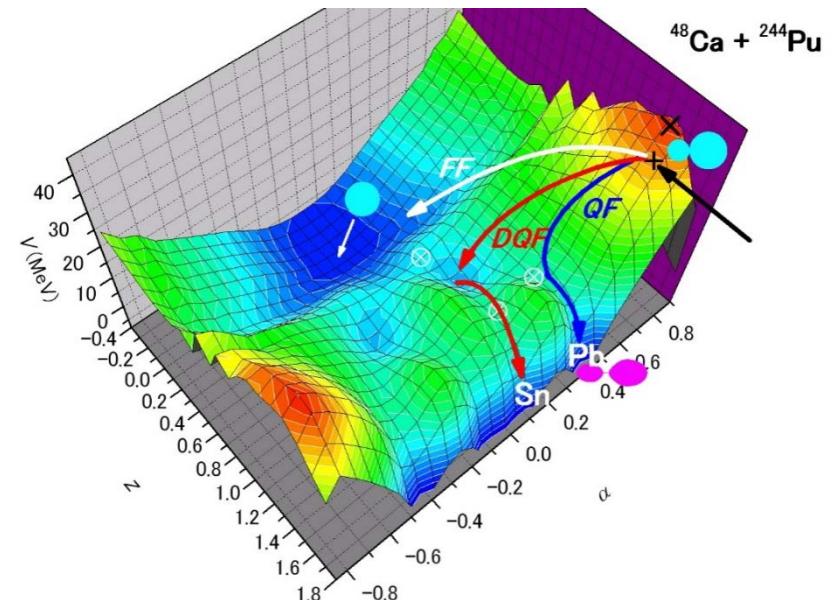
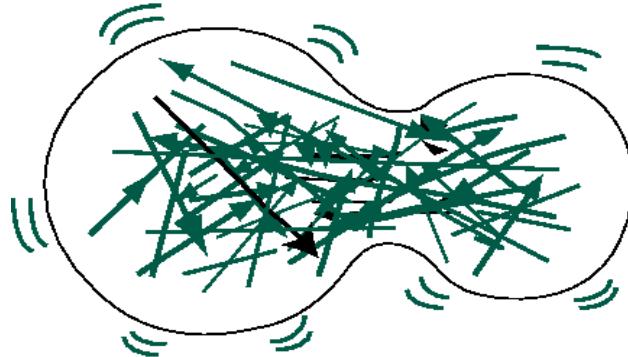
$D = \gamma T$  (Einstein relation)

(white noise; no memory)

nuclear reactions:

$q$  = the relative distance etc.

“atoms” = nucleonic d.o.f



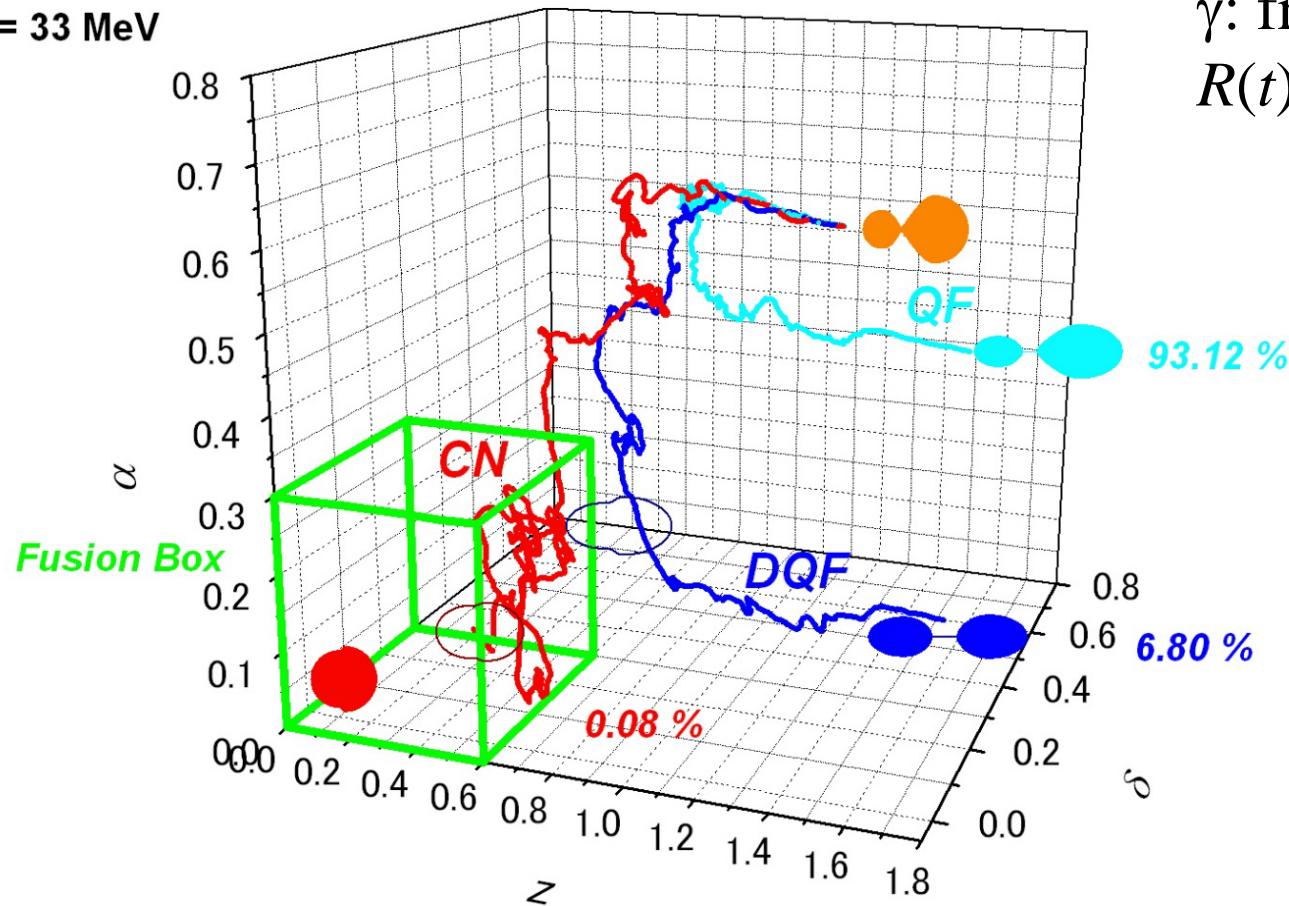
# Theory: Lagenvin approach

multi-dimensional extension of:

$$m \frac{d^2q}{dt^2} = -\frac{dV(q)}{dq} - \gamma \frac{dq}{dt} + R(t)$$



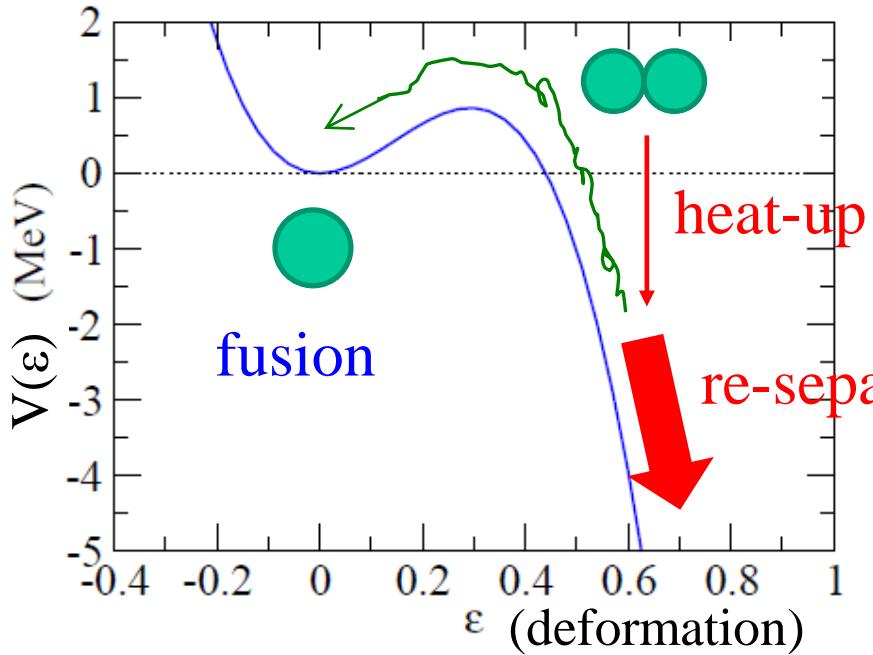
$E^* = 33 \text{ MeV}$



$\gamma$ : friction coefficient

$R(t)$ : random force

## Langevin approach



$$m \frac{d^2 q}{dt^2} = -\frac{dV(q)}{dq} - \gamma \frac{dq}{dt} + R(t)$$

## Theoretical issues

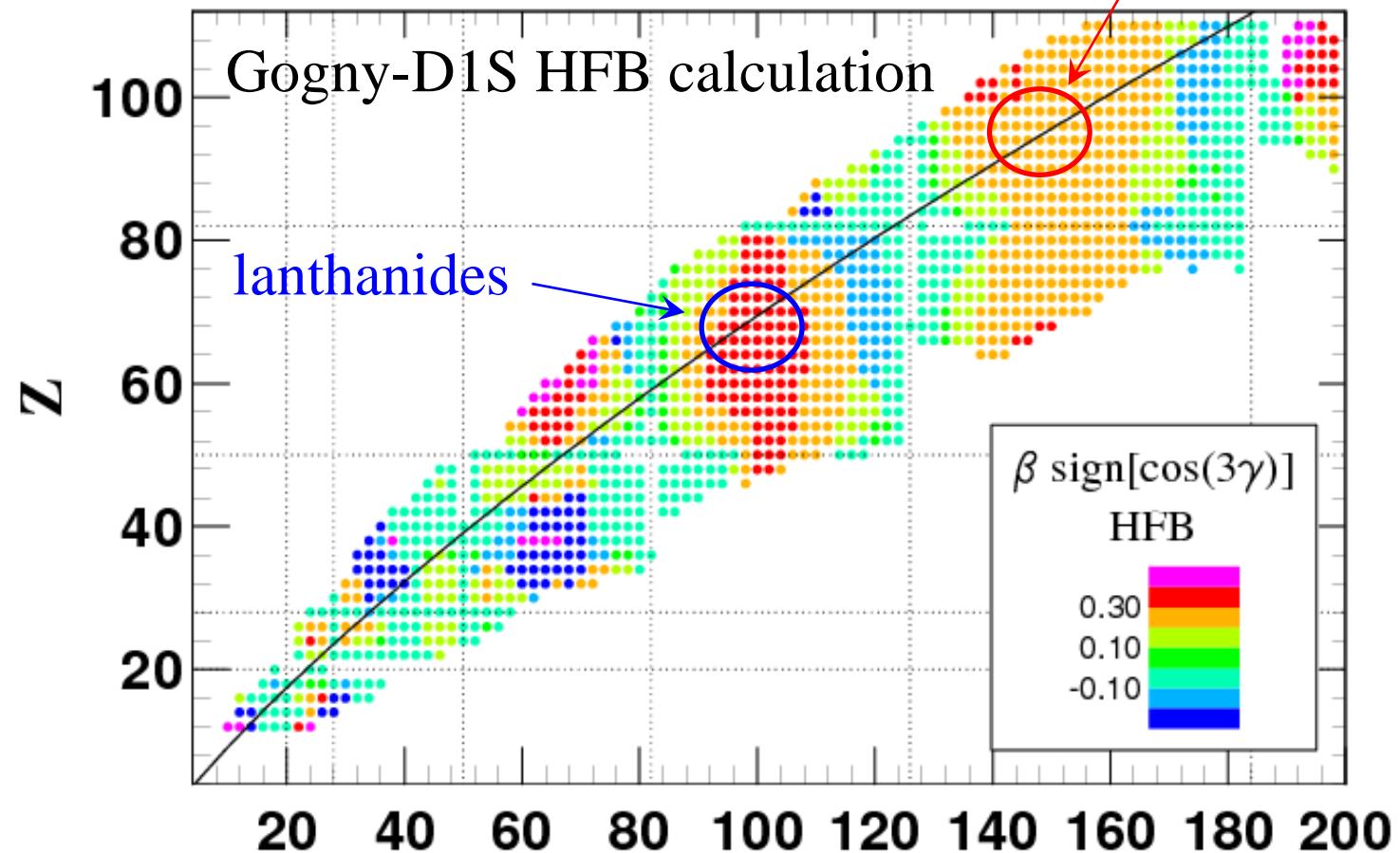
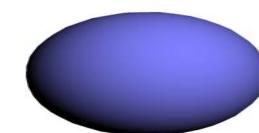
- ✓ how to thermalize? mechanisms?
- ✓ is thermal equilibrium OK?
- ✓ Is Markovian approximation OK?
- ✓ quantum effects?
- ✓ quantal-to-classical transitions (decoherence)?

}  microscopic models?

}  open quantum systems

## hot fusion: Nuclear Deformation

hot fusion:  $^{48}\text{Ca}$  + deformed target



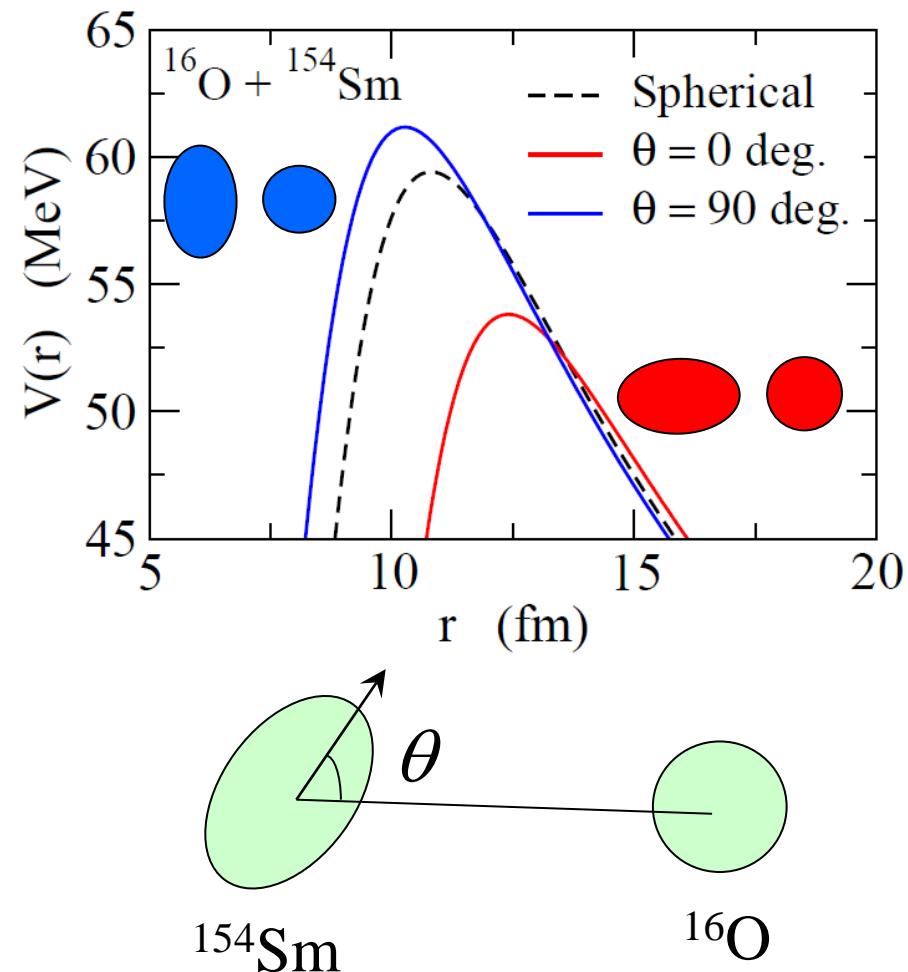
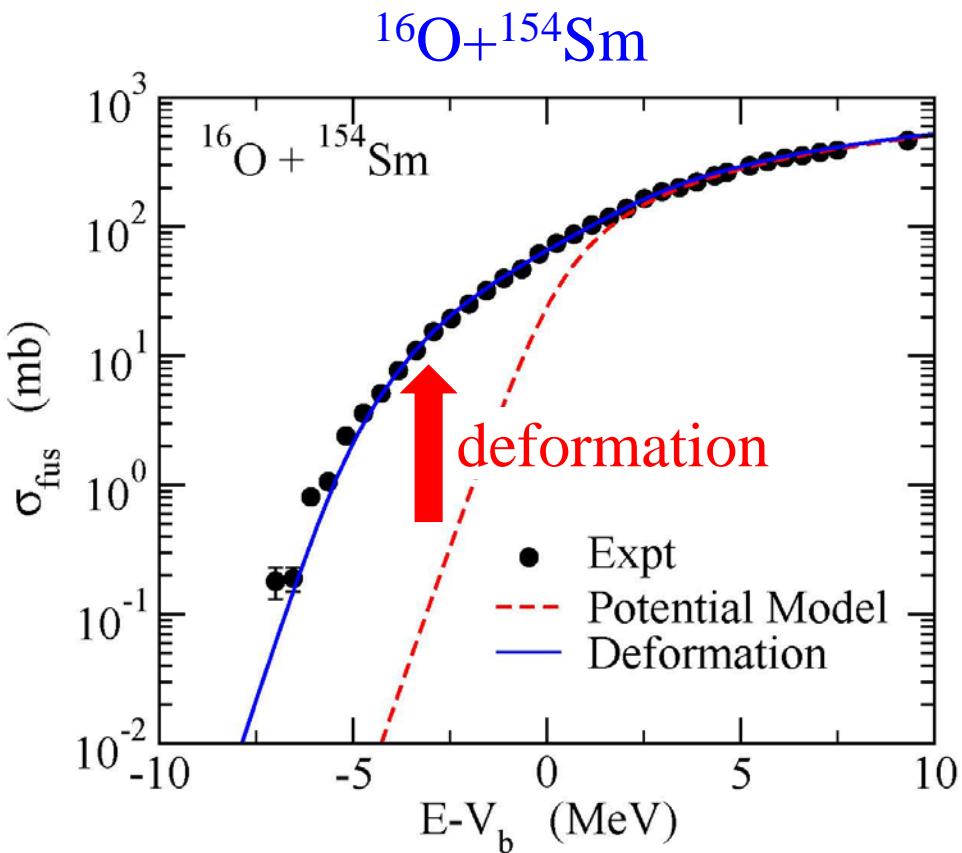
the role of deformation  
in heavy-ion reactions?

N

J.-P. Delaroche et al.,  
PRC81 ('10) 014303

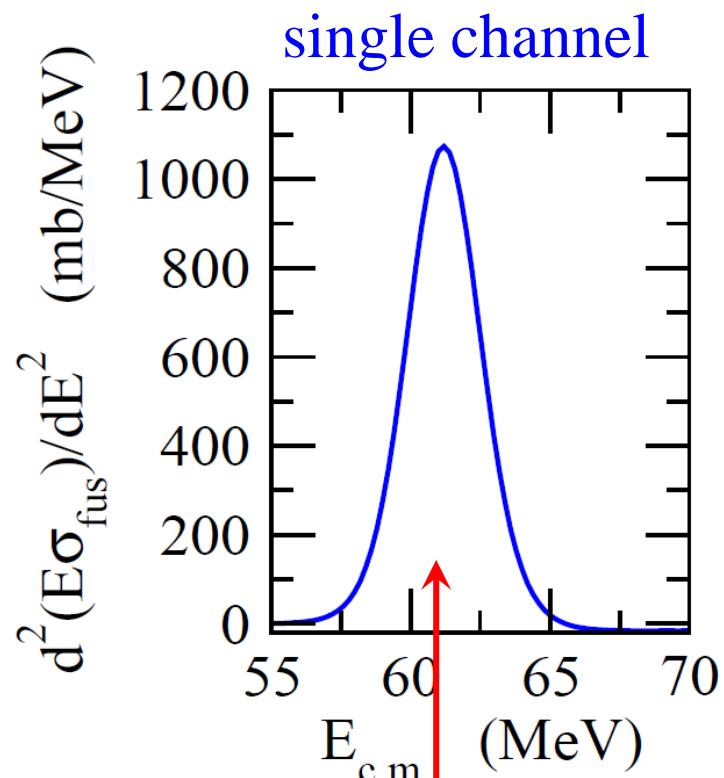
# Nuclear deformation and barrier distribution

Nuclear deformation → a large sub-barrier enhancement of fusion cross sections



## Fusion barrier distribution [Rowley, Satchler, Stelson, PLB254('91)]

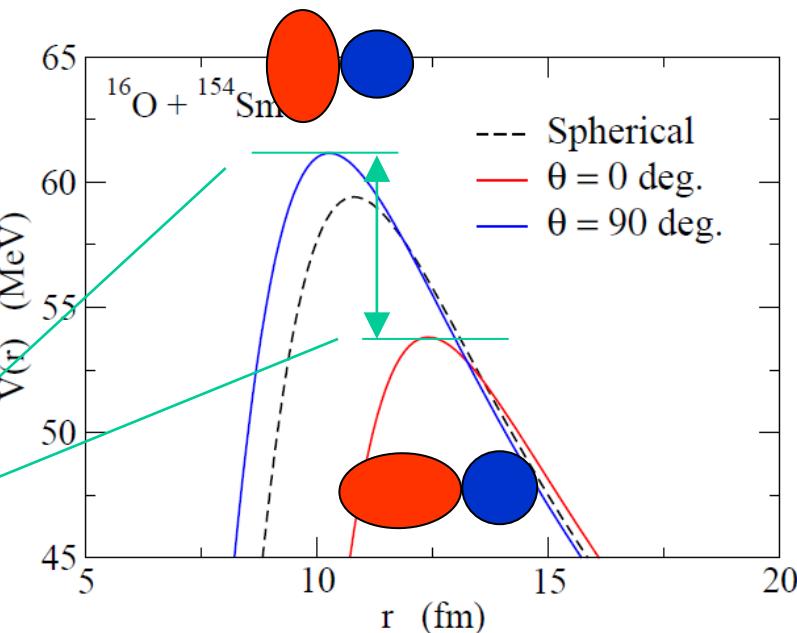
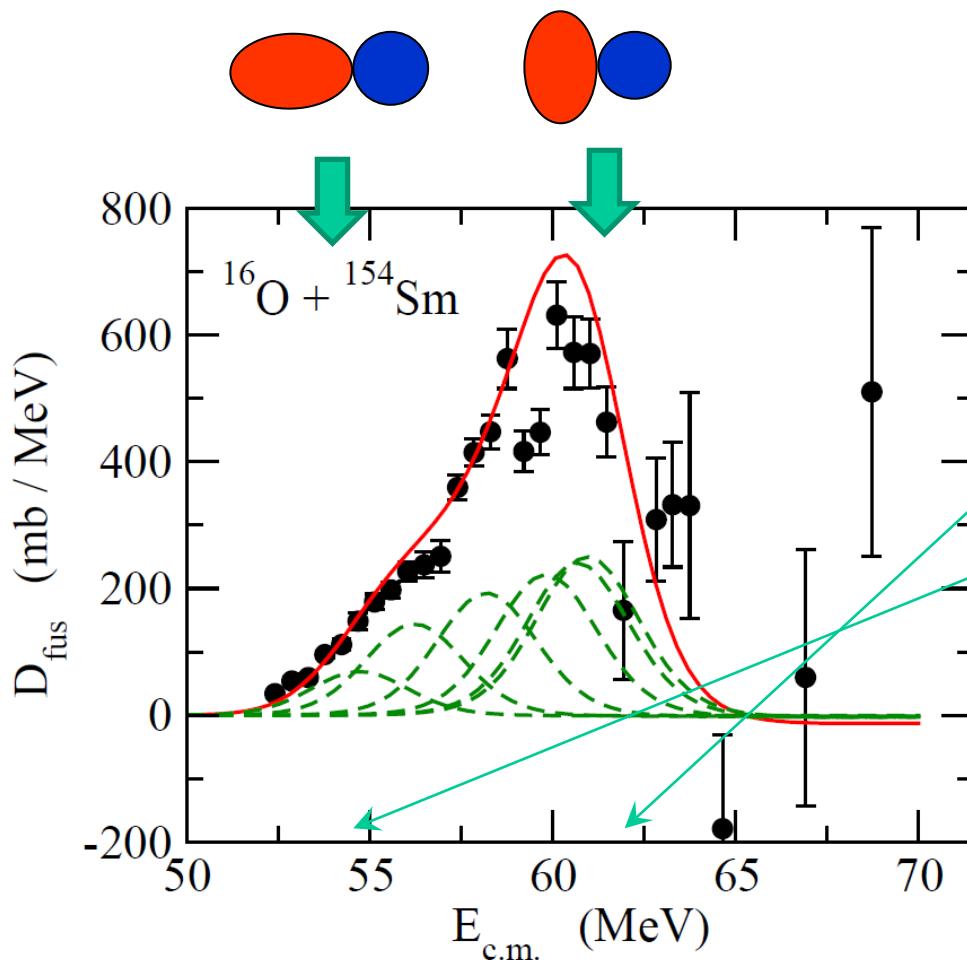
$$D_{\text{fus}}(E) = \frac{d^2(E\sigma_{\text{fus}})}{dE^2} \propto \frac{dP_{l=0}}{dE}$$



$V_b$

✓ Fusion barrier distribution (Rowley, Satchler, Stelson, PLB254('91))

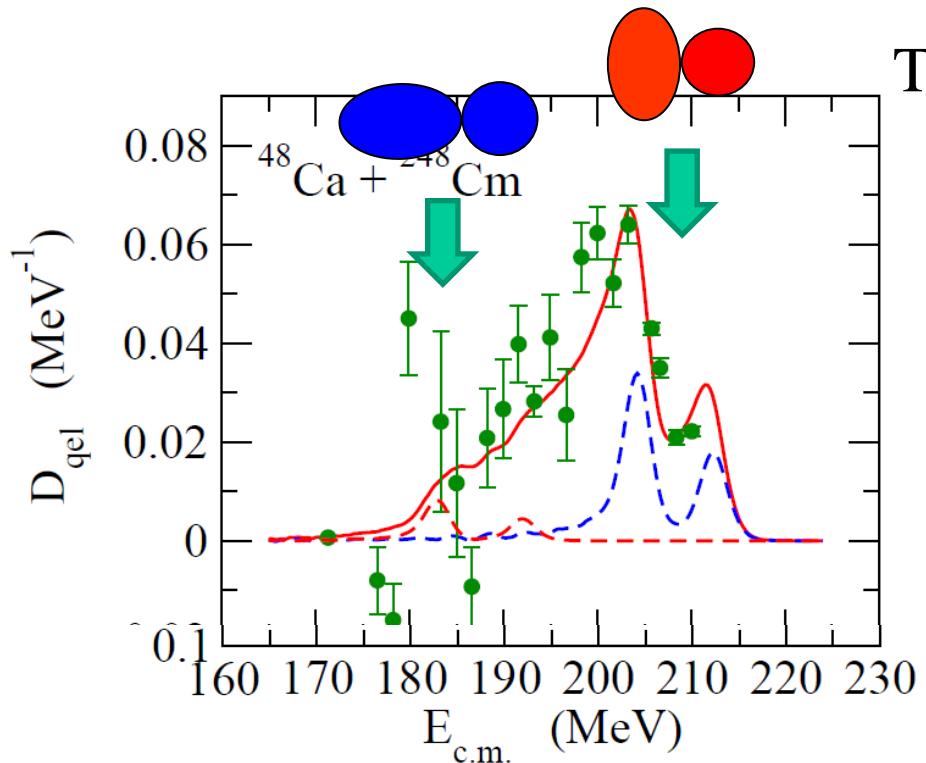
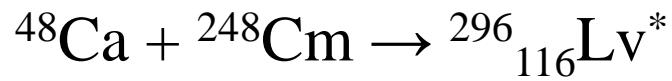
$$D_{\text{fus}}(E) = \frac{d^2(E\sigma_{\text{fus}})}{dE^2}$$



Data: J.R. Leigh et al.,  
PRC52 ('95) 3151

can be used to identify  
the side/tip collisions

## Application to hot fusion reactions

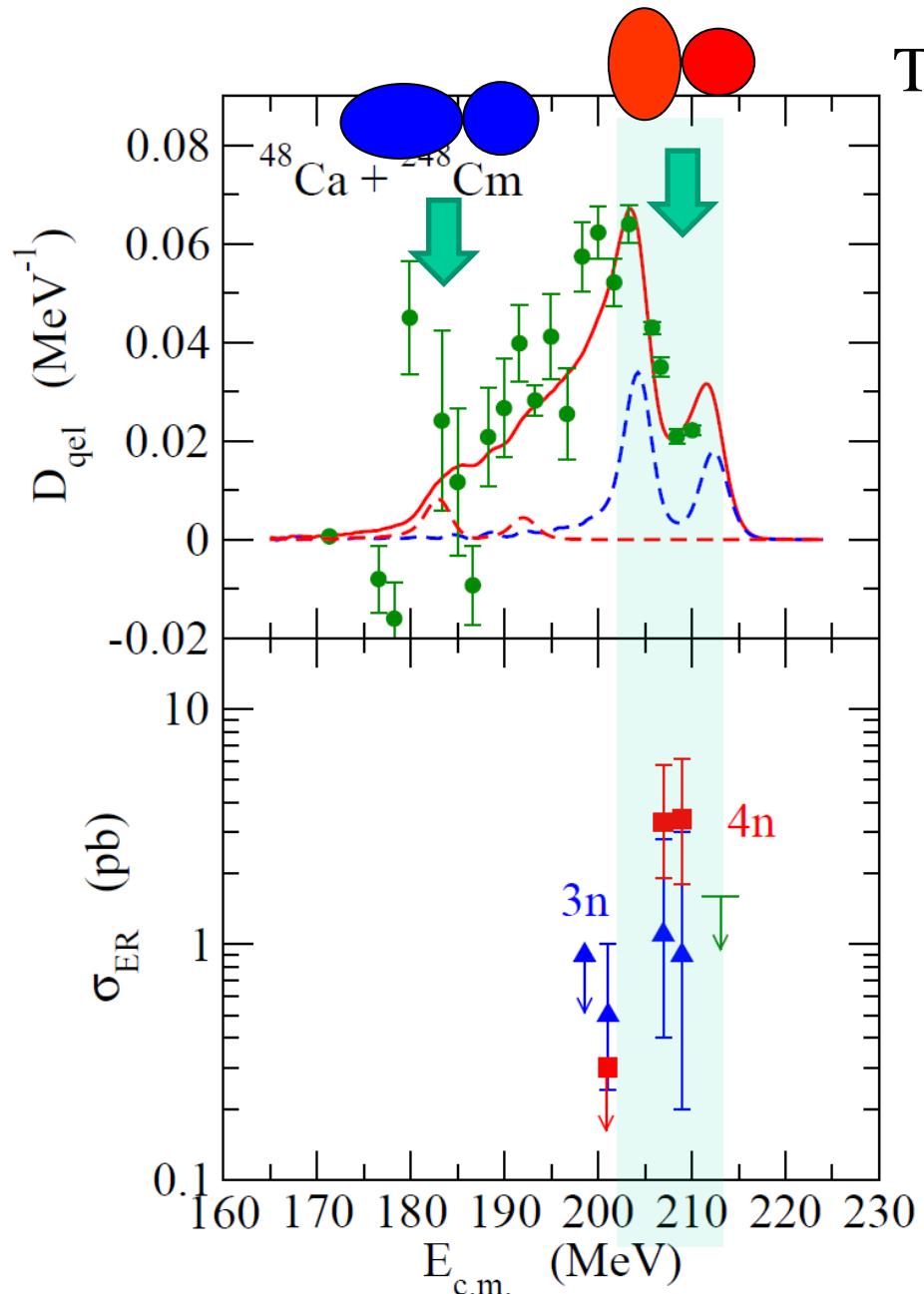
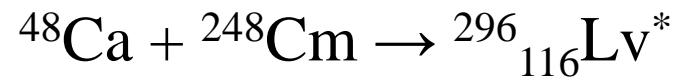


T. Tanaka,..., K.H., et al.,  
JPSJ 87 ('18) 014201  
PRL124 ('20) 052502



capture barrier distribution

## Application to hot fusion reactions



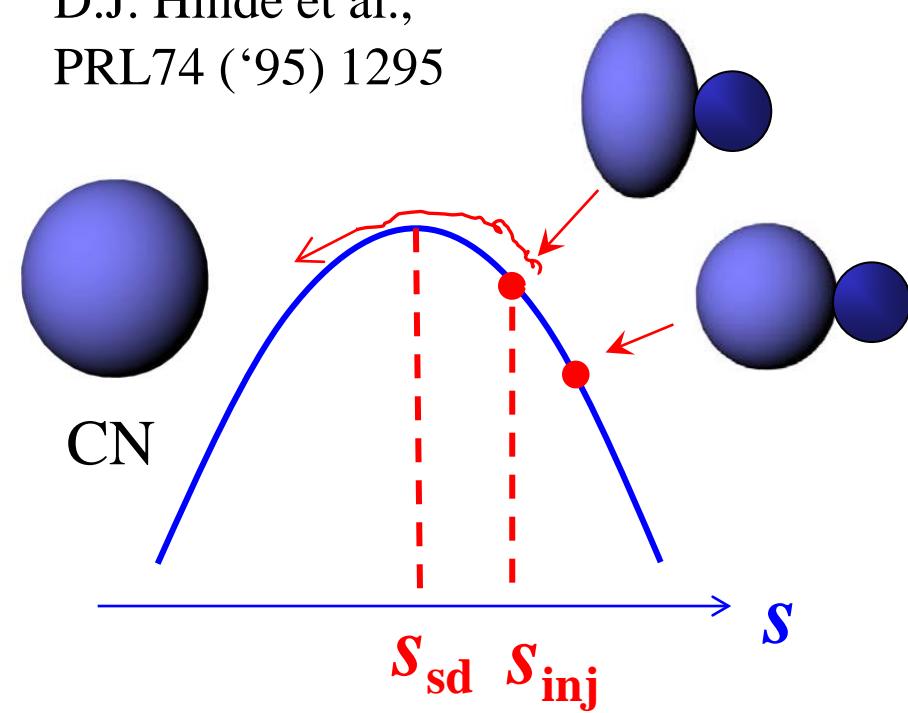
T. Tanaka,..., K.H., et al.,  
JPSJ 87 ('18) 014201  
PRL124 ('20) 052502



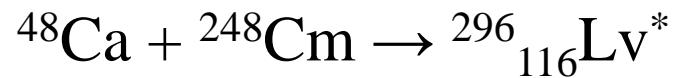
capture barrier distribution

cf. notion of compactness:

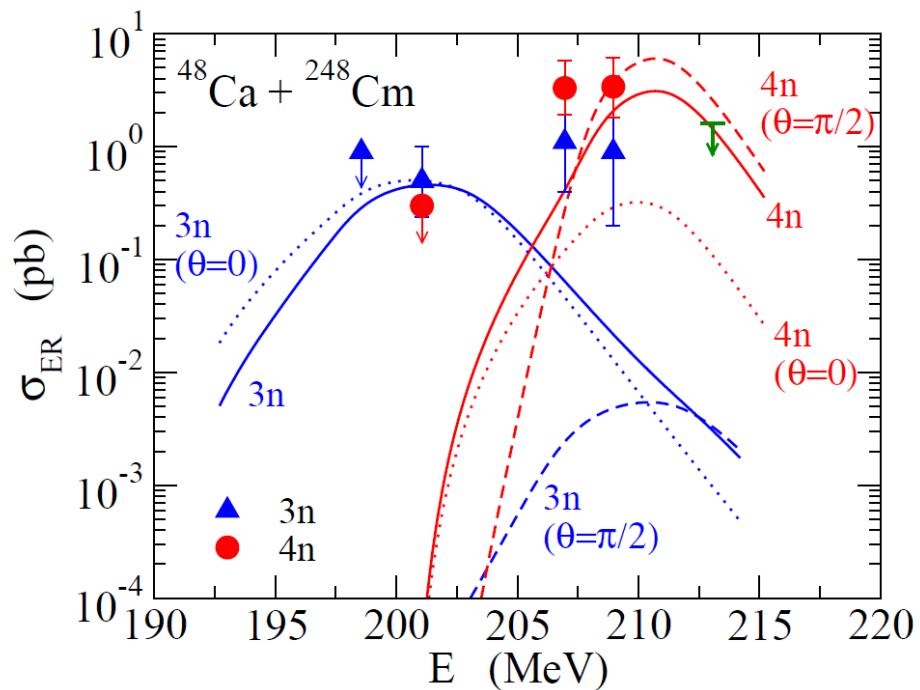
D.J. Hinde et al.,  
PRL74 ('95) 1295



## Application to hot fusion reactions



T. Tanaka,..., K.H., et al.,  
JPSJ 87 ('18) 014201  
PRL124 ('20) 052502

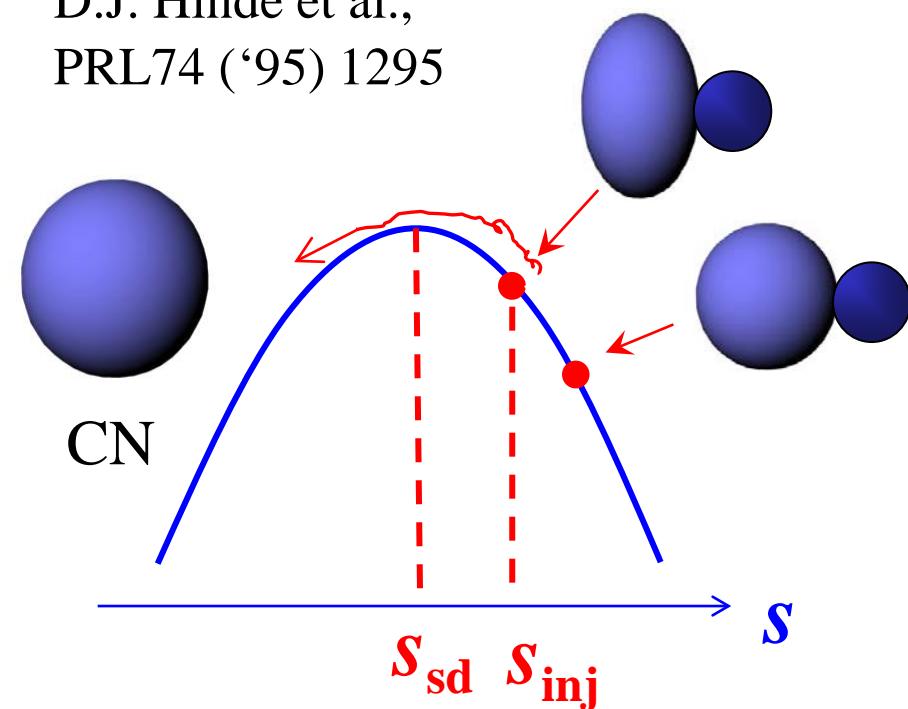


K. Hagino, PRC98 ('18) 014607

capture barrier distribution

cf. notion of compactness:

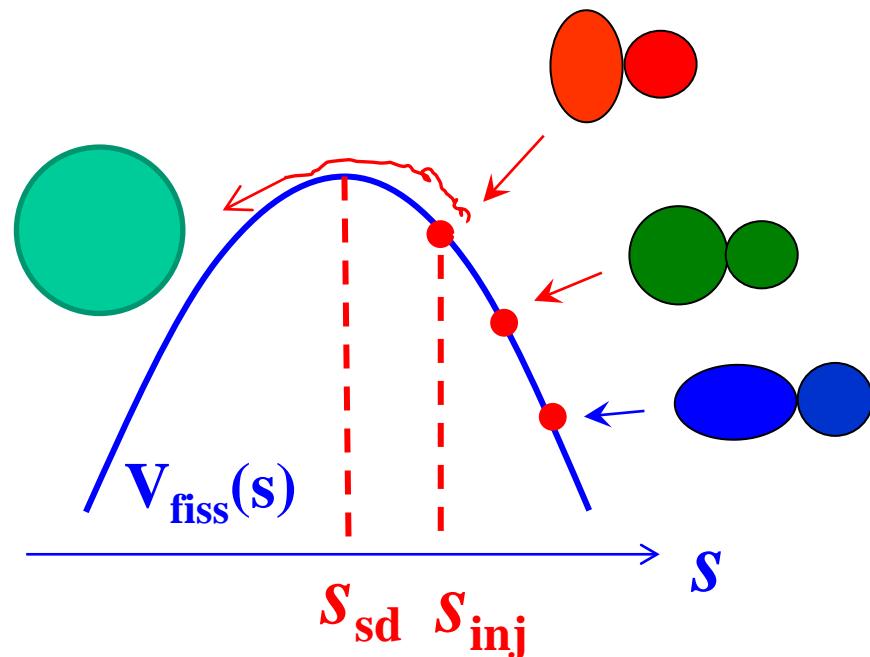
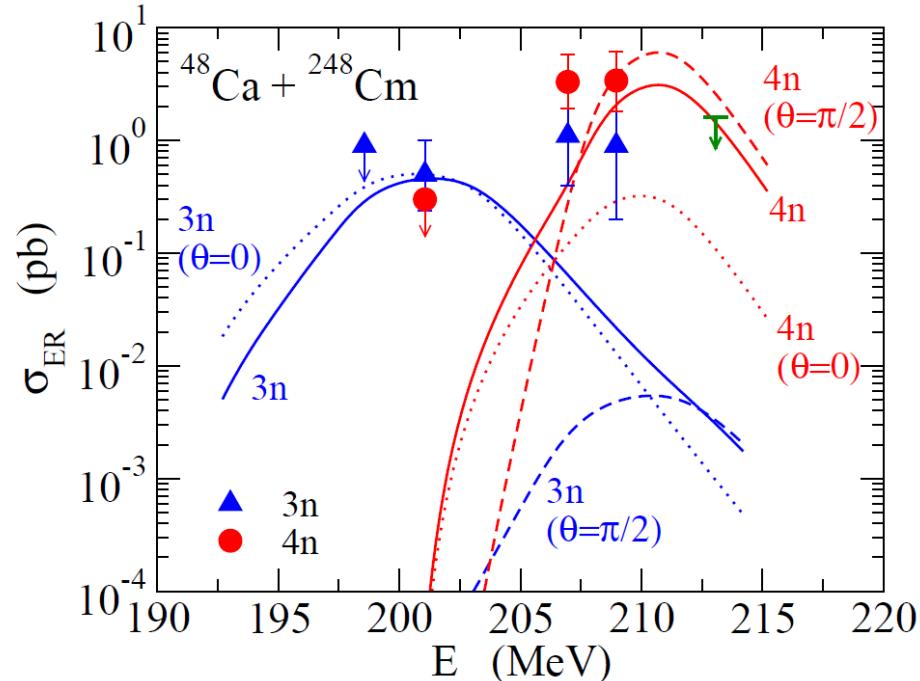
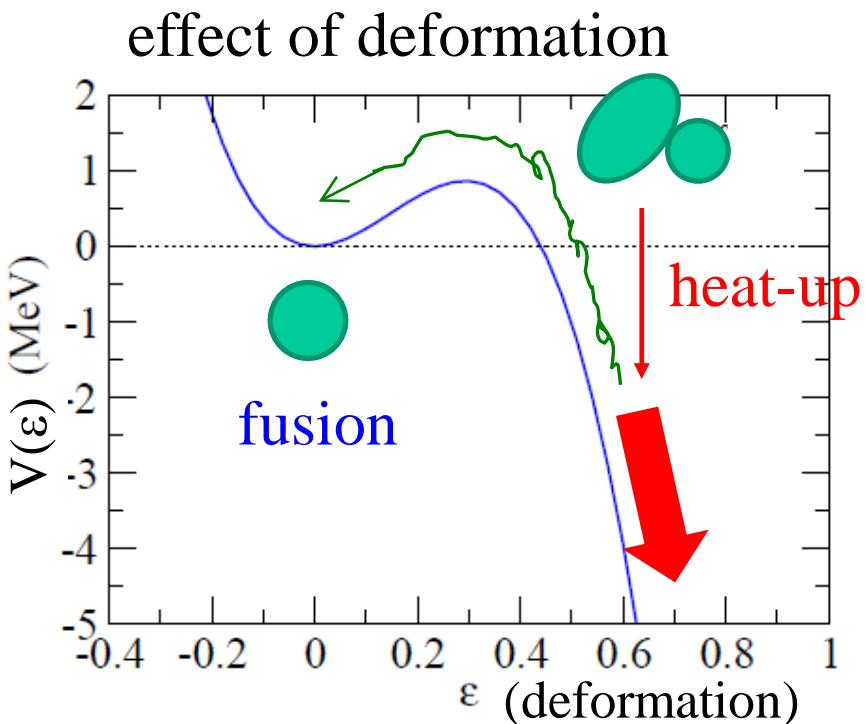
D.J. Hinde et al.,  
PRL74 ('95) 1295



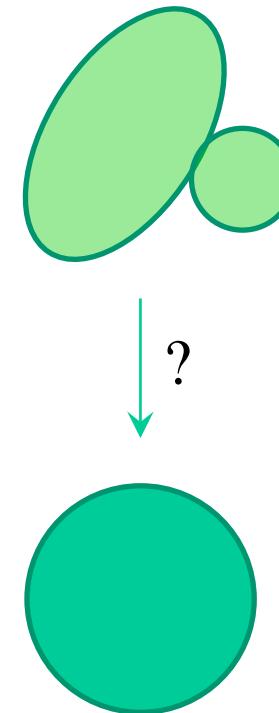
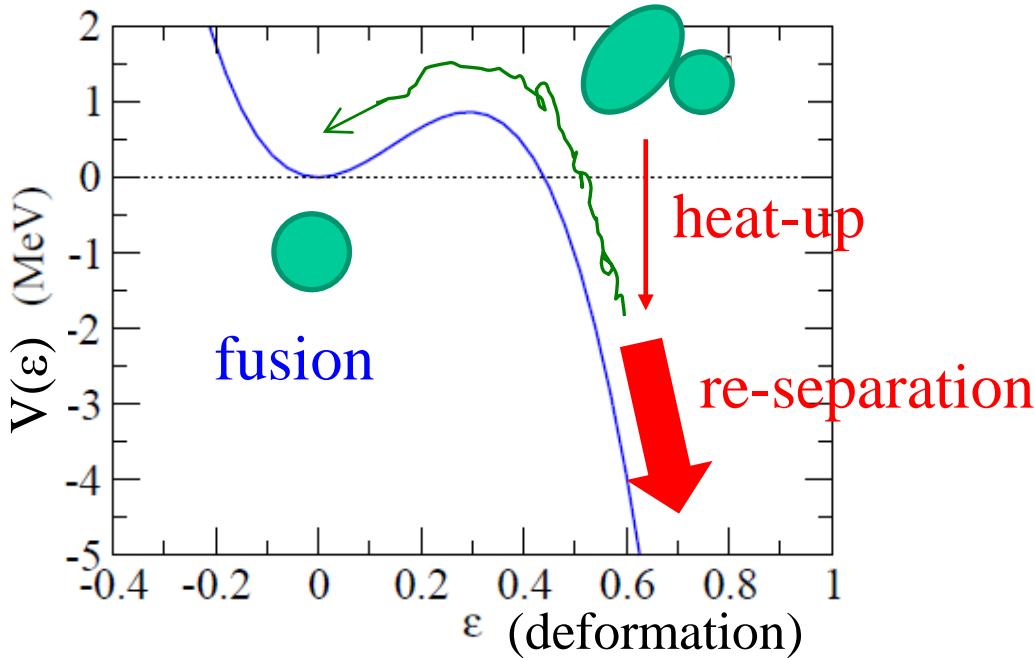
## the 2nd stage of fusion reaction

hot fusion:

$^{48}\text{Ca}$  + deformed target



## open problems



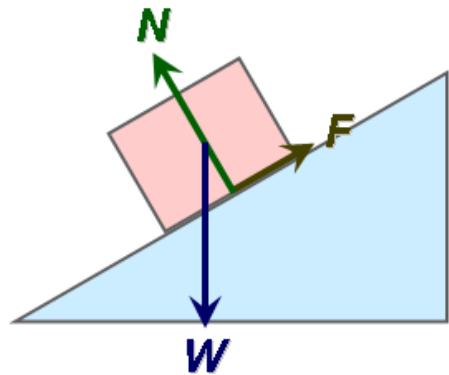
- how is the shape evolved to a compound nucleus?
- Deformation: a quantum effect  
how does the deformation disappear during heat-up?

quantum friction/open quantum systems

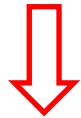
M. Tokieda and K.H., Ann. of Phys. 412 ('20) 168005.  
Front. in Phys. 8 ('20) 8.

# quantum friction

in quantum mechanics:

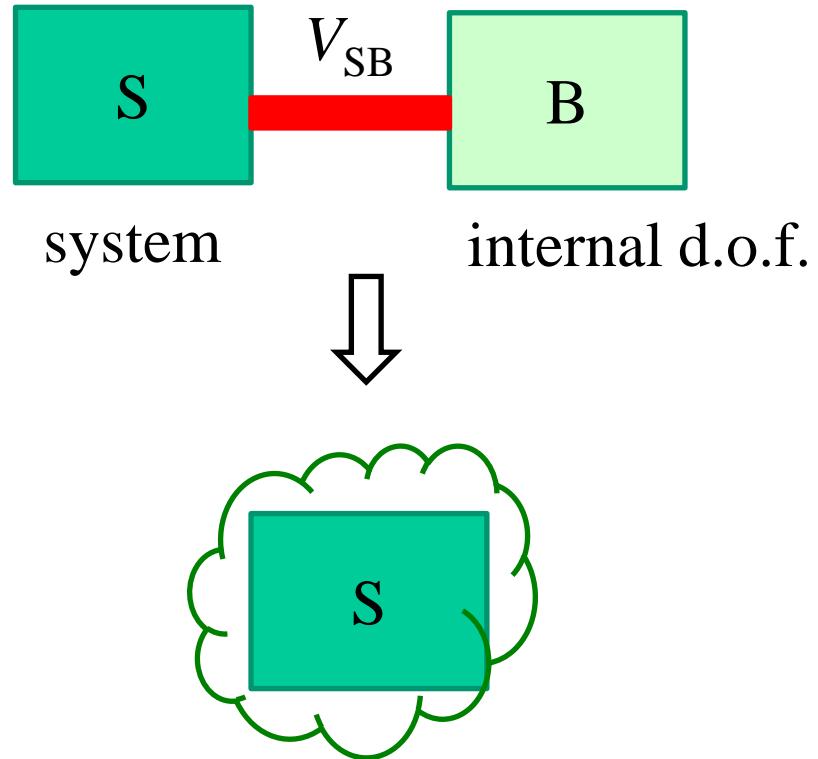


heat generation when a rigid body stops



the energy conversion from the rigid body to intrinsic d.o.f. (atoms)

quantum Langevin?

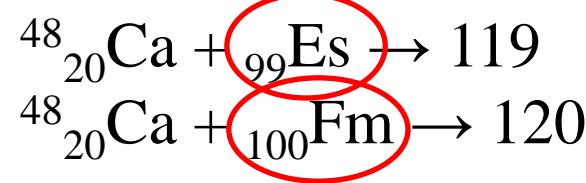


solve the whole H without introducing the quasi-particle

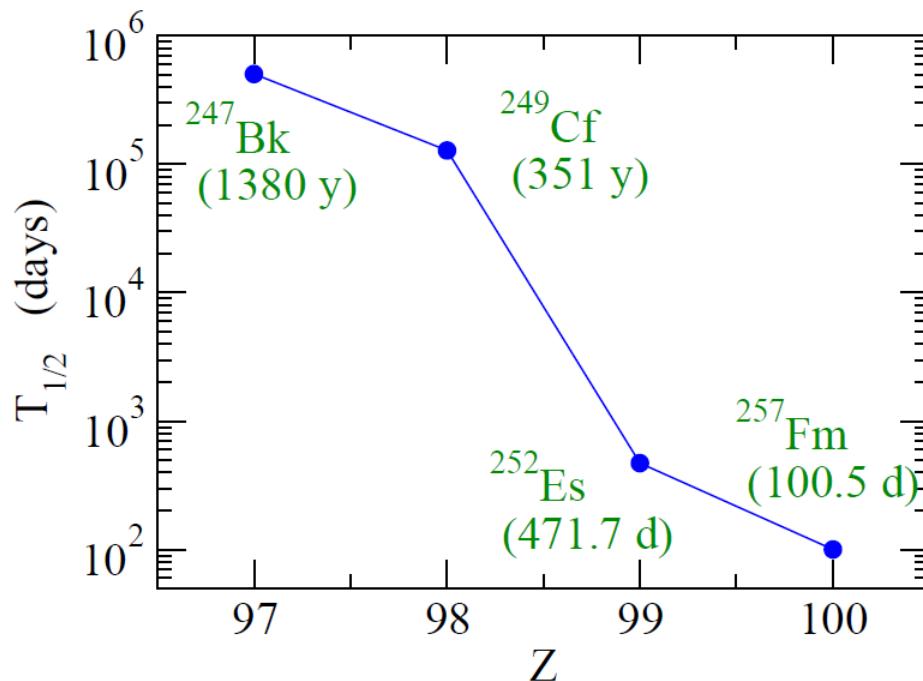
M. Tokieda and K.H. (2020)

## Hot fusion towards Z=119 and 120 nuclei

hot fusion reactions with  $^{48}\text{Ca}$ :



short lived → not available with sufficient amounts



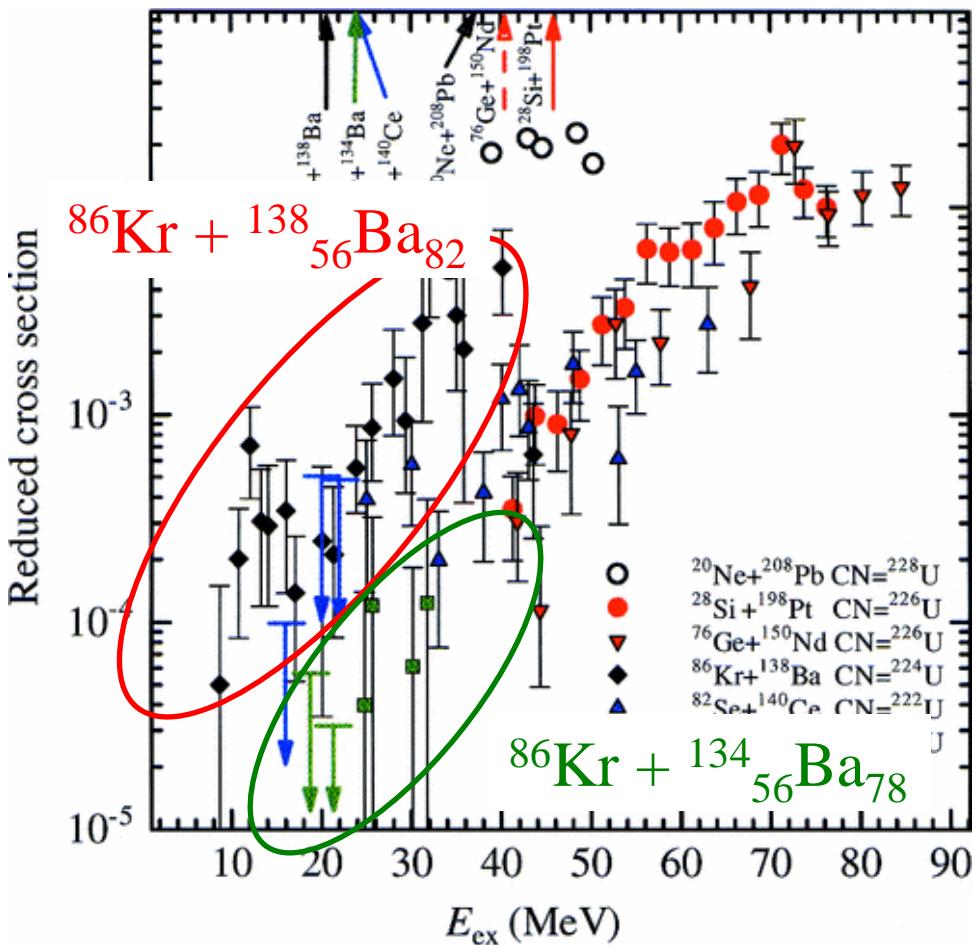
$^{48}\text{Ca} \rightarrow {}_{22}^{50}\text{Ti}, {}_{23}^{51}\text{V}, {}_{24}^{54}\text{Cr}$  projectiles

closed shell → open shells

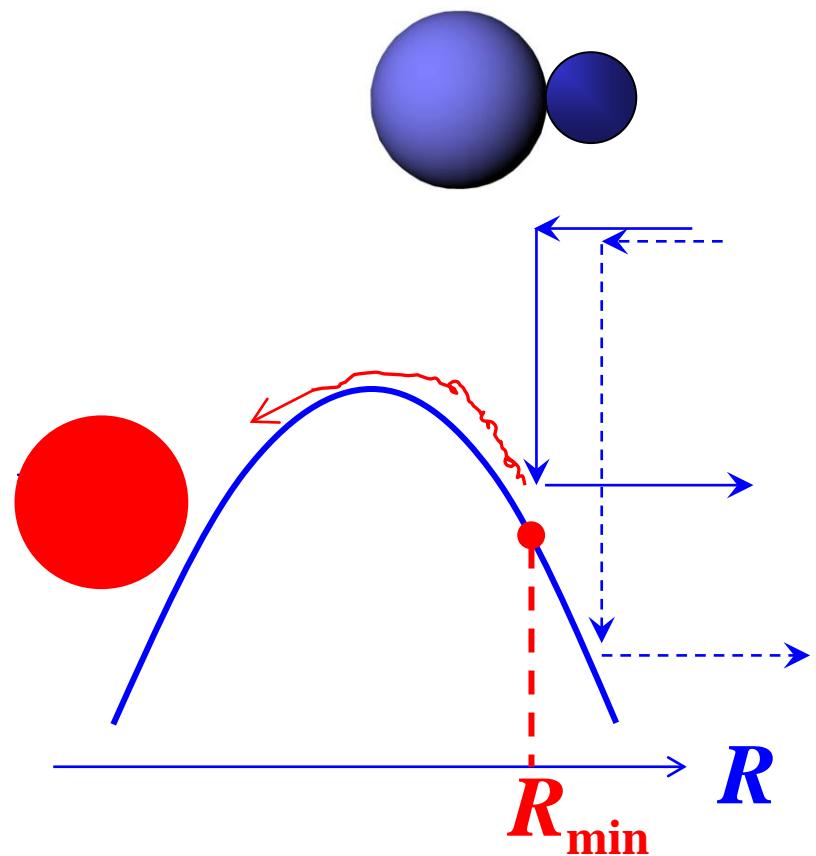
how much will cross sections be affected?

## Role of magicity

can proceed deeper  
with less friction



K. Satou, H. Ikezoe et al.,  
PRC73 ('06) 034609

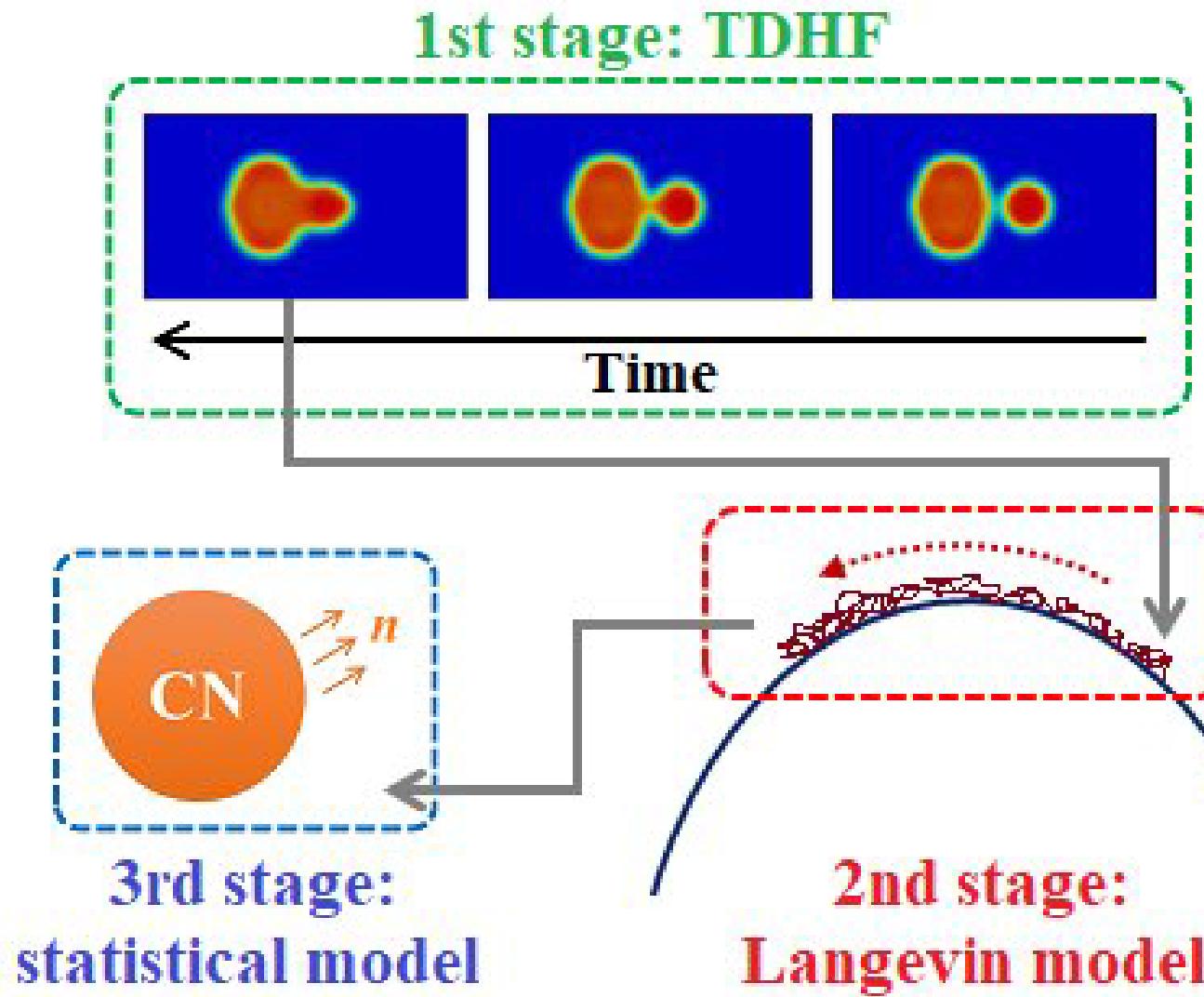


cf. P. Moller et al.,  
Z. Phys. A359 ('97) 251.

similar effect for  $^{48}\text{Ca}$ ?

# New hybrid model: TDHF + Langevin approach

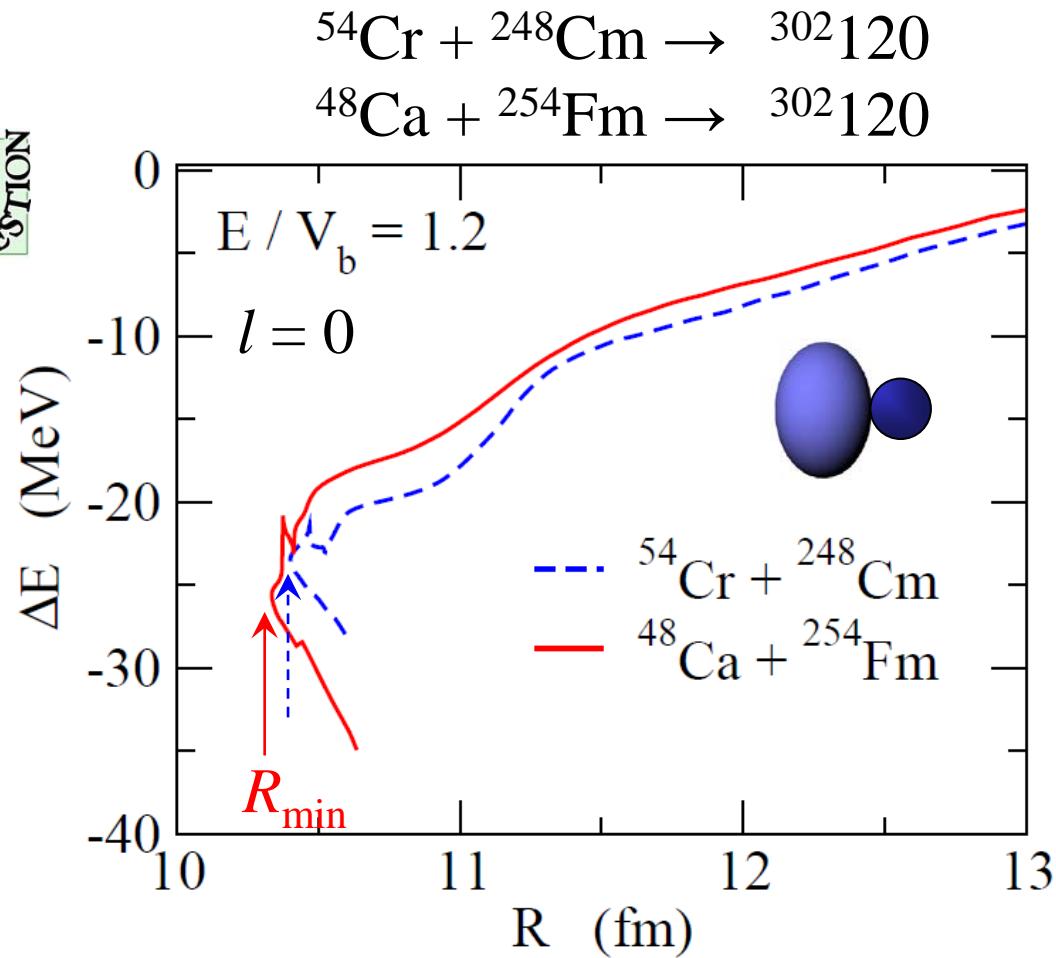
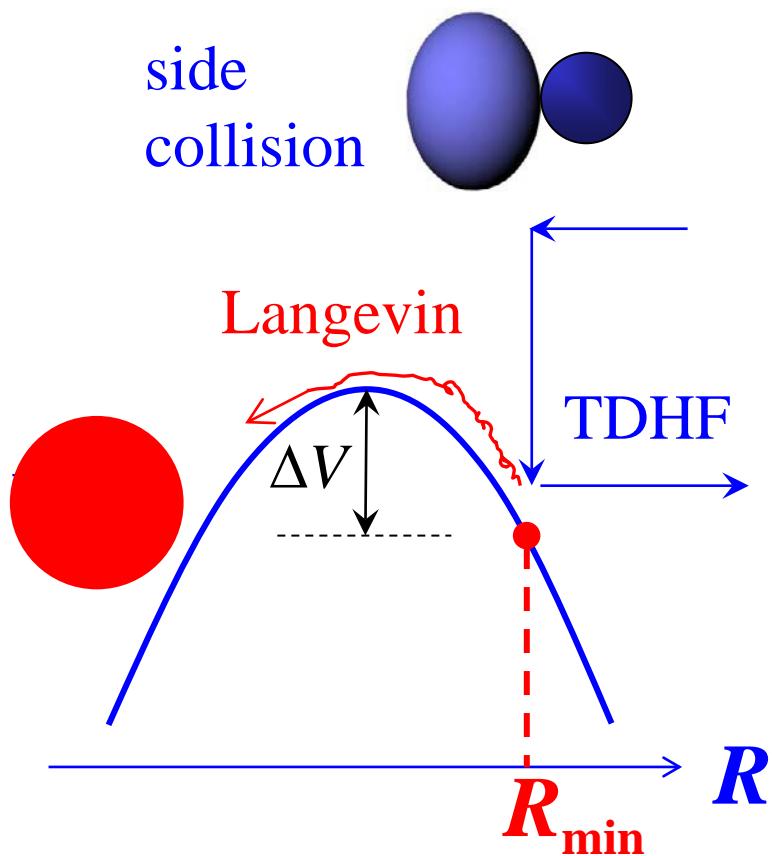
K. Sekizawa and K.H., PRC99 (2019) 051602(R)



## Furthermore : TDHF + Langevin approach and quantum friction

TDHF + Langevin approach :

K. Sekizawa and K. H.,  
PRC99 (2019) 051602(R)

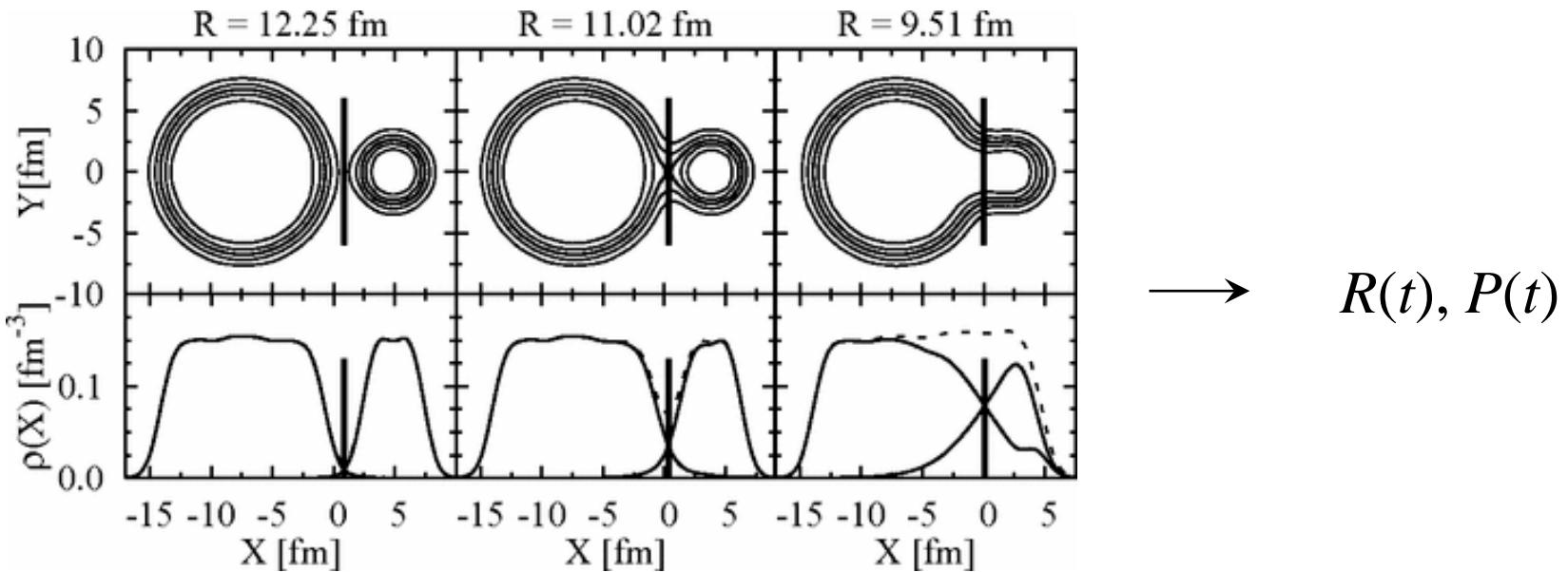


→ Langevin calculation

# Mapping TDHF onto a classical equation of motion

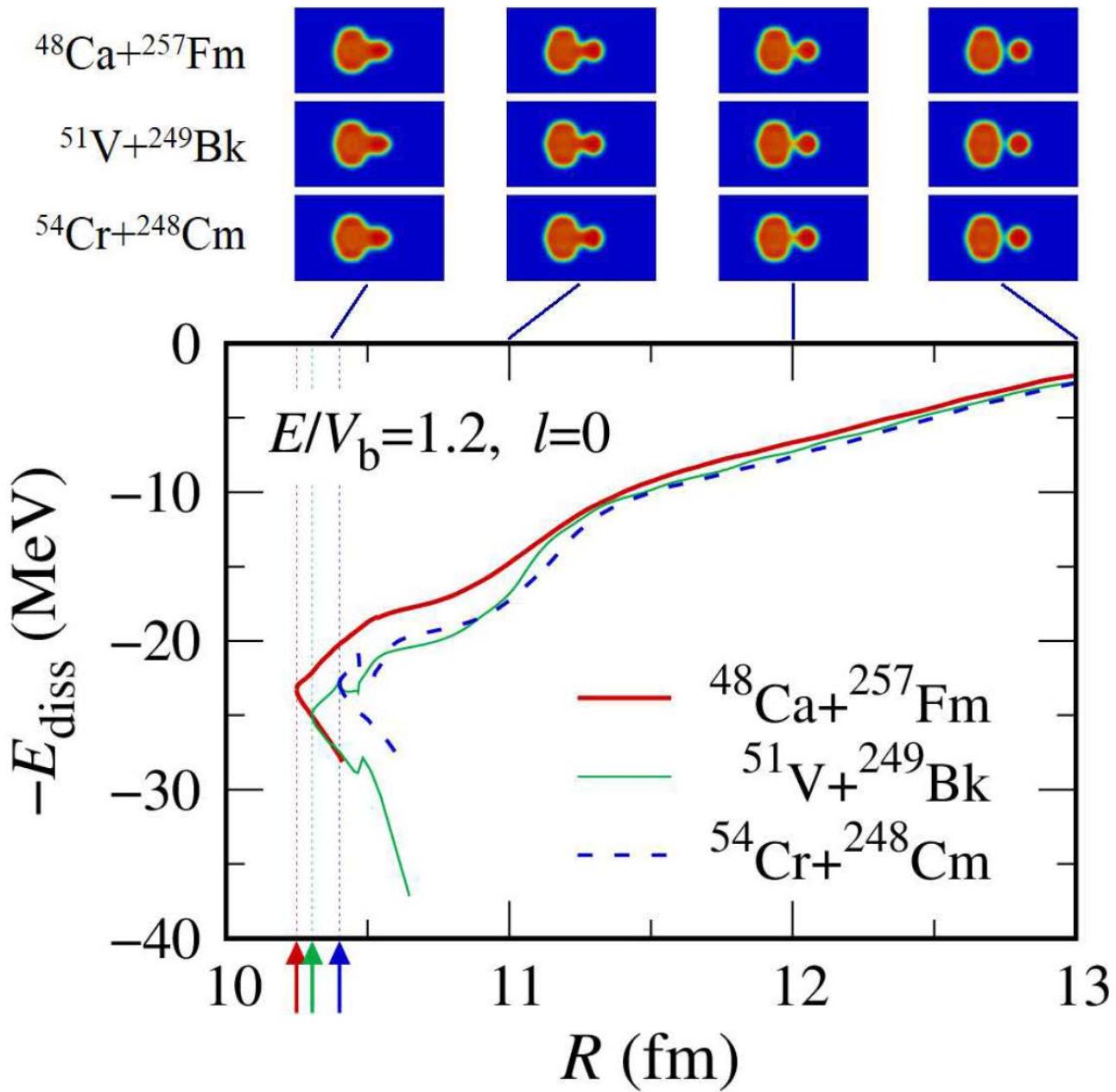
K. Washiyama and D. Lacroix, PRC78 ('08) 024610

## TDHF simulations



$$\dot{P} = -\frac{dV}{dR} - \frac{d}{dR} \left( \frac{P^2}{2\mu} \right) - \gamma \dot{R} \quad \rightarrow \quad V(R), \gamma(R)$$

$$\rightarrow \Delta E(t) = -E_{\text{diss}}(t) = \frac{P(t)^2}{2\mu(R(t))} + V(R(t)) - E_{\text{ini}}$$



# New model for fusion for SHE: TDHF + Langevin approach

K. Sekizawa and K.H., PRC99 (2019) 051602(R)



how special is  $^{48}\text{Ca}$  ?

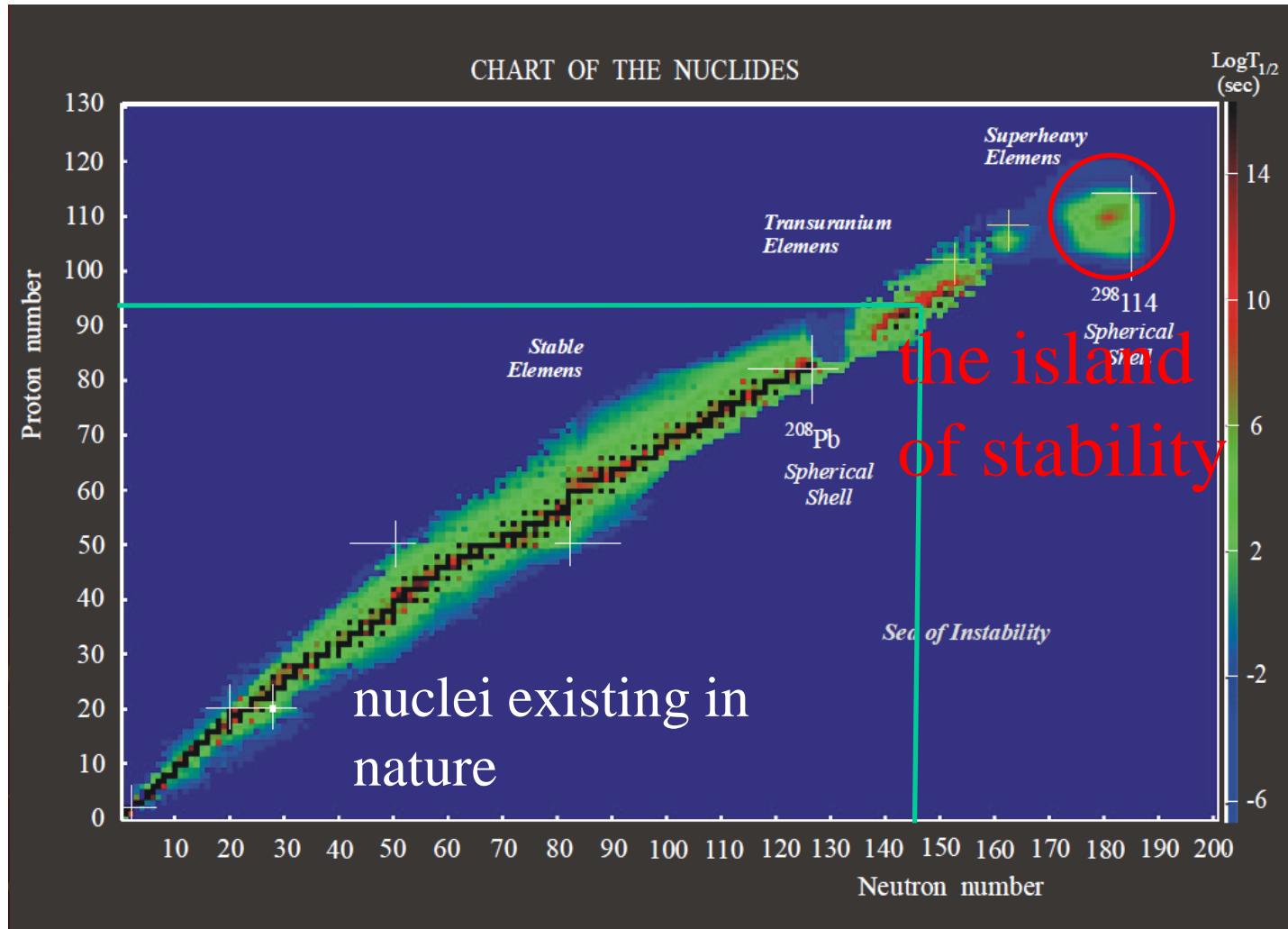
System	CN	$E^*$ (MeV)	$R_{\min}$ (fm)	$P_{\text{CN}}$ ( $\times 10^4$ )	$W_{\text{sur}}$ ( $\times 10^9$ )	$P_{\text{CN}} W_{\text{sur}}$ ( $\times 10^{13}$ )
$^{48}\text{Ca} + ^{254}\text{Fm}$	$^{302}120$	29.0	12.93	1.72	176	302
$^{54}\text{Cr} + ^{248}\text{Cm}$	$^{302}120$	33.2	13.09	1.89	1.31	2.47
$^{51}\text{V} + ^{249}\text{Bk}$	$^{300}120$	37.0	12.94	3.95	0.117	0.461
$^{48}\text{Ca} + ^{257}\text{Fm}$	$^{305}120$	30.5	12.94	2.49	0.729	1.82

$$P_{\text{ER}} = P_{\text{cap}} \cdot P_{\text{CN}} \cdot W_{\text{suv}}$$

similar  $P_{\text{CN}}$

- ✓ no special role of  $^{48}\text{Ca}$  in the entrance channel
- ✓ non- $^{48}\text{Ca}$  proj.: about 2 order of magnitude smaller  
due mainly to  $W_{\text{sur}}$

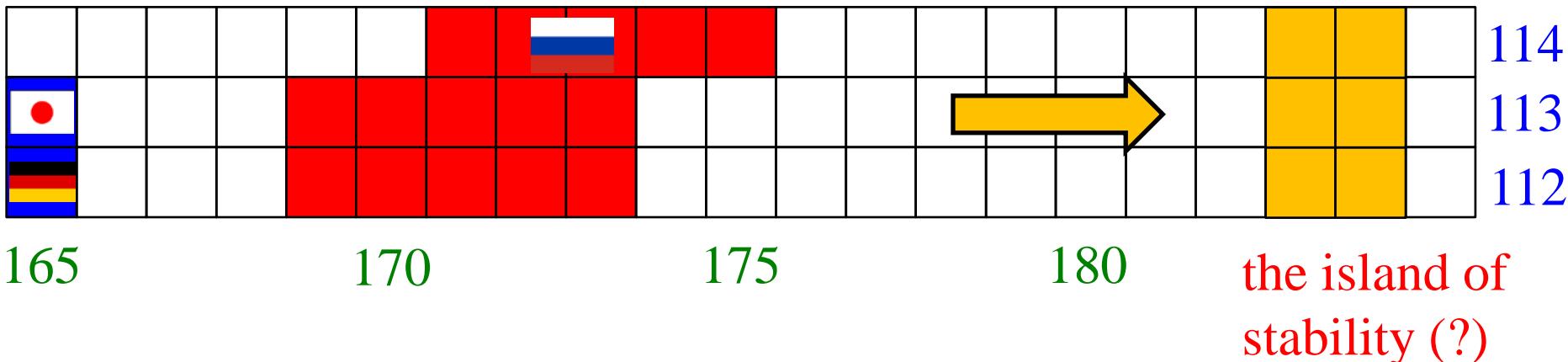
# Another important issue: physics of neutron-rich nuclei



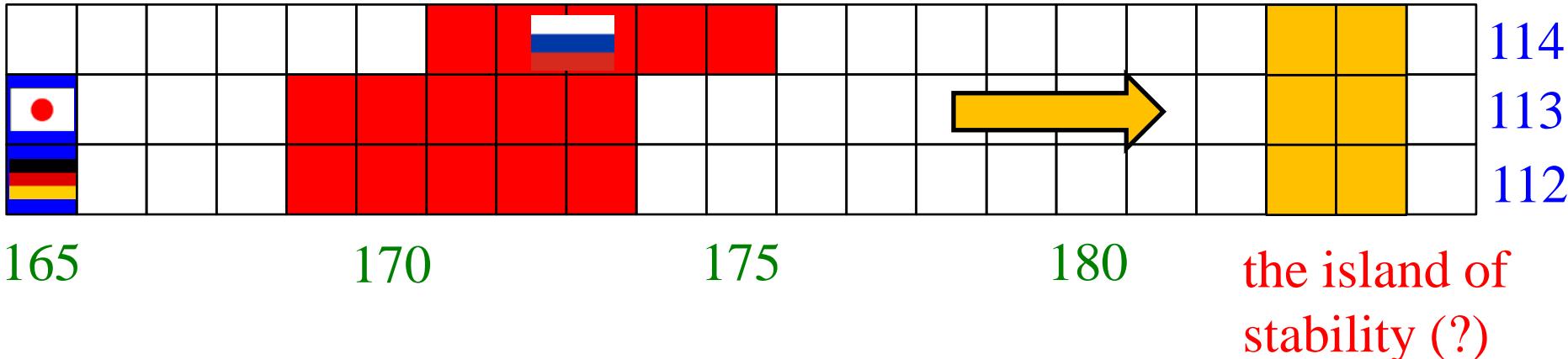
Yuri Oganessian

how to reach the island of stability?

## Fusion of unstable nuclei



# Fusion of unstable nuclei



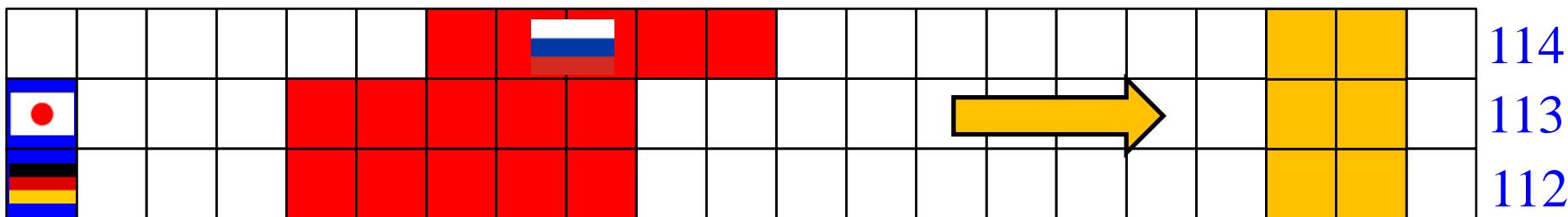
neutron-rich beams: indispensable

- how to deal with low beam intensity?
- reaction dynamics of neutron-rich beams?
  - ✓ capture: role of breakup and (multi-neutron) transfer?
  - ✓ diffusion: neutron emission during a shape evolution?
  - ✓ survival: validity of the statistical model?

structure of exotic nuclei

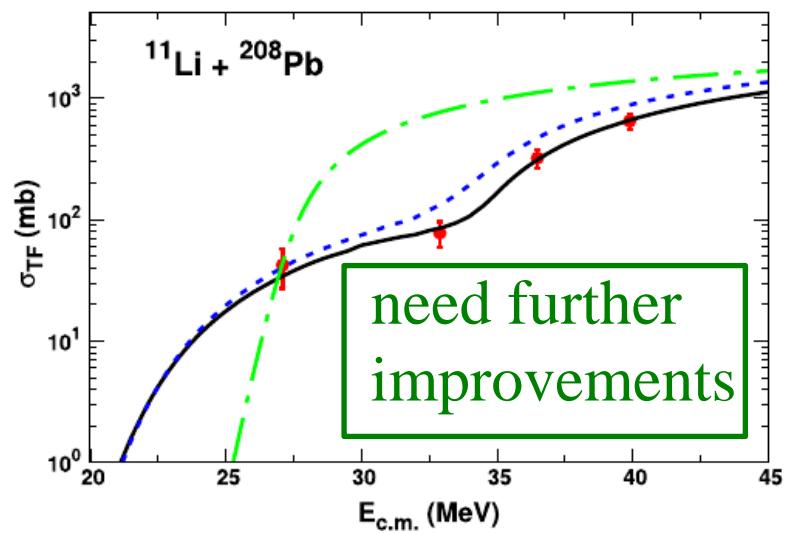
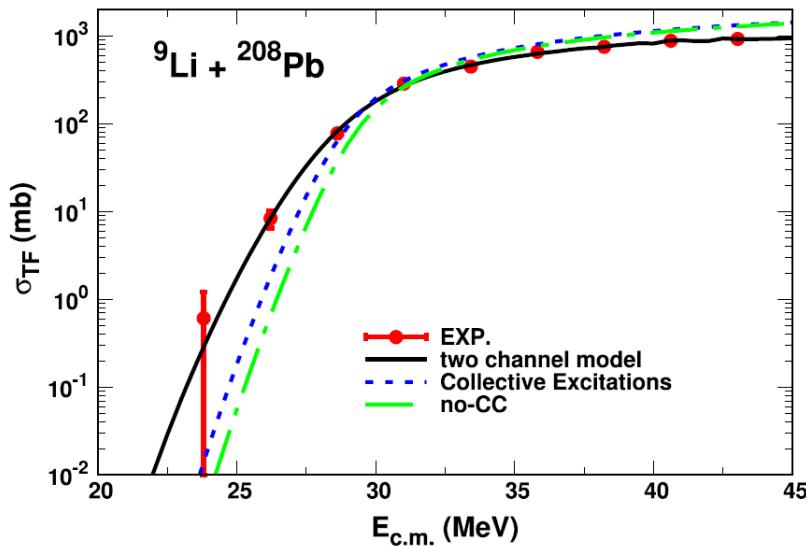
more studies are required

# Fusion of unstable nuclei



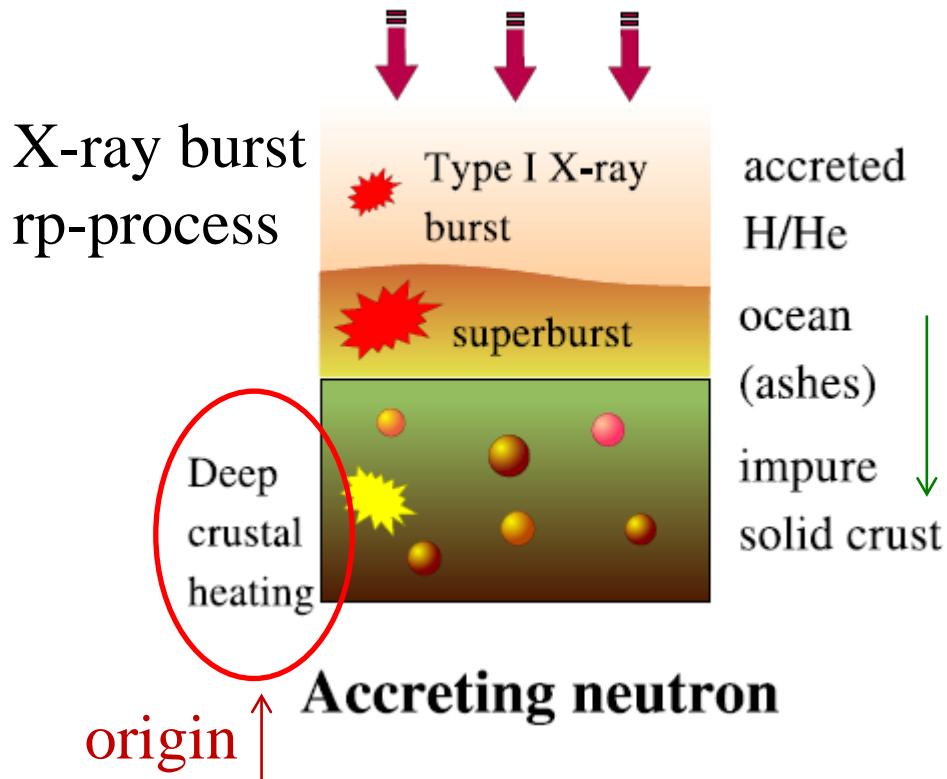
neutron-rich beams: indispensable  
→ reaction dynamics?

K.-S. Choi, K. Hagino et al.,  
Phys. Lett. B780 ('18) 455



simultaneous explanation for  ${}^9\text{Li} + {}^{208}\text{Pb}$  and  ${}^{11}\text{Li} + {}^{208}\text{Pb}$  with:  
 ${}^{11}\text{Li} + {}^{208}\text{Pb} \longleftrightarrow {}^9\text{Li} + {}^{210}\text{Pb} \longleftrightarrow {}^7\text{Li} + {}^{212}\text{Pb}$  transfer couplings

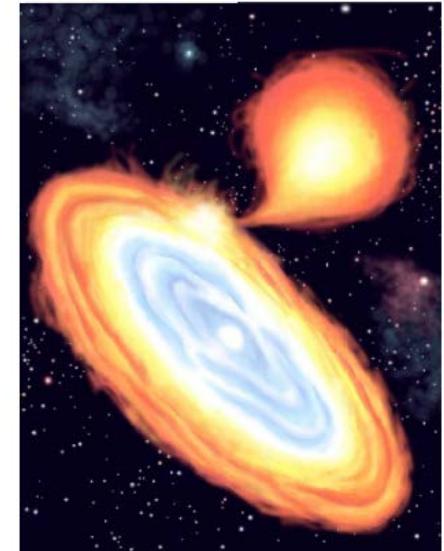
# fusion of neutron-rich nuclei in accreting (質量降着) neutron stars



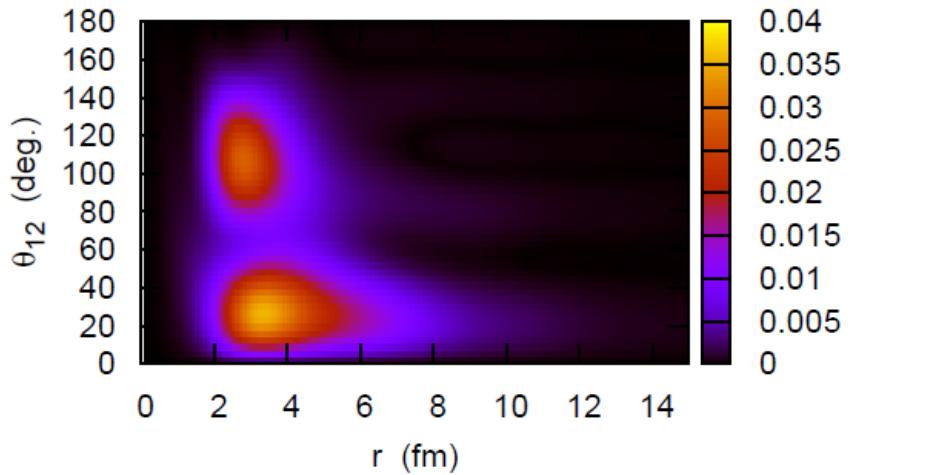
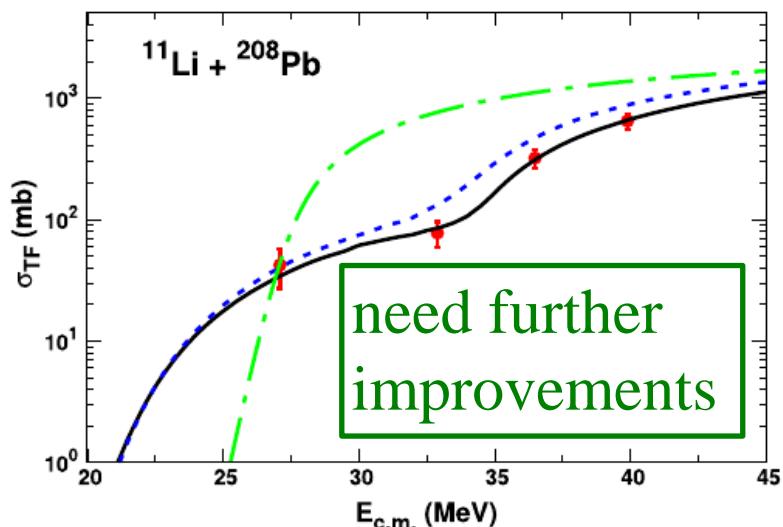
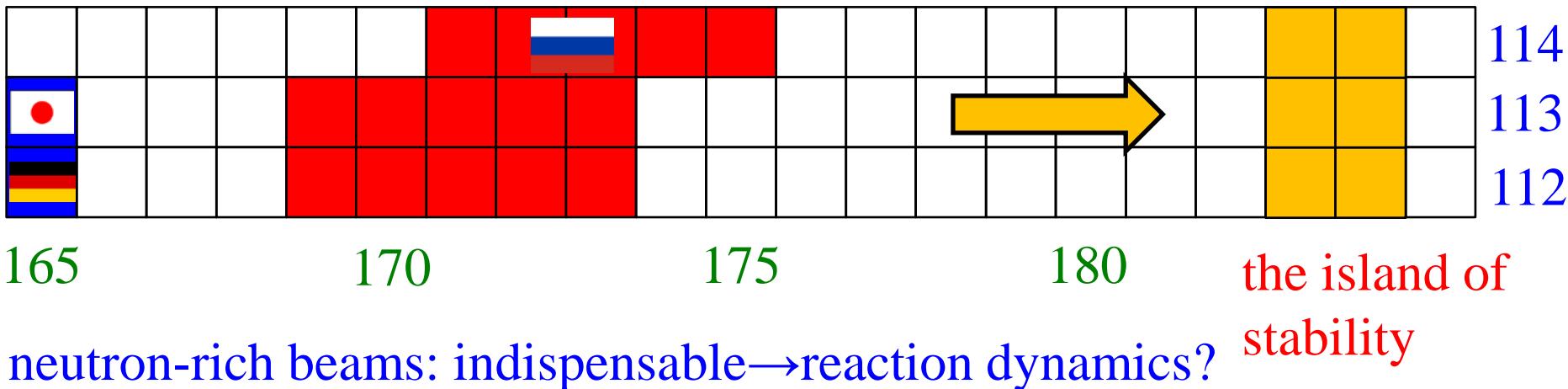
fusion of neutron-rich nuclei  
when Z becomes small enough



electron capture  
 $(A,Z) + e^- \rightarrow (A,Z-1) + \nu_e$   
towards neutron-rich nuclei



# Fusion of unstable nuclei



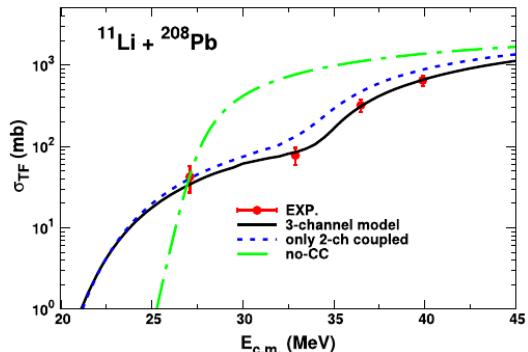
K.H. and H. Sagawa, PRC72('05)044321

K.-S. Choi, K. Hagino et al.,  
Phys. Lett. B780 ('18) 455

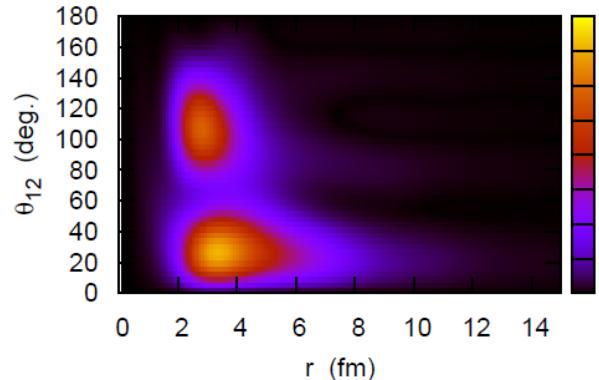
good understandings of the structure  
of neutron-rich nuclei is also important

# Physics of SHE

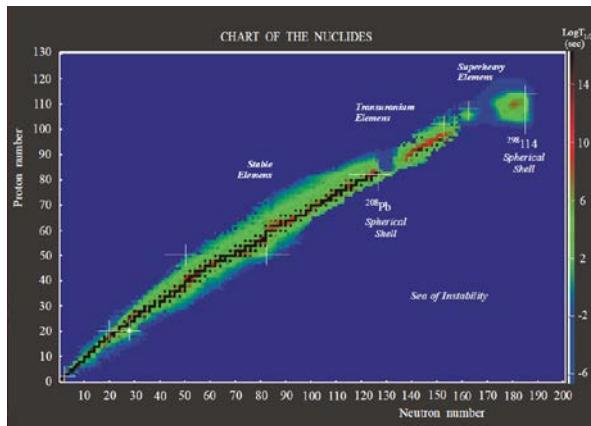
Reactions of n-rich nucl.



Structure of n-rich nucl.



- SHE
- the island of stability



open quantum systems (OQS)

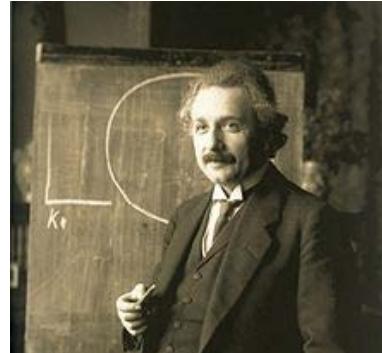
SHE + neutron-rich nuclei + OQS → new direction

# Chemistry of superheavy elements

Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
↓ Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	H												B	C	N	O	F	He	
2	Li	Be											Al	Si	P	S	Cl	Ne	
3	Na	Mg											In	Ge	As	Se	Br	Ar	
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Sn	As	Se	Br	Kr	
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
6	Cs	Ba	La	*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	Rn	
7	Fr	Ra	Ac	*	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og
	*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
	*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 S	100 Fm	101 Md	102 No	103 Lr				

- Are they here in the periodic table?
- Does Nh show the same chemical properties as B, Al, Ga, In, and Tl?

relativistic effect : important for large Z



$$E = mc^2$$

Solution of the Dirac equation (relativistic quantum mechanics) for a hydrogen-like atom:

$$E_{1S} = mc^2 \sqrt{1 - (Z\alpha)^2} \sim mc^2 \left( 1 - \frac{(Z\alpha)^2}{2} - \frac{(Z\alpha)^4}{8} + \dots \right)$$

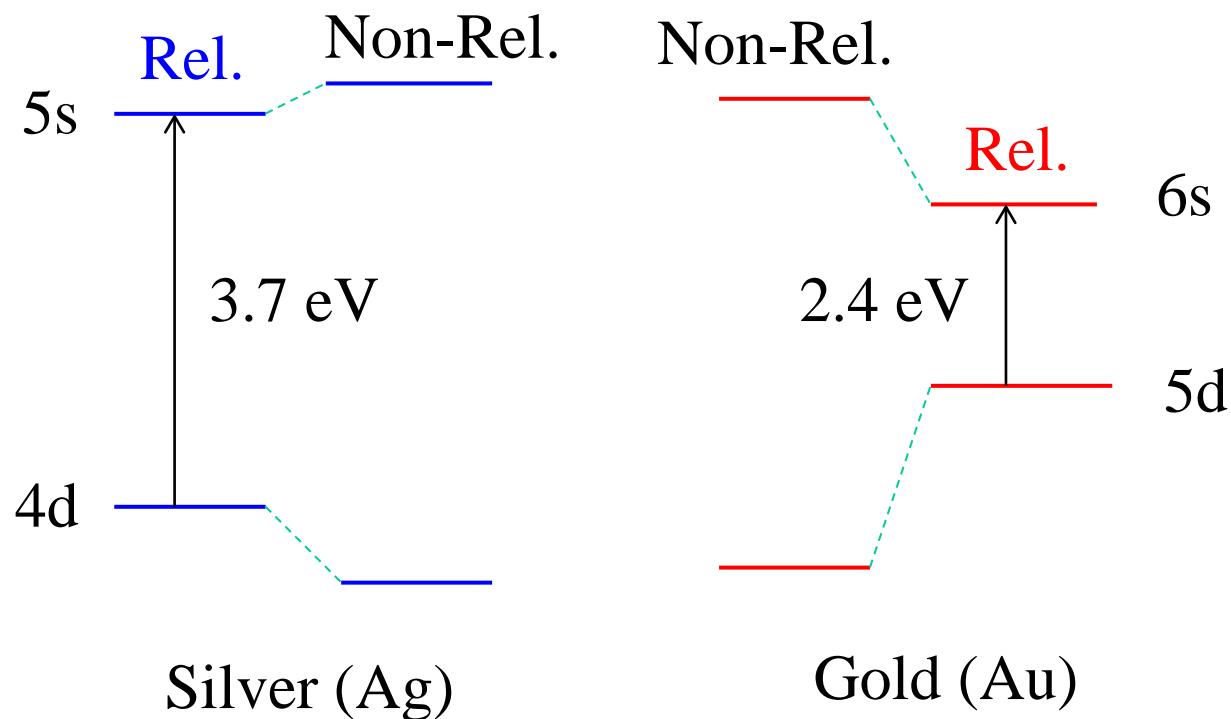
relativistic effect

# Famous example of relativistic effects: the color of gold

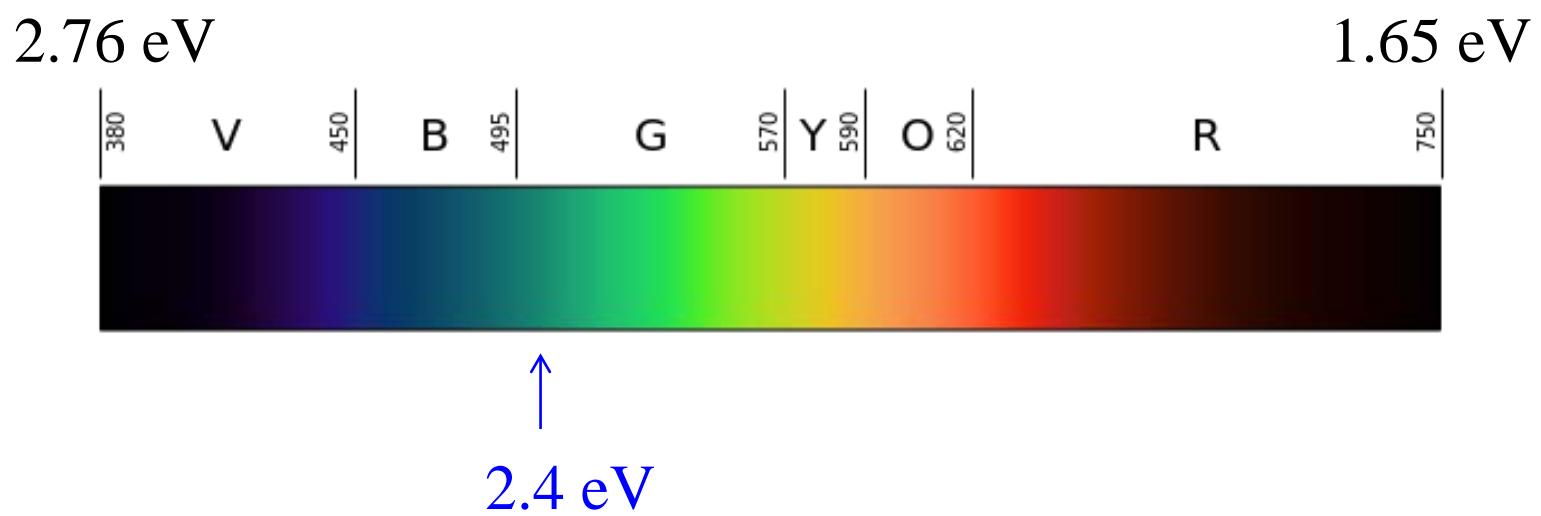
1	1 H																2 He	
2	3 Li	4 Be															10 Ne	
3	11 Na	12 Mg															18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra		104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Uut	114 Fl	115 Uup	116 Lv	117 Uus	118 Uuo

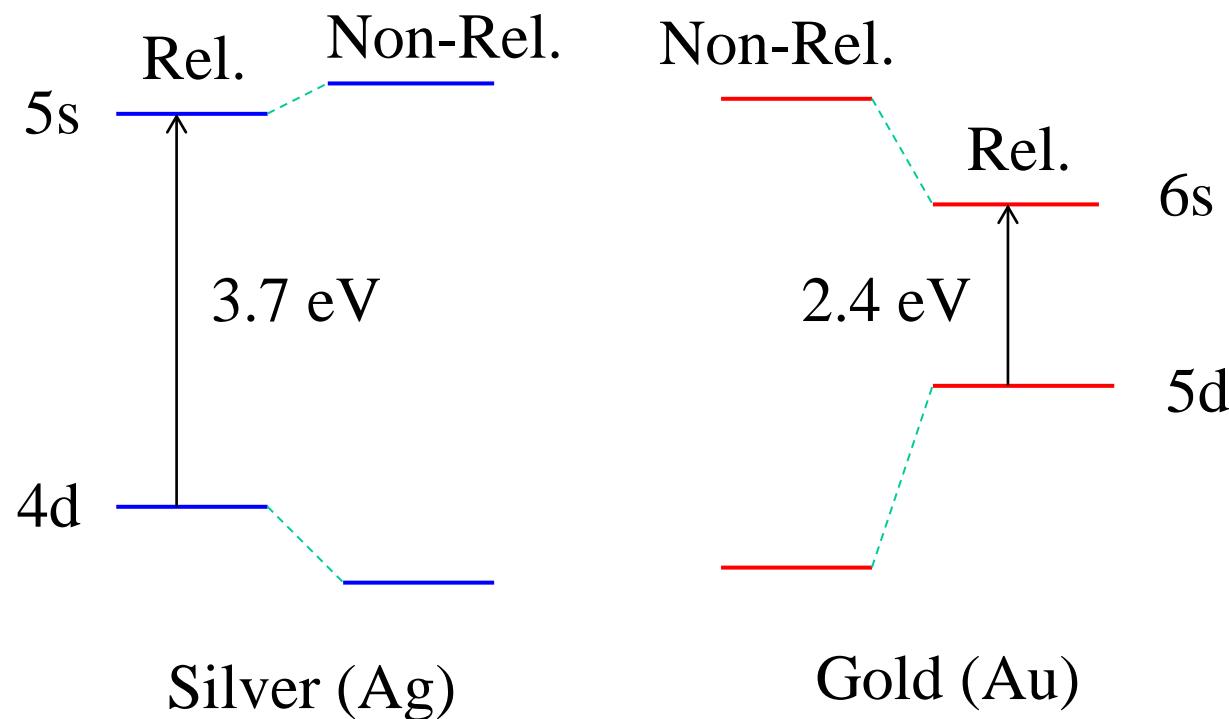


Gold looked like silver if there was no relativistic effects!

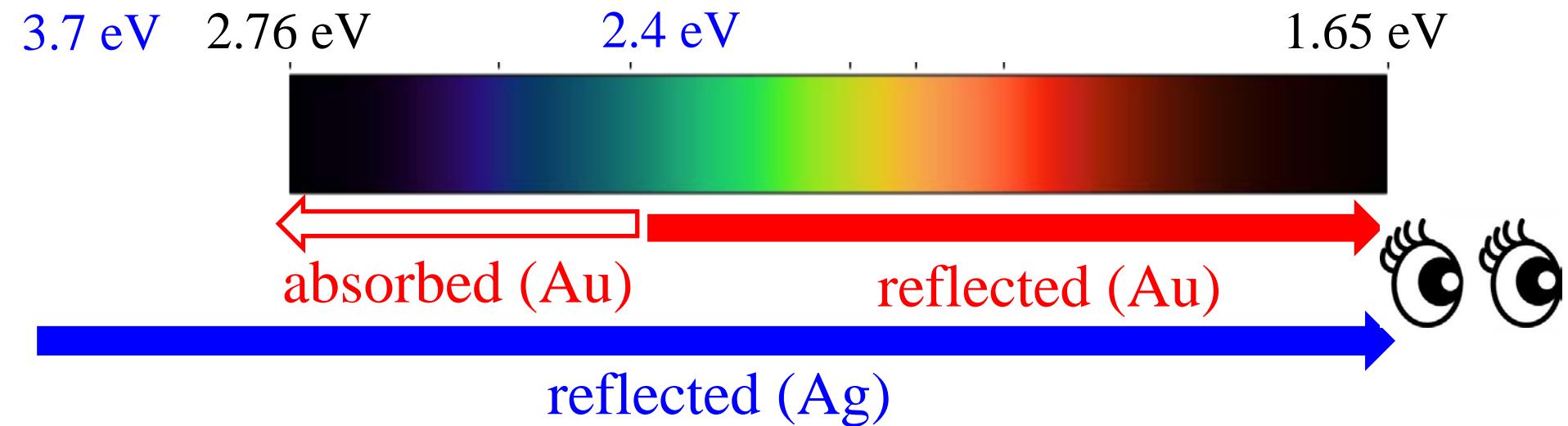


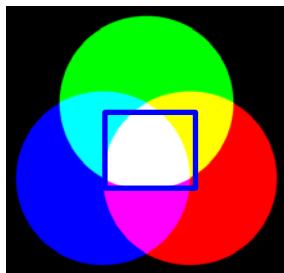
cf. visible spectrum



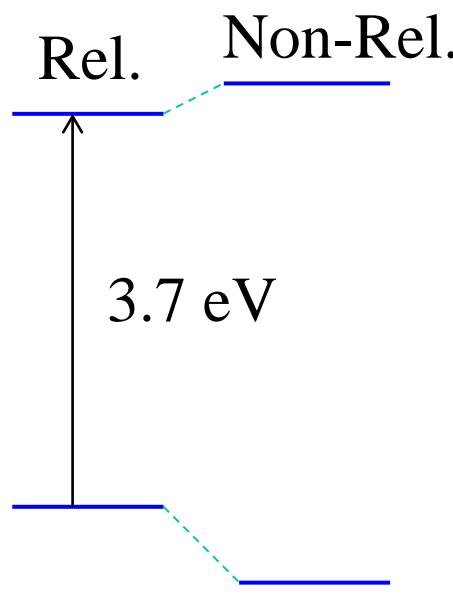


cf. visible spectrum

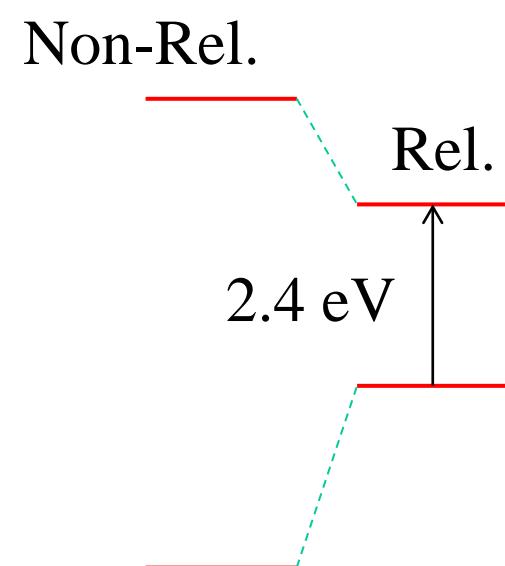




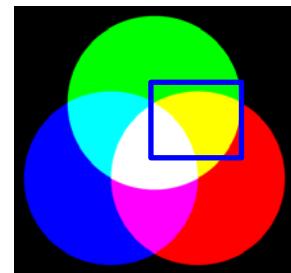
no color  
absorbed



Silver (Ag)



Gold (Au)



blue: absorbed



Ag



Au

# Chemistry of superheavy elements

Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
↓ Period																		
1	1 H																2 He	
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
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6	55 Cs	56 Ba		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra		104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Uut	114 Fl	115 Uup	116 Lv	117 Uus	118 Uuo

Lanthanides

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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Actinides

89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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How do the relativistic effects alter the periodic table for SHE?

→ a big open question

# Summary

## SHE: quantum many-body systems with a strong Coulomb field

physics

113 <b>Nh</b> nihonium	115 <b>Mc</b> moscovium
117 <b>Ts</b> tennessine	118 <b>Og</b> oganesson

reaction dynamics

- ✓ quantum friction
- ✓ neutron-rich nuclei

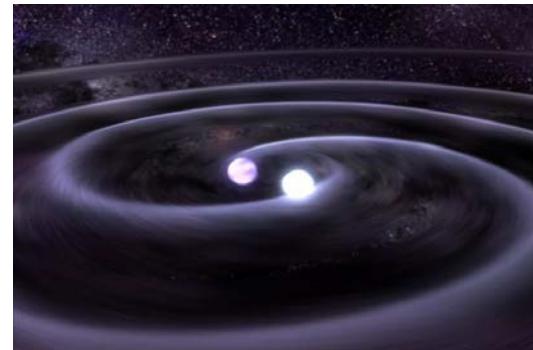
chemistry

Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
↓ Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
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5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	57 La	*	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Os	81 Tl	82 Pb	83 Bi	84 Po	85 At	
7	87 Fr	88 Ra	89 Ac	*	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
				*	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	
				*	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	



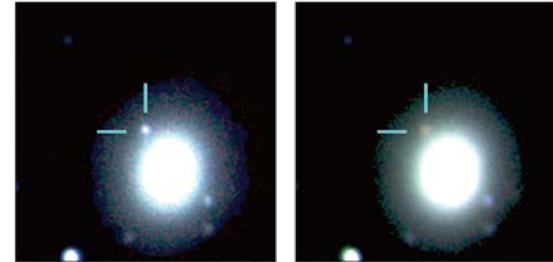
International Year  
of the Periodic Table  
of Chemical Elements

astronomy



2017.08.18-19

2017.08.24-25



- ✓ origin of elements
- ✓ r-process
- ✓ kilonova

interdisciplinary SHE science



# New method for open quantum systems

M. Tokieda and K.H., Ann. of Phys., in press ('19)

$$H_{\text{tot}} = H_S + \sum_i \hbar \omega_i a_i^\dagger a_i + h(q) \sum_i d_i (a_i^\dagger + a_i)$$



system      environment    coupling

naiive coupled-channels equations:

$$\Psi_{\text{tot}}(q, t) = \sum_{\{n_i\}} \psi_{\{n_i\}}(q, t) |\{n_i\}\rangle; \quad |\{n_i\}\rangle = \prod_i \frac{1}{\sqrt{n_i!}} (a_i^\dagger)^{n_i} |0\rangle$$

$$\rightarrow \langle \{n_i\} | i \hbar \frac{\partial}{\partial t} | \Psi_{\text{tot}} \rangle = \langle \{n_i\} | H_{\text{tot}} | \Psi_{\text{tot}} \rangle$$

→ coupled-channels eqs. for  $\psi_{\{n_i\}}(q, t)$

difficult when the number of environmental osc. modes is large

$$H_{\text{tot}} = H_S + \sum_i \hbar \omega_i a_i^\dagger a_i + h(q) \sum_i d_i (a_i^\dagger + a_i)$$

→ introduce more efficient basis

$$e^{-i\omega t} \sim \sum_{k=1}^K \eta_k(\omega) u_k(t)$$

exp.      basis  
coef.

cf. correlation function

$$L(t) = \int_{-\infty}^{\infty} d\omega \frac{J(\omega)}{1 - e^{-\beta \hbar \omega}} e^{-i\omega t}$$

cf. 階層型運動方程式

Y. Tanimura and R. Kubo,  
J. Phys. Soc. Jpn 58 ('89)101

$$b_k^\dagger = \sum_i \frac{d_i}{\hbar} \eta_k(\omega_i) a_i^\dagger$$

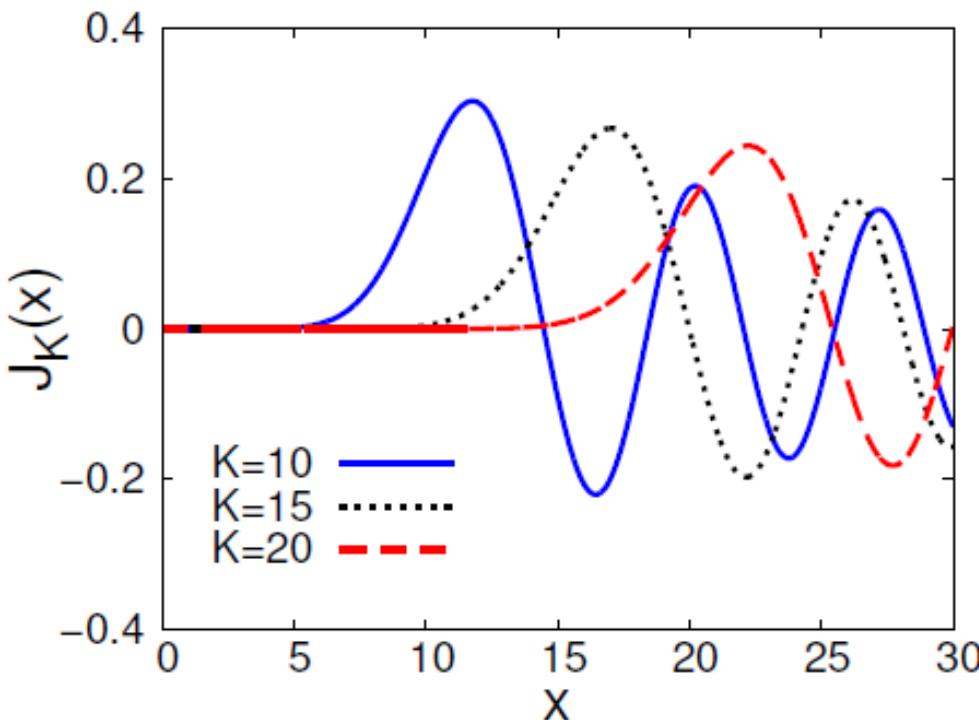
cf. Lanczos method

$$\Psi_{\text{tot}}(q, t) = \sum_{\{\tilde{n}_k\}} \tilde{\psi}_{\{\tilde{n}_k\}}(q, t) |\{\tilde{n}_k\}\rangle; \quad |\{\tilde{n}_k\}\rangle = \prod_{k=1}^K \frac{1}{\sqrt{\tilde{n}_k!}} (b_k^\dagger)^{\tilde{n}_k} |0\rangle$$

$$e^{-i\omega t} \sim \sum_{k=1}^K \eta_k(\omega) u_k(t) \quad \longrightarrow \quad b_k^\dagger = \sum_i \frac{d_i}{\hbar} \eta_k(\omega_i) a_i^\dagger$$

in actual calc.: expansion with Bessel function (Jacobi-Anger identity):

$$e^{-i\omega t} = J_0(\Omega t) + 2 \sum_{k=1}^{\infty} (-i)^k T_k \left( \frac{\omega}{\Omega} \right) J_k(\Omega t)$$



Chebyshev polynomials

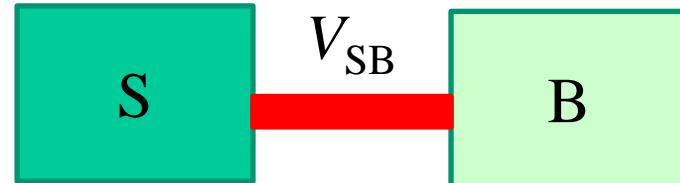
large  $k$ : does not contribute  
when  $t$  is small

$$J_k(x) = \sum_{s=0}^{\infty} \frac{(-1)^s}{s!(s+k)!} \left( \frac{x}{2} \right)^{k+2s}$$

## modelling of open quantum systems

i) system + bath

$$H = H_S + H_B + V_{SB}$$



- ✓ Caldeira-Leggett
- ✓ Feynmann-Vernon

solution:

a) eliminate B (bath)  
→ eff. action for S  
(influence functional)

b)  $\rho_S = Tr_B[\rho]$   
→  $i\hbar\dot{\rho}_S = \dots$

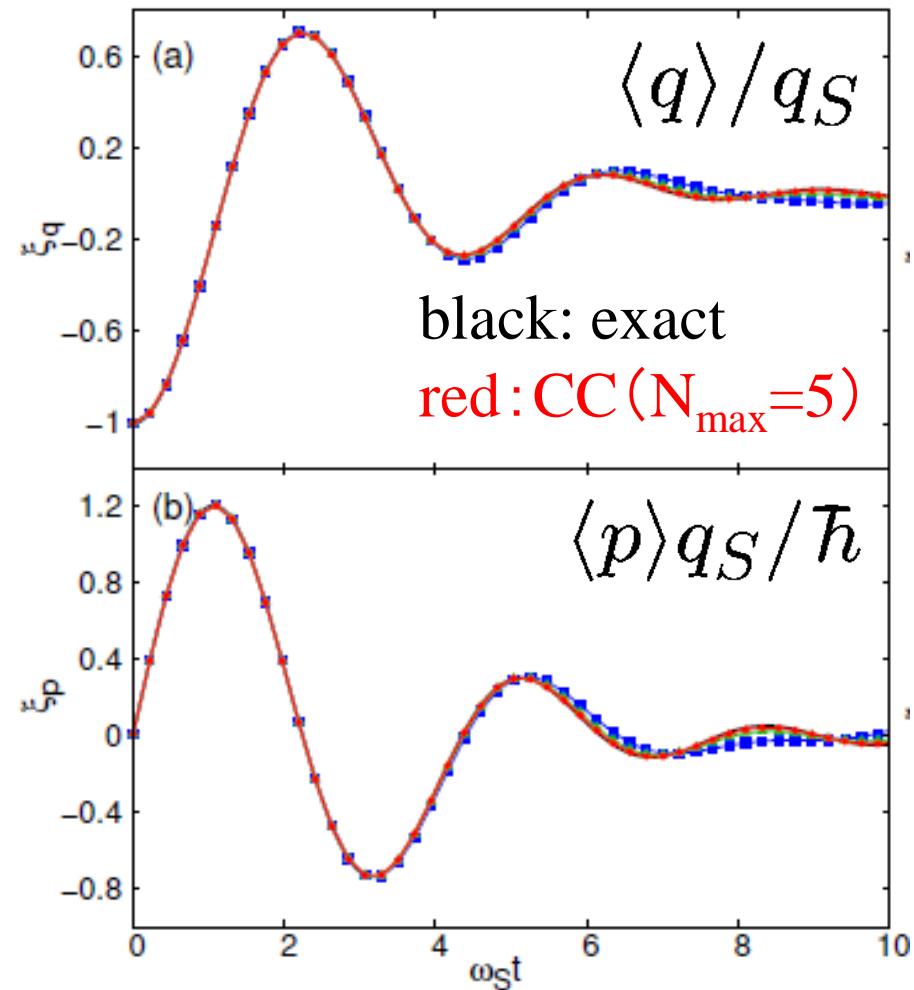
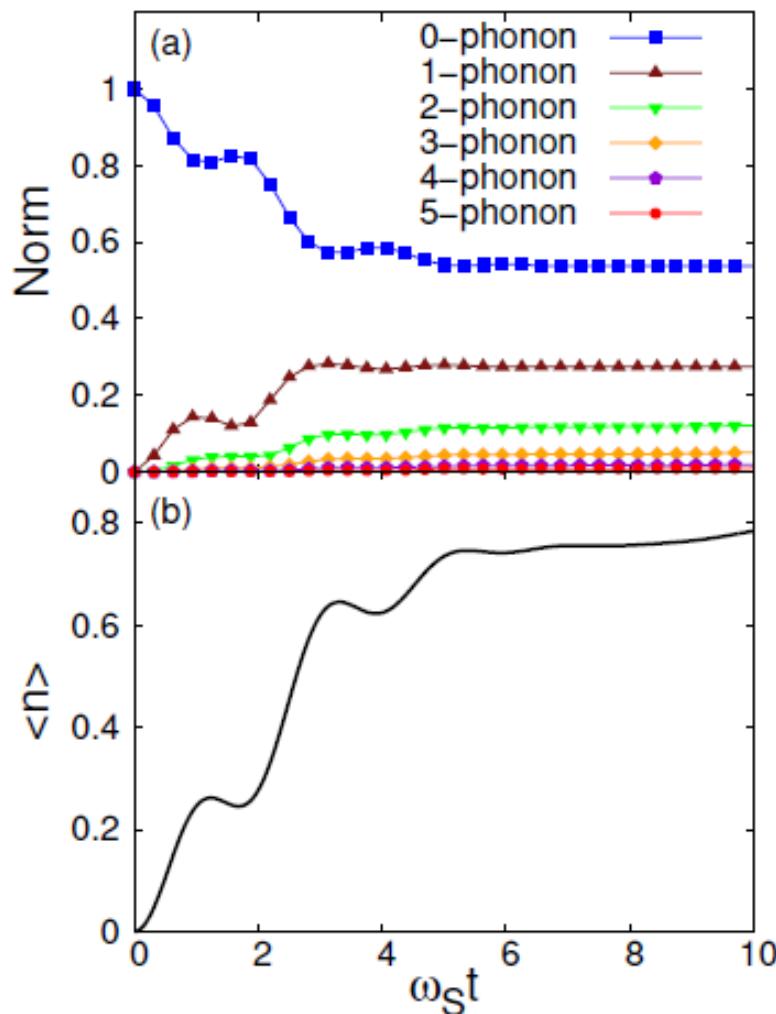
Markovian approximation  
→ Lindblad equation

c) expand the tot. wf with the eigen states of  $H_B$  (coupled-channels eq.)

M. Tokieda and K.H.,  
Ann. of Phys. 412 ('20) 168005.  
Front. in Phys. 8 ('20) 8.

# application to damped oscillation: HO + environment (conti. spectrum)

expansion with Bessel functions up to K=20



## modelling of open quantum systems

i) system + bath

$$H = H_S + H_B + V_{SB}$$

- ✓ Caldeira-Leggett
- ✓ Feynmann-Vernon

solution:

- a) eff. action for S  
(influence functional)
- b) Lindlad eq.
- c) coupled-channels eqs.

- microscopic
- but, hard to solve

ii) quantum friction model

construct a Hamiltonian which leads to

$$\frac{d}{dt}\langle p \rangle = -\langle V'(x) \rangle - \gamma \langle p \rangle$$

- ✓ E. Kanai, PTP 3 ('48)
- ✓ M.D. Kostin, JCP 57 ('72)
- ✓ K. Albrecht, PLB56 ('75)
- ✓ K.-K. Kan & J.-J. Griffin,  
PLB50 ('74)
- ✓ A. Bulgac, S. Jin, and I. Stetcu,  
PRC100, 014615 (2019)

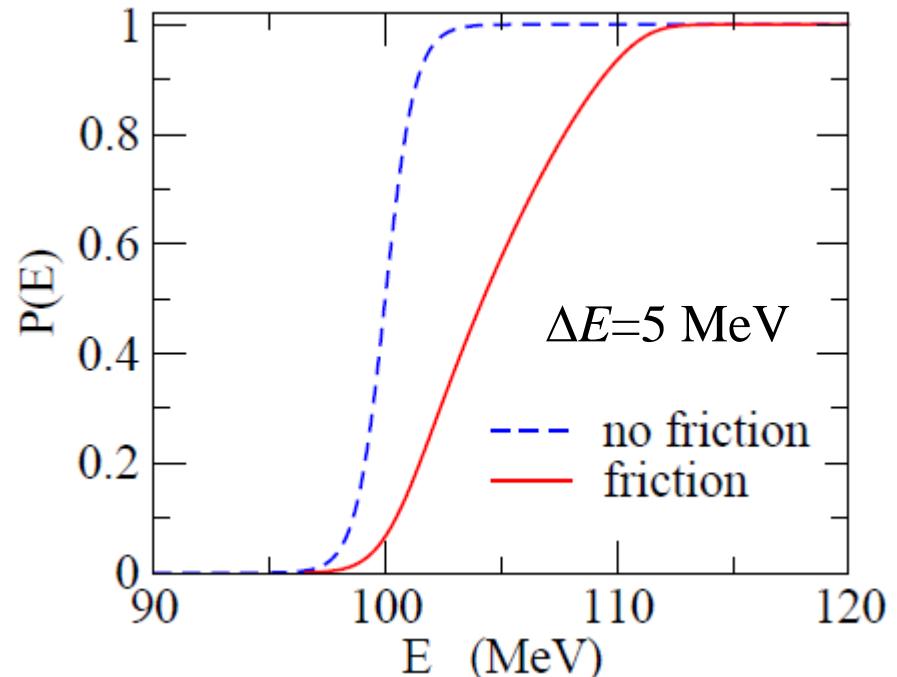
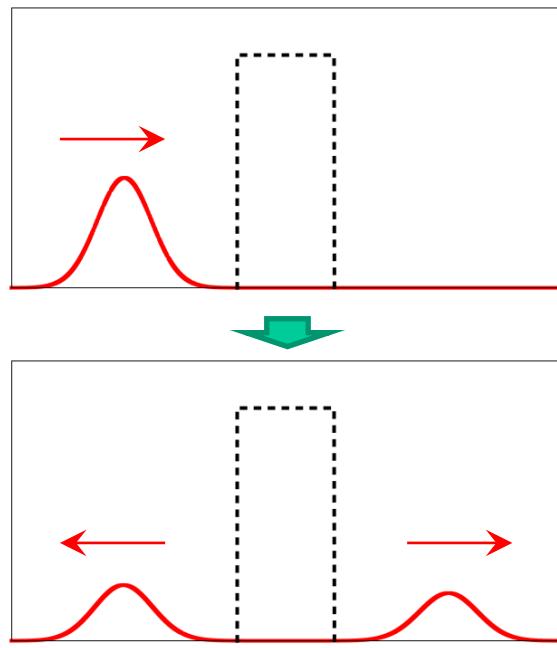
Quantum friction      classical eq. of motion       $\dot{p} = -V'(x) - \gamma p$

a quantization: Kanai model   E. Kanai, PTP 3 (1948) 440

$$H = \frac{p^2}{2m} + V(x) \rightarrow \frac{\pi^2}{2m} e^{-\gamma t} + e^{\gamma t} V(x) \quad (\pi = e^{\gamma t} p)$$

$$\longrightarrow \frac{d}{dt} \langle p \rangle = -\langle V'(x) \rangle - \gamma \langle p \rangle$$

time-dep. wave packet approach



## modelling of open quantum systems

i) system + bath

$$H = H_S + H_B + V_{SB}$$

- ✓ Caldeira-Leggett
- ✓ Feynmann-Vernon
- microscopic
- but, hard to solve

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