

Advanced Nuclear Physics

Nuclear Theory Group,
Tohoku University
Kouichi Hagino

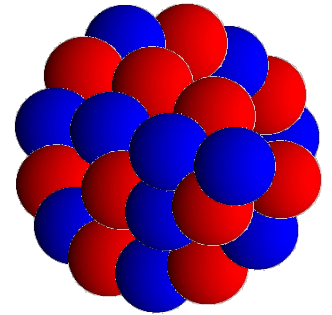
原子核理論特論

東北大学
原子核理論研究室
萩野浩一

Contents

Nuclei: many-body systems of nucleons (protons and neutrons)

→ *Nuclear Many-Body Problems
with strong interaction*



(Low-energy) Nuclear Physics

to understand rich nature of atomic nuclei starting from nucleon-nucleon interactions

- size, mass, density, shape
- excitations
- decays
- nuclear reactions

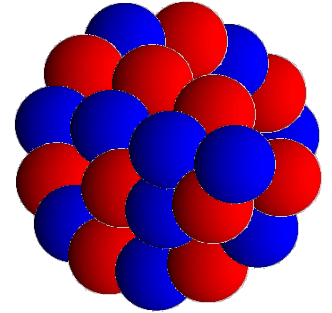
two kinds of particle: protons and neutrons

Contents

Nuclei: many-body systems of nucleons (protons and neutrons)

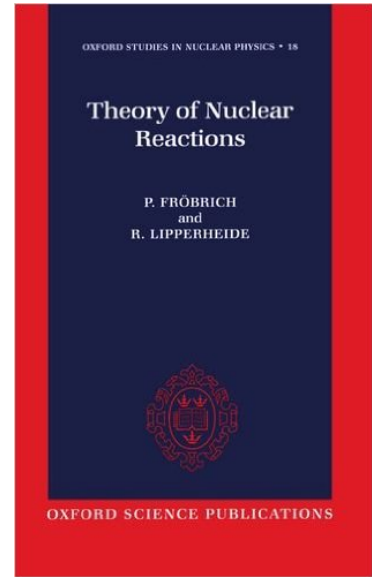
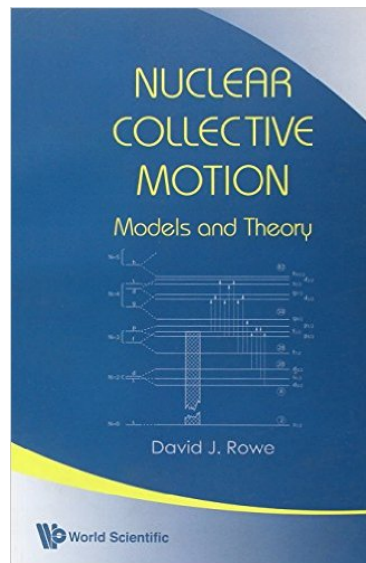
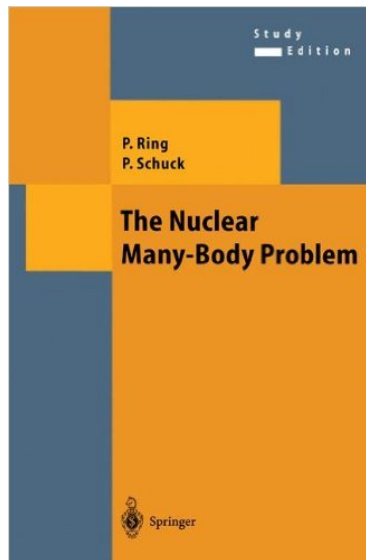
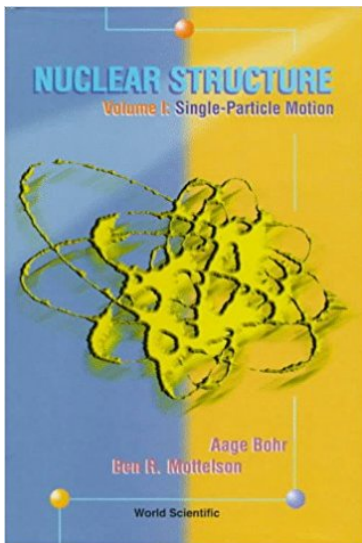
→ *Nuclear Many-Body Problems*

this lecture: microscopic descriptions of atomic nuclei



- Liquid drop model
- Single-particle motion and shell structure
- **Hartree-Fock approximation**
- Bruckner theory
- Pairing correlations and superfluid Nuclei
- nucleosynthesis and physics of neutron-rich nuclei
- **Random phase approximation (RPA)**
- **Nuclear reactions and superheavy elements
(physics of Nihonium)**

References



Bohr-Mottelson

Ring-Schuck

Rowe

Frobrich
-Lipperheide

Lecture notes:

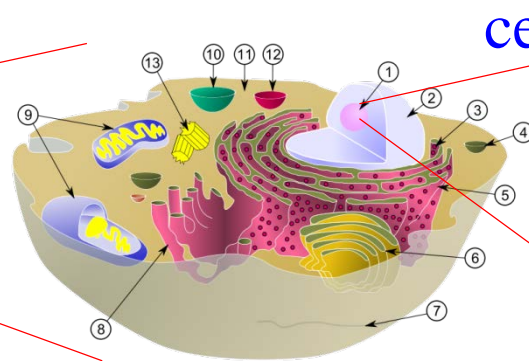
<http://www.nucl.phys.tohoku.ac.jp/~hagino/lecture.html>

(Tohoku University → Physics → Nuclear Theory
→ Kouichi Hagino → Lectures)

Introduction: atoms and atomic nuclei

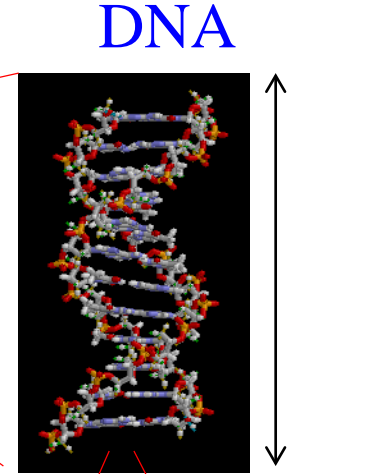


~ 50 cm



cells

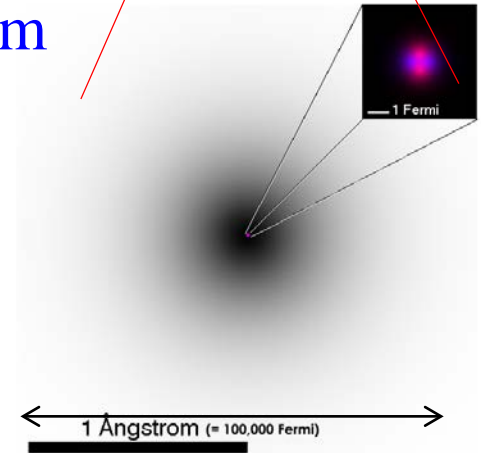
~ $\mu\text{m} = 10^{-6} \text{ m}$



DNA

~ 10^{-8} m

atom



1 Ångstrom (= 100,000 Fermi)

~ 10^{-10} m

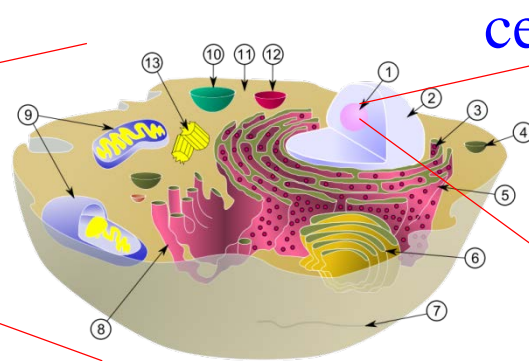
everything is made of atoms.



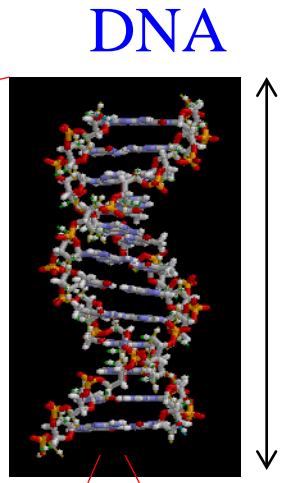
Introduction: atoms and atomic nuclei



~ 50 cm



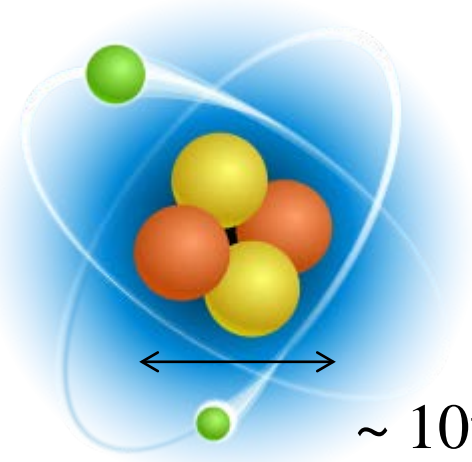
cells



DNA

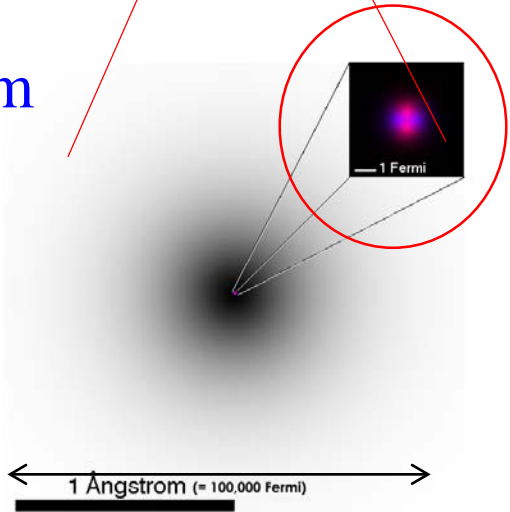
~ 10⁻⁸ m

atomic nucleus



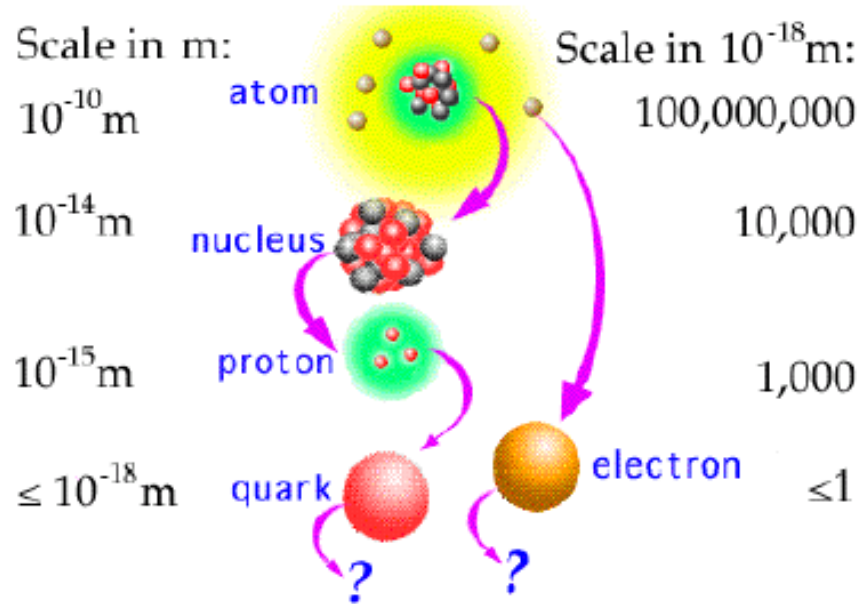
~ 10⁻¹⁵ m

atom



~ 10⁻¹⁰ m

Nuclear Physics



$$1 \text{ fm} = 10^{-15} \text{ m}$$

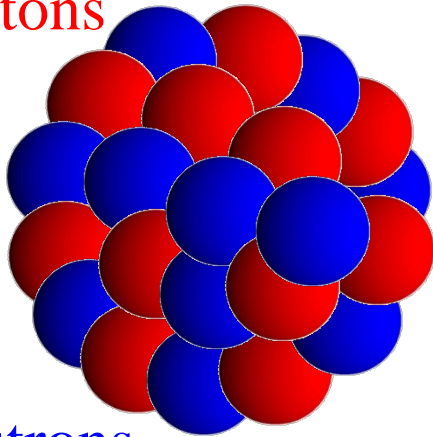
Nucleus as a *quantum many body system*

Basic ingredients:

	charge	mass (MeV)	spin, parity
Proton	+e	938.256	$1/2^+$
Neutron	0	939.550	$1/2^+$

(note) $n \rightarrow p + e^- + \bar{\nu}_e$ (10.4 min)

protons

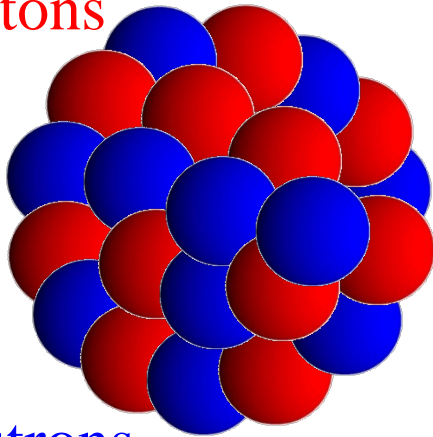


neutrons

- Nucleons are not stopping inside a nucleus.
(they move relatively freely)
- Yet, they are not completely independent.
A nucleus keeps its shape while nucleons influence among themselves so that a nucleon does not escape.

a self-bound system

protons

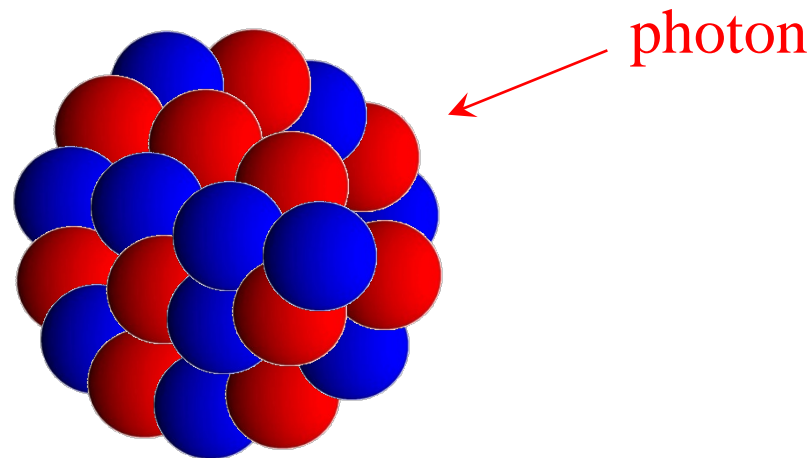


neutrons

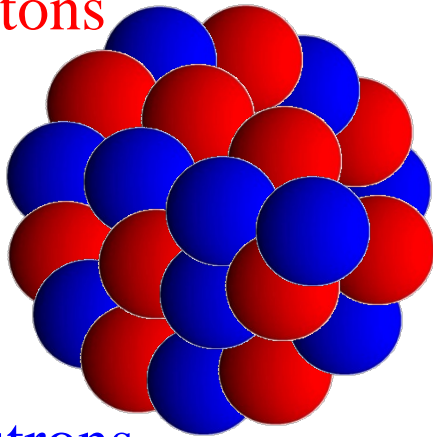
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a self-bound system

What happens if a photon is absorbed into a nucleus?
- one nucleon simply starts moving faster?



protons

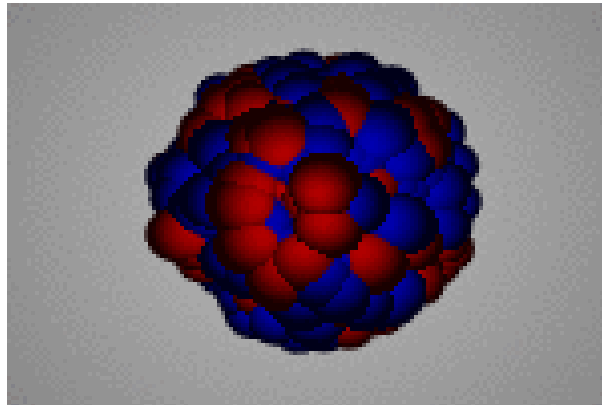


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a self-bound system

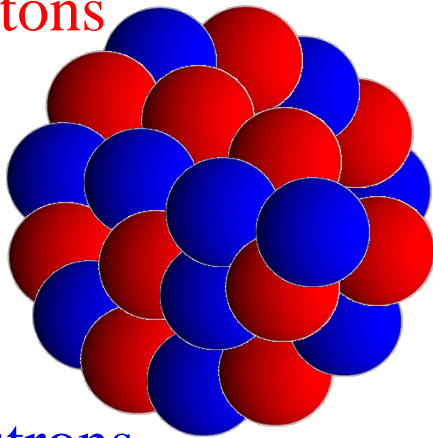
What happens if a photon is absorbed into a nucleus?
- one nucleon simply starts moving faster?



Very coherent
motion can happen

Collective motions

protons

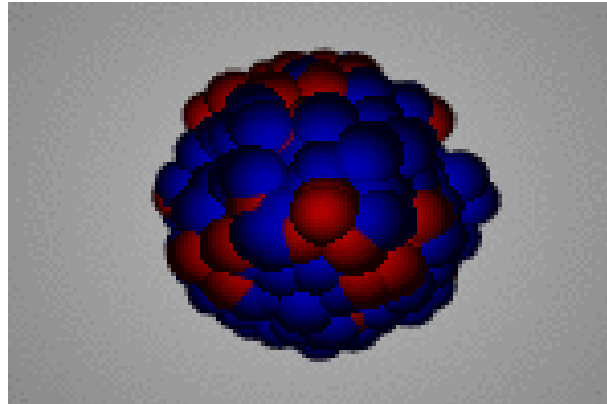
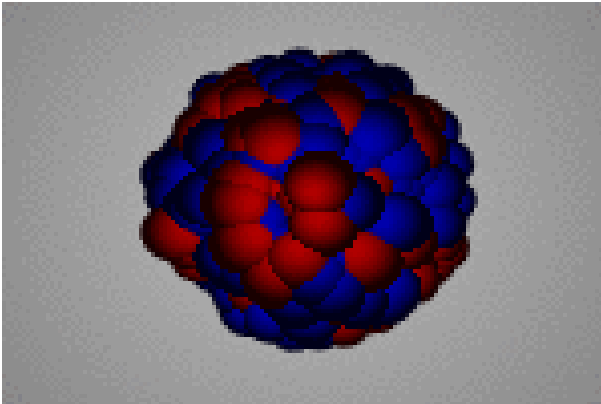


neutrons

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A nucleus keeps its shape while nucleons influence among themselves so that a nucleon does not escape.

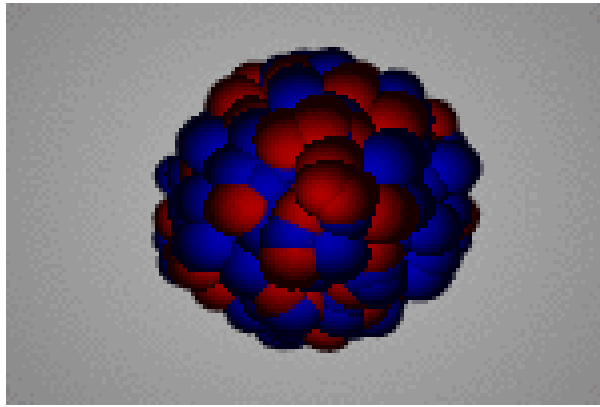
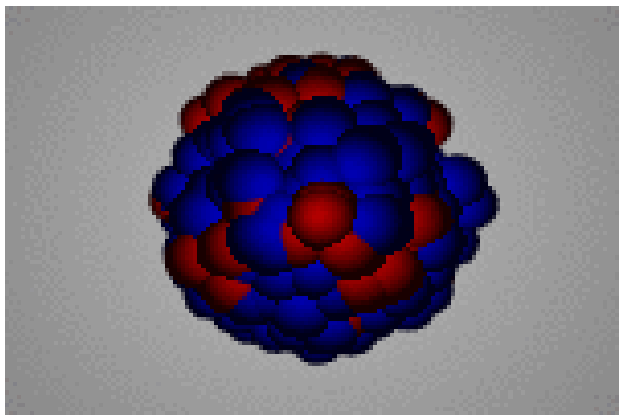
a self-bound system

What happens if a photon is absorbed into a nucleus?
- one nucleon simply starts moving faster?



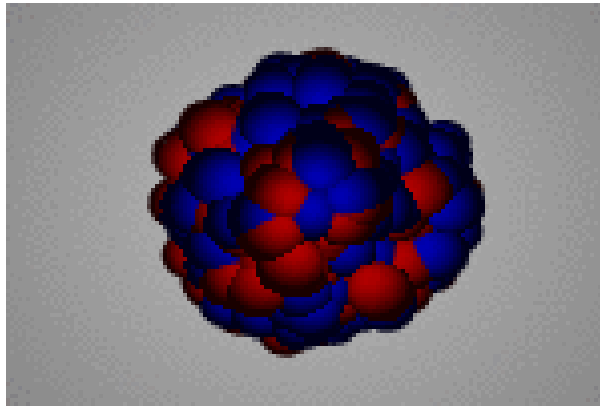
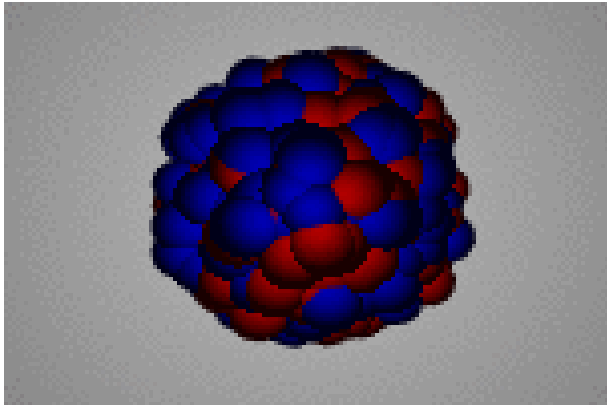
Very coherent
motion can happen

Collective motions

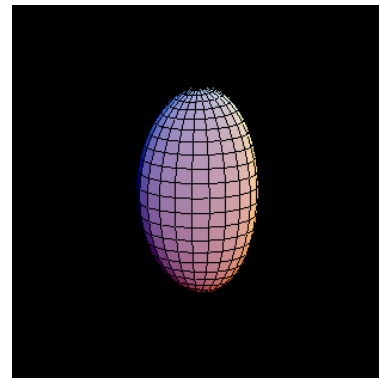
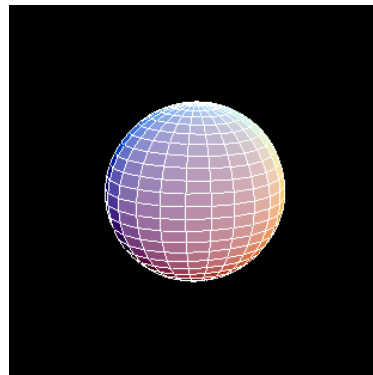
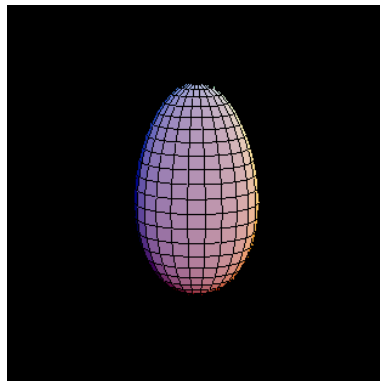
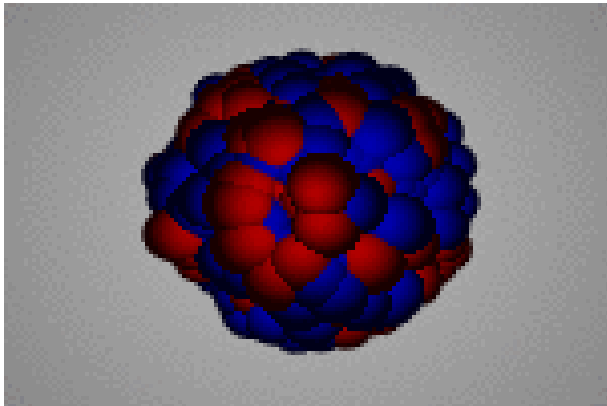


Very coherent
motion can happen

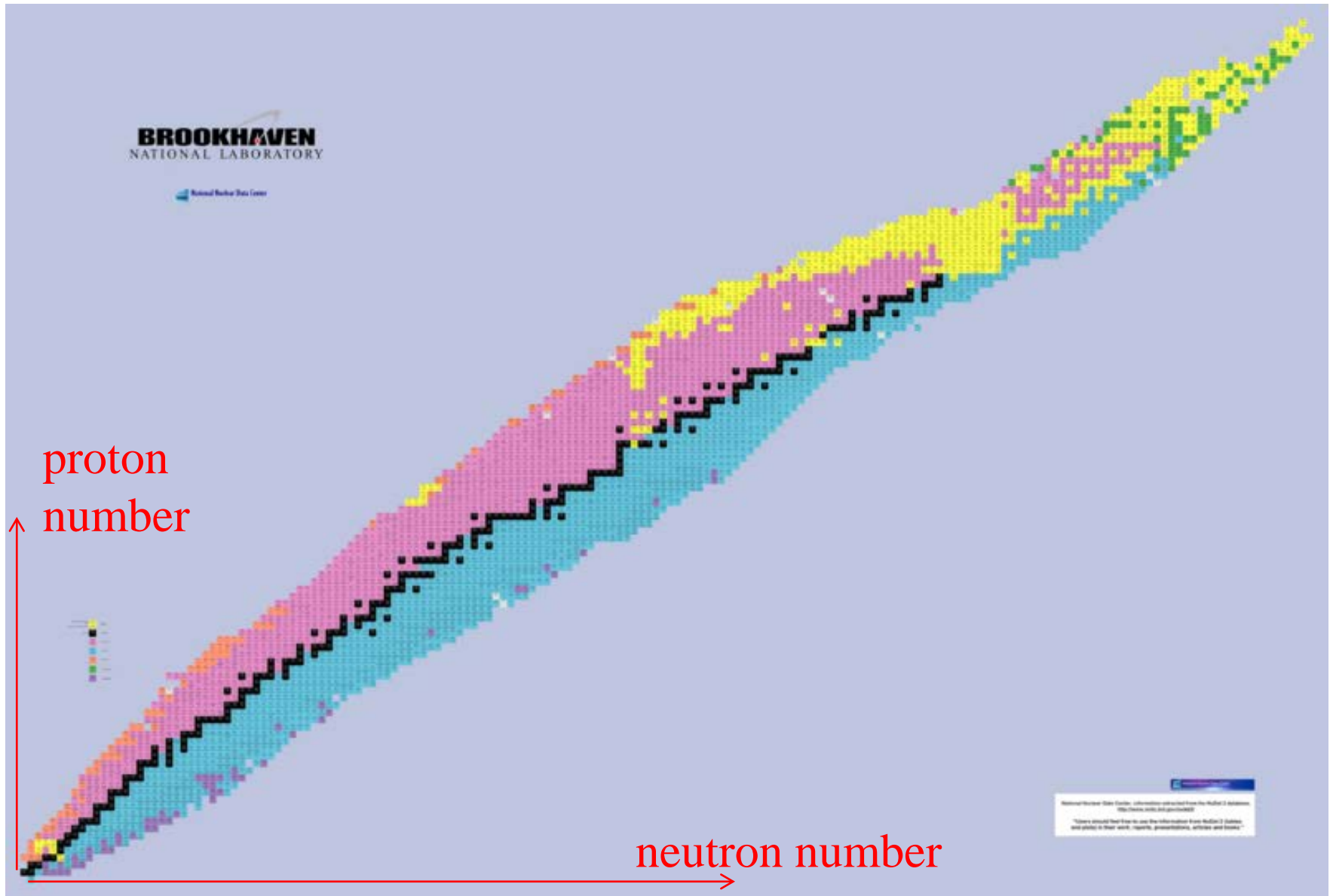
Collective motions



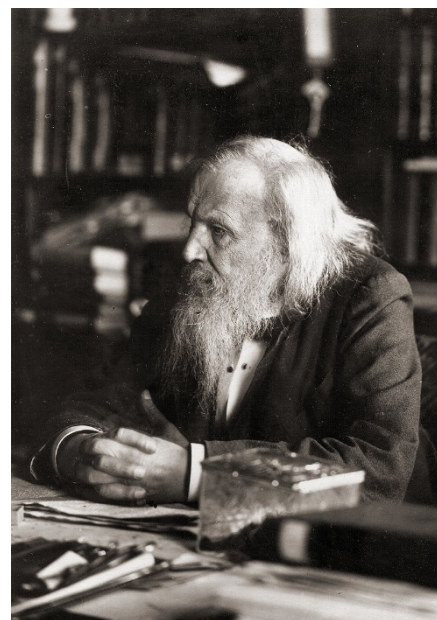
a variety of
motions
→ very rich!



Nuclear Chart: 2D map of atomic nuclei



Periodic Table of elements (1869)

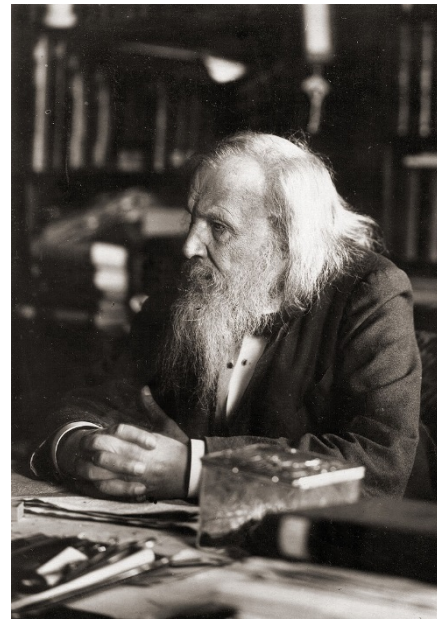


Mendeleev
(1834-1907)

Group →	1	2											13	14	15	16	17	18
↓ Period																		
1	1 H																	2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba	57 La	* 72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	89 Ac	* 104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
				* 58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	
				* 90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	

International Year
of the Periodic Table
of Chemical Elements

Periodic Table of elements (1869)



Mendeleev
(1834-1907)

Group	5	6	7
↓			
1			
2			
3			

	13	14	15	16	17	18
						2 He
	5 B	6 C	7 N	8 O	9 F	10 Ne
	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar

Taka and Fuji (a grand-daughter of Mendeleev)

1947

FAMILY HISTORY 0 FAMILY HISTORY 1 FAMILY HISTORY 2

FAMILY HISTORY
ファミリーヒストリー


▷ 00A 0 ▷ 0A 1 ▷ 1A 2

1920 1925 1931 1955 1980 1990

6	55 Cs	56 Ba	57 La *	72 Hf	73 Ta	74 W
7	87 Fr	88 Ra	89 Ac *	104 Rf *	105 Db	106 Sg
			* 58 Ce	59 Pr	60 Nd	
			* 90 Th	91 Pa	92 U	

Periodic Table of elements (1869)

protons only, no information on neutrons

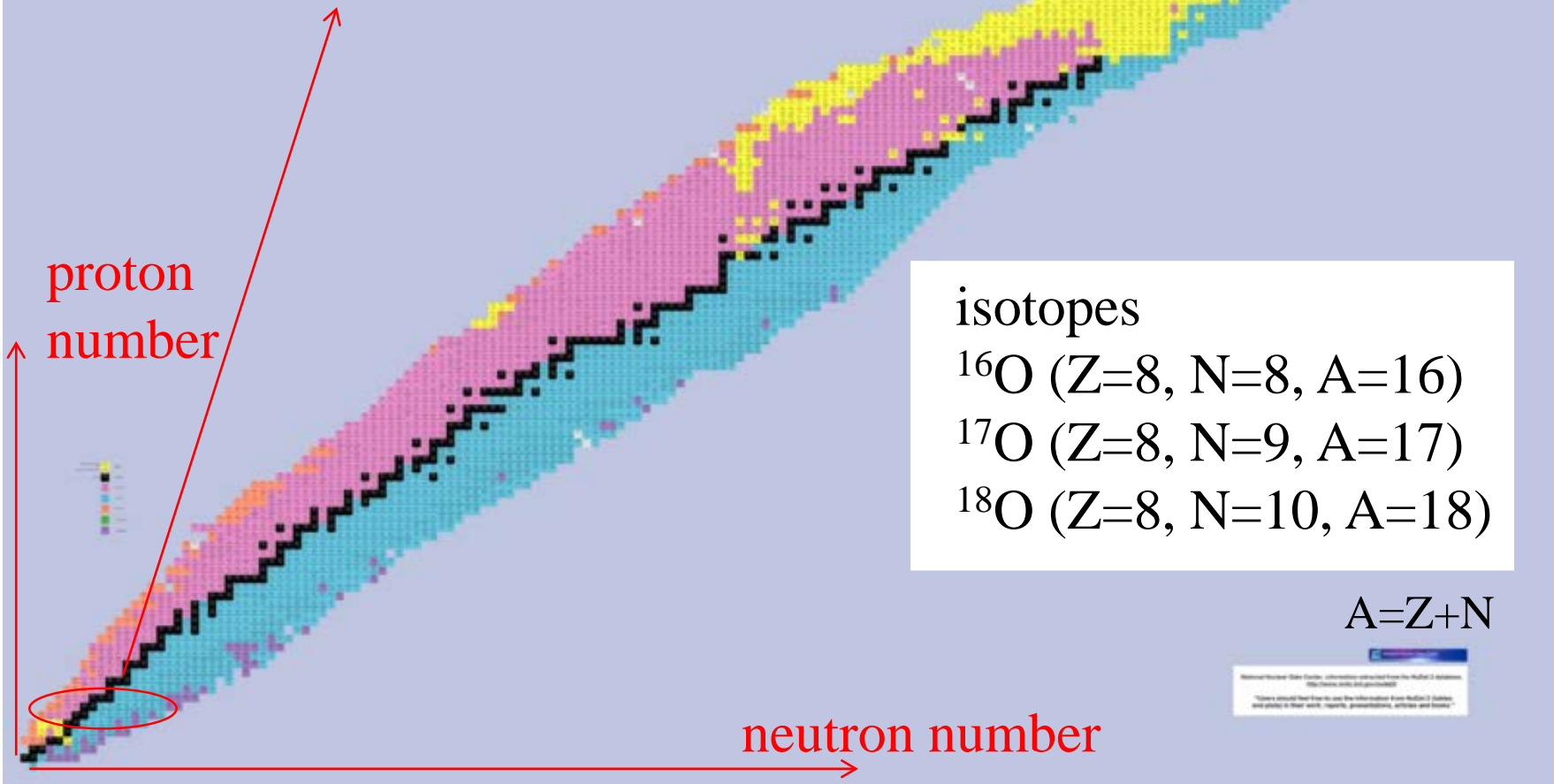


2019 IYPT
 International Year
 of the Periodic Table
 of Chemical Elements

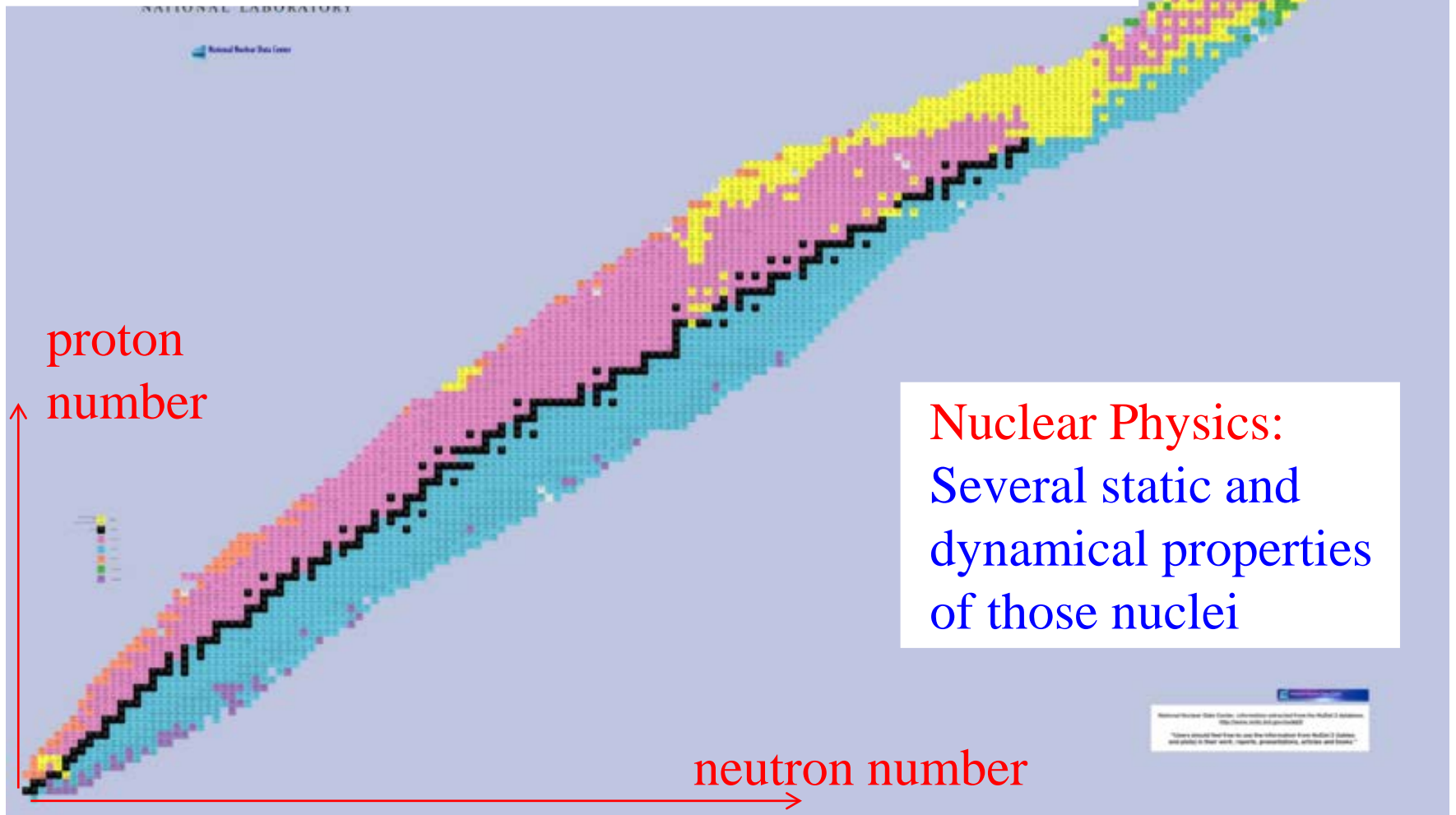
Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
Period ↓	1																		2	
1	1 H																			2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F		10 Ne	
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4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br		36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I		54 Xe	
6	55 Cs	56 Ba	57 La *	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At		86 Rn	
7	87 Fr	88 Ra	89 Ac *	104 Rf *	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts		118 Og	
				* 58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
				* 90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			

Nuclear Chart: an extended version of periodic table

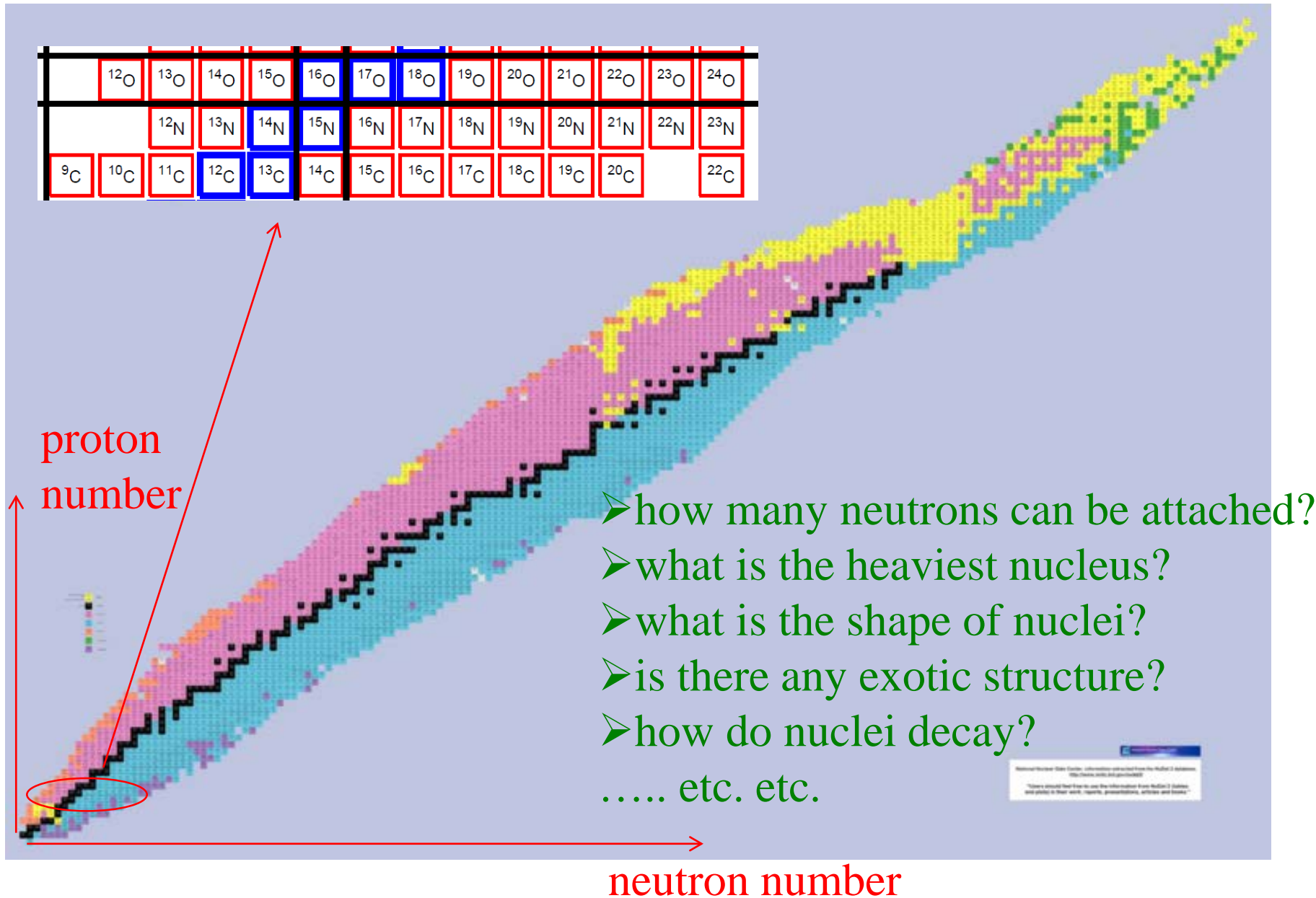
	^{12}O	^{13}O	^{14}O	^{15}O	^{16}O	^{17}O	^{18}O	^{19}O	^{20}O	^{21}O	^{22}O	^{23}O	^{24}O
		^{12}N	^{13}N	^{14}N	^{15}N	^{16}N	^{17}N	^{18}N	^{19}N	^{20}N	^{21}N	^{22}N	^{23}N
^9C	^{10}C	^{11}C	^{12}C	^{13}C	^{14}C	^{15}C	^{16}C	^{17}C	^{18}C	^{19}C	^{20}C		^{22}C



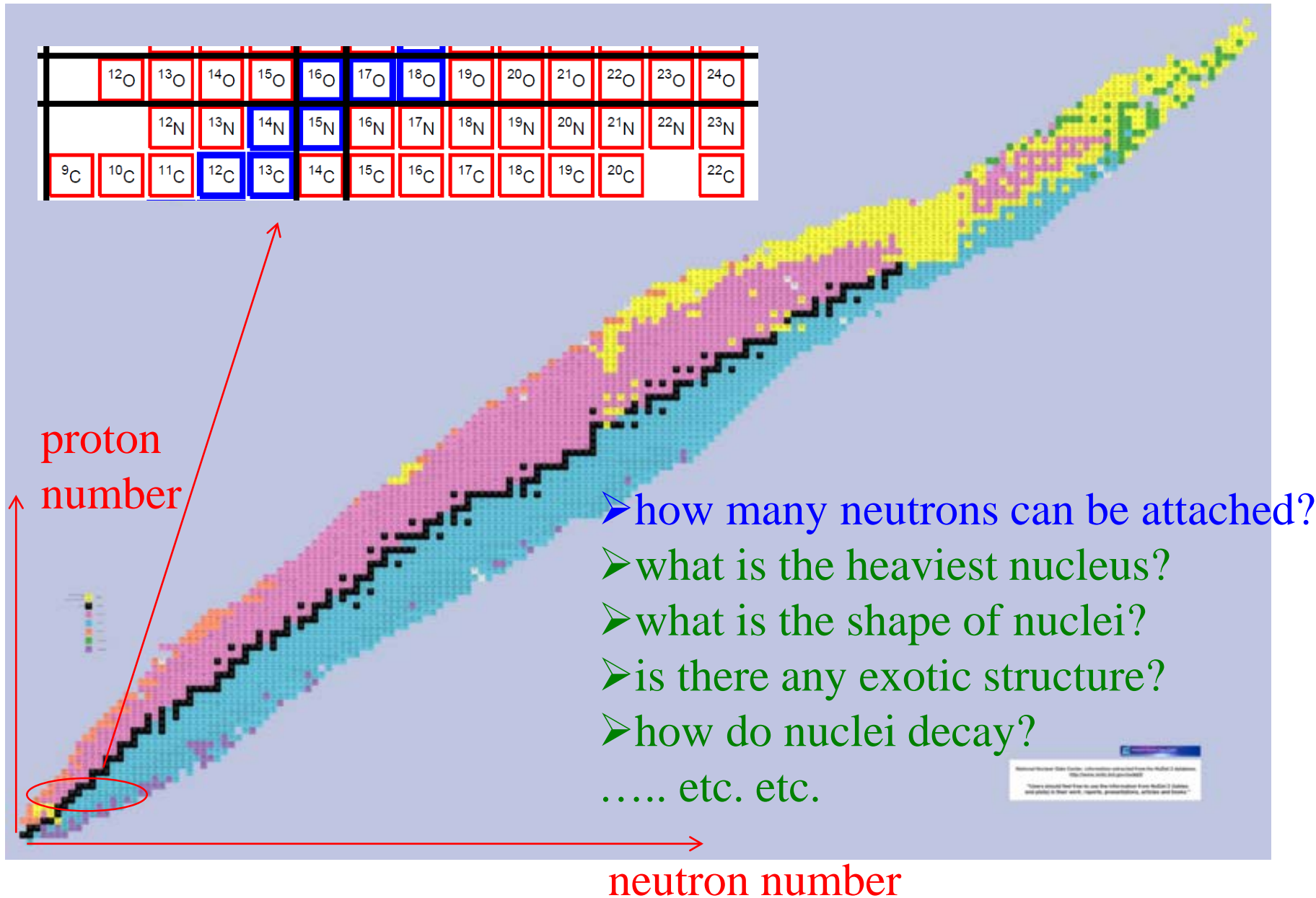
- Stable nuclei in nature: 287
- Nuclei artificially synthesized : about 3,000
- Nuclei predicted : about 7,000 ~ 10,000



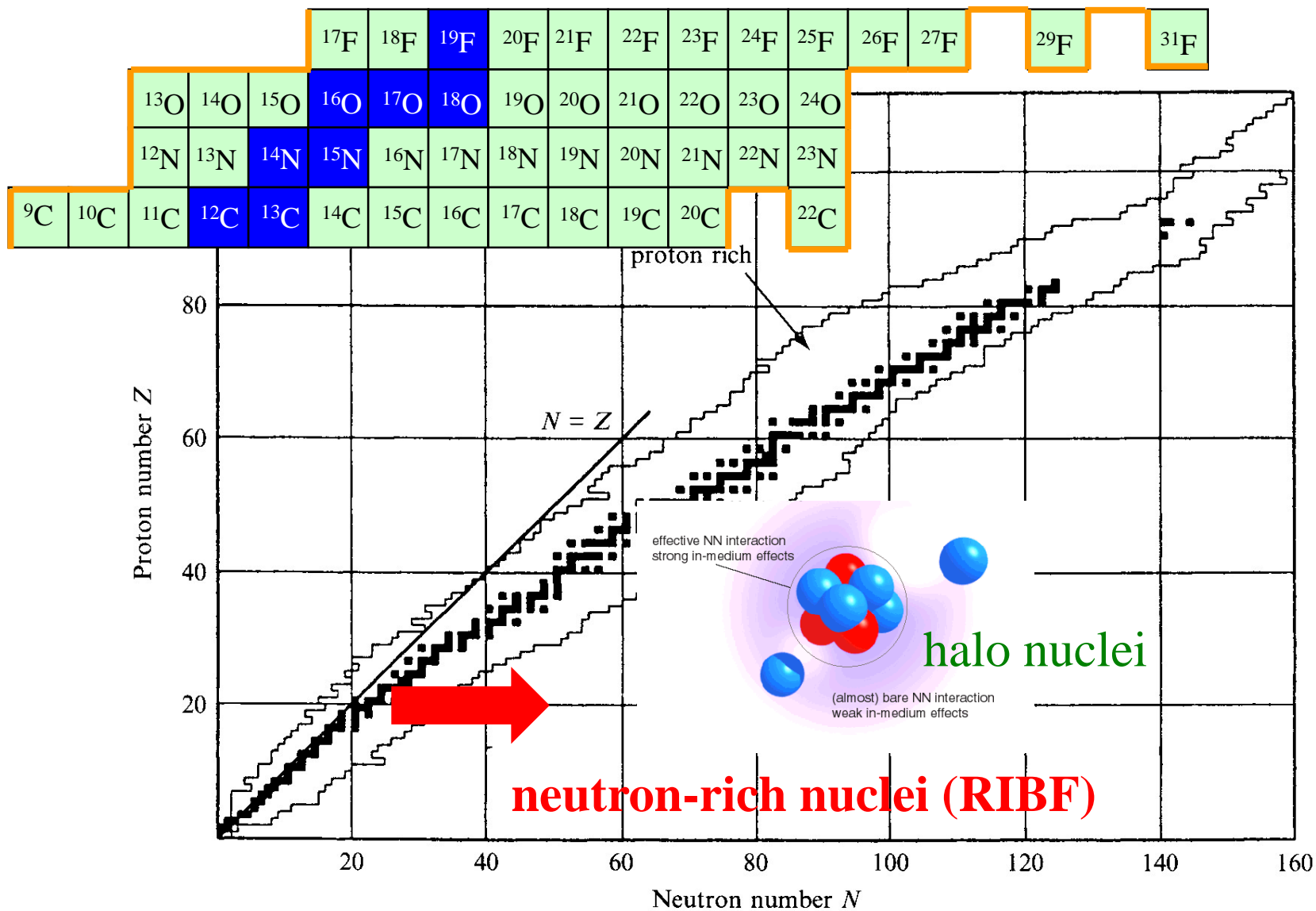
Nuclear Chart: an extended version of periodic table



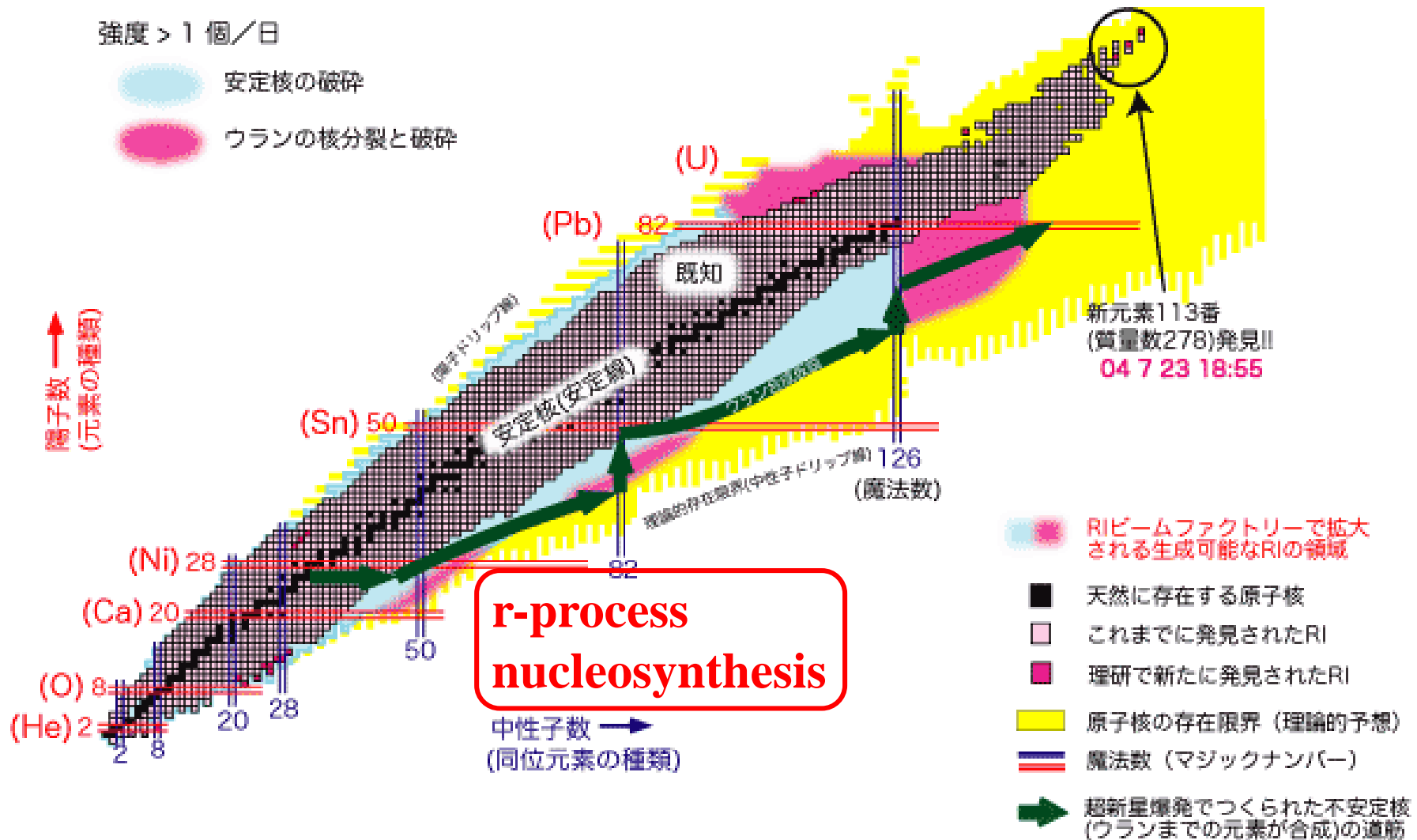
Nuclear Chart: an extended version of periodic table



Extension of nuclear chart: frontier of nuclear physics



Neutron-rich nuclei (RIBF at RIKEN)

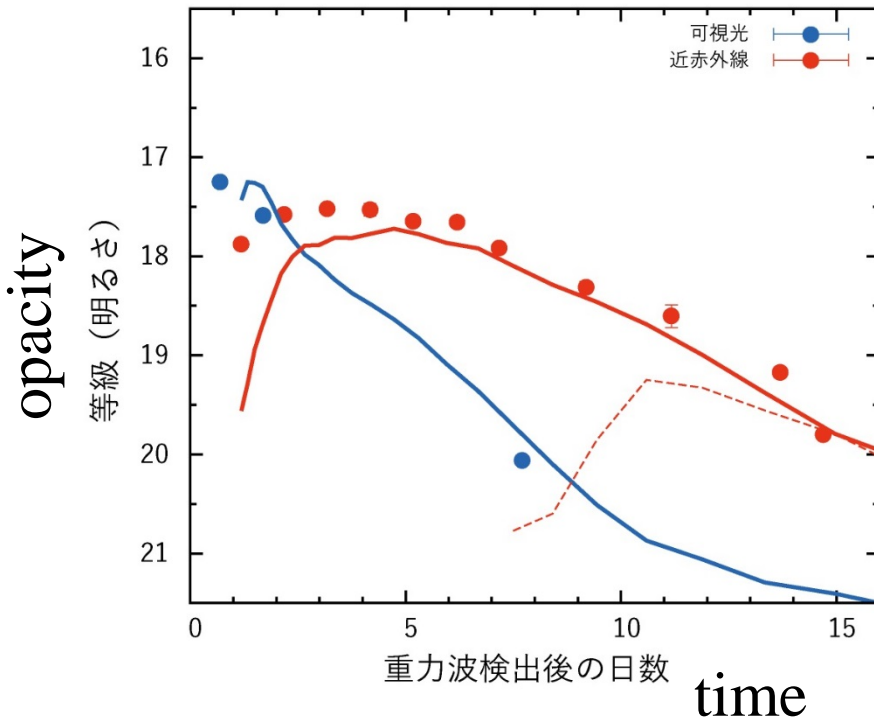
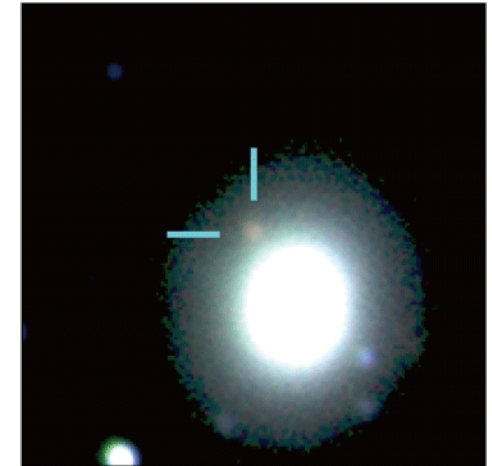
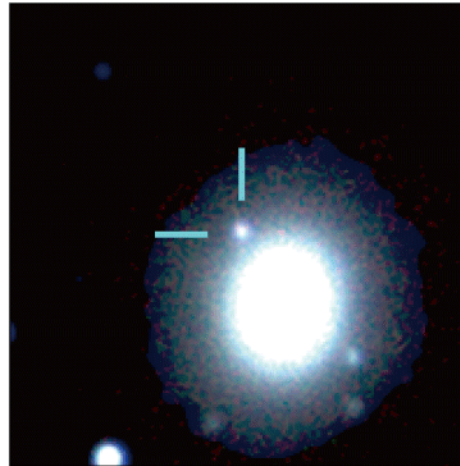


October 17, 2017

Ele.-mag. wave from the source of GW

2017.08.18-19

2017.08.24-25

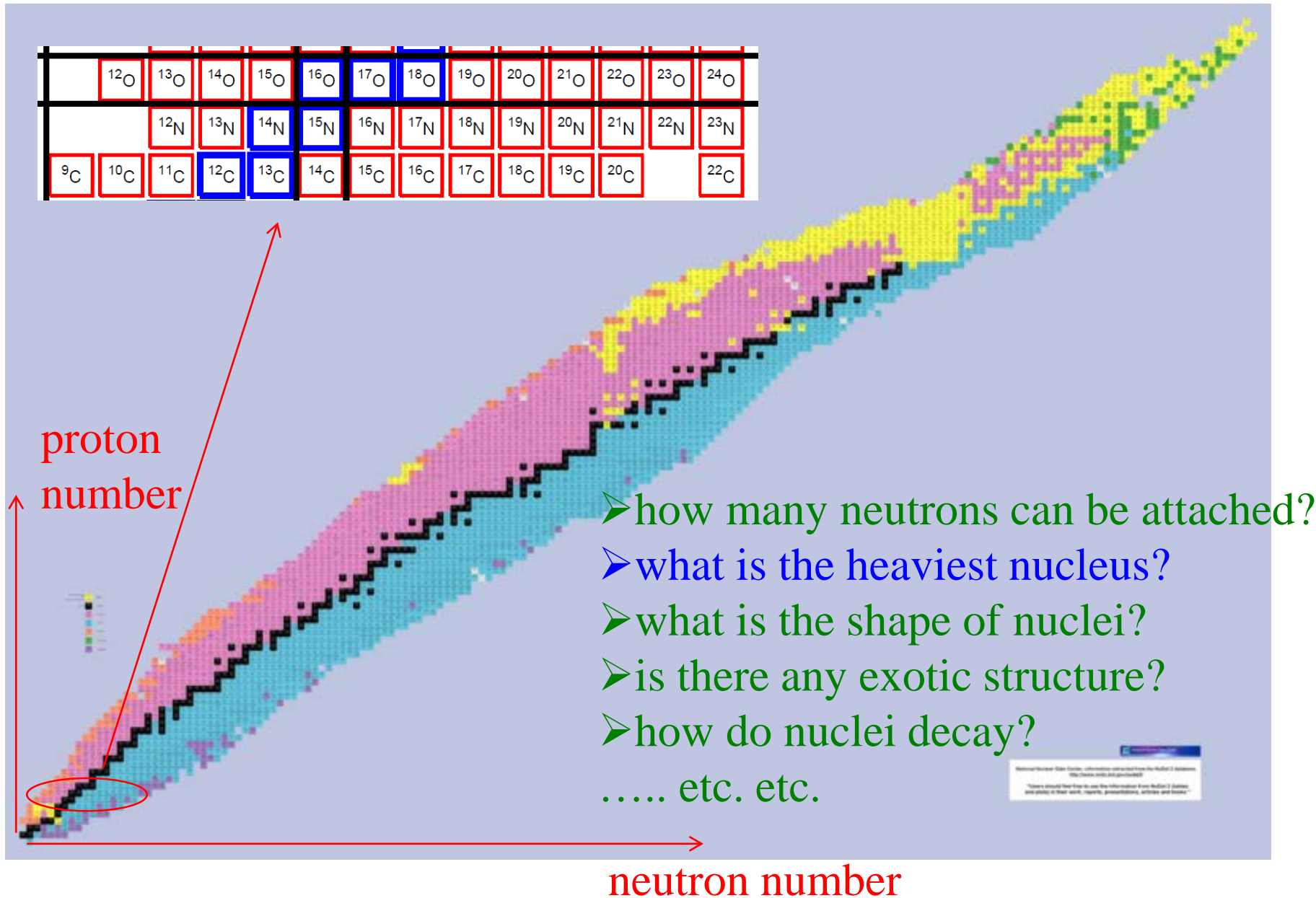


solid lines : with r-process nucleosynthesis
dashed line : no r-process nucleosynthesis

M. Tanaka et al.,
Astron. Soc. Jpn. 69 ('17) 102

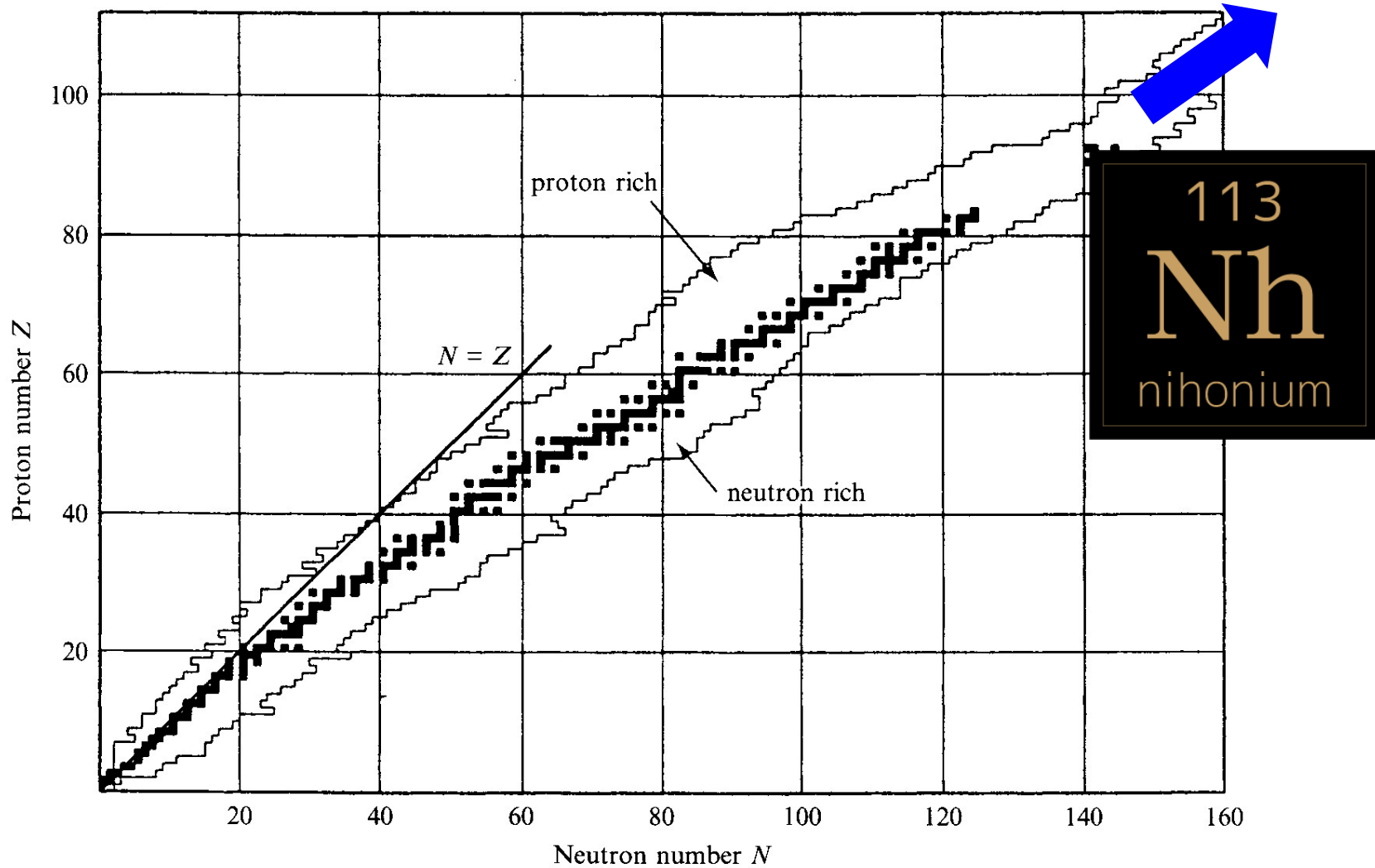
<http://www.cfca.nao.ac.jp/pr/20171016>

Nuclear Chart: an extended version of periodic table

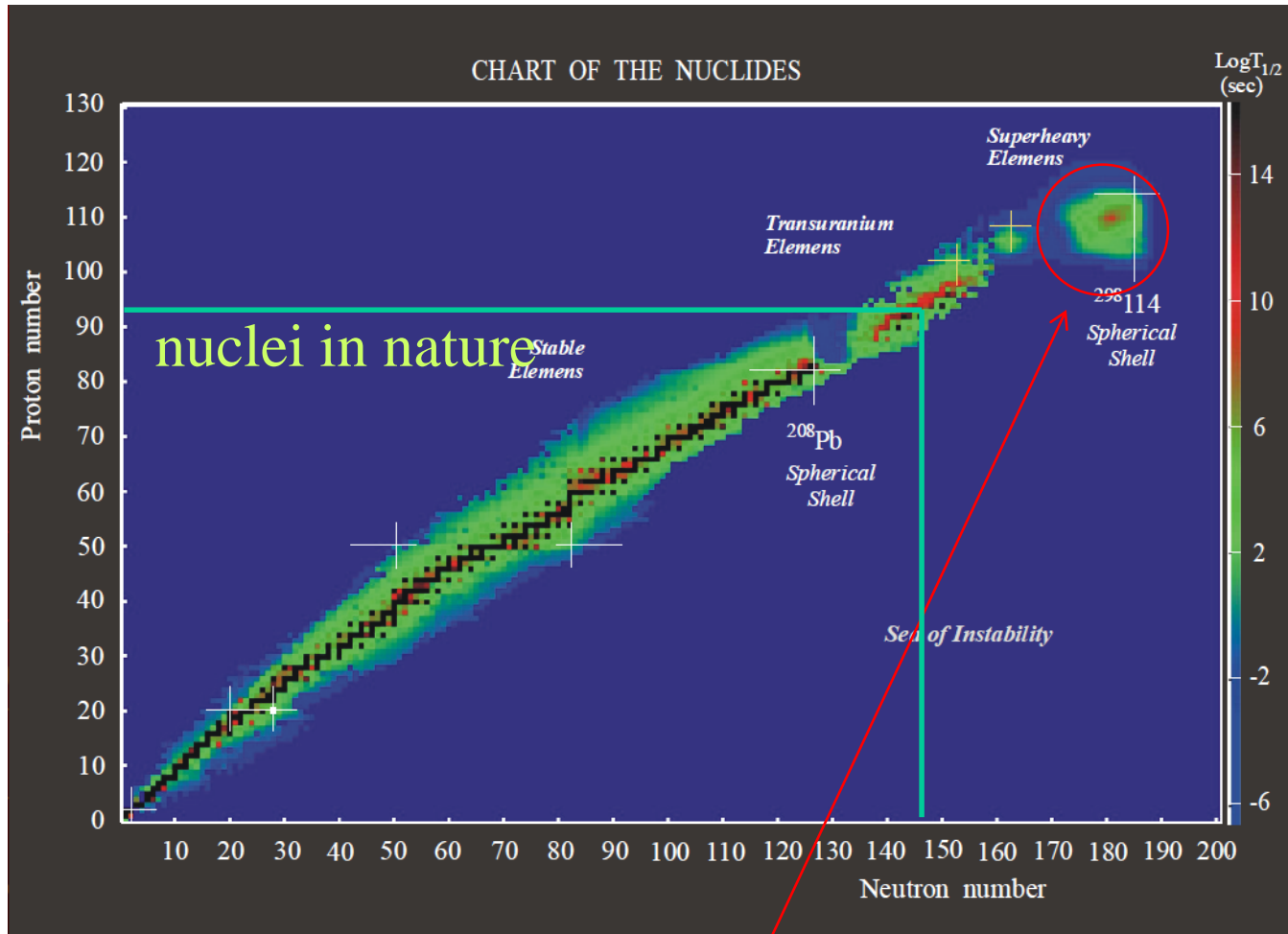


Extension of nuclear chart: frontier of nuclear physics

superheavy
elements



Prediction of island of stability: an important motivation of SHE study

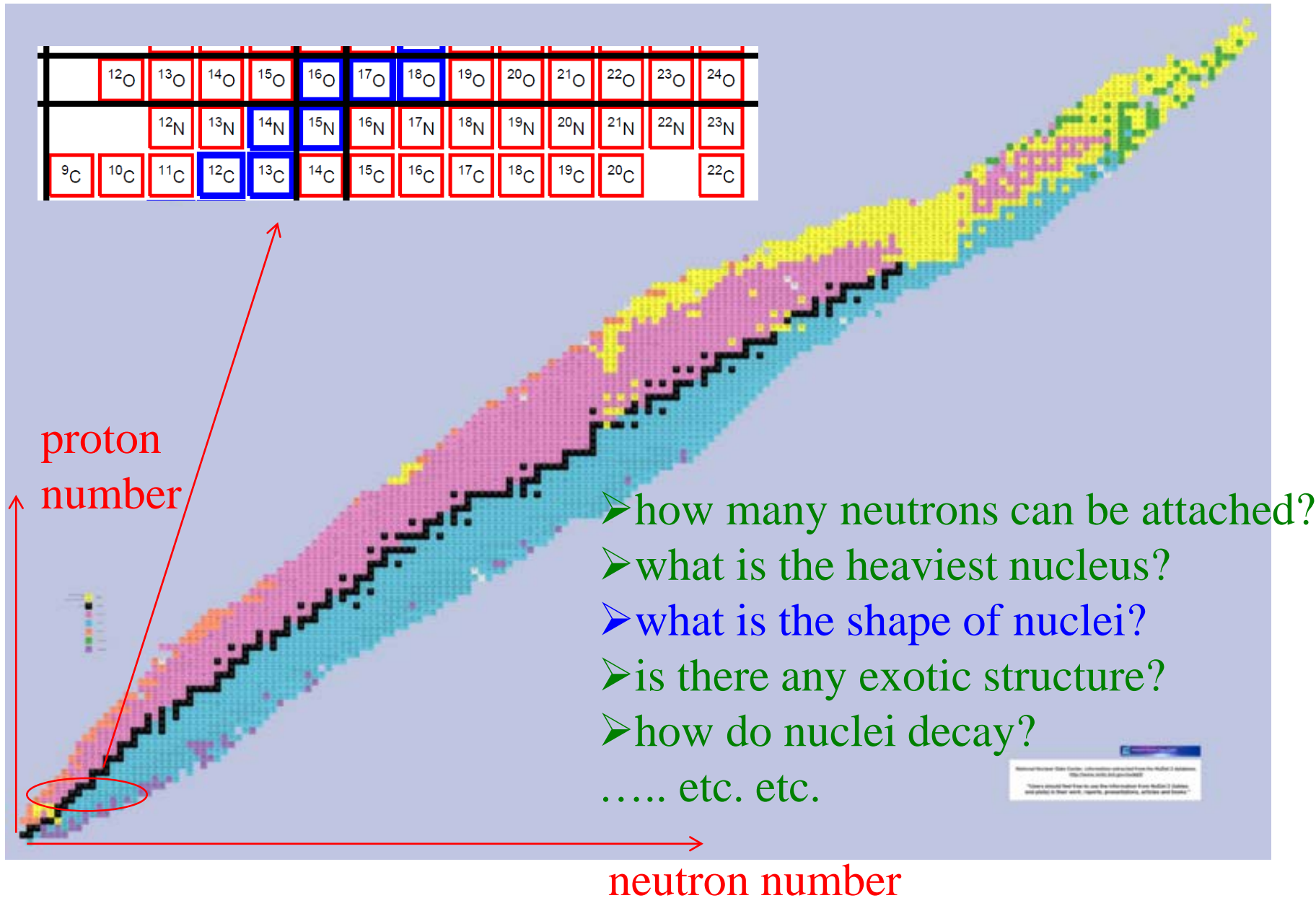


Yuri Oganessian

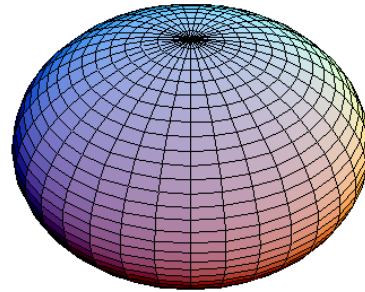
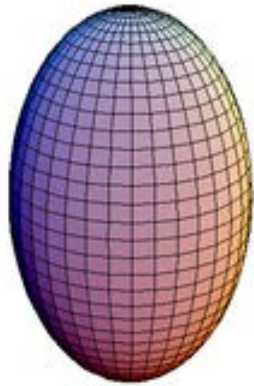
island of stability around Z=114, N=184

W.D. Myers and W.J. Swiatecki (1966), A. Sobiczewski et al. (1966)

Nuclear Chart: an extended version of periodic table



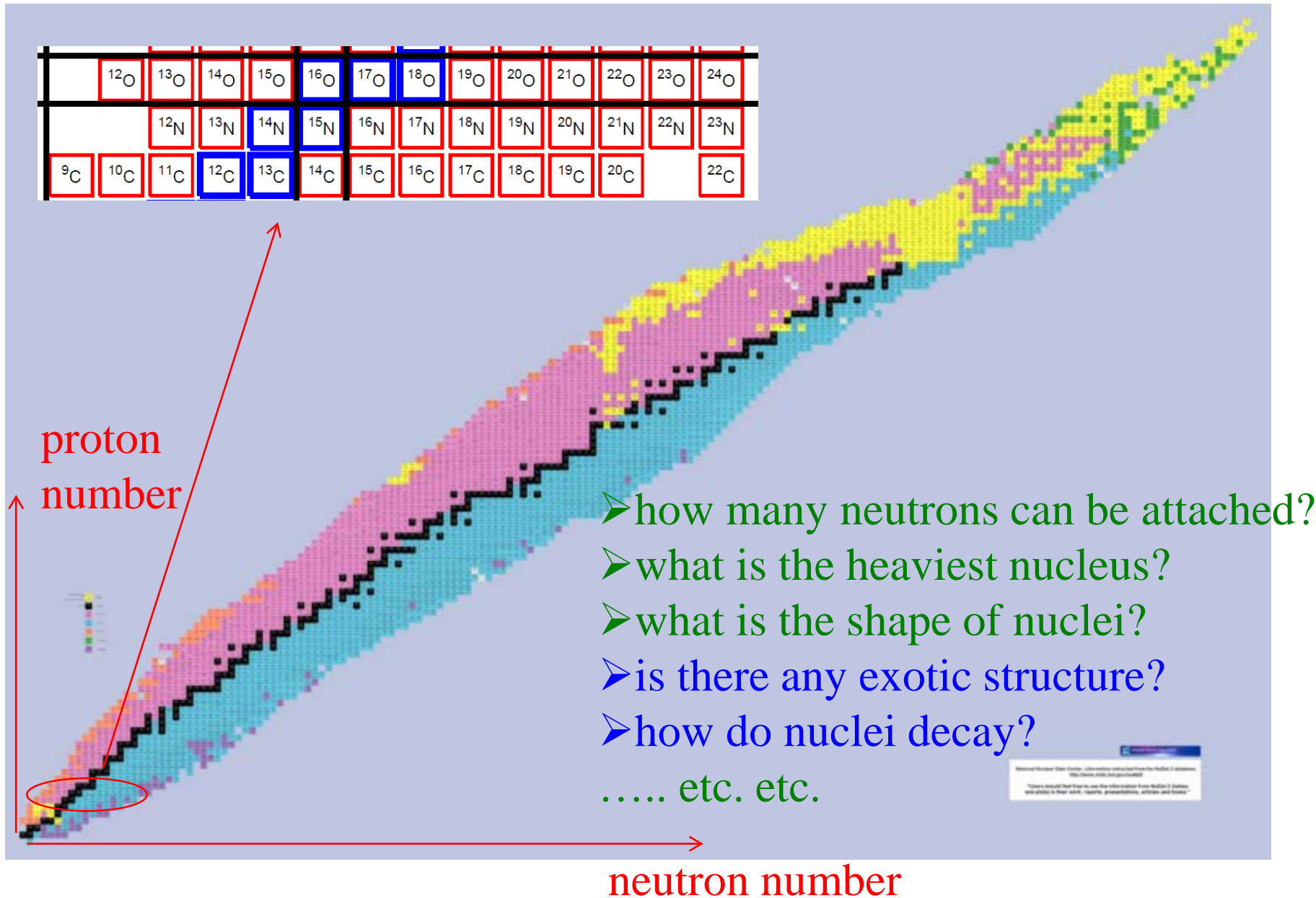
a nucleus is not always spherical



Quantum shape
dynamics

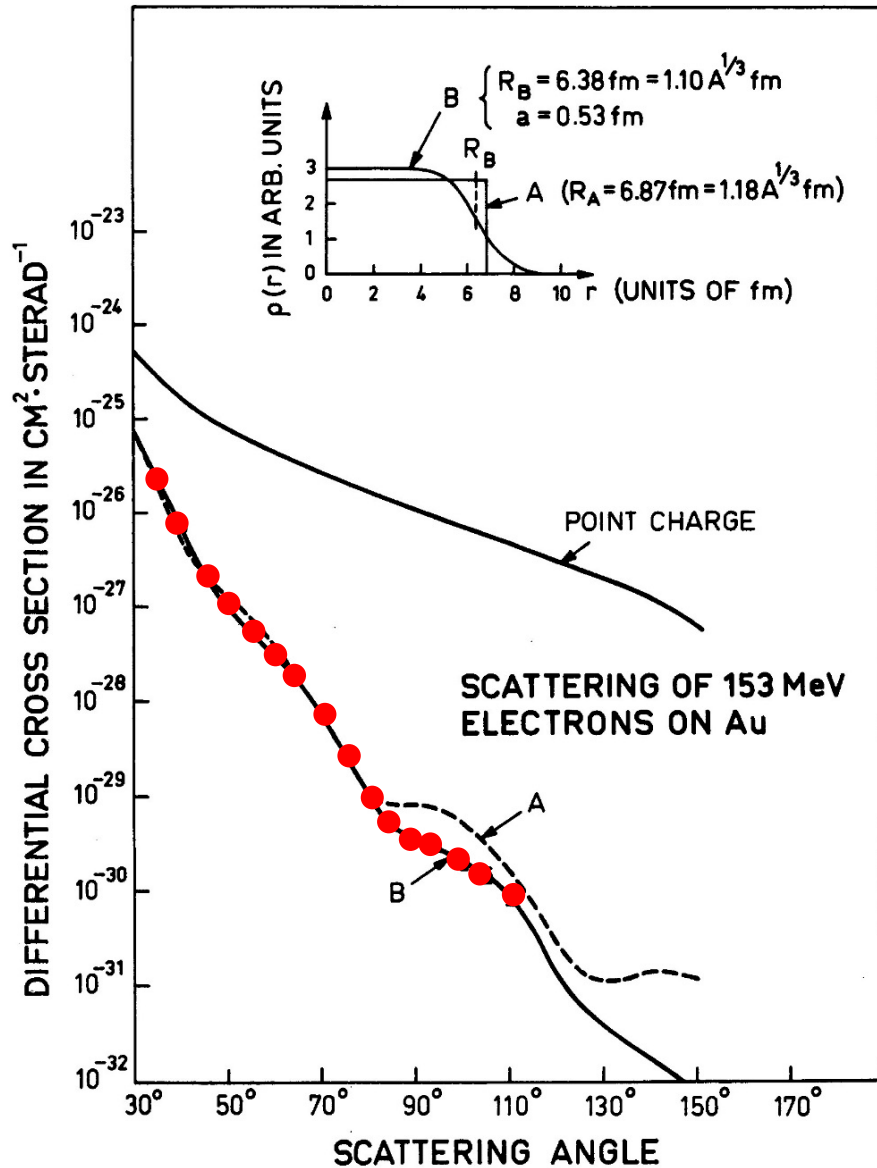
「形の量子論」

Nuclear Chart: an extended version of periodic table



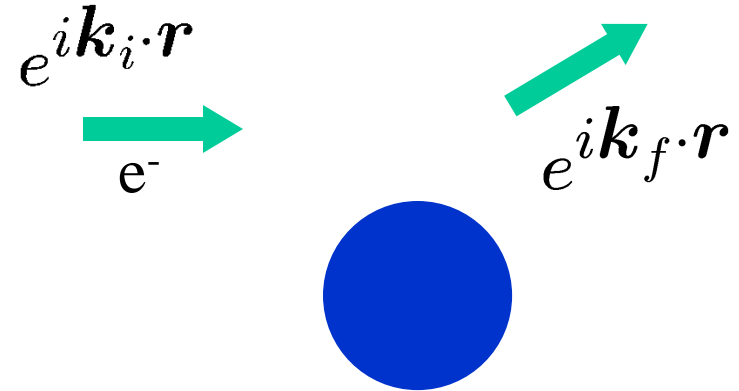
Density Distribution

Density Distribution



High energy electron scattering

Born approximation:



$$\frac{d\sigma}{d\Omega} = \frac{Z_P^2 e^4}{(4E \sin^2 \theta/2)^2} |F(\mathbf{q})|^2$$

Form factor

$$F(\mathbf{q}) = \int e^{-i\mathbf{q} \cdot \mathbf{r}} \rho(\mathbf{r}) d\mathbf{r}$$

(Fourier transform of the density)

Born approximation

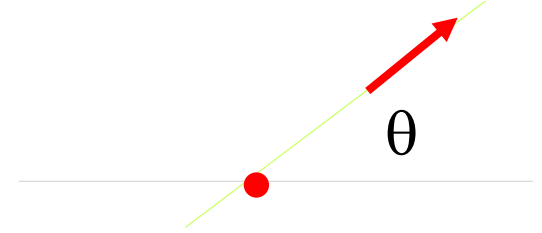
$$\psi_i(\mathbf{r}) = e^{i\mathbf{p}_i \cdot \mathbf{r} / \hbar}$$



$V(r)$



$$\psi_f(\mathbf{r}) = e^{i\mathbf{p}_f \cdot \mathbf{r} / \hbar}$$



$$W_{fi} = \frac{2\pi}{\hbar} |\langle \psi_f | V | \psi_i \rangle|^2 \rho(E_f)$$

incident flux: $j_{\text{inc}} = \rho_i v = p_i / \mu$



$$\sigma = \frac{W_{fi}}{j_{\text{inc}}} = \int d\Omega \frac{d\sigma}{d\Omega}$$

Electron scattering

$$V(r) = -e^2 \int dr' \frac{\rho(r')}{|r - r'|}$$

$$\frac{d\sigma}{d\Omega} = \frac{Z_P^2 e^4}{(4E \sin^2 \theta/2)^2} |F(q)|^2$$

$$= \left(\frac{d\sigma_{\text{Ruth}}}{d\Omega} \right) |F(q)|^2$$

Form factor

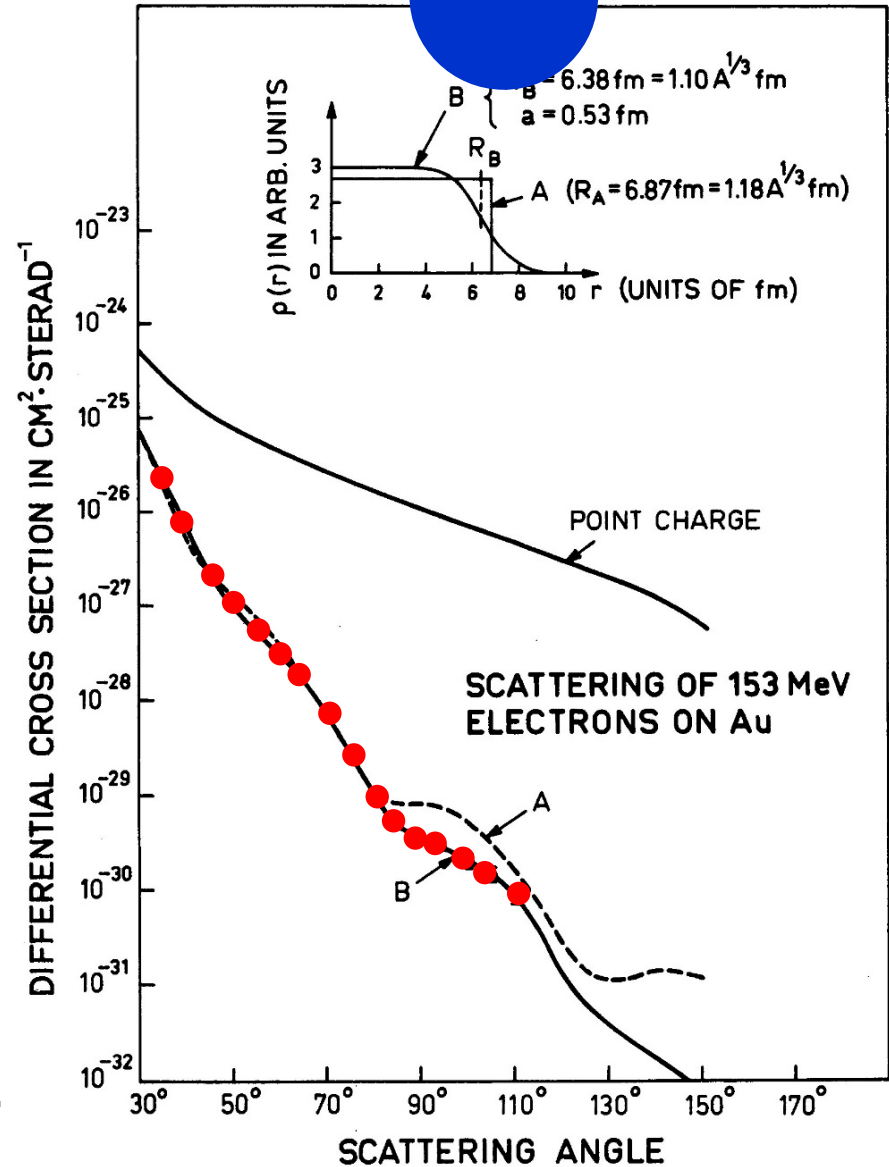
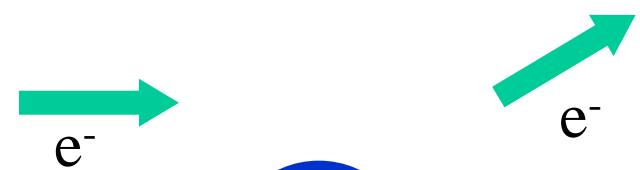
$$F(q) = \int e^{-i\mathbf{q} \cdot \mathbf{r}} \rho(r) dr$$

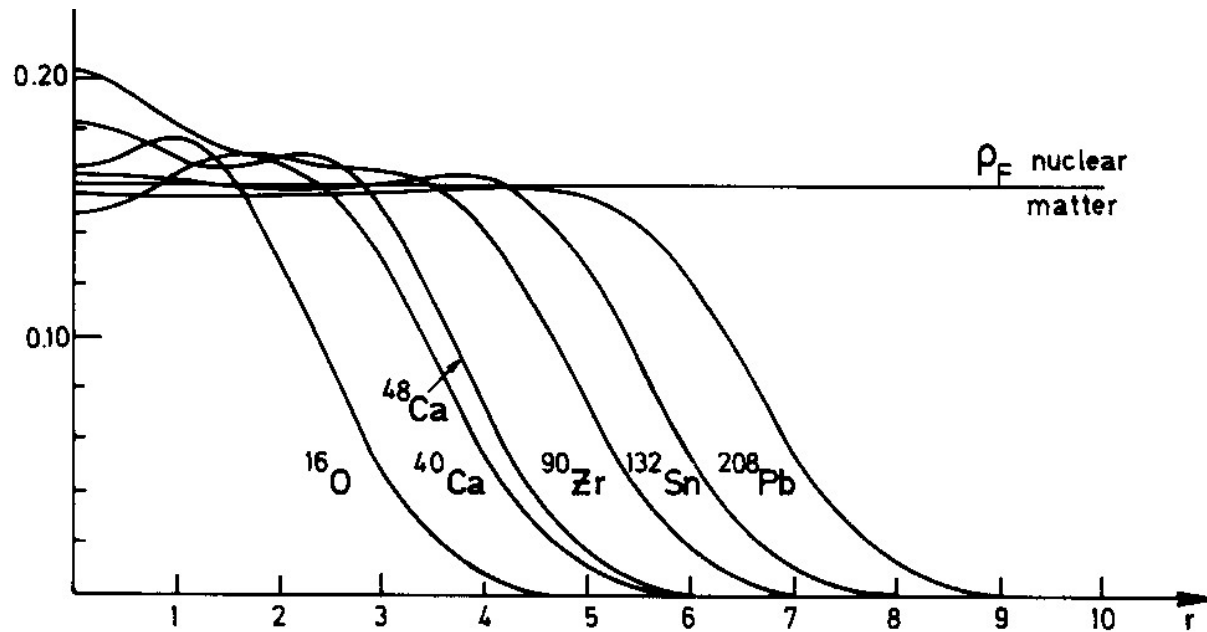
* relativistic correction:

$$\frac{d\sigma_{\text{Ruth}}}{d\Omega} \rightarrow \frac{d\sigma_{\text{Mott}}}{d\Omega}$$

$$= \frac{d\sigma_{\text{Ruth}}}{d\Omega} \cdot \left(1 - \frac{v^2}{c^2} \sin^2 \frac{\theta}{2} \right)$$

$$\sim \frac{d\sigma_{\text{Ruth}}}{d\Omega} \cdot \cos^2 \frac{\theta}{2} \quad (v \rightarrow c)$$





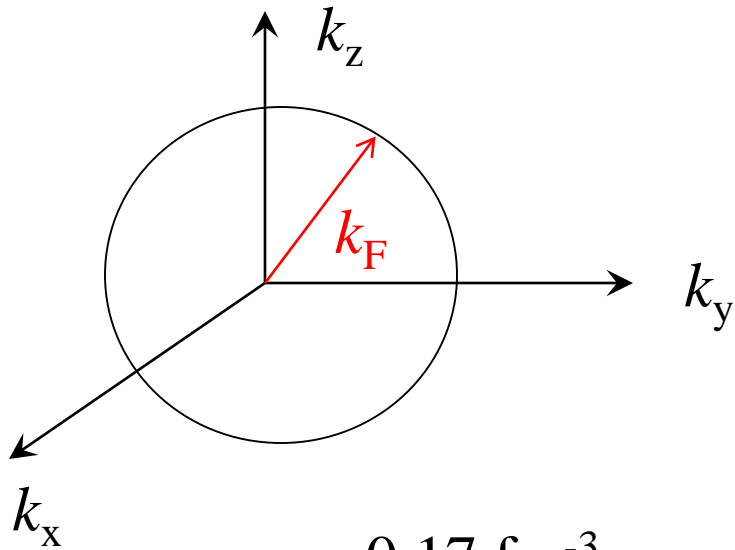
Fermi distribution

$$\rho(r) = \frac{\rho_0}{1 + \exp[(r - R_0)/a]}$$

$$\begin{aligned} \rho_0 &\sim 0.17 \text{ (fm}^{-3}\text{)} && \leftarrow \text{Saturation property} \\ R_0 &\sim 1.1 \times A^{1/3} \text{ (fm)} && \text{property} \\ a &\sim 0.57 \text{ (fm)} \end{aligned}$$

Momentum Distribution

Fermi gas approximation



$$\begin{aligned}\rho &= 2 \times 2 \times 4\pi \int_0^{k_F} \frac{k^2 dk}{(2\pi)^3} \\ &= \frac{2}{3\pi^2} k_F^3\end{aligned}$$

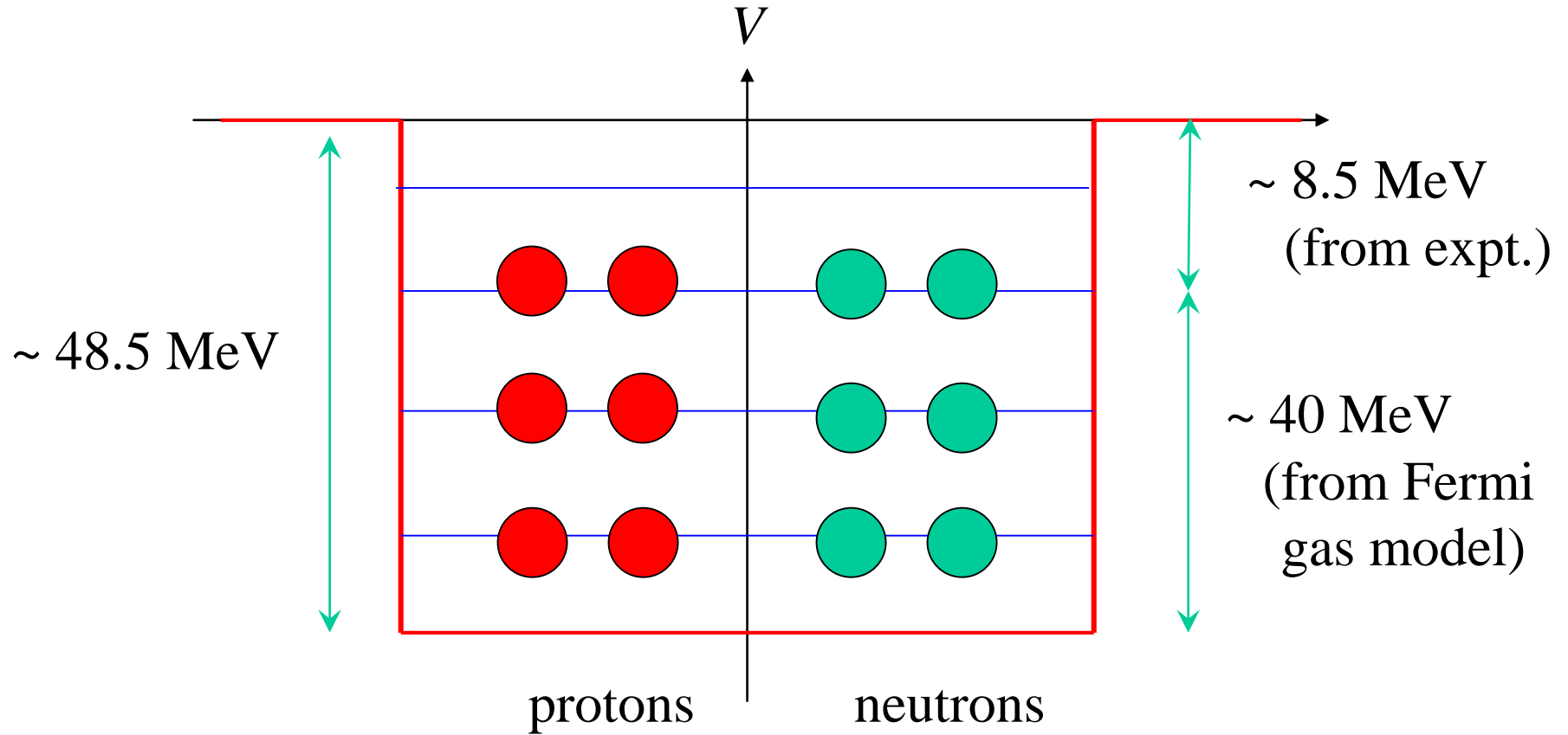
(note: spin-isospin degeneracy)

$$\rho = 0.17 \text{ fm}^{-3} \longrightarrow k_F \sim 1.36 \text{ fm}^{-1}$$

$$\iff \frac{v_F}{c} = \frac{k_F \cdot \hbar c}{mc^2} = 0.285$$

$$\text{Fermi energy: } \epsilon_F = \frac{k_F^2 \hbar^2}{2m} \sim 37 \text{ (MeV)}$$

A potential for nucleons inside a nucleus



Discussion: Electron scattering

$$V(r) = -e^2 \int dr' \frac{\rho(r')}{|r - r'|}$$

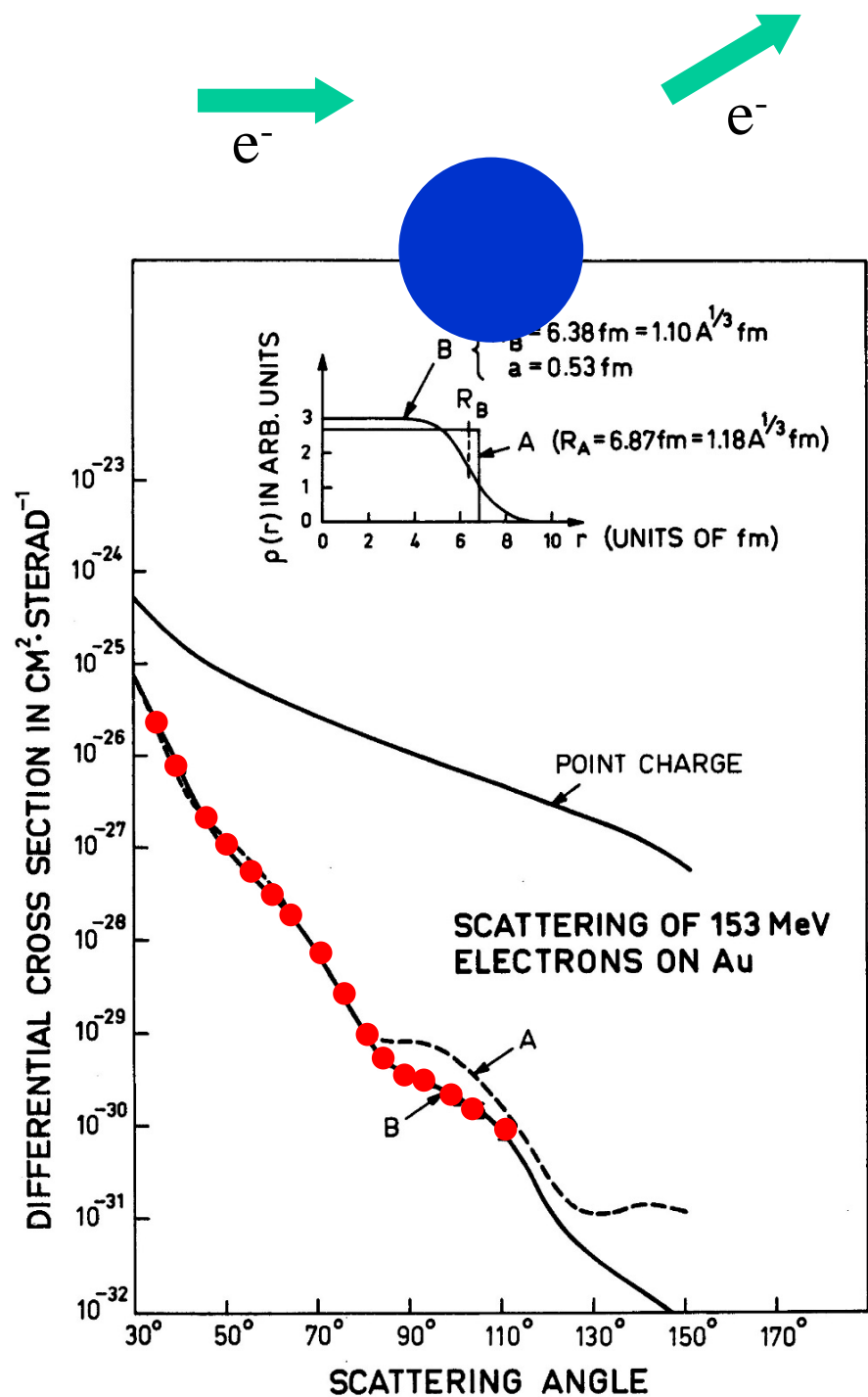
$$\frac{d\sigma}{d\Omega} = \frac{Z_P^2 e^4}{(4E \sin^2 \theta/2)^2} |F(\mathbf{q})|^2$$

$$= \left(\frac{d\sigma_{\text{Ruth}}}{d\Omega} \right) |F(\mathbf{q})|^2$$

Form factor

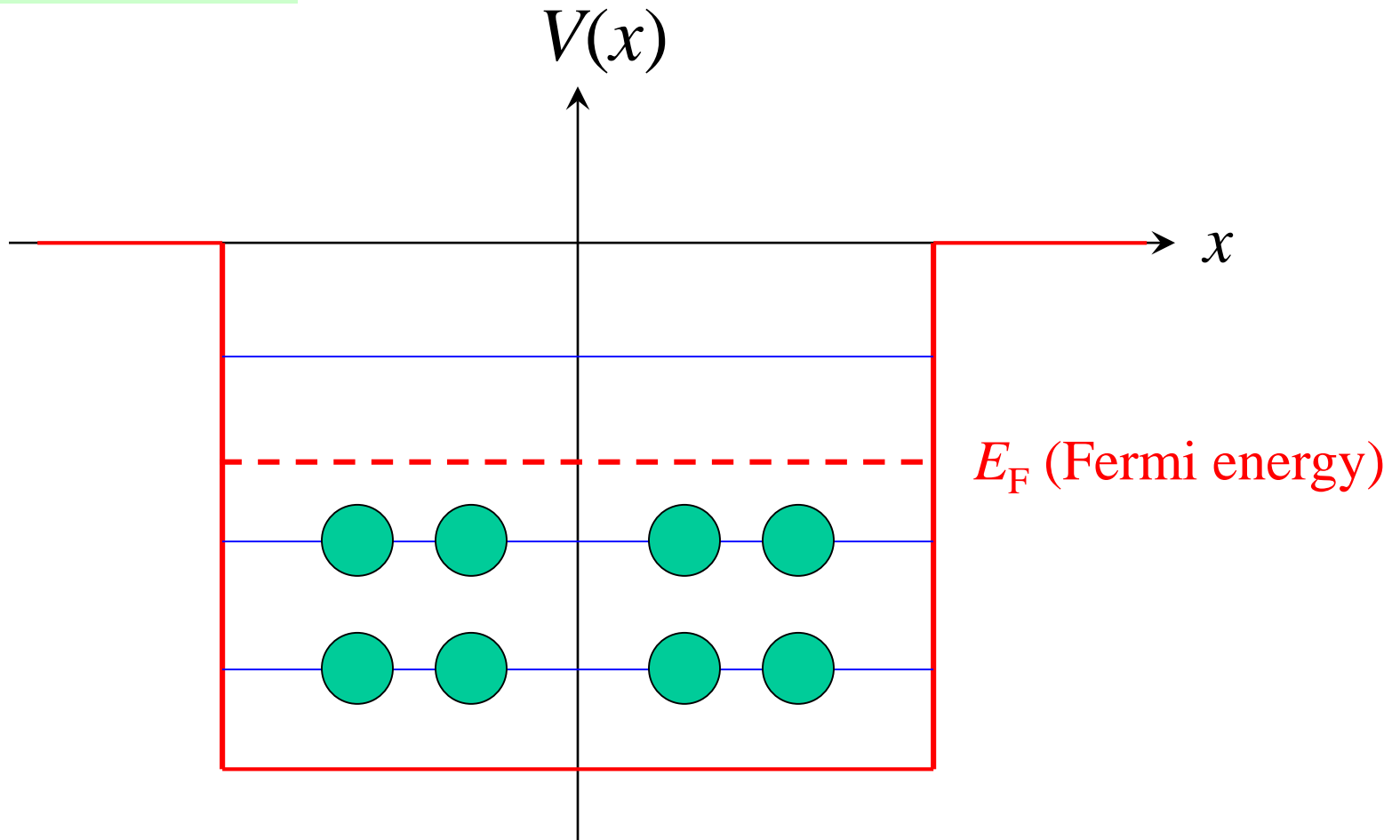
$$F(\mathbf{q}) = \int e^{-i\mathbf{q}\cdot\mathbf{r}} \rho(r) dr$$

Why do cross sections decrease for an extended density distribution?



Appendix

Fermi gas model

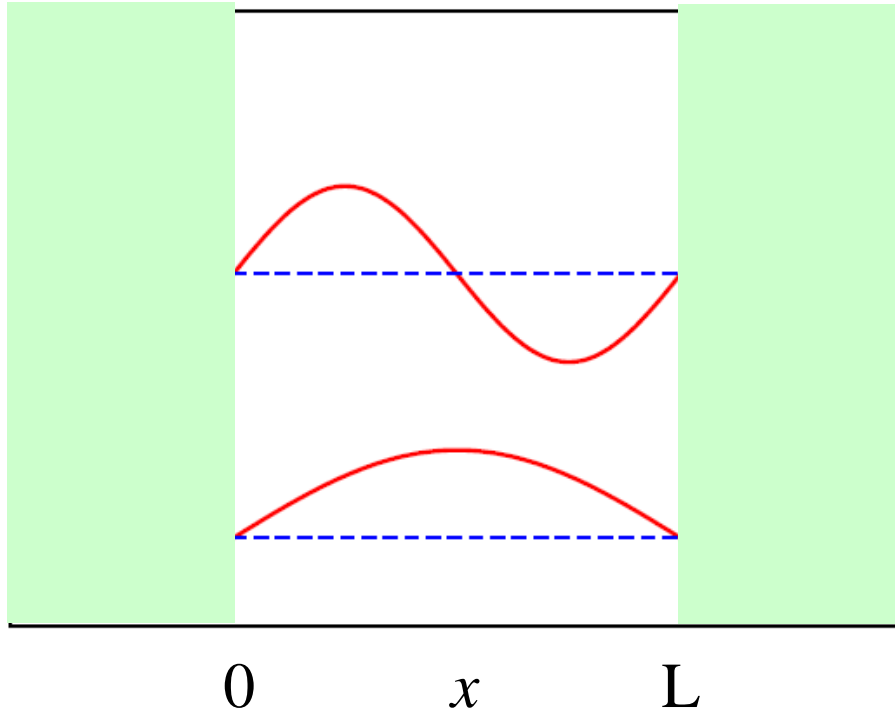


What is the relation between E_F and the particle number?

→ Fermi gas model

Fermi gas model

non-interaction many Fermion system (with no external potential)



put infinite walls at $x = 0$ and $x = L$:

$$\rightarrow \psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right)$$

$$E_n = \frac{\hbar^2}{2m} \left(\frac{n\pi}{L}\right)^2$$

$(n = 1, 2, \dots)$

three-dimensional case:

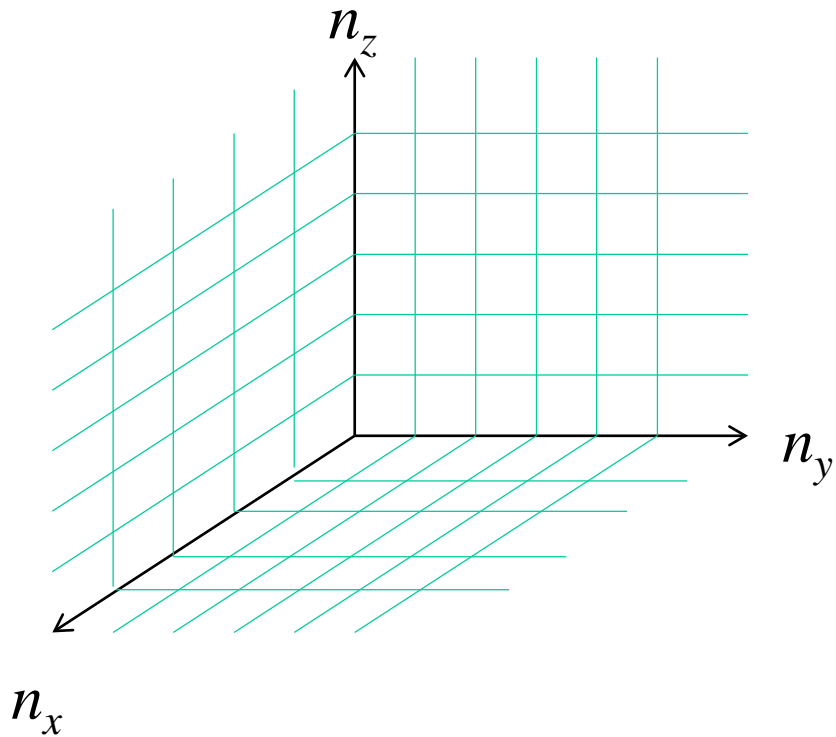
$$\psi_{n_x n_y n_z}(x, y, z) = \left(\frac{2}{L}\right)^{3/2} \sin\left(\frac{n_x \pi}{L}x\right) \sin\left(\frac{n_y \pi}{L}y\right) \sin\left(\frac{n_z \pi}{L}z\right)$$

$$E_{n_x n_y n_z} = \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 (n_x^2 + n_y^2 + n_z^2)$$

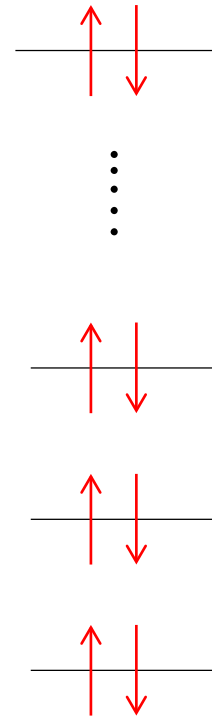
Fermi gas model

$$\psi_{n_x n_y n_z}(x) = \left(\frac{2}{L}\right)^{3/2} \sin\left(\frac{n_x \pi}{L} x\right) \sin\left(\frac{n_y \pi}{L} y\right) \sin\left(\frac{n_z \pi}{L} z\right)$$

$$E_{n_x n_y n_z} = \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 (n_x^2 + n_y^2 + n_z^2)$$



N particles



(2,1,1), (1,2,1)

(1,1,2)

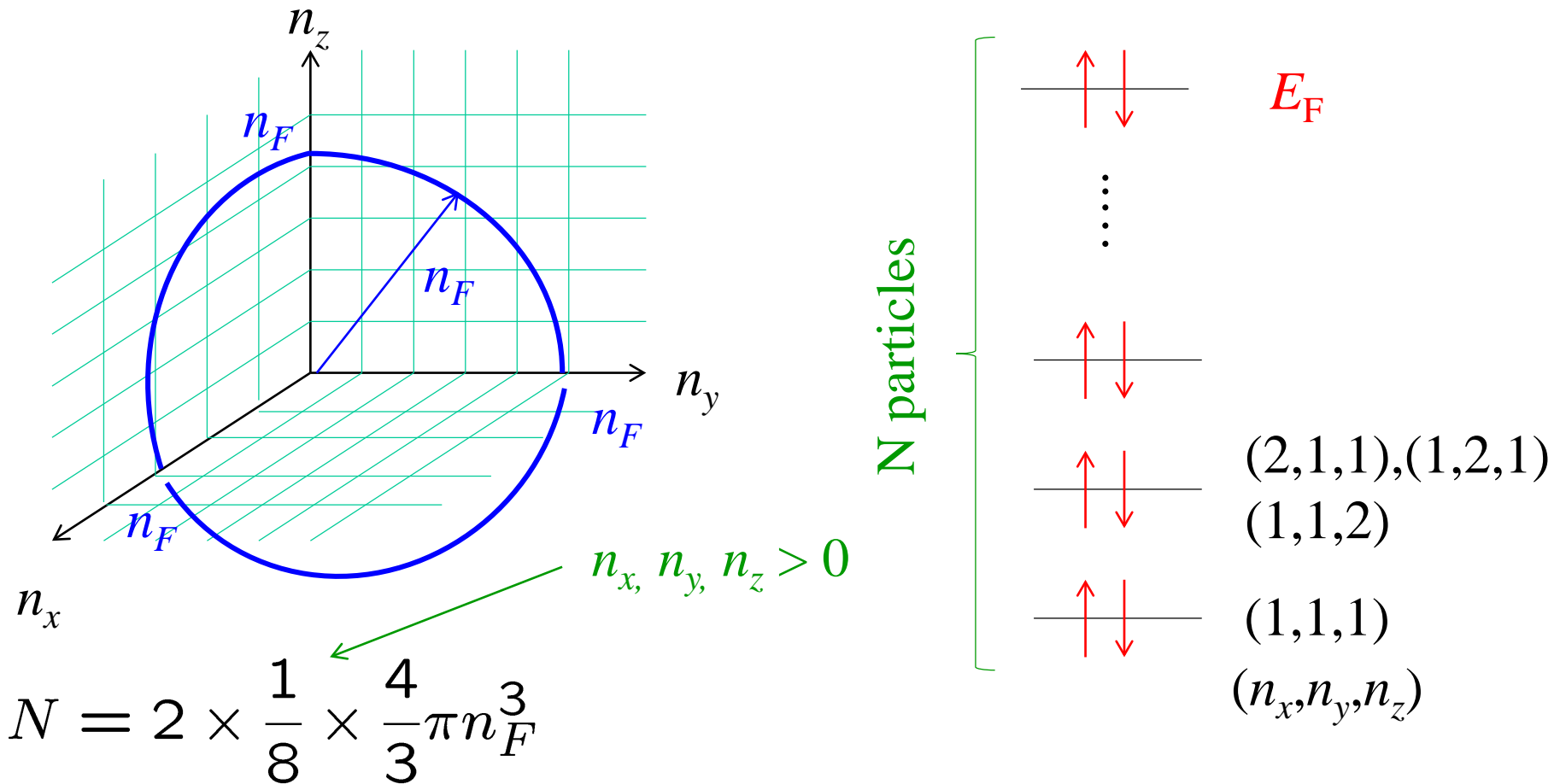
(1,1,1)

(n_x, n_y, n_z)

Fermi gas model

the highest energy:
$$E_F = \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 (n_x^2 + n_y^2 + n_z^2) \equiv \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 n_F^2$$

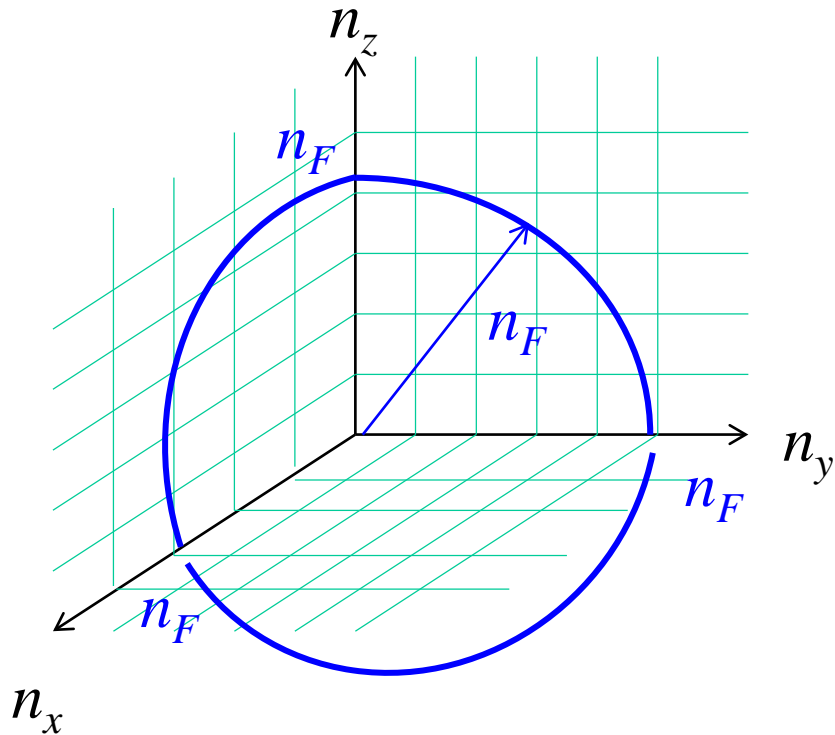
$$\longrightarrow n_F^2 = \frac{2mE_F}{\hbar^2 \pi^2} L^2$$



Fermi gas model

the highest energy: $E_F = \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 (n_x^2 + n_y^2 + n_z^2) \equiv \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 n_F^2$

$$\longrightarrow n_F^2 = \frac{2mE_F}{\hbar^2 \pi^2} L^2$$



$$N = 2 \times \frac{1}{8} \times \frac{4}{3} \pi n_F^3 = \frac{\pi}{3} \left(\frac{2mE_F}{\hbar^2 \pi^2} L^2 \right)^{3/2} = \frac{\pi}{3} \left(\frac{2mE_F}{\hbar^2 \pi^2} \right)^{3/2} L^3$$

Fermi gas model

$$N = 2 \times \frac{1}{8} \times \frac{4}{3} \pi n_F^3 = \frac{\pi}{3} \left(\frac{2mE_F}{\hbar^2 \pi^2} L^2 \right)^{3/2} = \frac{\pi}{3} \left(\frac{2mE_F}{\hbar^2 \pi^2} \right)^{3/2} L^3$$

→

$$\rho = \frac{N}{V} = \frac{N}{L^3} = \frac{\pi}{3} \left(\frac{2mE_F}{\hbar^2 \pi^2} \right)^{3/2}$$

or

$$E_F = \frac{\pi^2 \hbar^2}{2m} \left(\frac{3}{\pi} \rho \right)^{2/3}$$

Fermi gas model

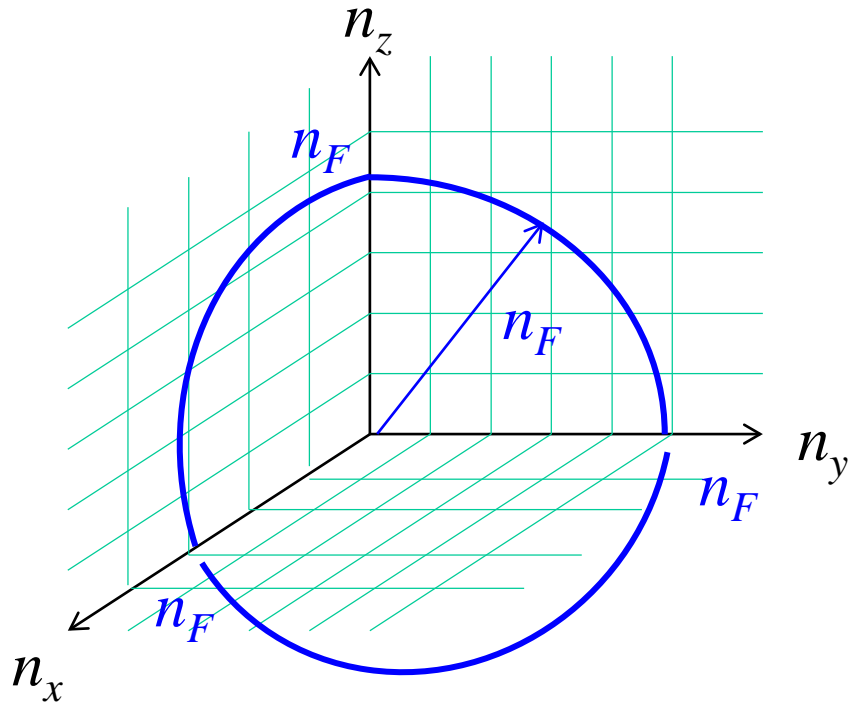
total energy

$$\begin{aligned} E_{\text{tot}} &= 2 \times \frac{1}{8} \int_{|\mathbf{n}| \leq n_F} E_{n_x n_y n_z} d^3 n \\ &= 2 \times \frac{1}{8} \int_{|\mathbf{n}| \leq n_F} \frac{\pi^2 \hbar^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2) d^3 n \\ &= \dots = \frac{\hbar^2 \pi^3}{10mL^2} n_F^5 \end{aligned}$$

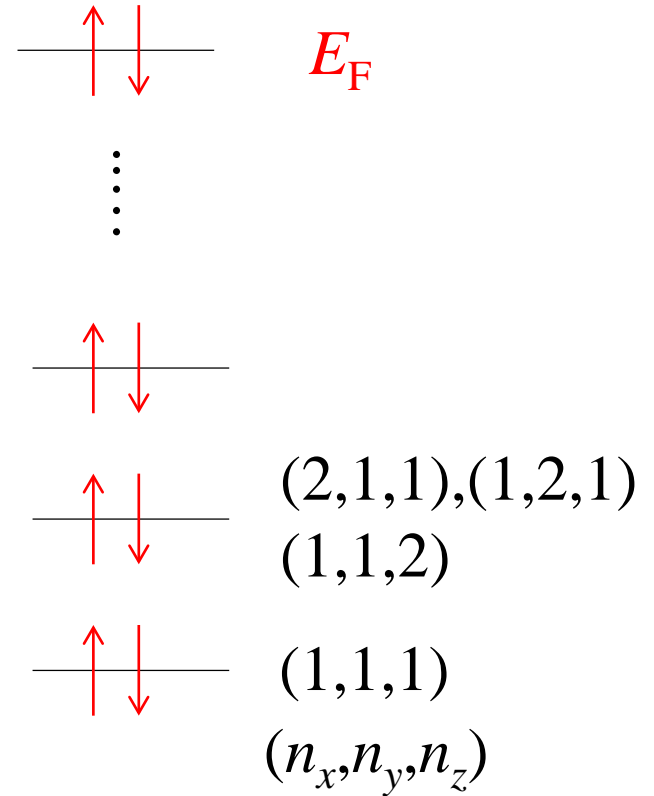
$$N = 2 \times \frac{1}{8} \times \frac{4}{3} \pi n_F^3$$

$$\longrightarrow E_{\text{tot}} = \frac{3}{5} N \cdot \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{2/3} = \frac{3}{5} N E_F$$

Fermi gas model



N particles



$$N = 2 \times \frac{1}{8} \times \frac{4}{3} \pi n_F^3$$

$$E_{\text{tot}} = 2 \times \frac{1}{8} \int_{|\mathbf{n}| \leq n_F} E_{n_x n_y n_z} d^3 n$$

$$= \dots = \frac{3}{5} N \cdot \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{2/3}$$