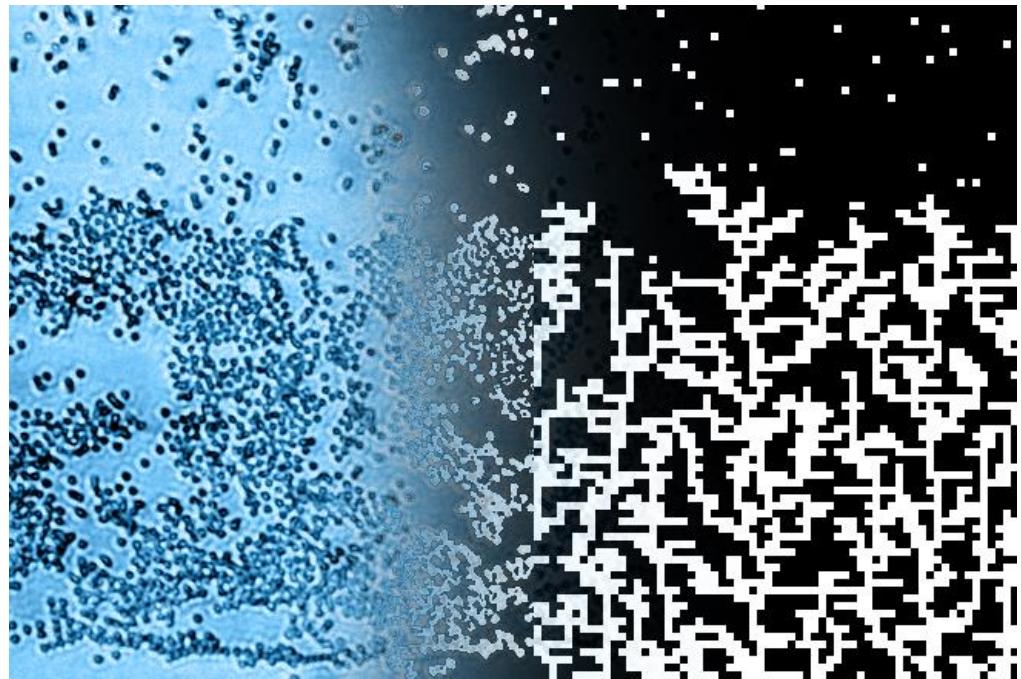


Growth Processes in Evaporating Drops

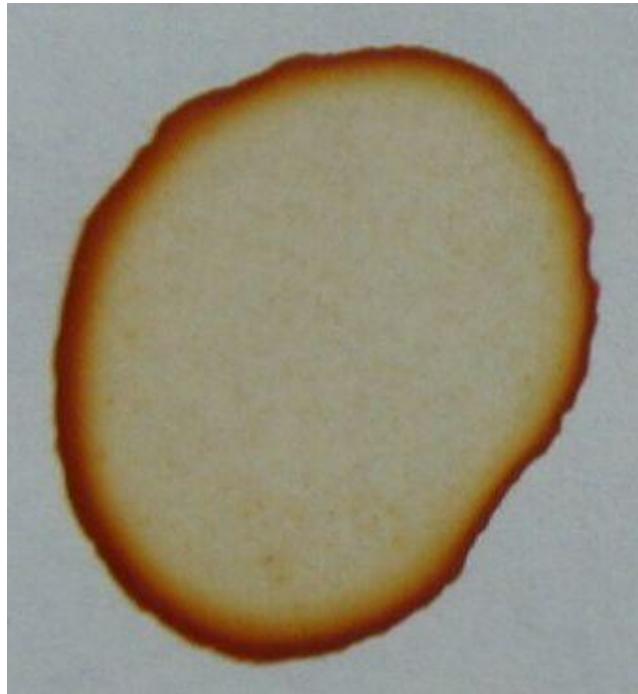


**Peter Yunker (Harvard, UPenn, GT)
Tim Still, Matt Lohr,
Arjun Yodh, Doug Durian (UPenn)
Alexei Borodin (MIT)**

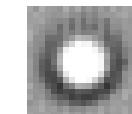
P.J. Yunker, T. Still, M.A. Lohr, and A.G. Yodh, *Nature* 476, 308 (2011).
P.J. Yunker, *et al.*, PRL (2012).
P.J. Yunker, *et al.*, PRL (2013). P.J. Yunker, *et al.*, Physics Today (2013).



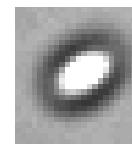
Growth process depends on particle shape



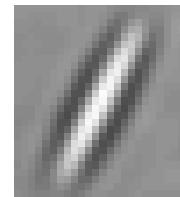
Coffee-ring effect



Poisson Process



KPZ Process

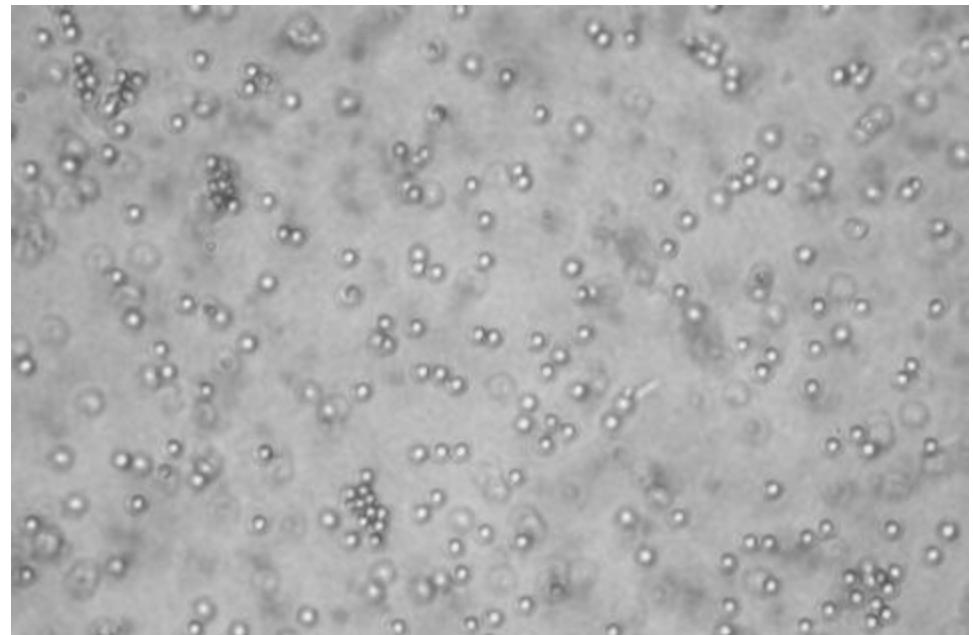
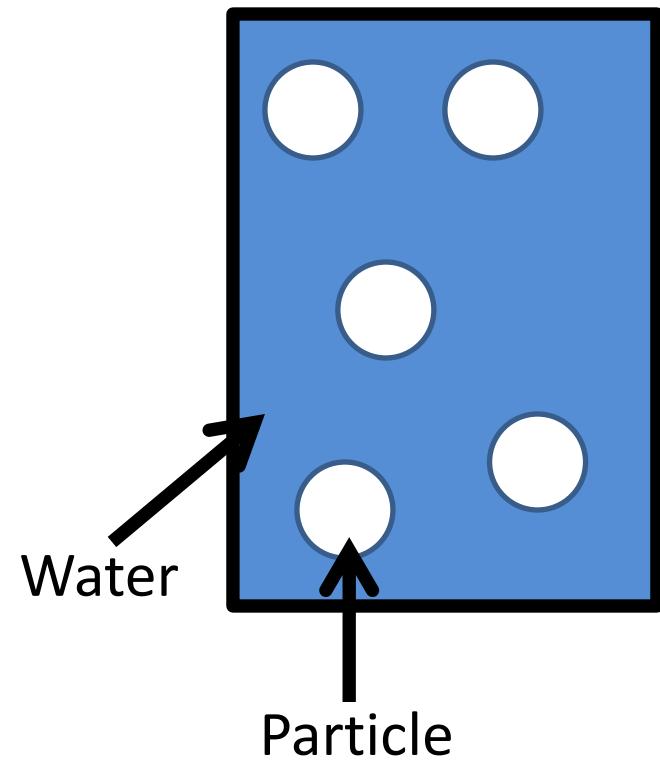


Colloidal Matthew Effect

Which growth processes?

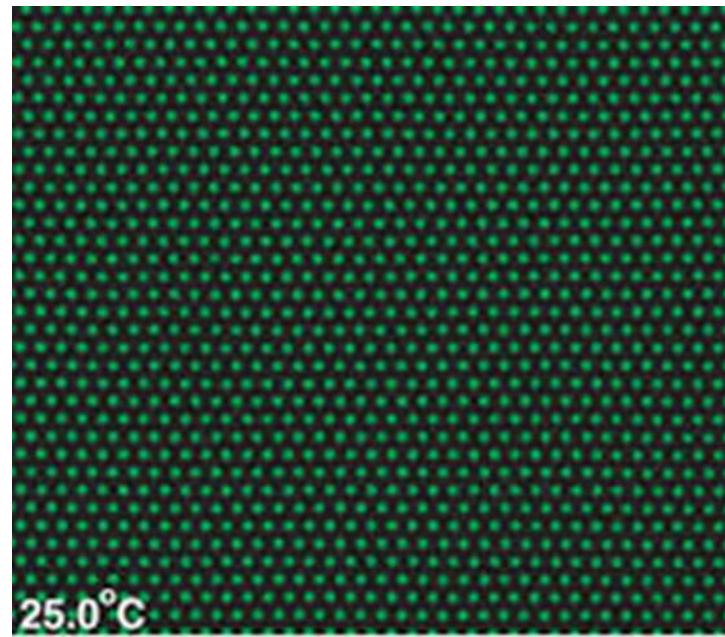
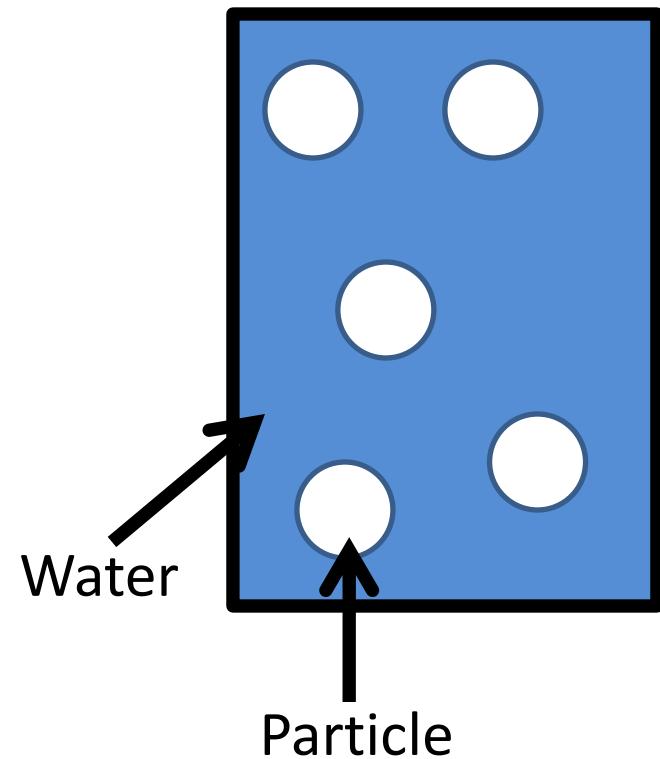
Why particle shape?

Colloidal particles: a convenient experimental tool



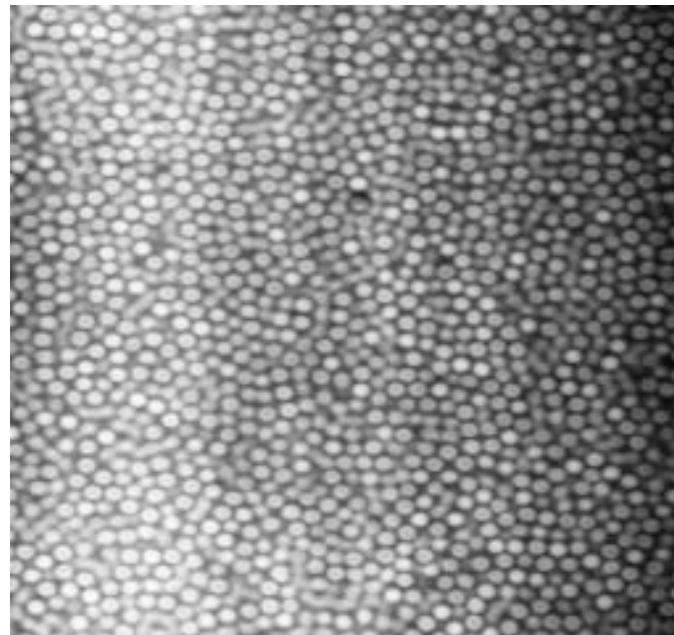
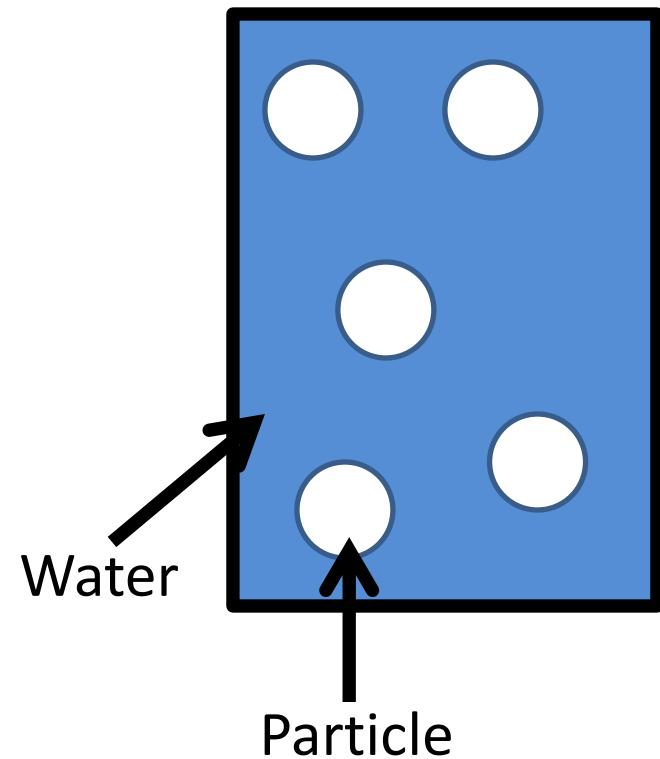
Colloidal Fluid

Colloidal particles: a convenient experimental tool



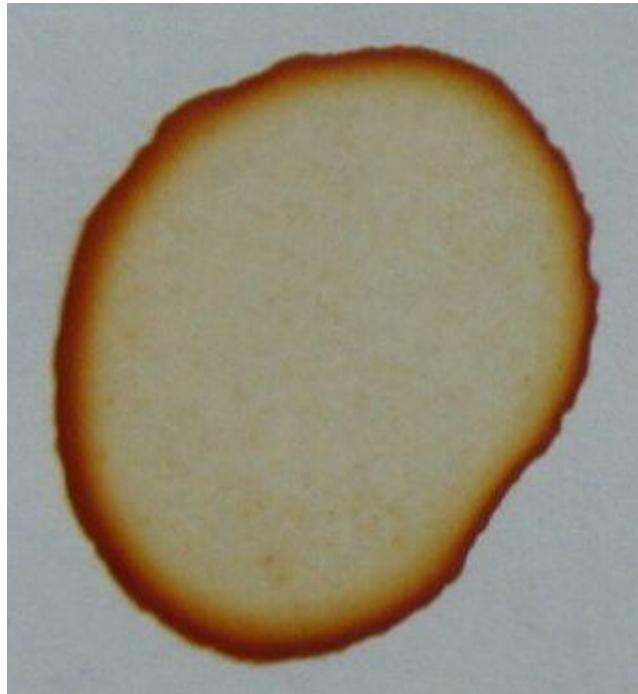
Colloidal Crystal

Colloidal particles: a convenient experimental tool

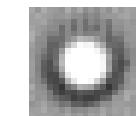


Colloidal Glass

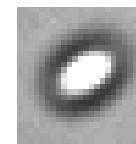
Growth process depends on particle shape



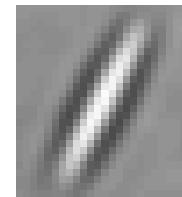
Coffee-ring effect



Poisson Process



KPZ Process



Colloidal Matthew Effect

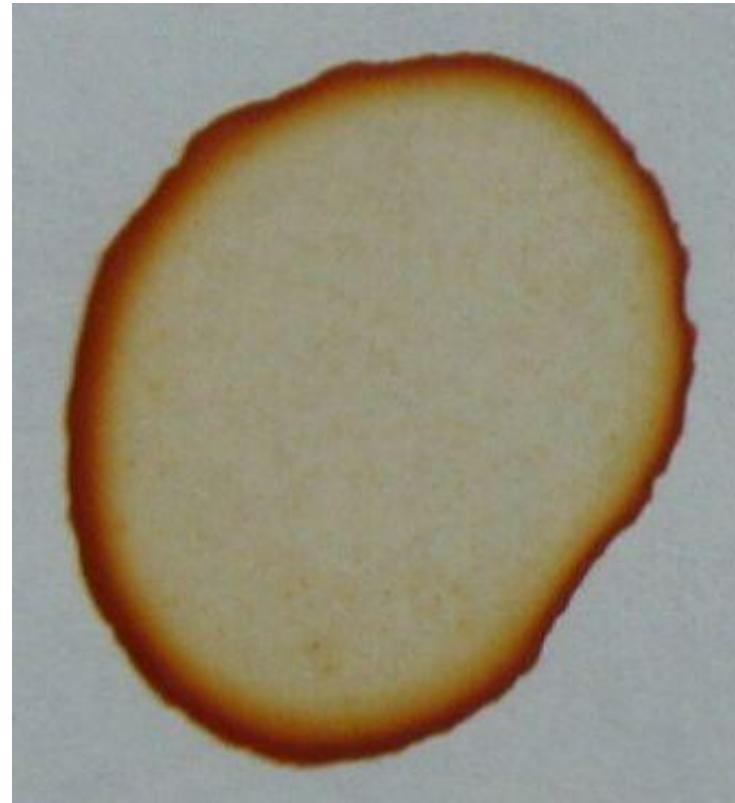
Which growth processes?

Why particle shape?

The Coffee Ring Effect



The Coffee Ring Effect



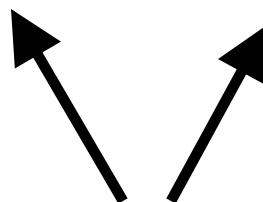
Robert D. Deegan, Olgica Bakajin, T.F. Dupont, G.Huber, Sidney R. Nagel , Thomas A. Witten, *Nature* (1997).

Drop Edges Pinned During Evaporation

Edges Unpinned – Diameter Decreases



Edges Pinned – Contact Angle Decreases



Drop edges pinned

Convective Fluid Flow from Middle of Drop to Edges

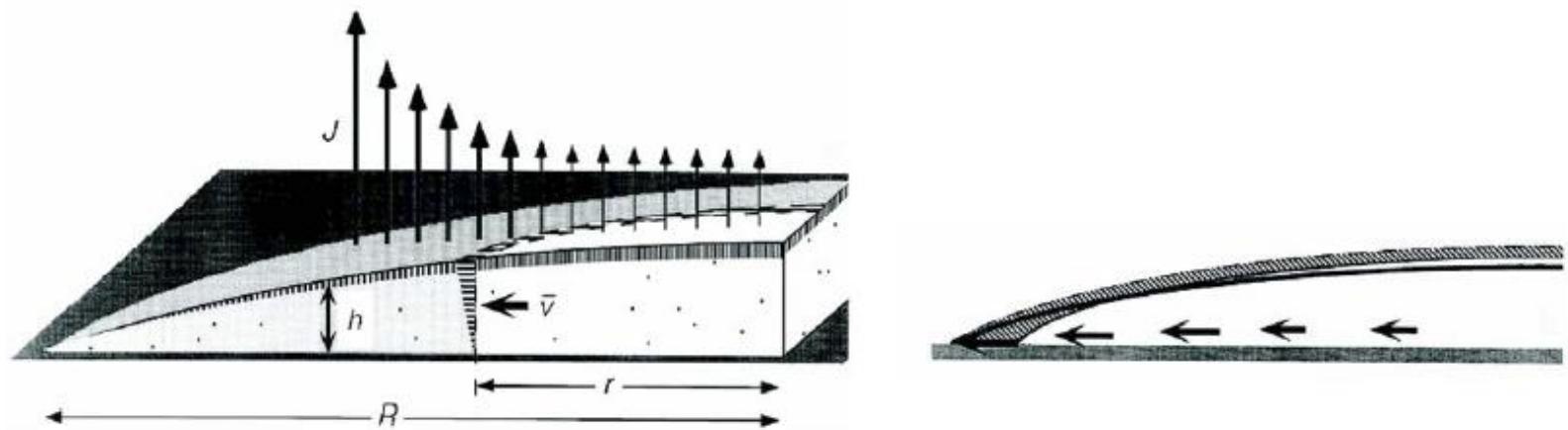
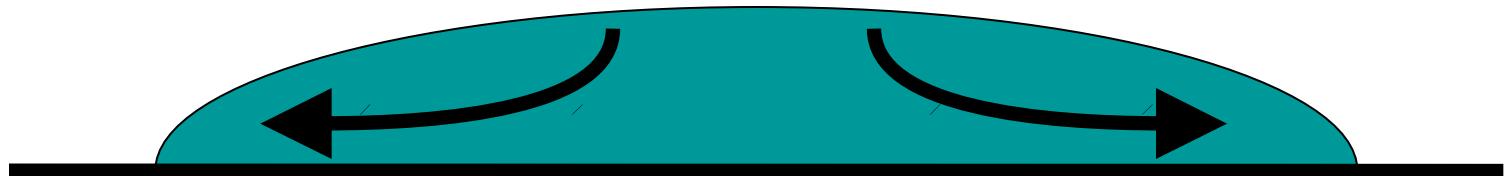
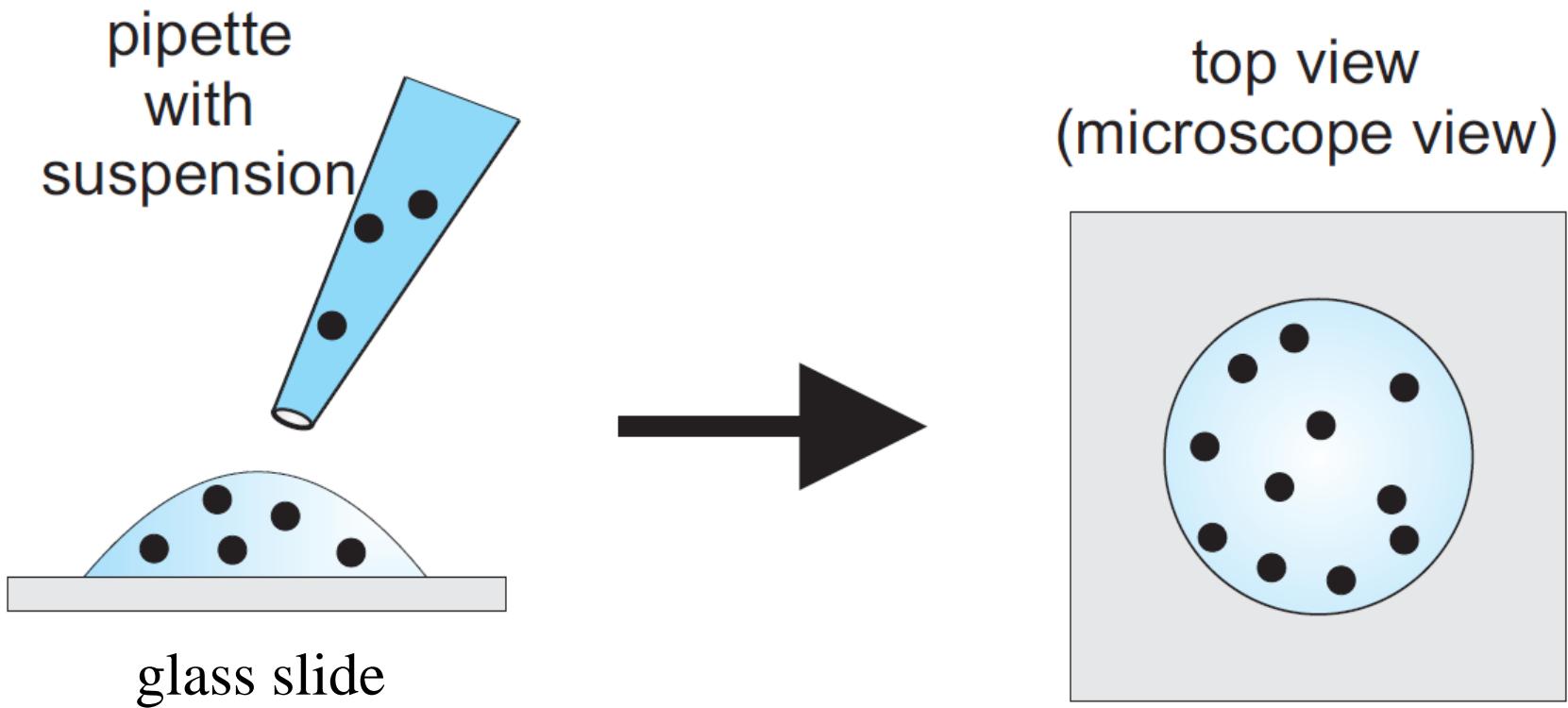


Figure from R. D. Deegan et al., Nature 389, 827 (1997).

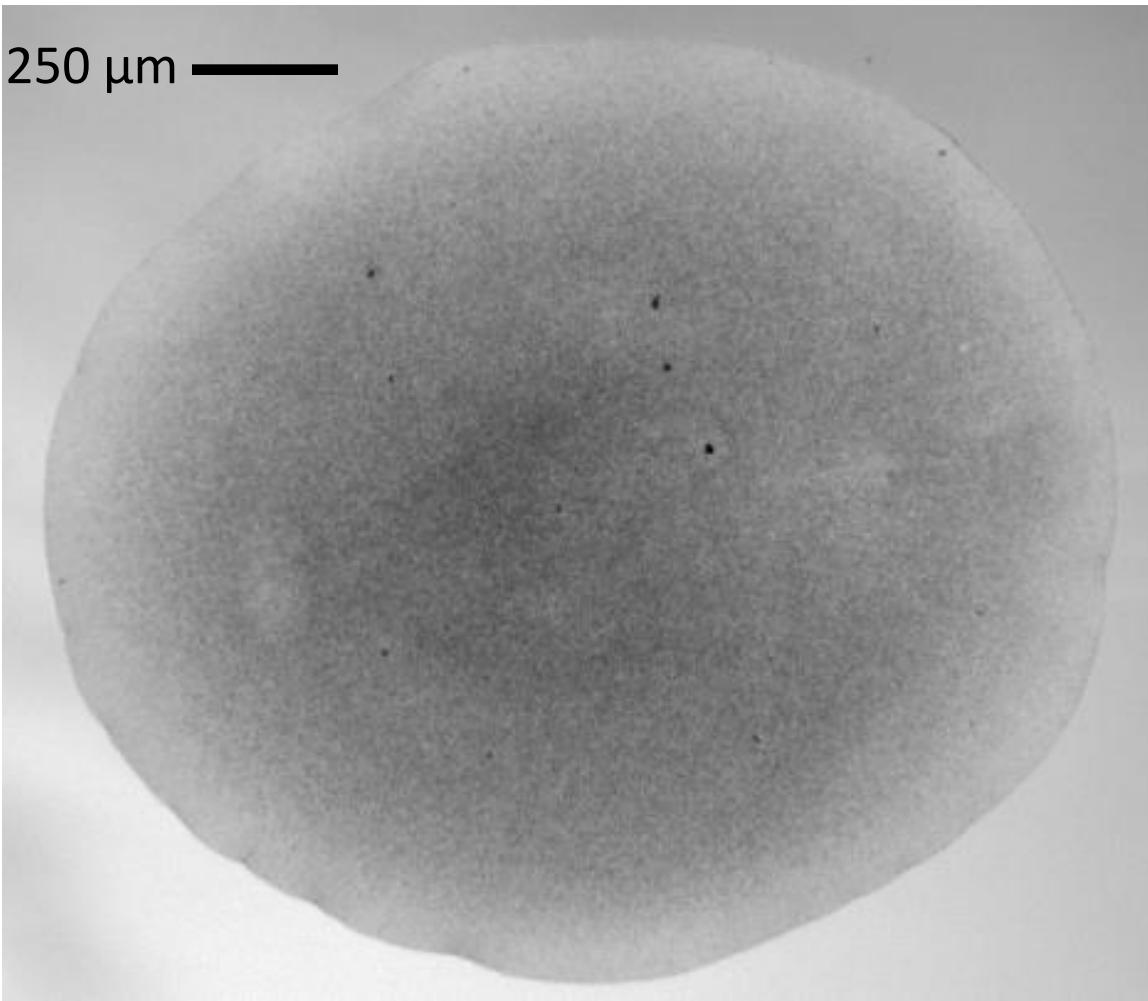


Basic Experiment



Drying Drops Containing Polystyrene Spheres

(20x faster) 250 μm —

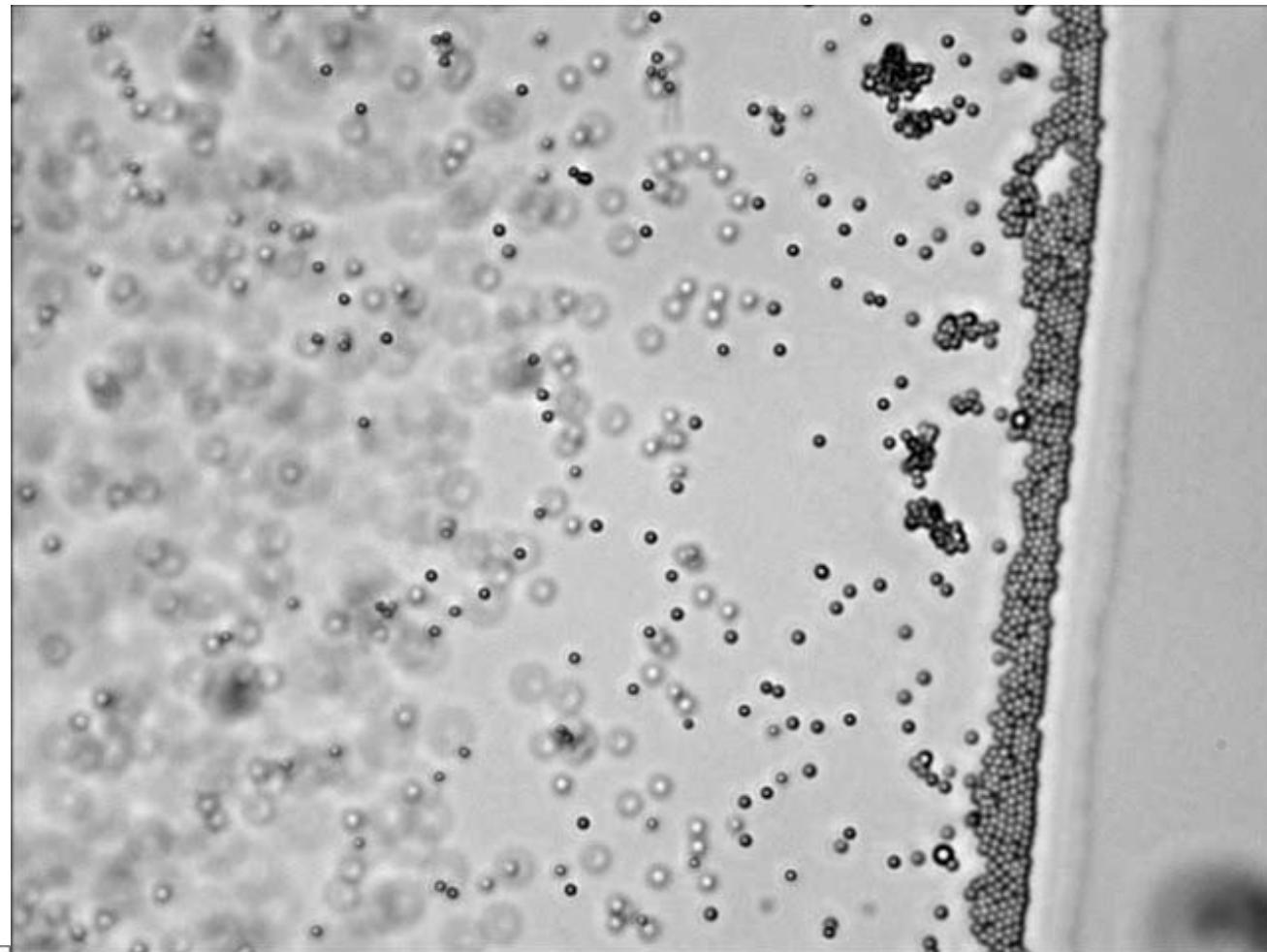


Initial
Volume
Fraction
 $\phi = 0.005$



Drying Drops Containing Polystyrene Spheres

10 μm

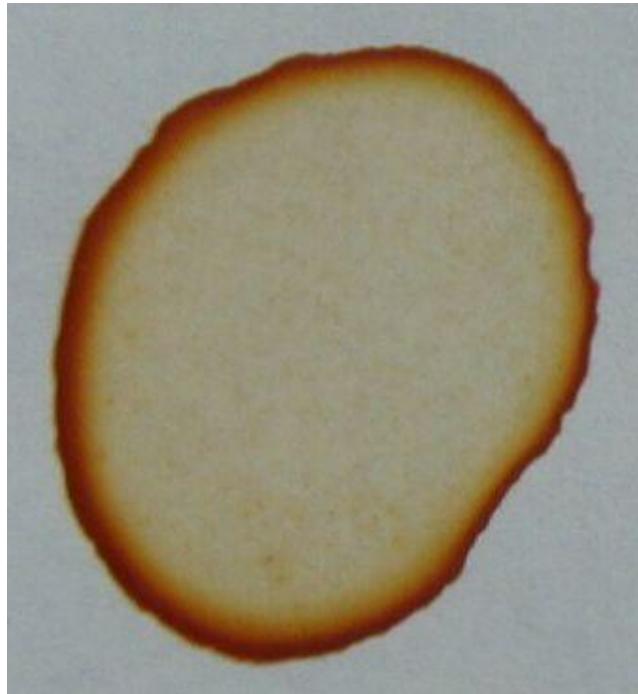


(10x faster)

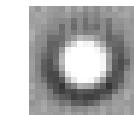
Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

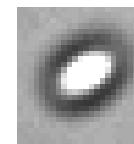
Growth process depends on particle shape



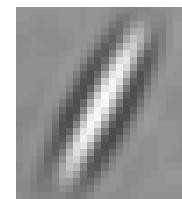
Coffee-ring effect



Poisson Process



KPZ Process

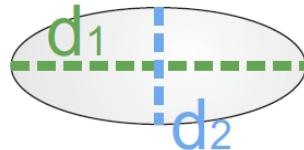
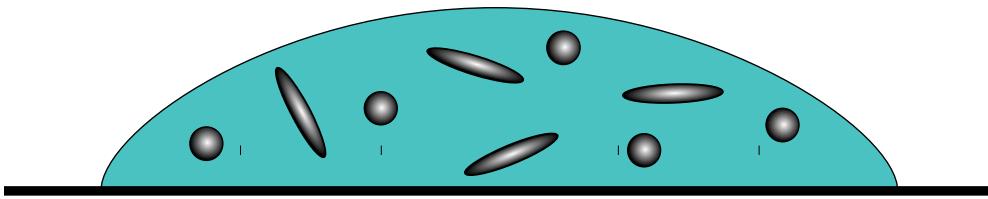


Colloidal Matthew Effect

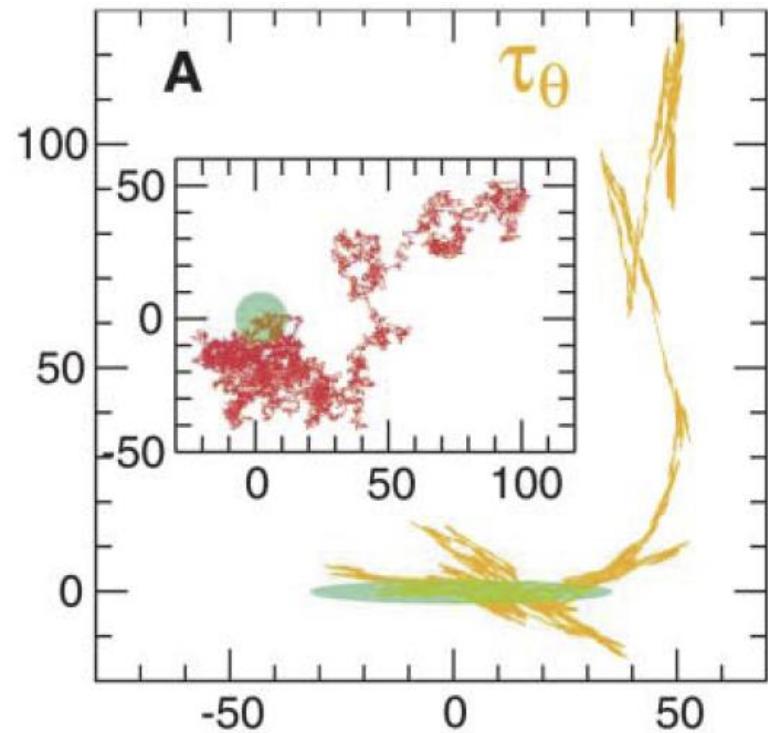
Which growth processes?

Why particle shape?

Why Does Particle Shape Matter?



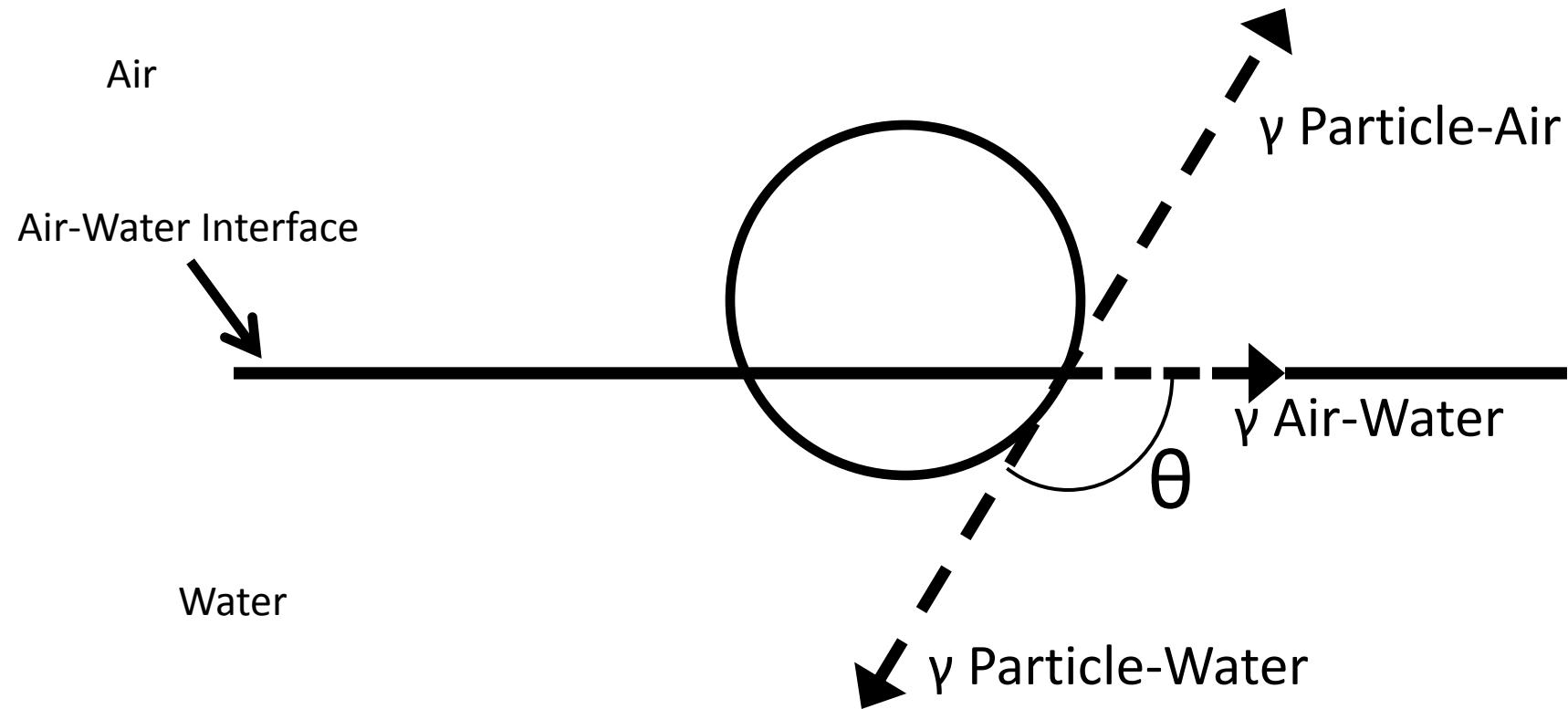
$$\epsilon = d_1/d_2$$



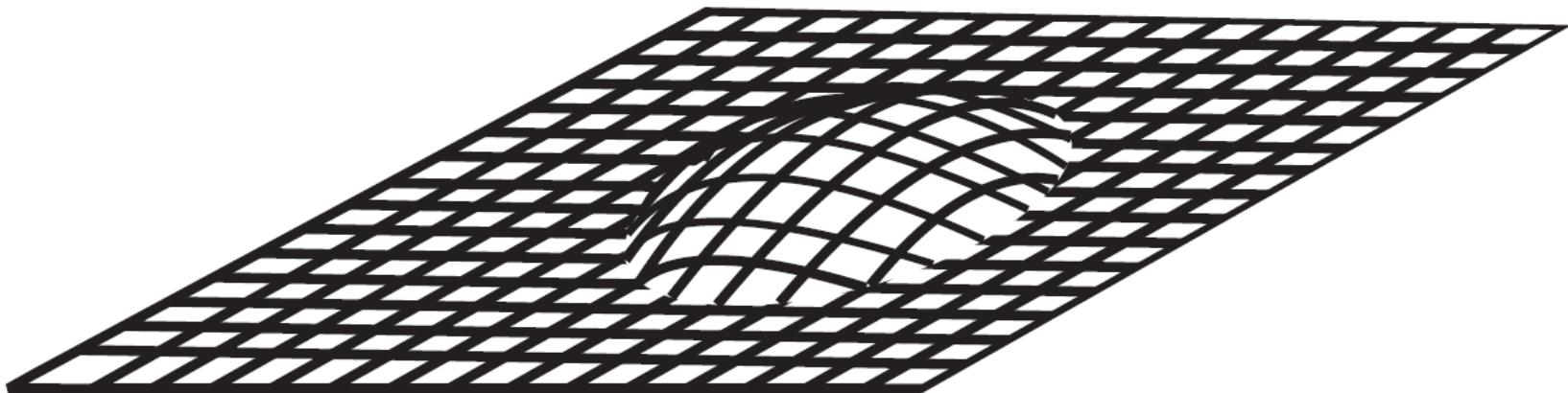
Han, Y., Alsayed, A.M., Nobili, M., Zhang, J., Lubensky, T.C., & Yodh, A.G., Brownian motion of an ellipsoid. *Science* (2006).

**“Spheres” and “Ellipsoids” at low volume fraction
will have fairly similar bulk behaviors.**

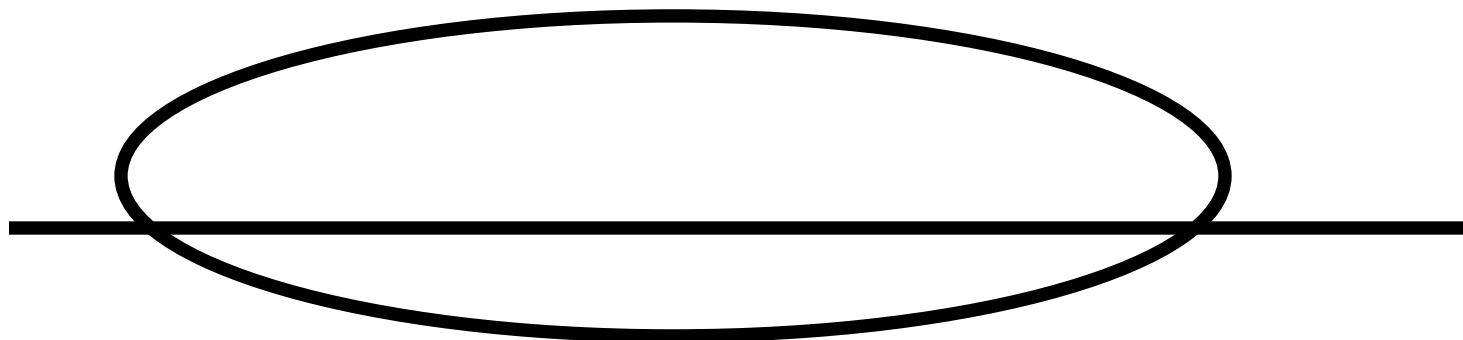
Interfacial tensions balance at equilibrium contact angle



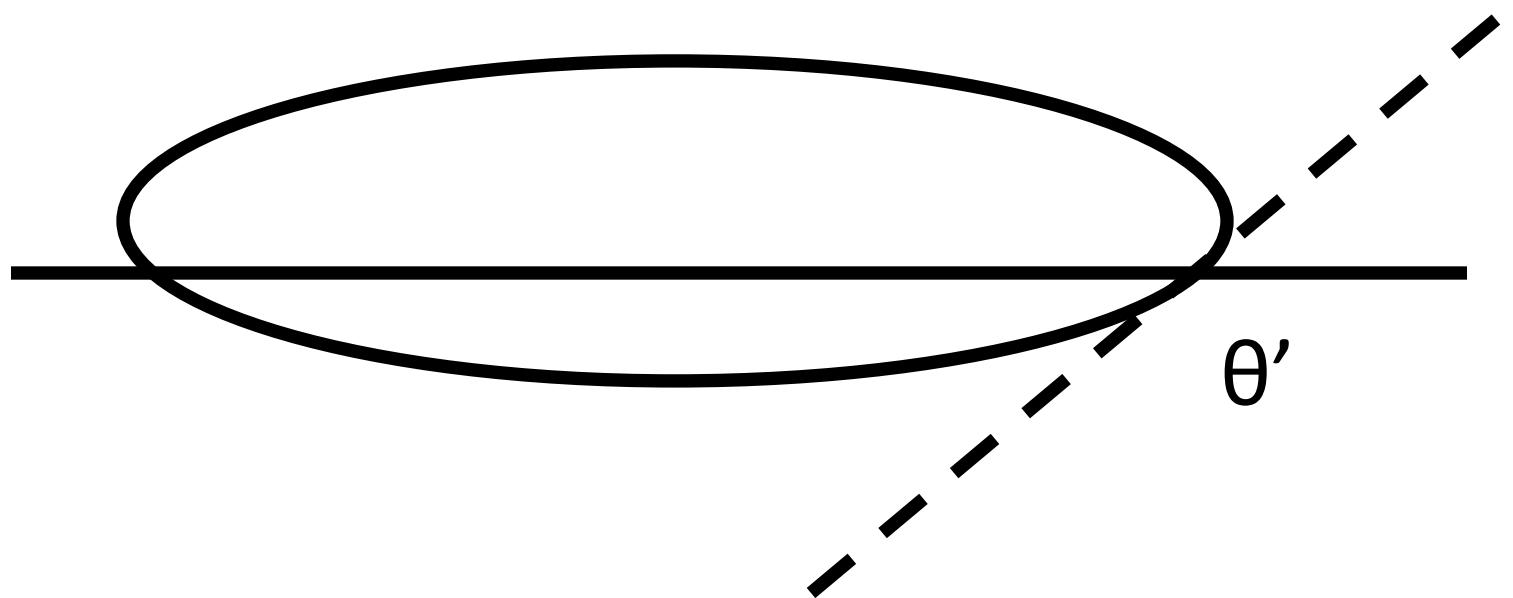
Interfacial tensions balance at equilibrium contact angle



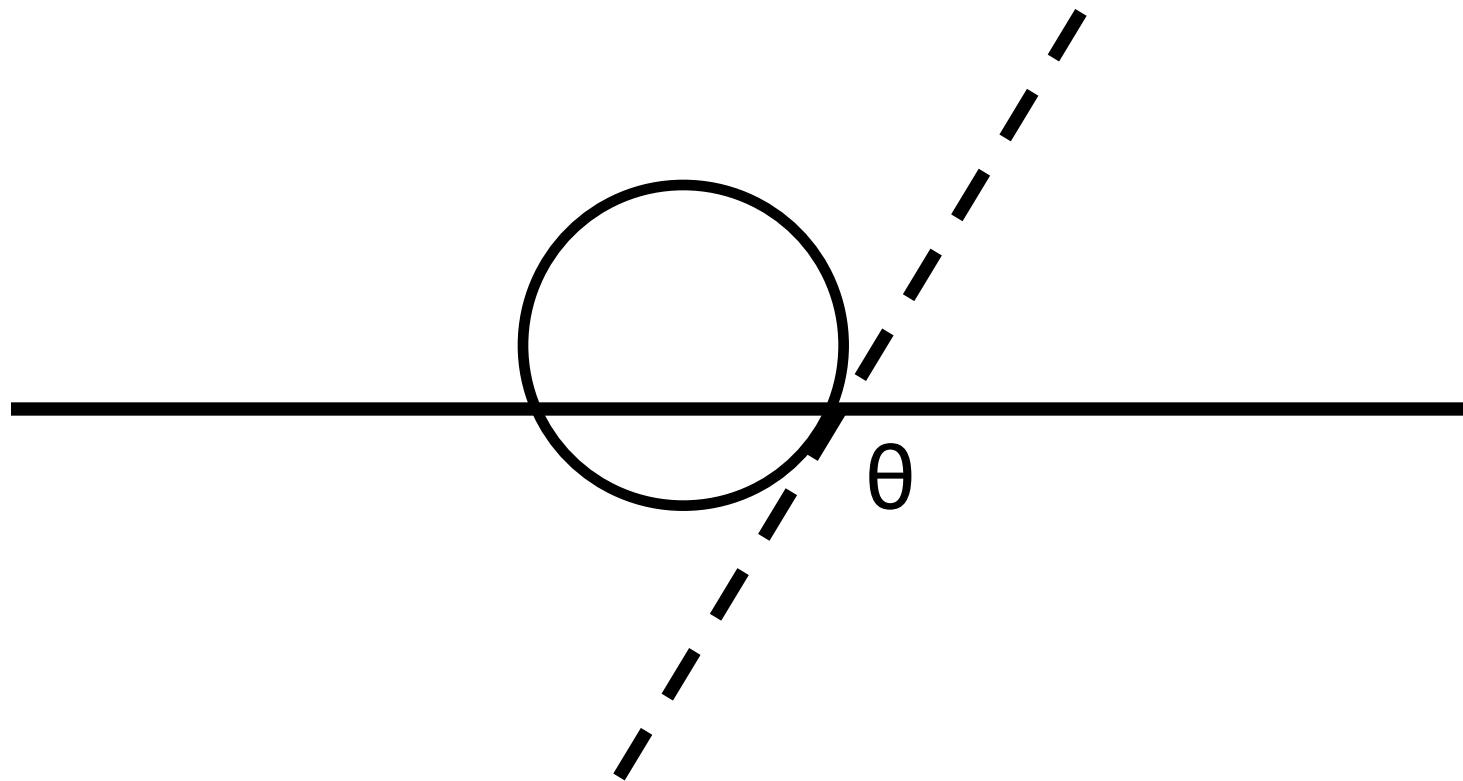
Interfacial tensions balance at equilibrium contact angle



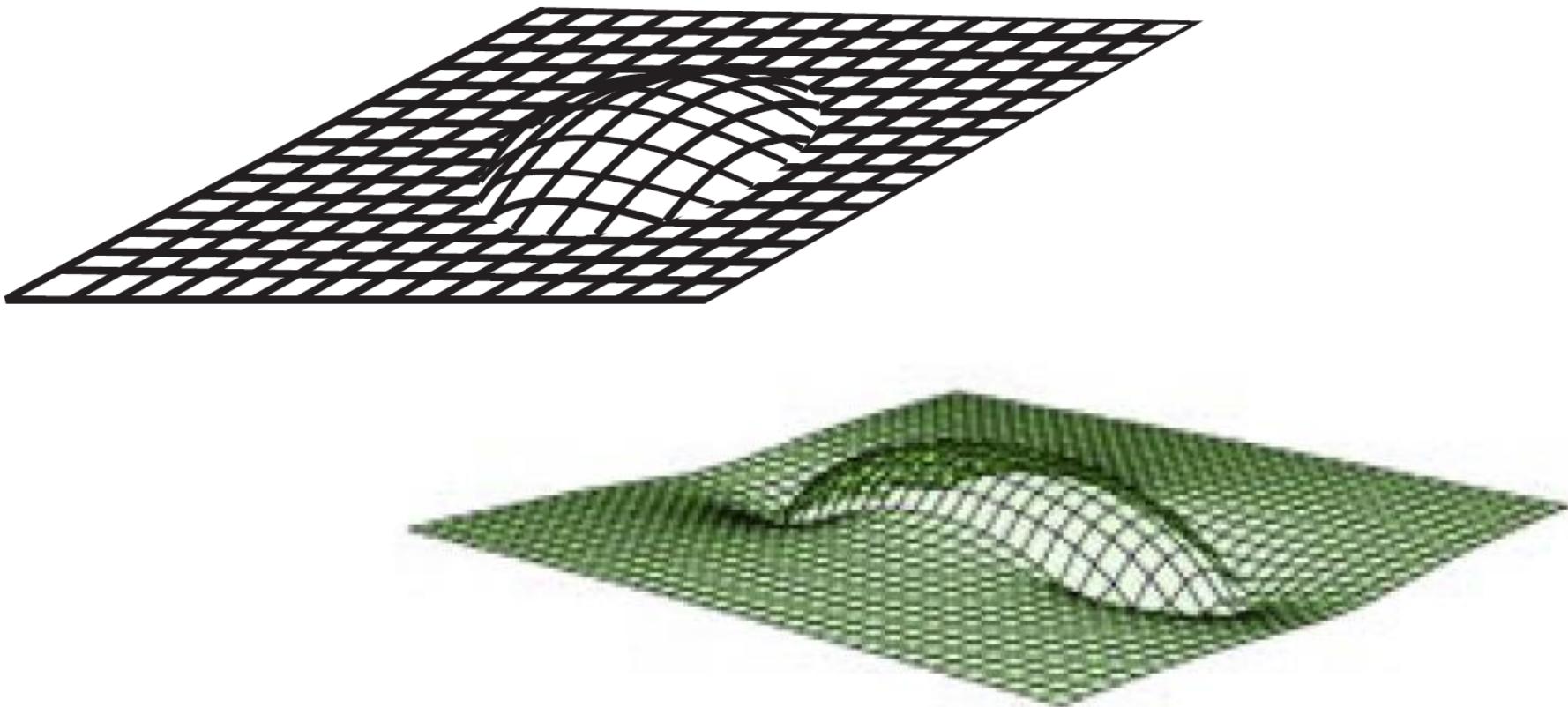
Interfacial tensions balance at equilibrium contact angle



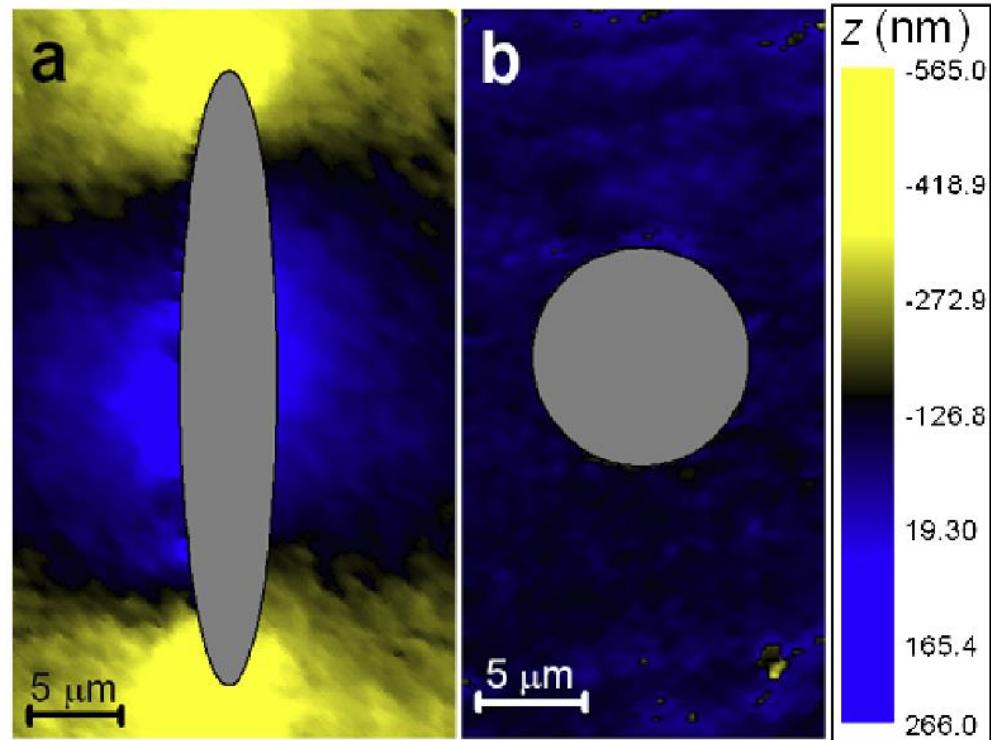
Interfacial tensions balance at equilibrium contact angle



Interfacial tensions balance at equilibrium contact angle

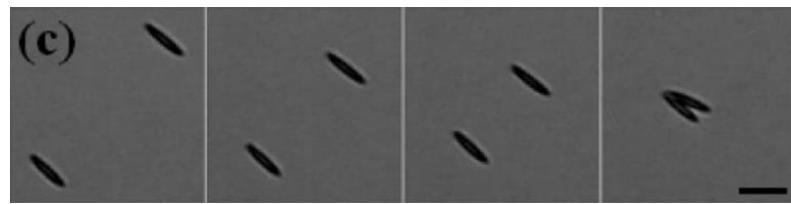
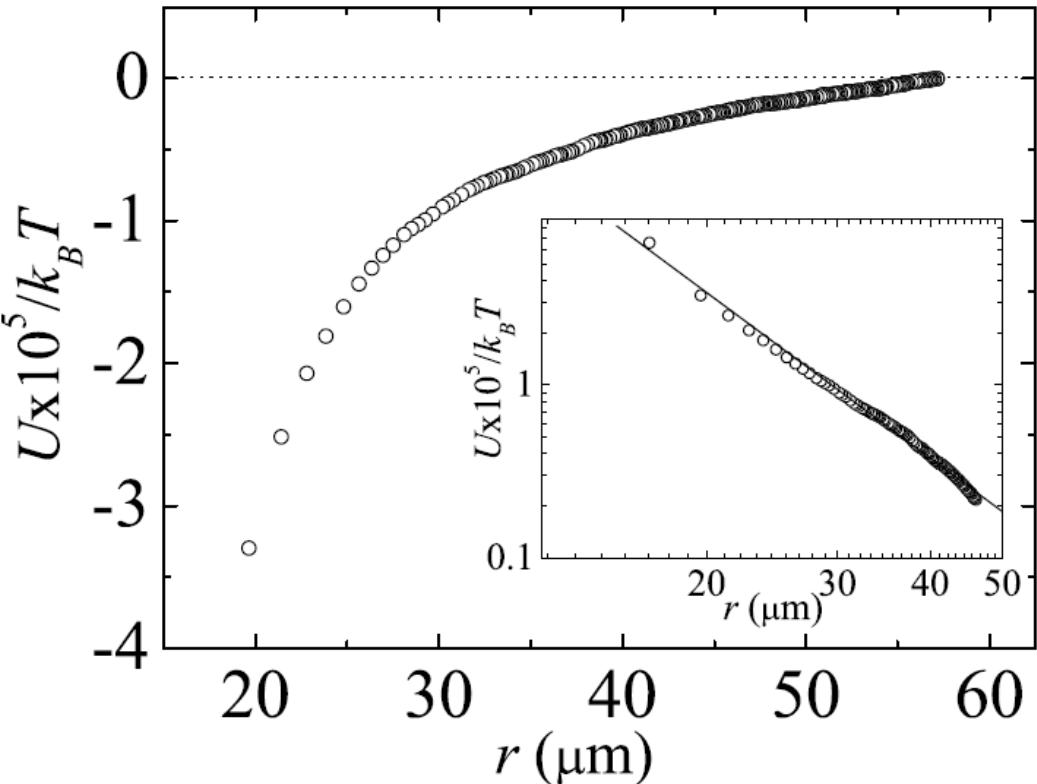


Shape Anisotropy Deforms Interface



Loudet, J.C., Alsayed, A.M., Zhang, J., and Yodh, A.G., *Phys Rev Lett* (2005); Loudet J.C., Yodh A.G., Pouliquen B., *Phys Rev Lett* (2006); Kralchevsky, Paunov, Ivanov and Nagayama, *J. Coll. Inter. Sci.*, (1992)

Interfacial Forces Depend on Particle Shape

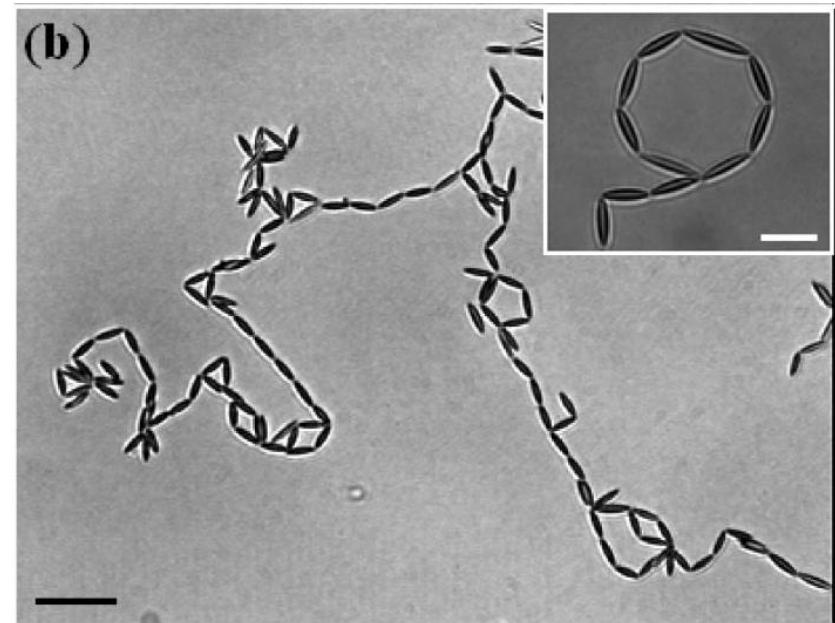
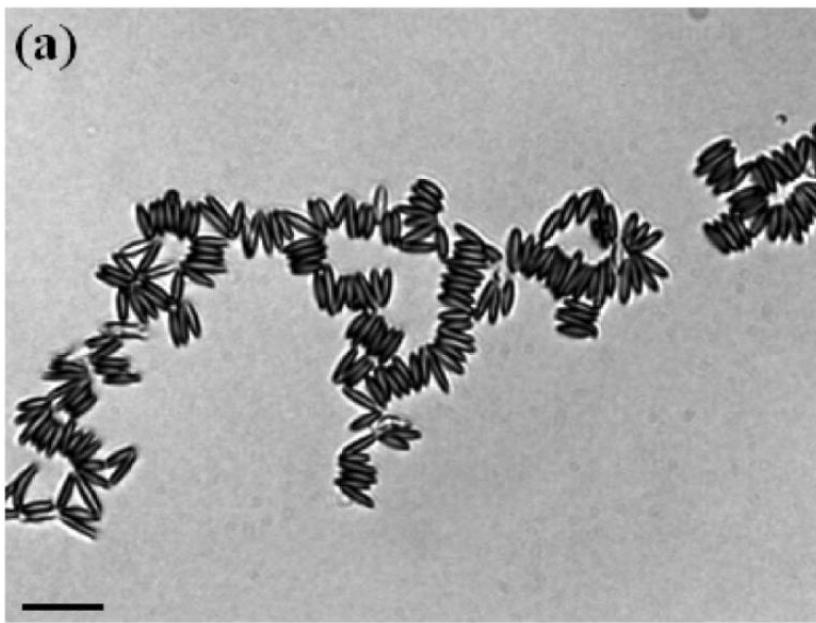


$$U \propto r^2(\varepsilon - 1)^2$$

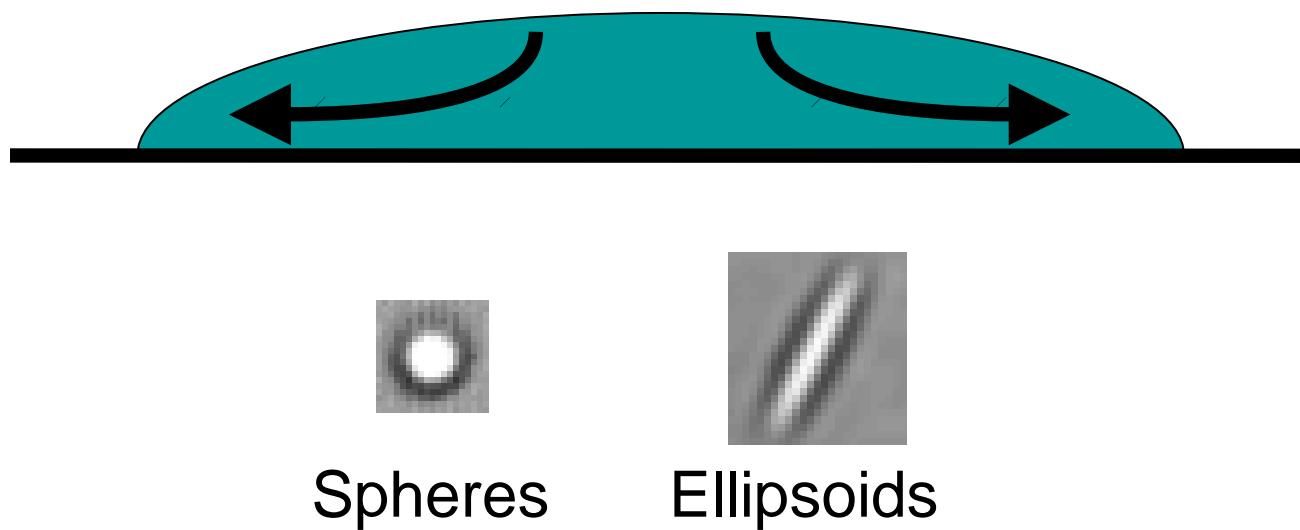
Loudet, J.C., Alsayed, A.M., Zhang, J., and Yodh, A.G., *Phys Rev Lett* (2005);
Loudet J.C., Yodh A.G., Pouliquen B., *Phys Rev Lett* (2006); Kralchevsky, Paunov,
Ivanov and Nagayama, *J. Coll. Inter. Sci.*, (1992)



Highly anisotropic ellipsoids prefer tip-to-tip or side-to-side

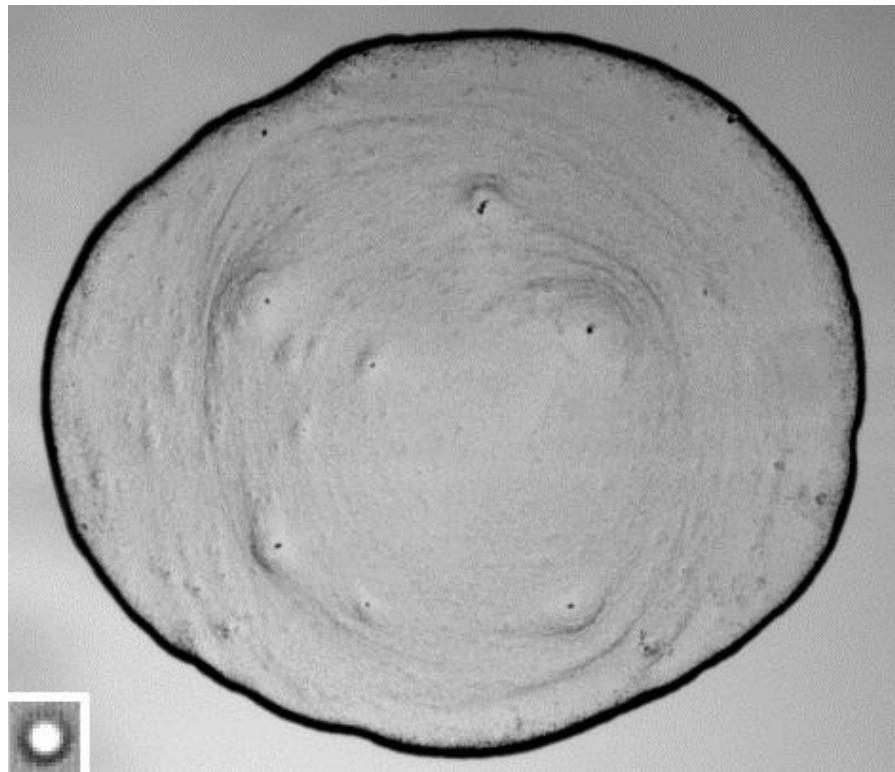


Outward flows drive coffee ring-effect

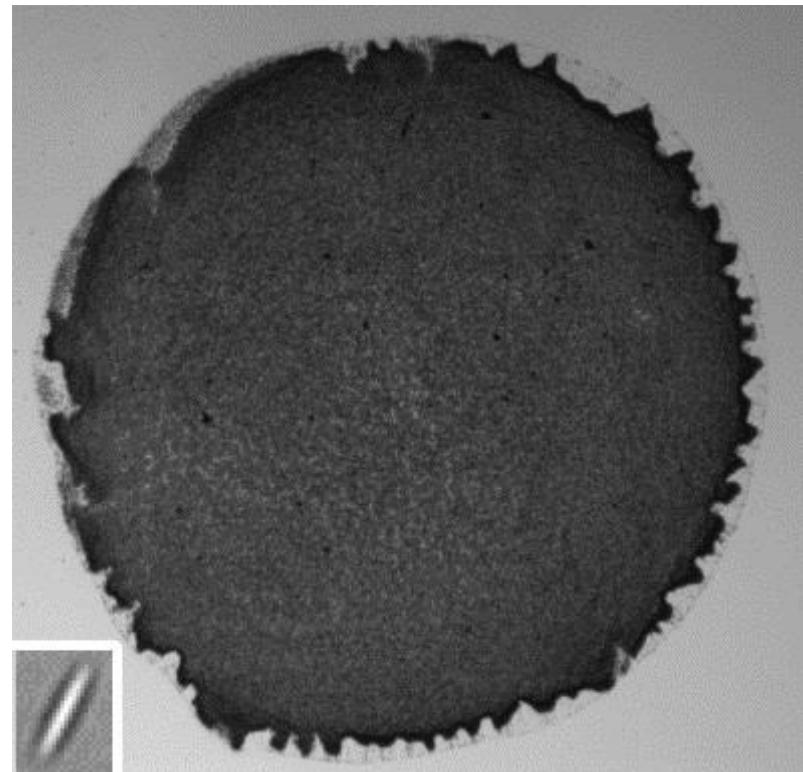


Robert D. Deegan, Olgica Bakajin, T.F. Dupont, G.Huber, Sidney R. Nagel , Thomas A. Witten, *Nature* (1997).

Coffee Rings for Spheres but not Ellipsoids!



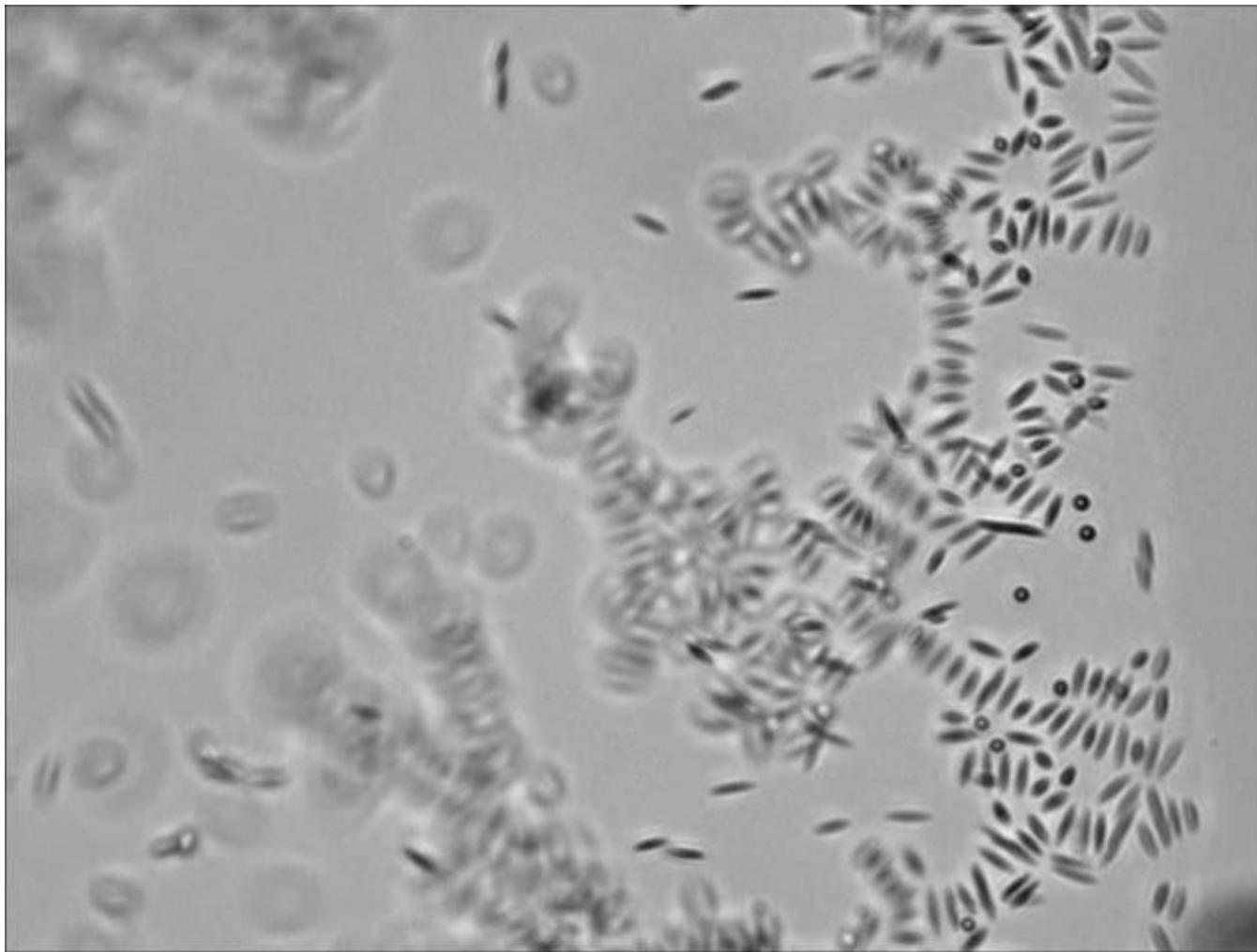
250 μ m —



Initial
Volume
Fraction
 $\phi = 0.005$

Yunker, P.J., Still, T., Lohr, M.A., and Yodh, A.G., *Nature* (2011).

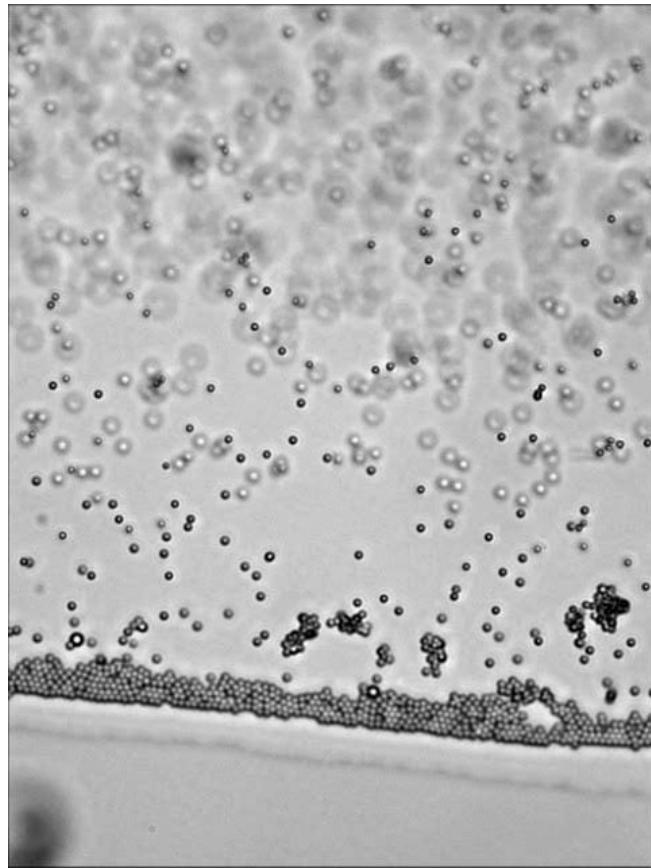
Ellipsoids form loosely-packed, open network



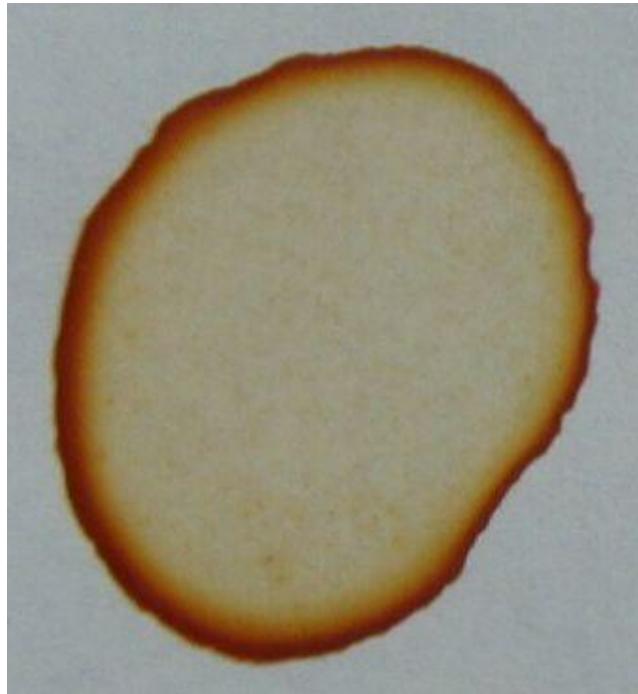
5 μm —

Drop Edge

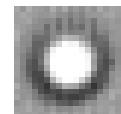
Reminiscent of growth process simulation



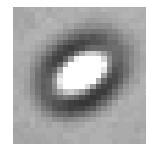
Growth process depends on particle shape



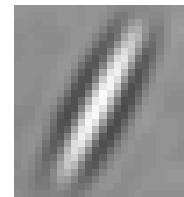
Coffee-ring effect



Poisson Process



KPZ Process

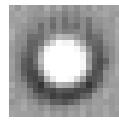
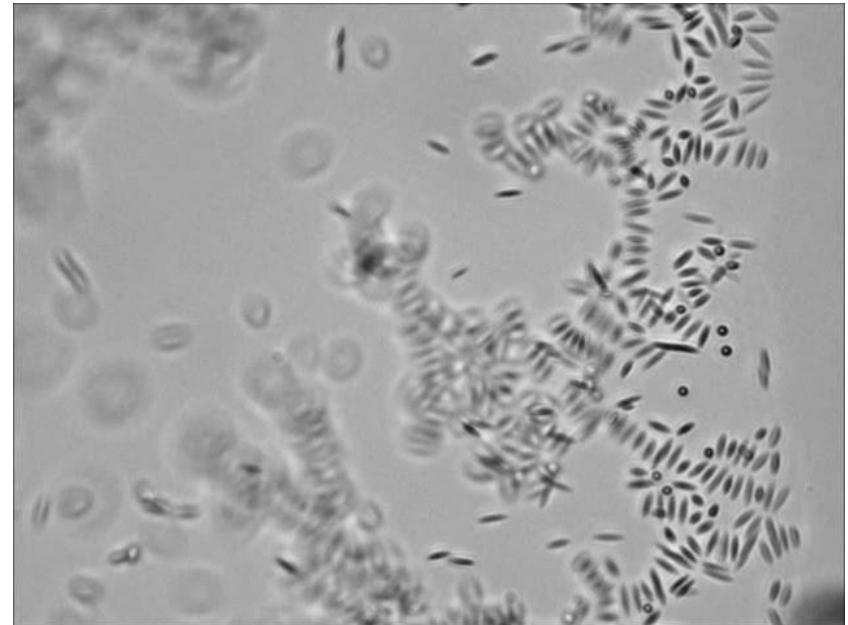
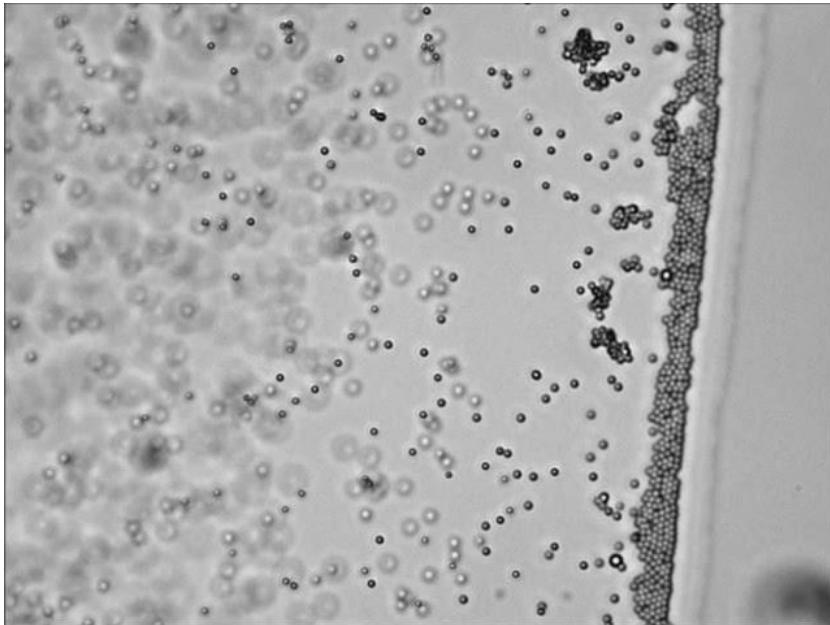


Colloidal Matthew Effect

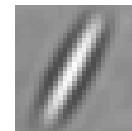
Which growth processes?

Why particle shape?

How do deposits grow during evaporation?



Spheres

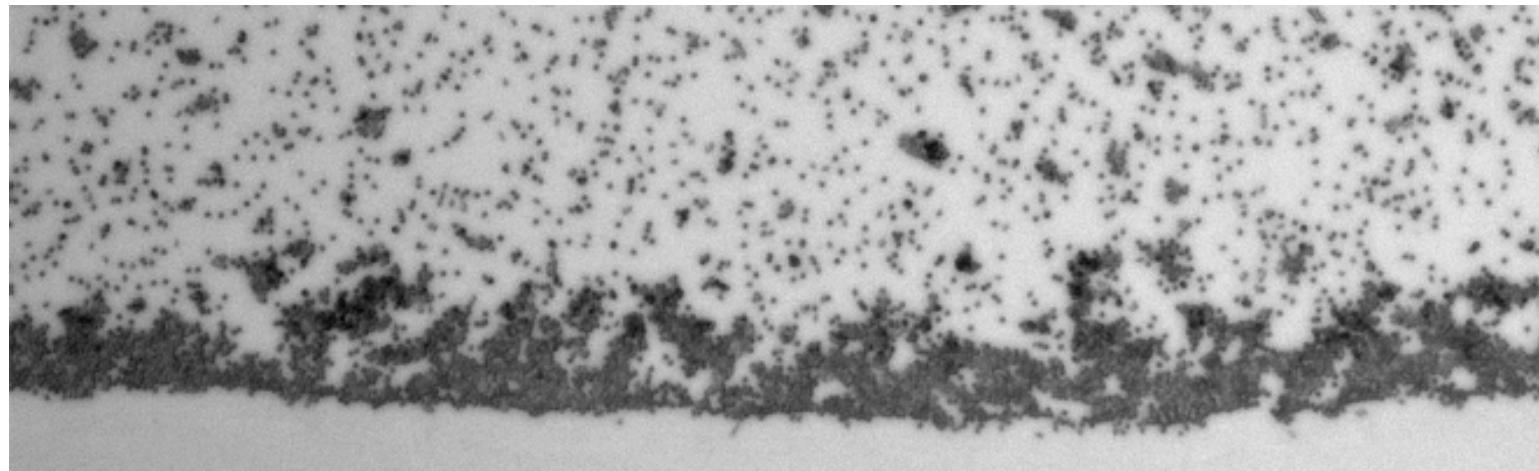


Ellipsoids

What types of growth processes occur?

How does growth process depend on particle shape?

Deposit Characterization



25 μm —

Deposit Characterization

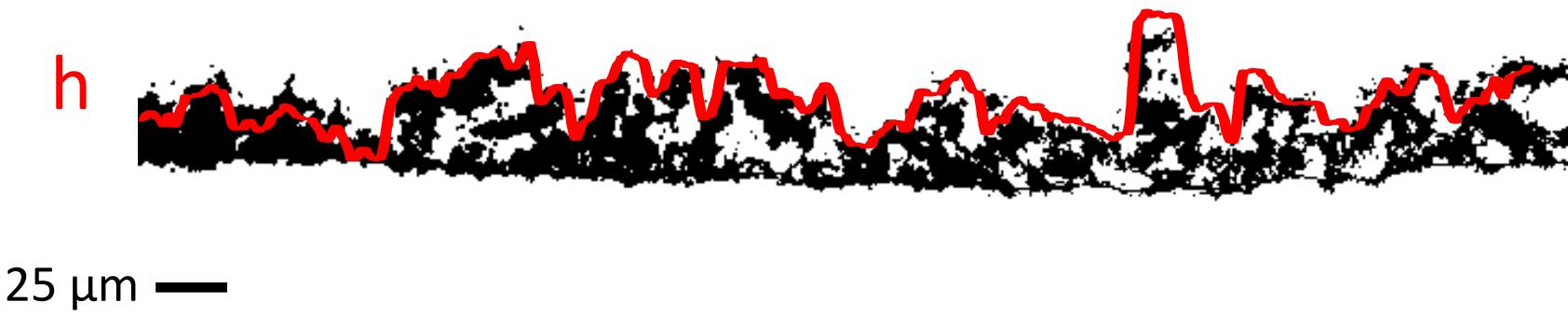


25 μm —

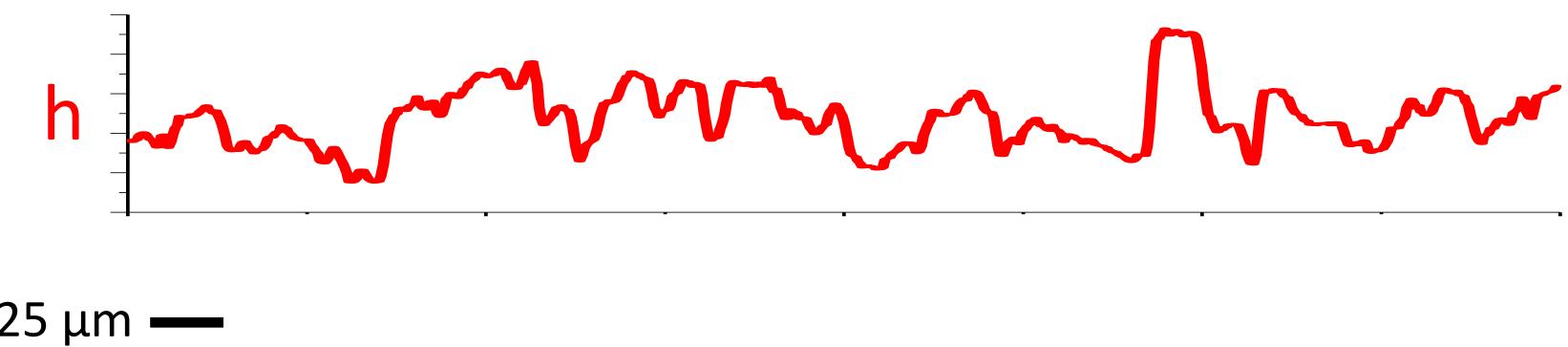
Deposit Characterization



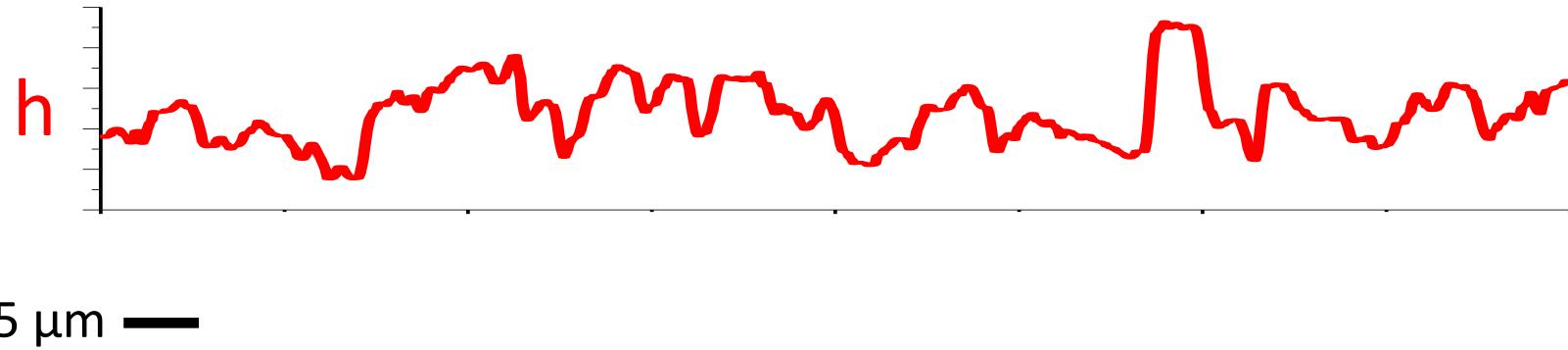
Deposit Characterization



Deposit Characterization



Deposit Characterization



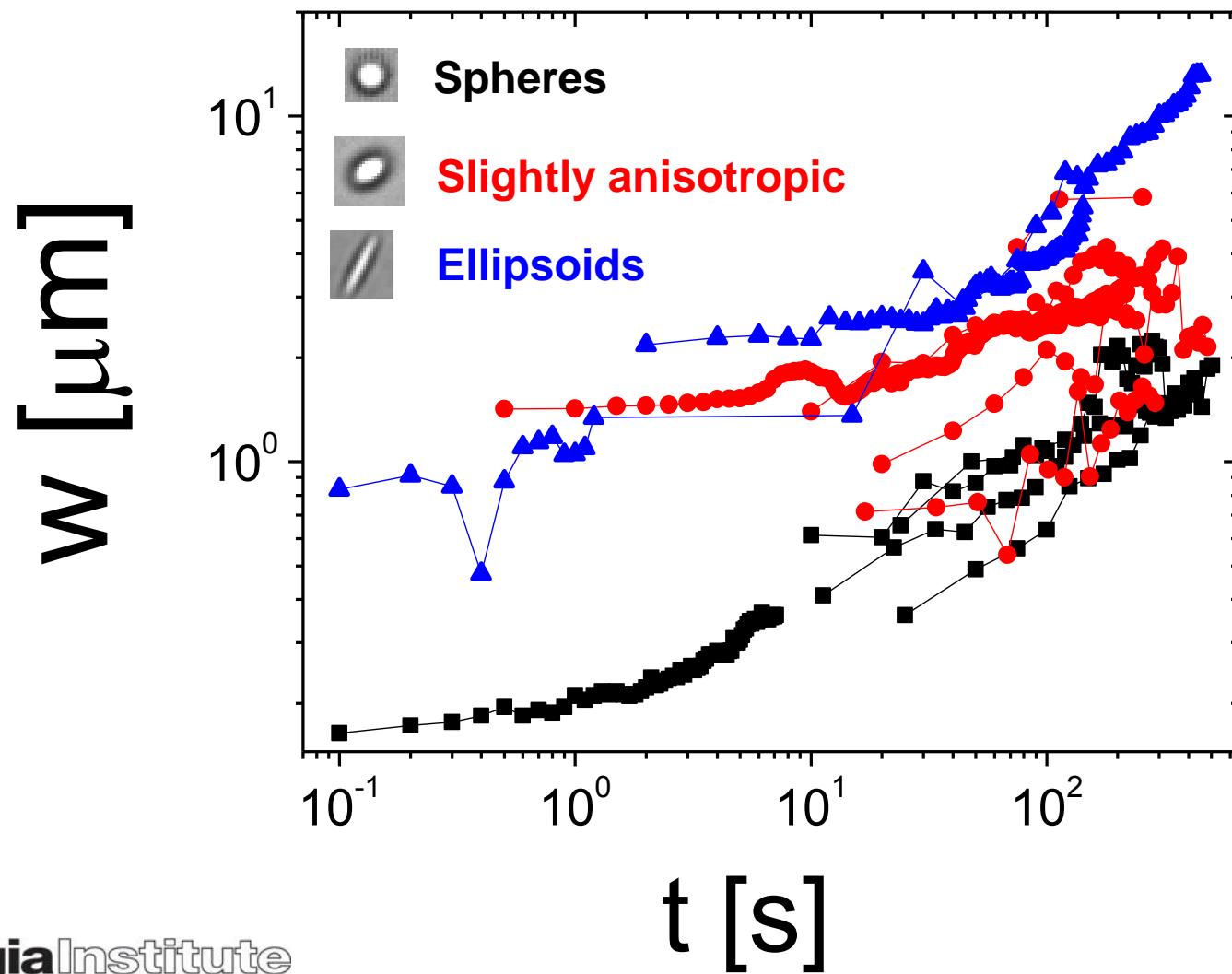
$$\text{Width, } w = \sqrt{\langle (h - \langle h \rangle)^2 \rangle} \sim t^\beta$$

Slightly anisotropic particles

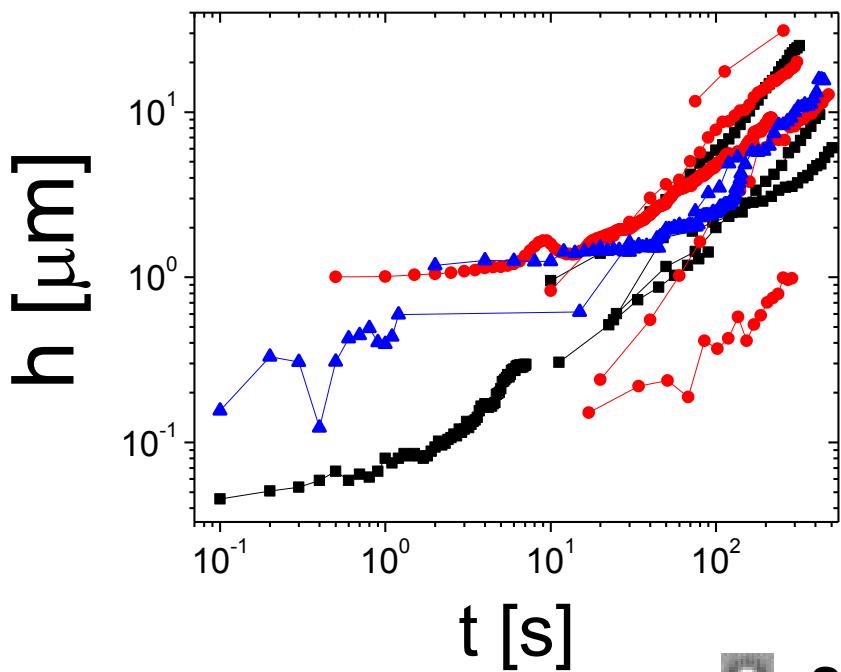
$$\alpha = 1.2$$



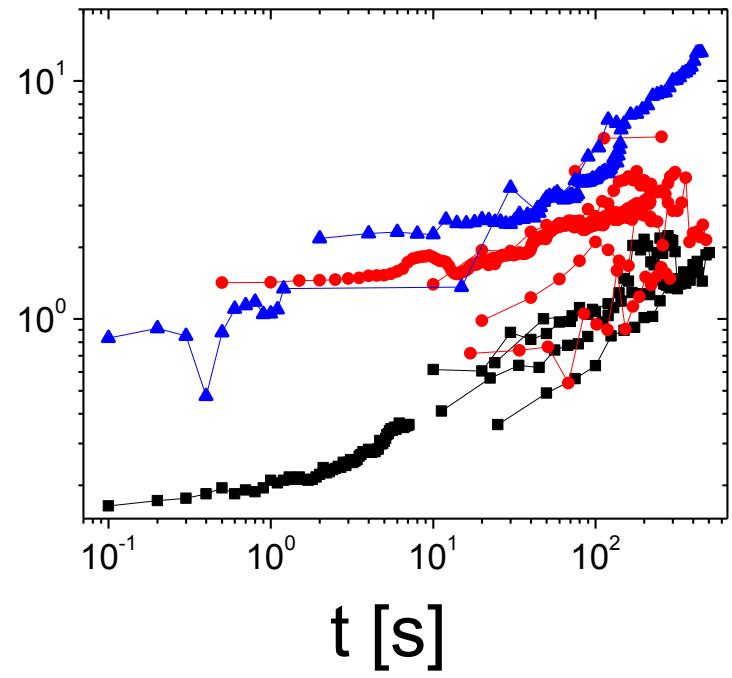
Deposit width increases over time



Use h instead of t



h [μm]



w [μm]

t [s]

Spheres



Slightly anisotropic



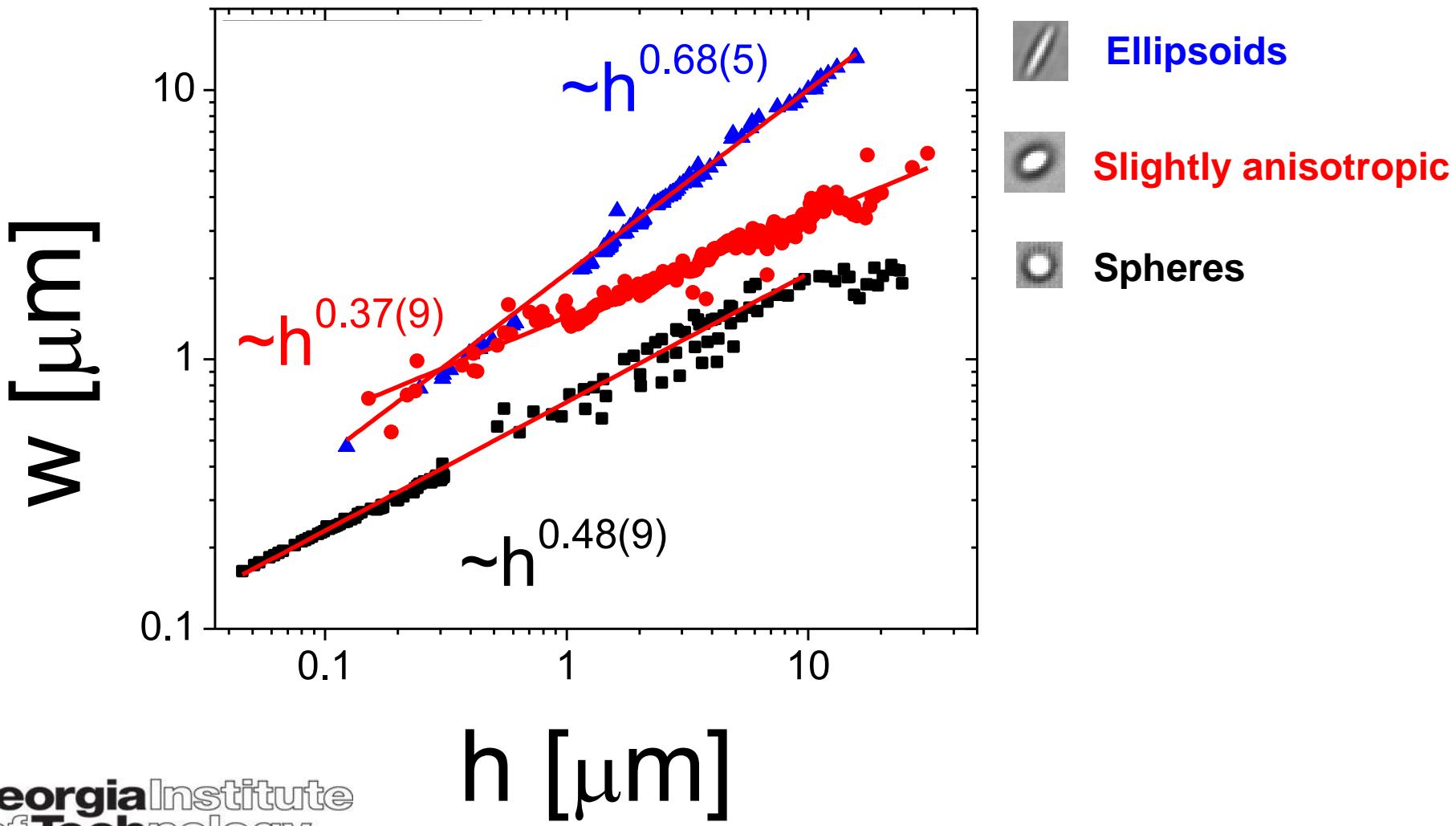
Ellipsoids



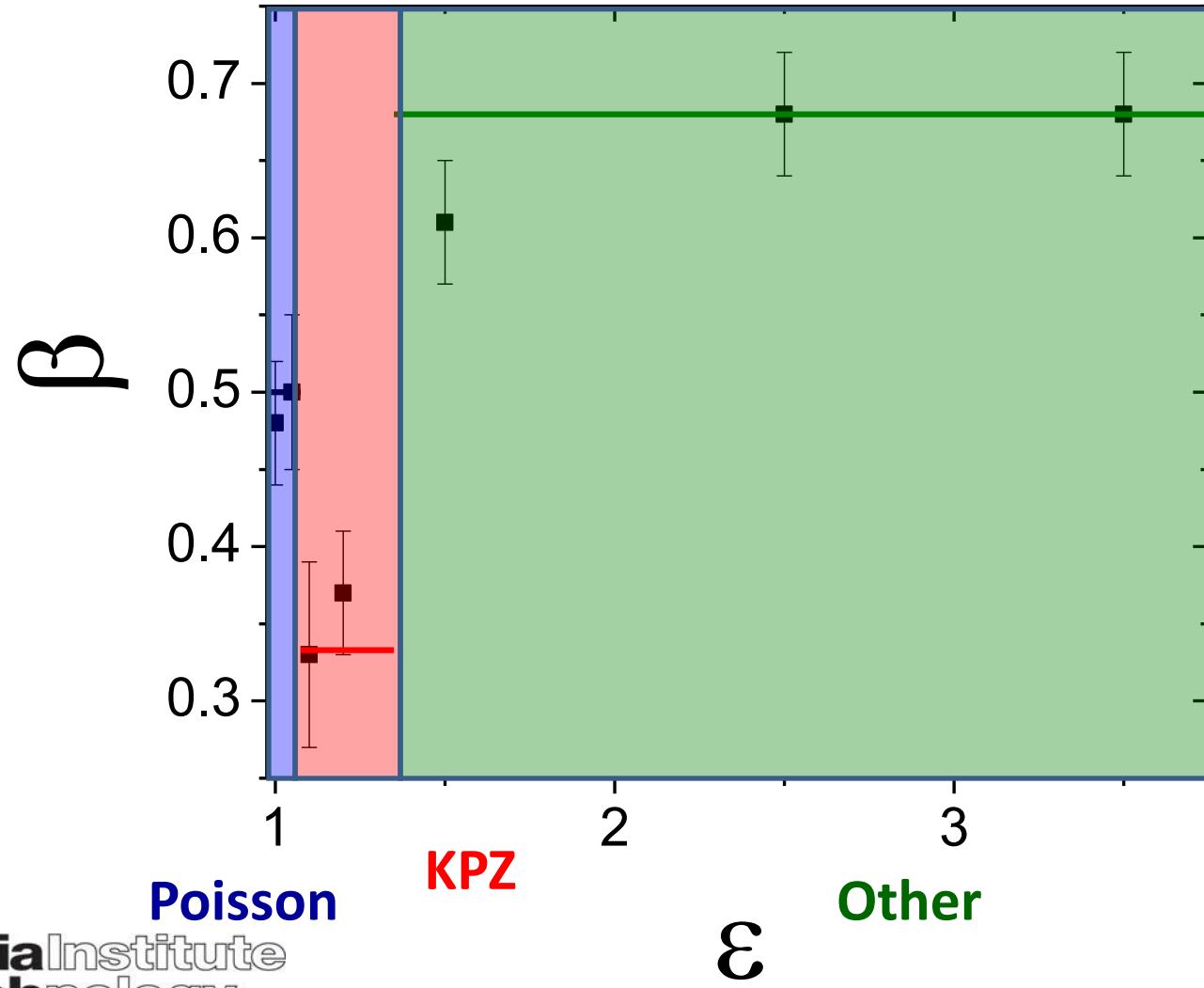
Georgia Institute
of Technology



Data collapse into distinct trends based on particle shape

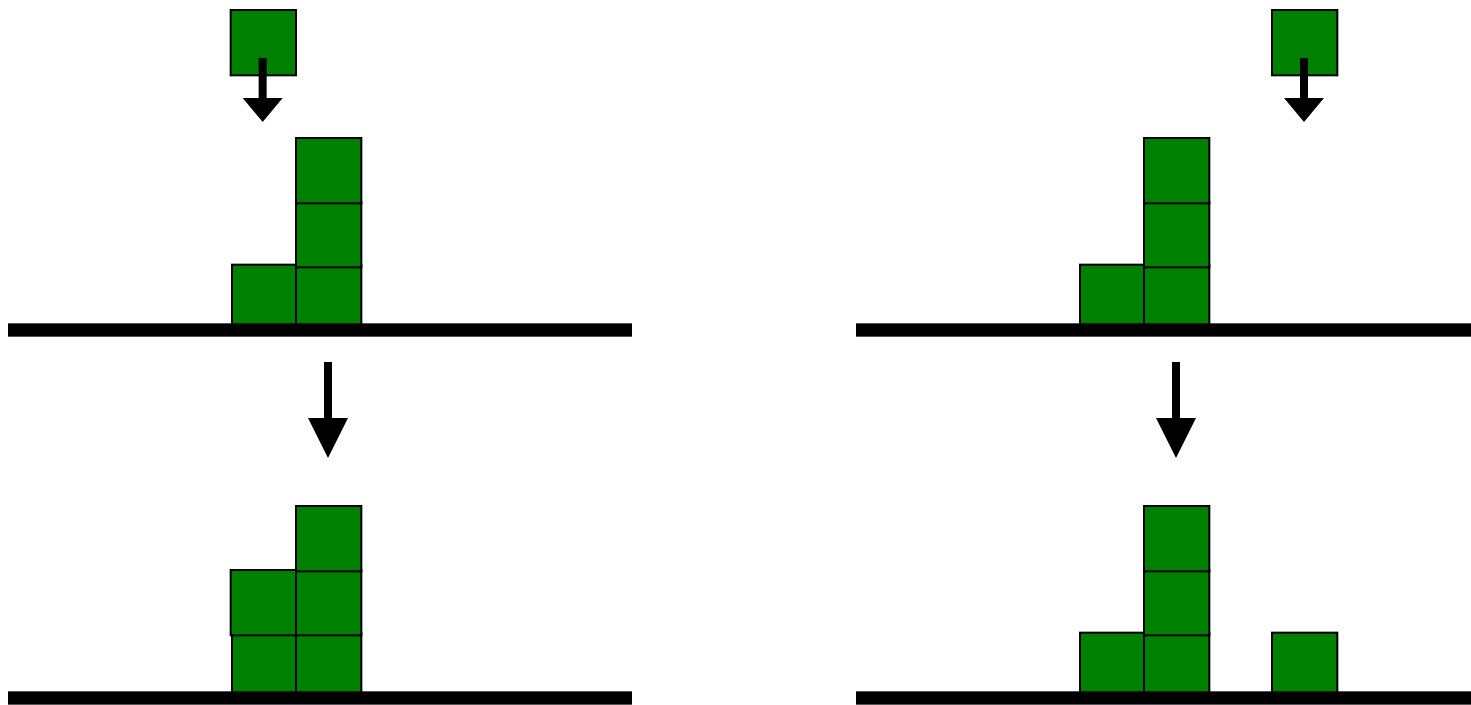


Three distinct regimes

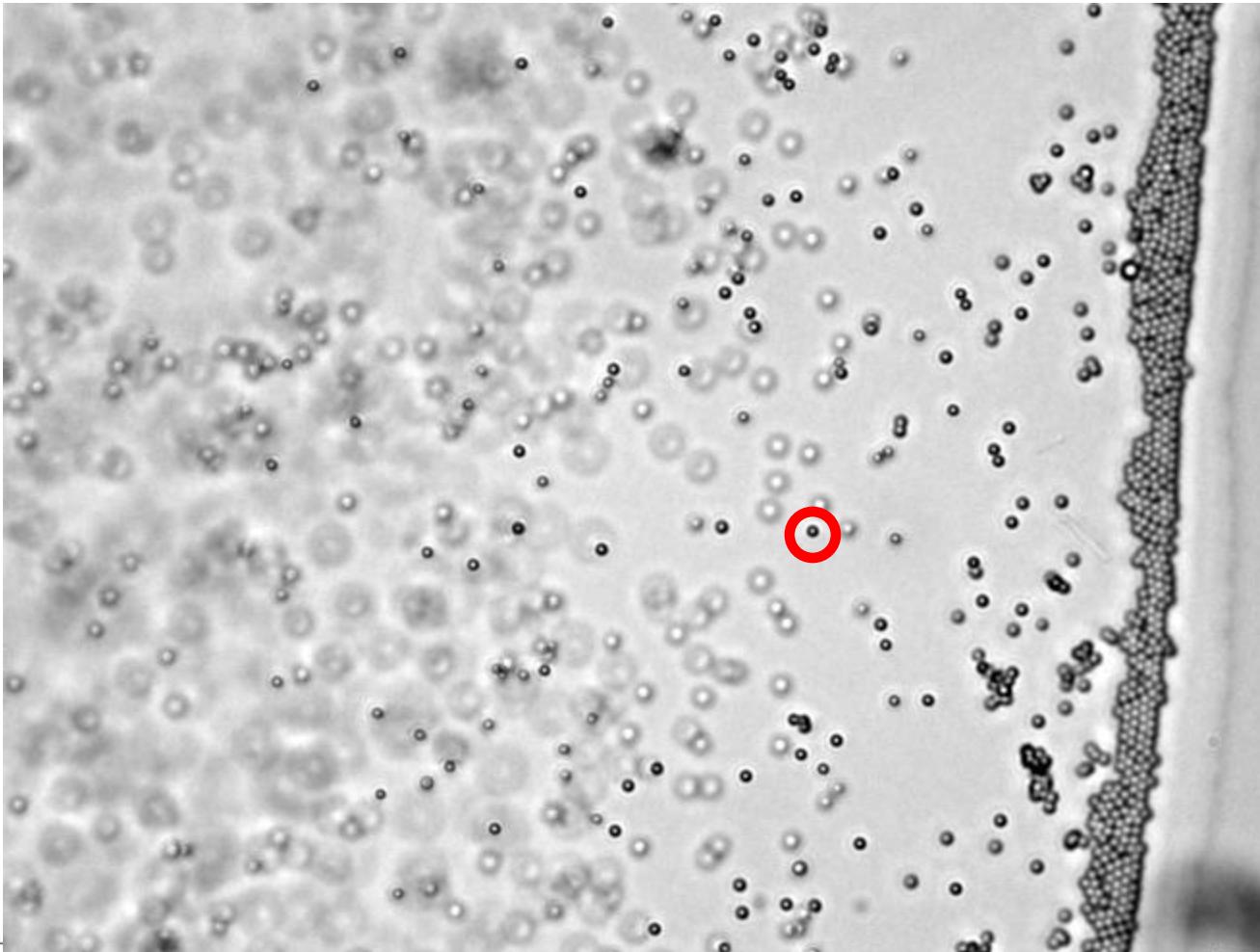


Spheres
 $\varepsilon = 1.0, 1.05$
Poisson Process

Random deposition – Poisson process



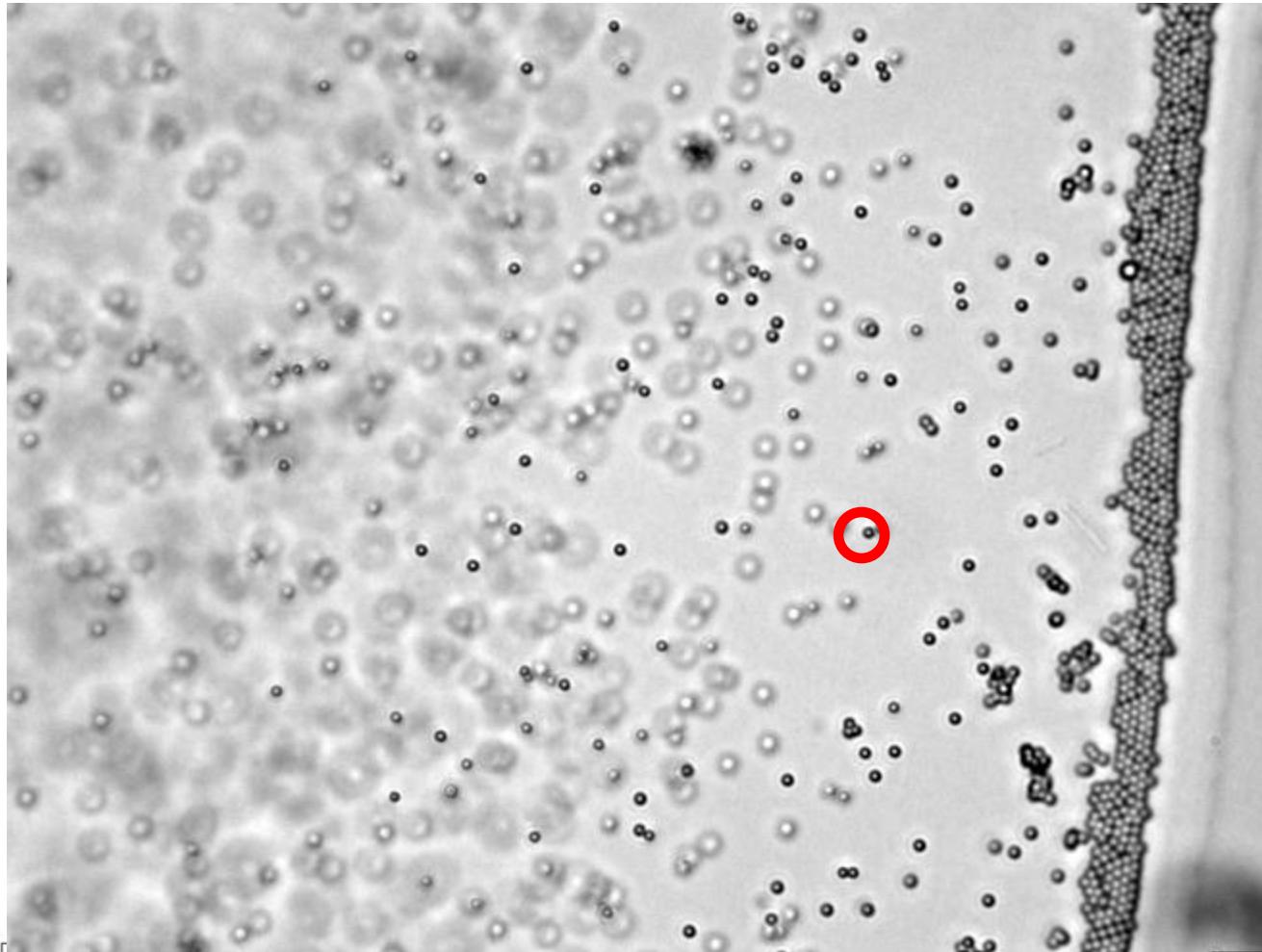
Spheres densely pack at drop edge



Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

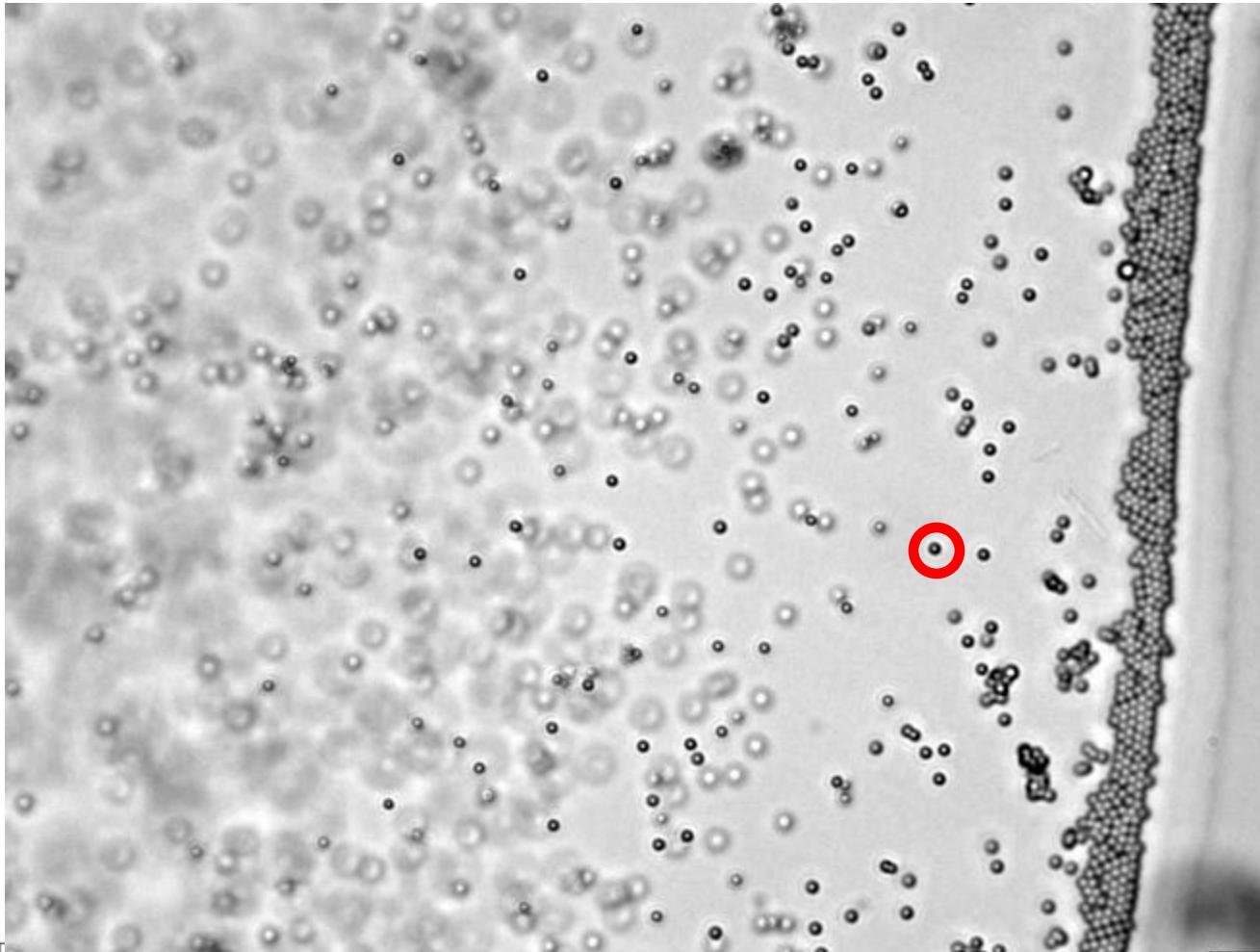
Spheres densely pack at drop edge



Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

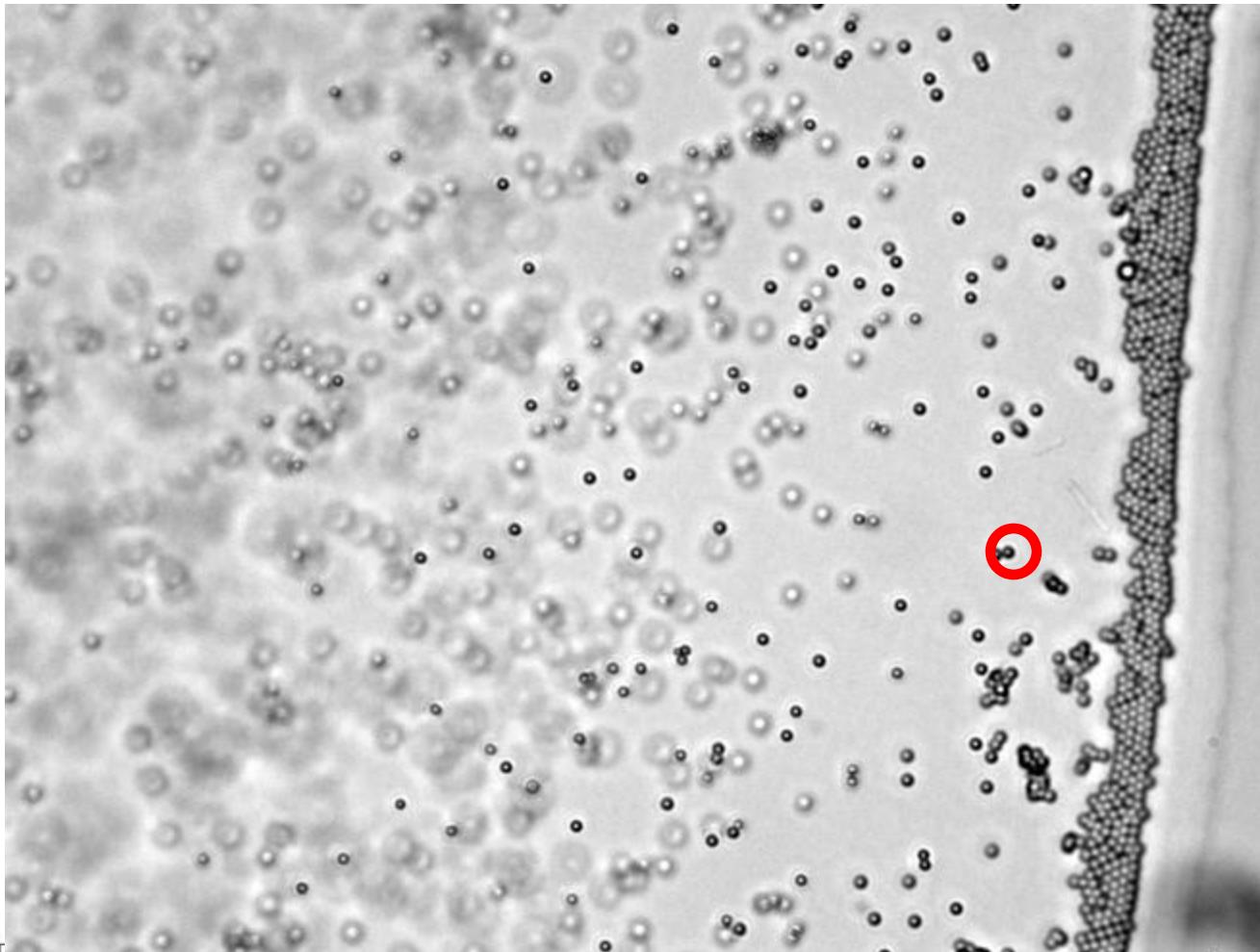
Spheres densely pack at drop edge



Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

Spheres densely pack at drop edge



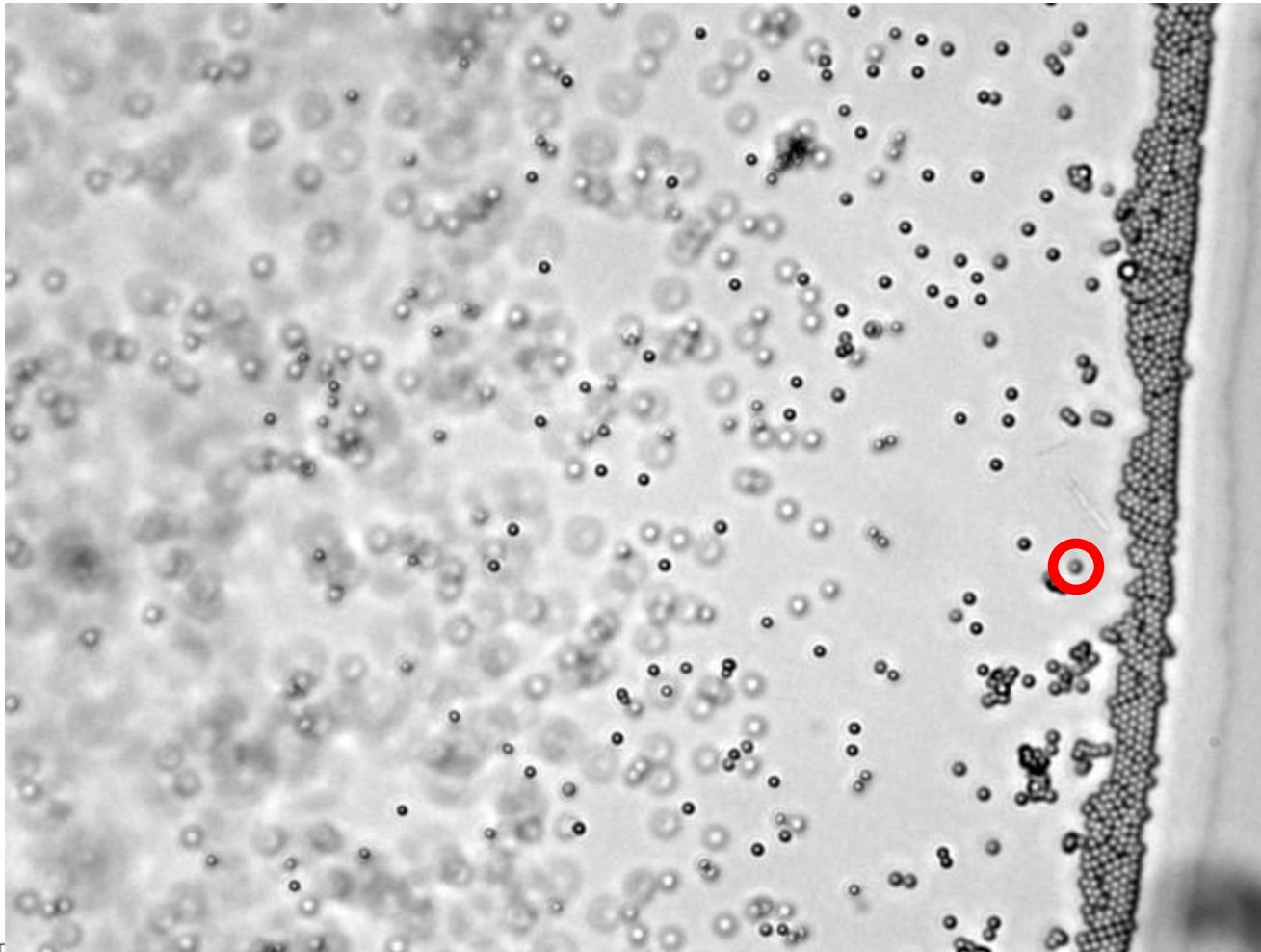
Drop Edge

Initial
Volume
Fraction
 $\phi = 0.005$

5 μm —

t = 4 s

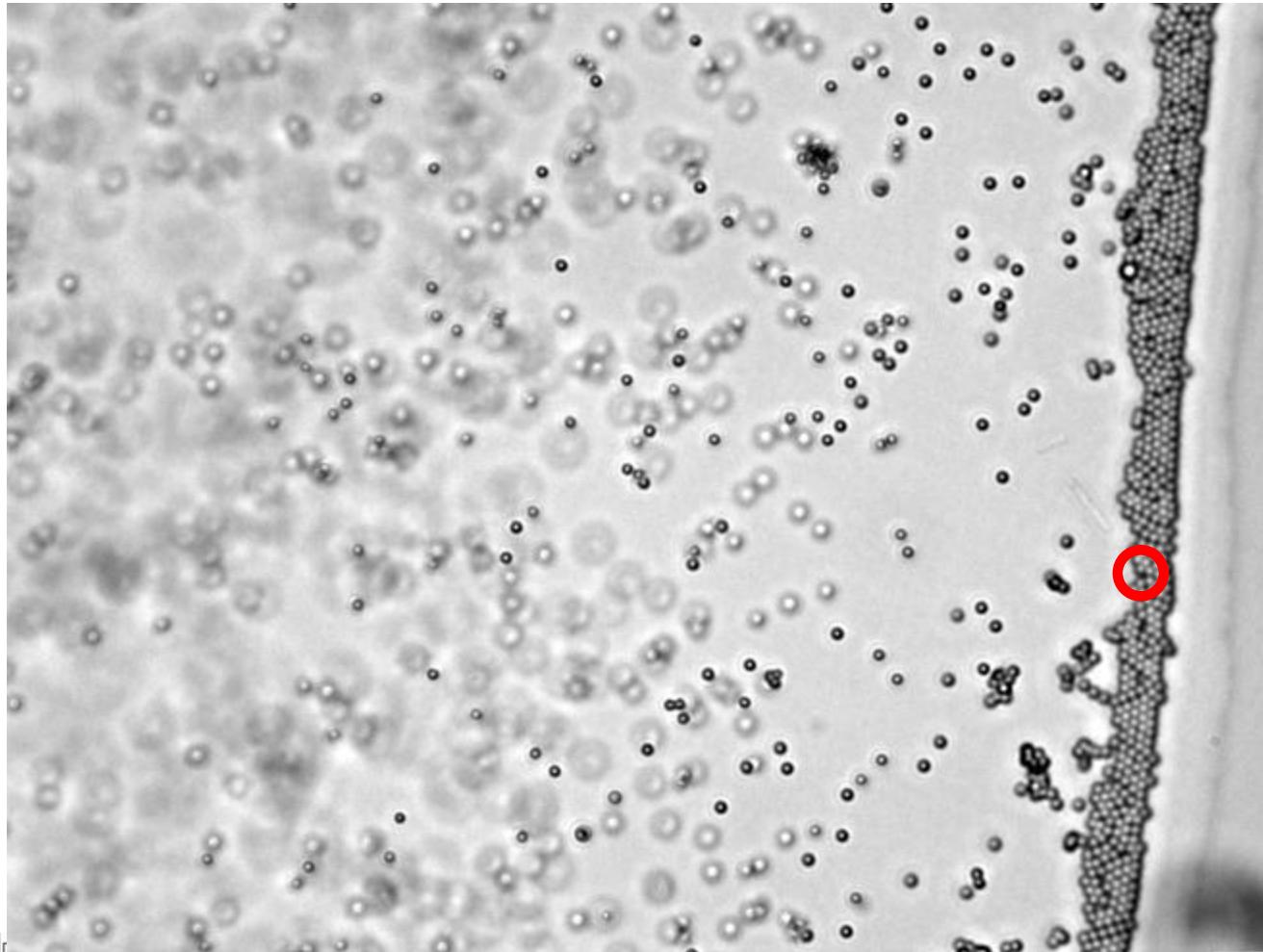
Spheres densely pack at drop edge



Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

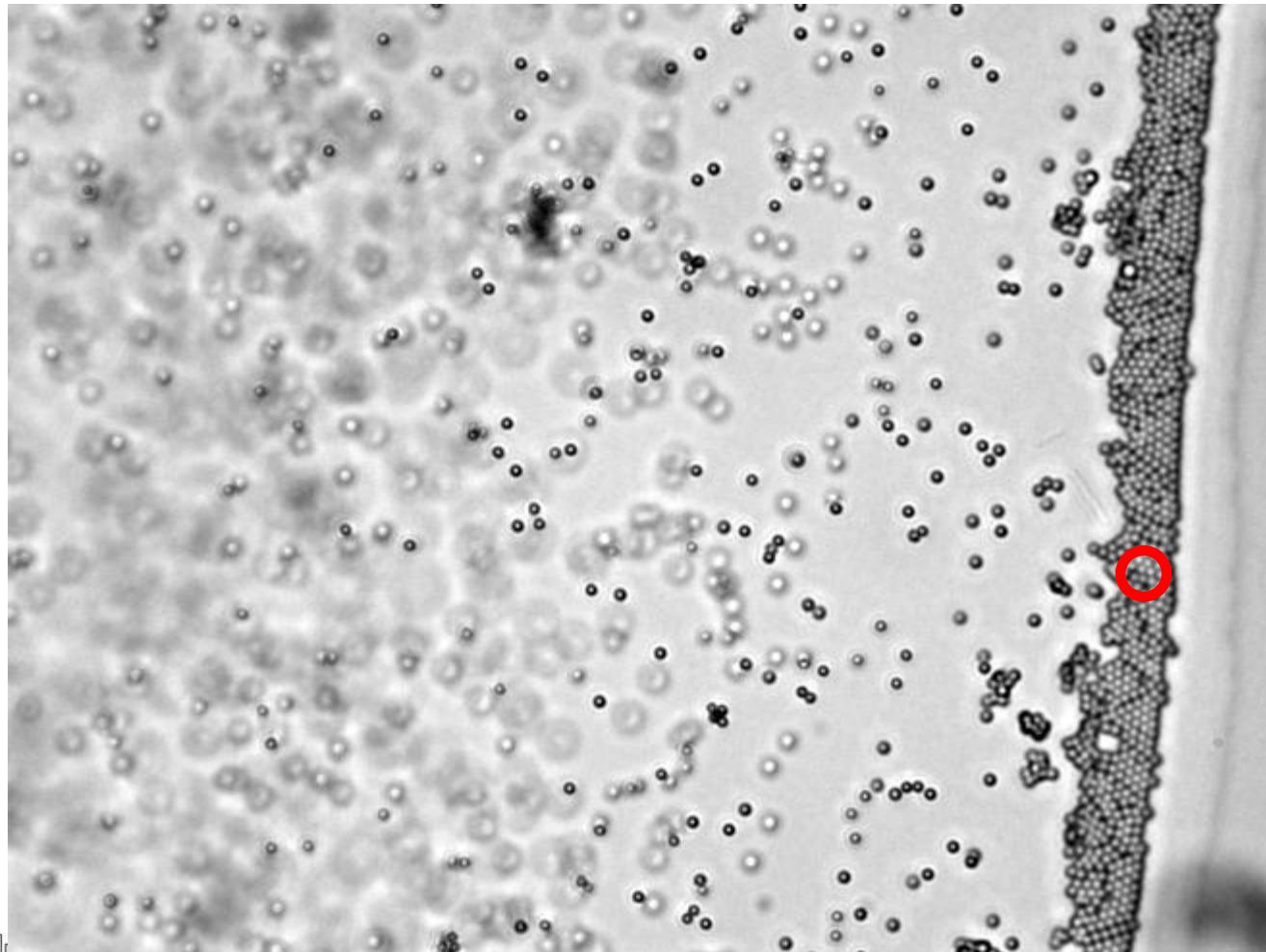
Spheres densely pack at drop edge



Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

Spheres densely pack at drop edge

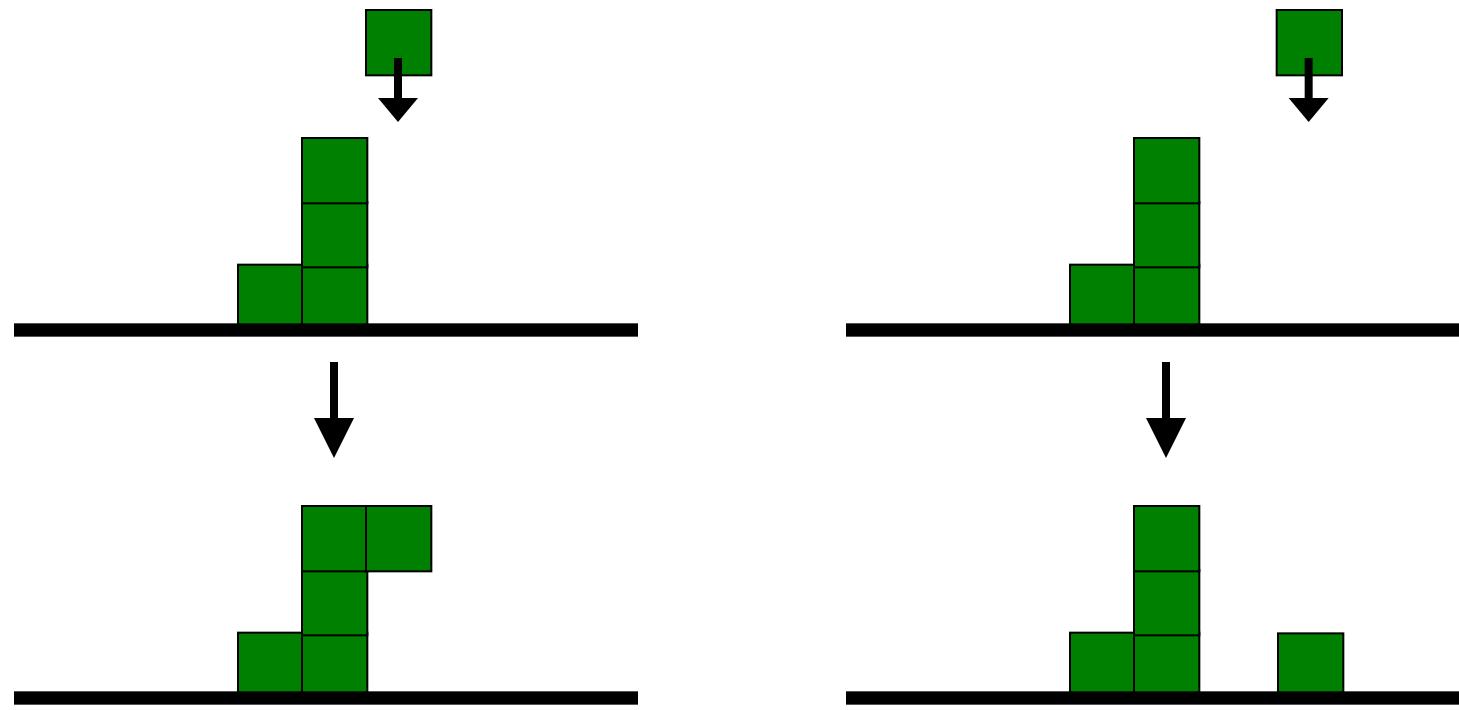


Drop
Edge

Initial
Volume
Fraction
 $\phi = 0.005$

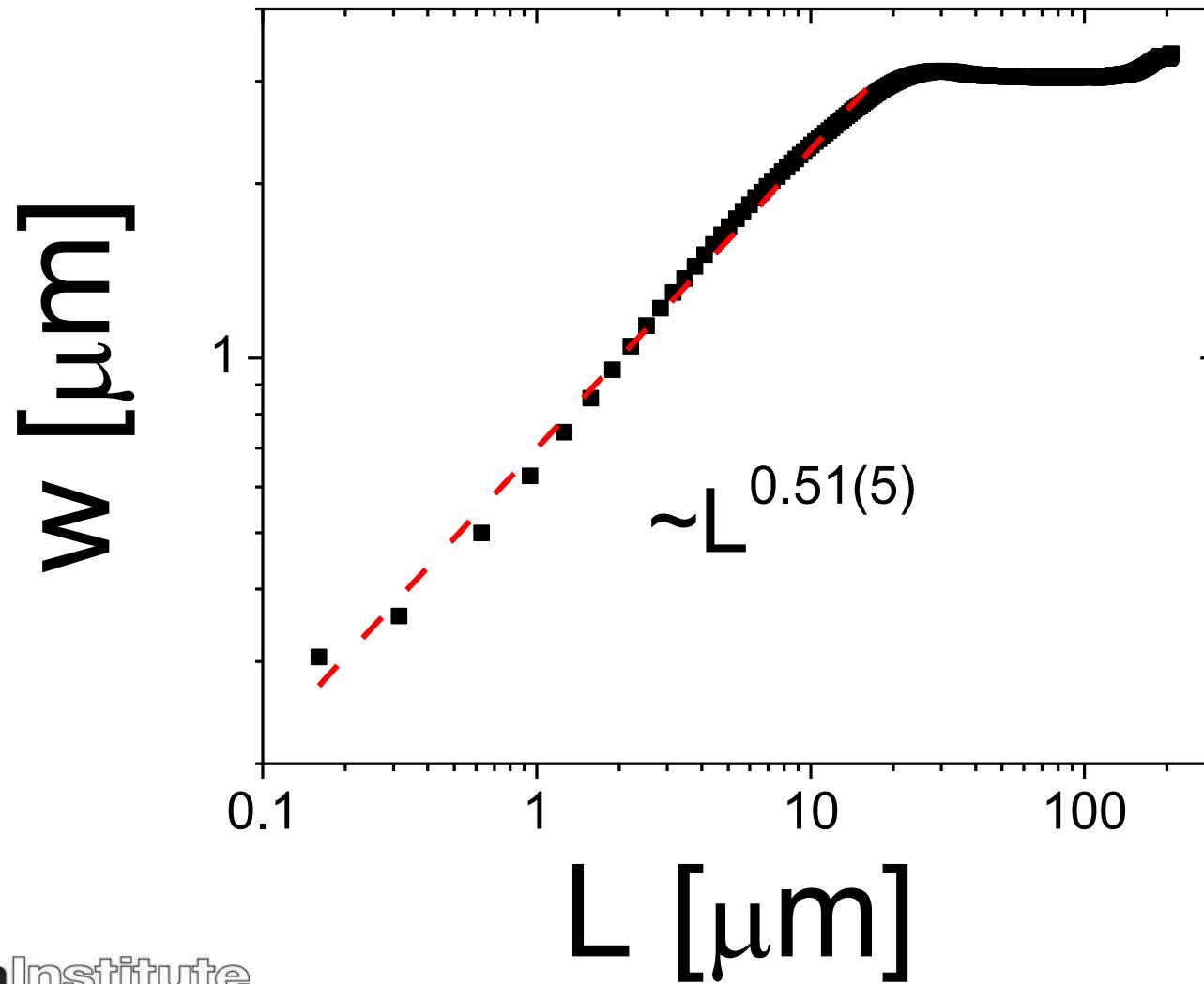
Slightly anisotropic particles
 $\epsilon = 1.1, 1.2$
KPZ Process

Slightly stretched particle deposition similar to ballistic deposition

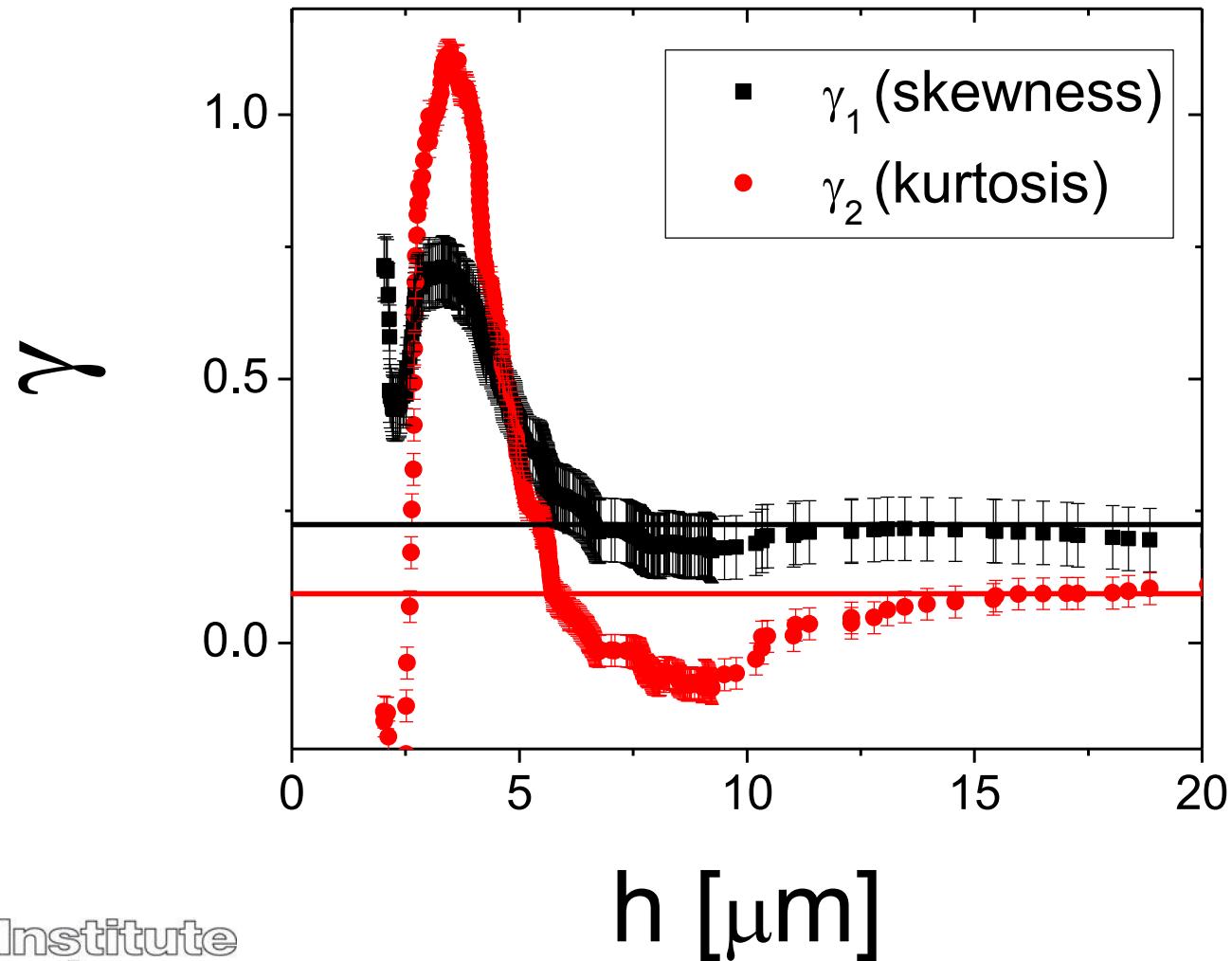


Interparticle interaction is $>> 1 k_B T$ for nearest neighbors only

Roughness exponent α agrees with KPZ



Skewness and Kurtosis approach KPZ values



Very anisotropic ellipsoids

$$\epsilon > 1.2$$

Colloidal Matthew Effect



Particle-rich regions get richer

What type of growth process is this?

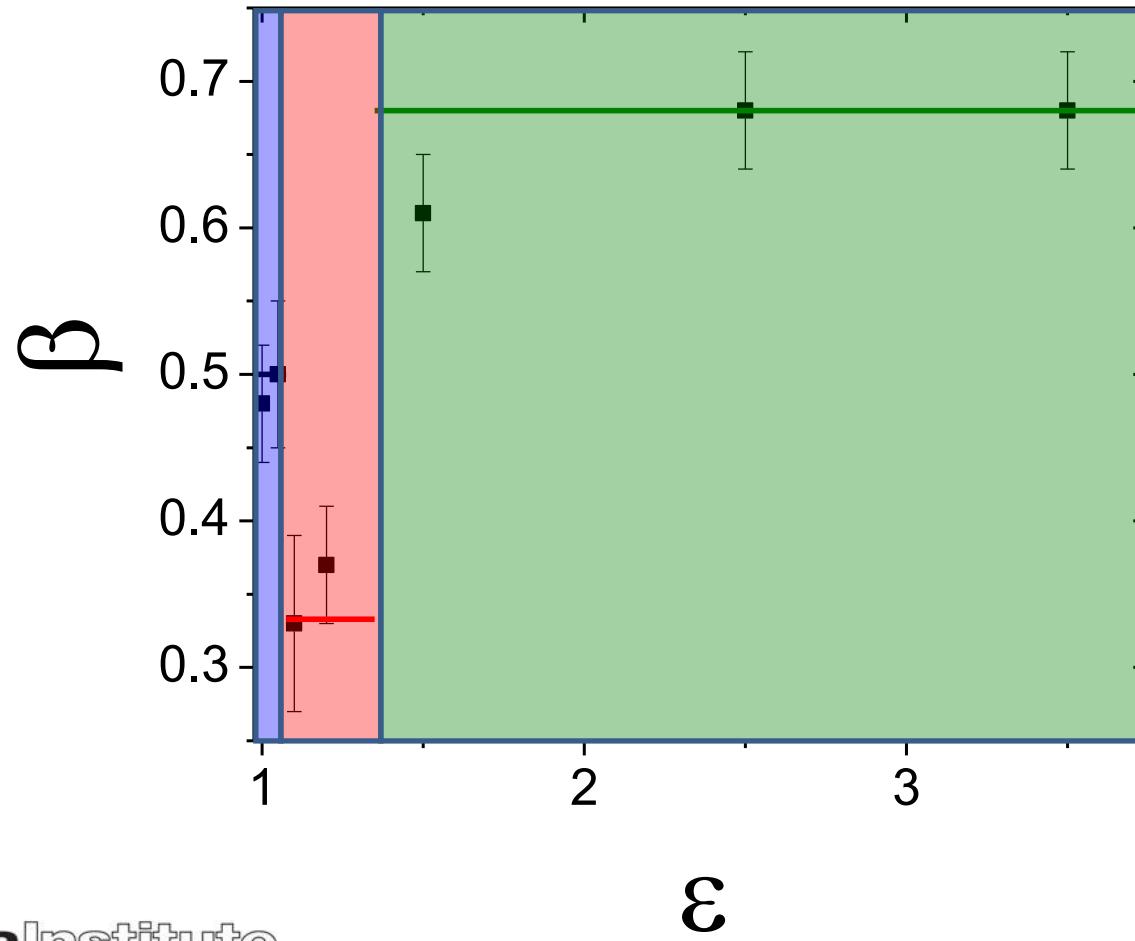
- Anomalous roughening?
 - Nicoli, Cuerno, and Castro PRL (2013)
- KPZ?
 - Oliveira, Aarao Reis arXiv:1401.0696 (2014)
- KPZQ?
 - Dias, Araujo, Telo da Gama, arXiv:1407.2374 (2014)

Drop Edge

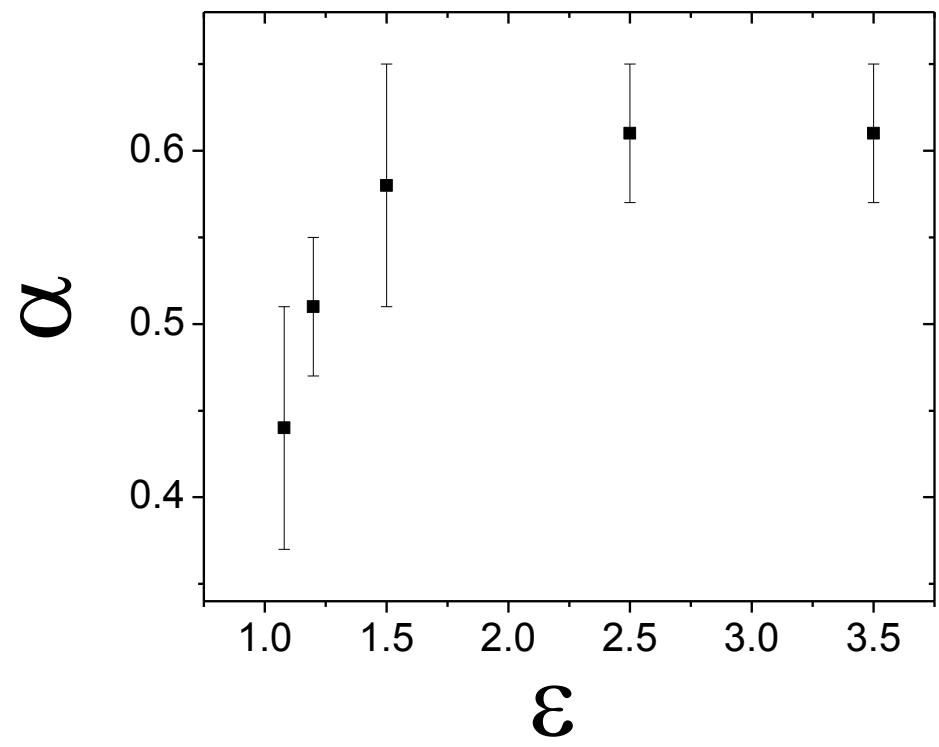
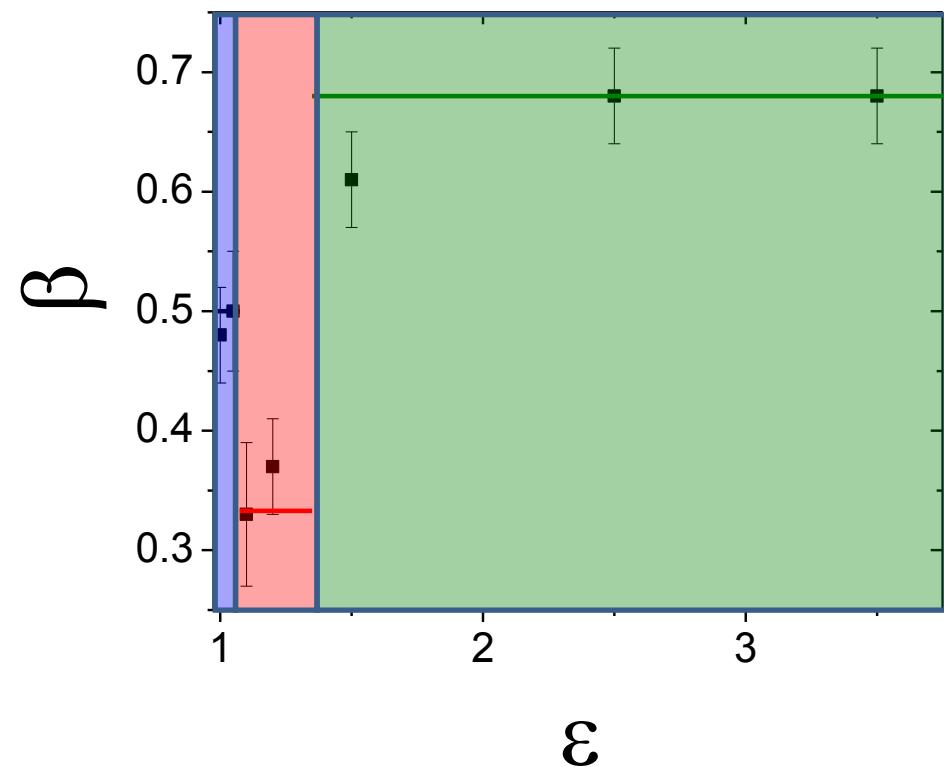
10 μm —



Must explain abrupt transition



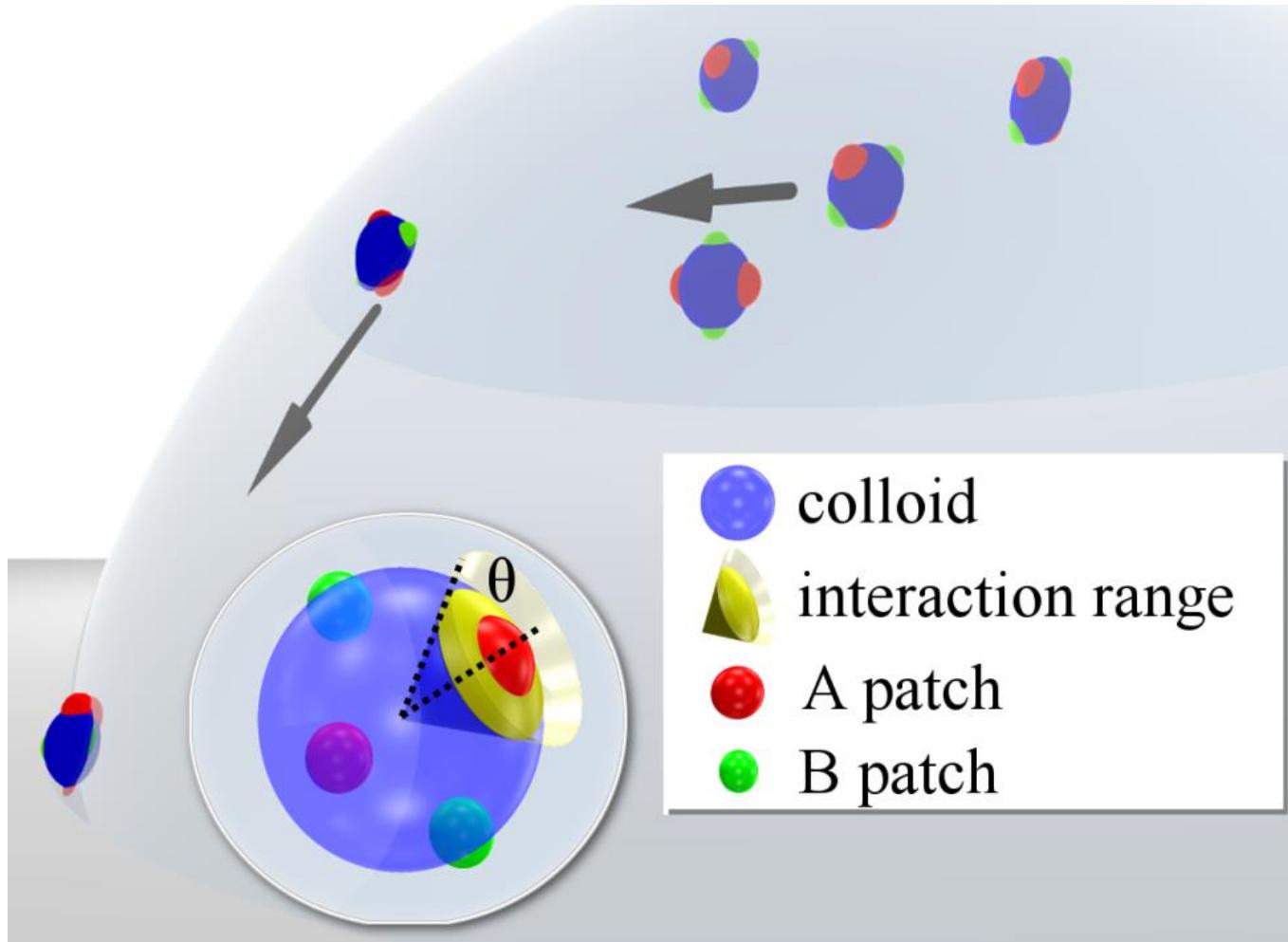
Must explain abrupt transition



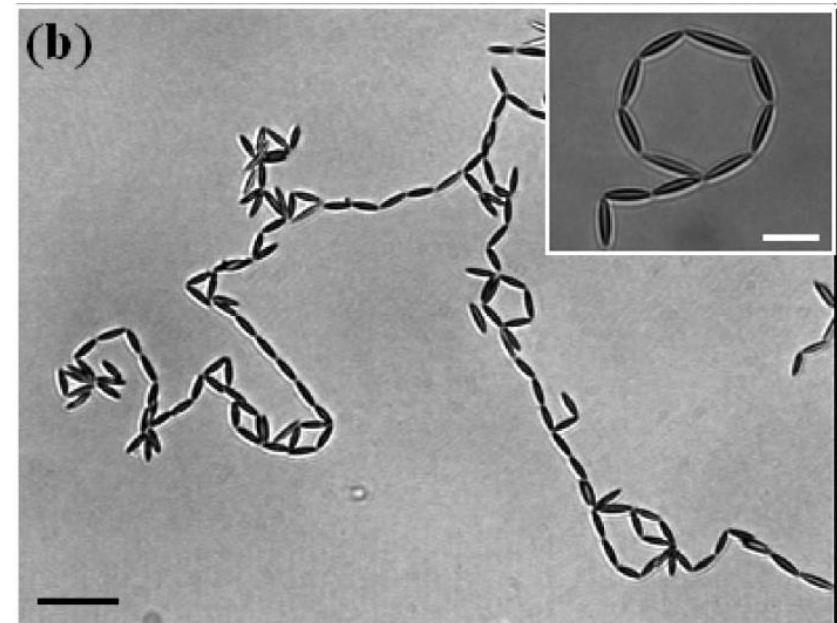
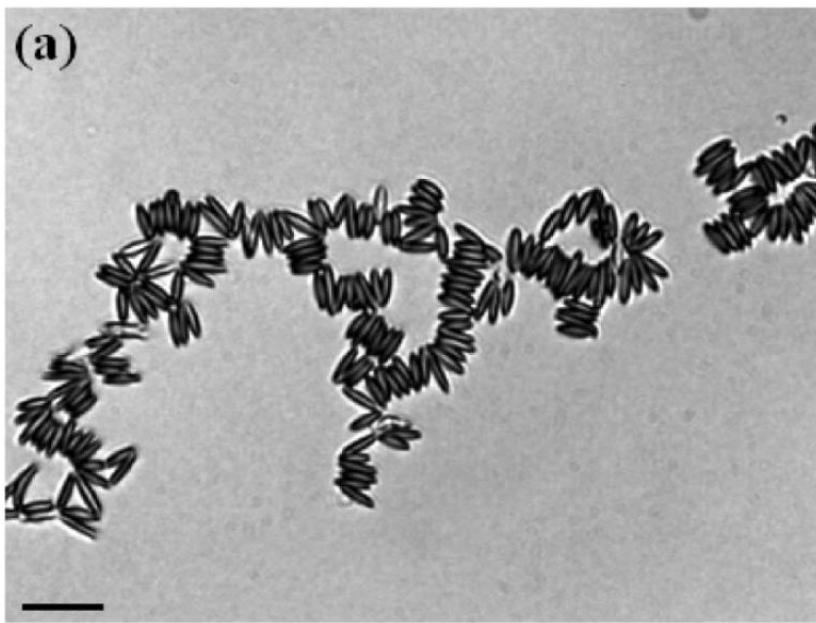
Minimal simulation?



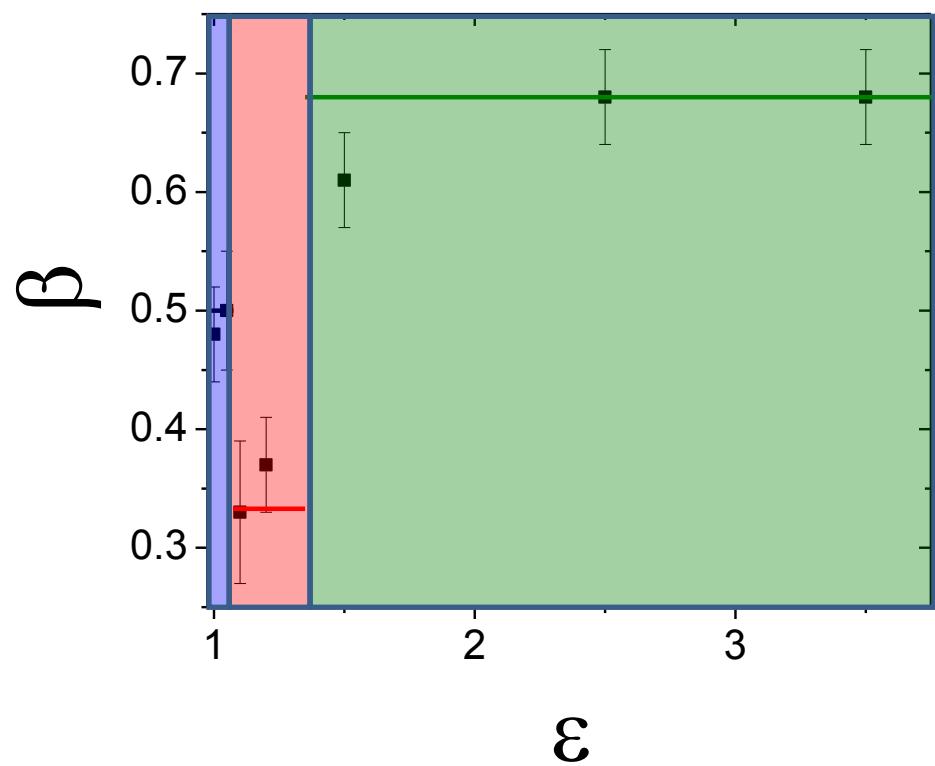
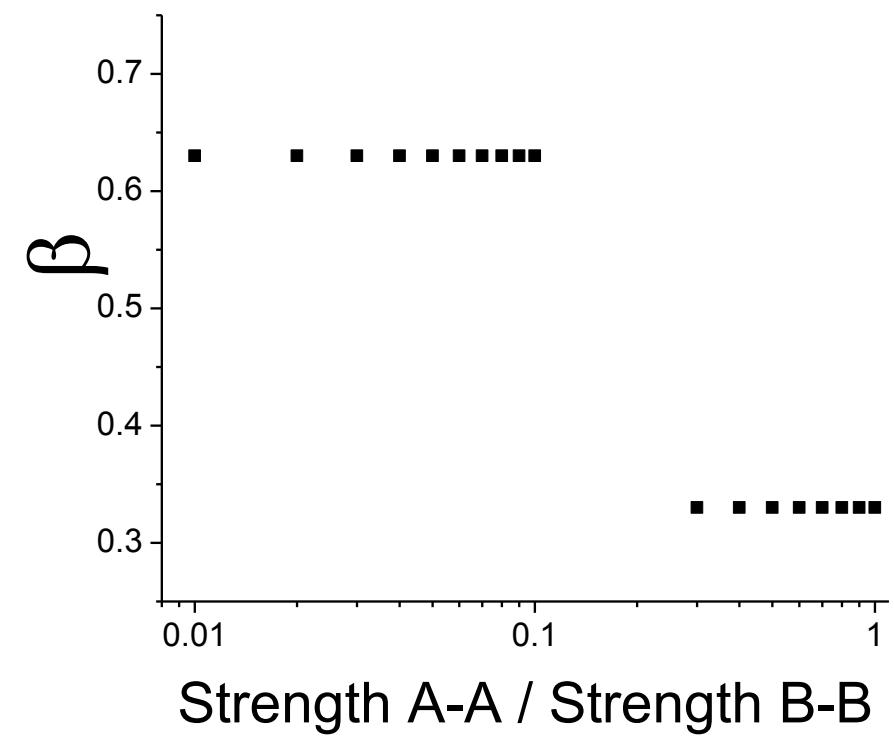
Simulation of patchy colloids



Highly anisotropic ellipsoids prefer tip-to-tip or side-to-side

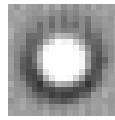


Abrupt transition occurs

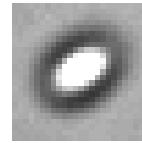


Summary

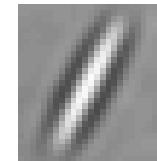
- Changing particle shape selects 3 distinct growth processes



Poisson
Process



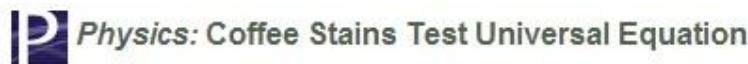
KPZ
Process



Colloidal Matthew
Effect
(other universal
process?)

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- Collaborators: Alexei Borodin, Doug J. Durian, Tom Lubensky, Matthew A. Lohr, Tim Still



January 18, 2013



An equation that describes a wide array of phenomena can be directly tested by wa
the equivalent of a drying coffee drip.

[Focus on Phys. Rev. Lett. **110**, 035501 (2013)]

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