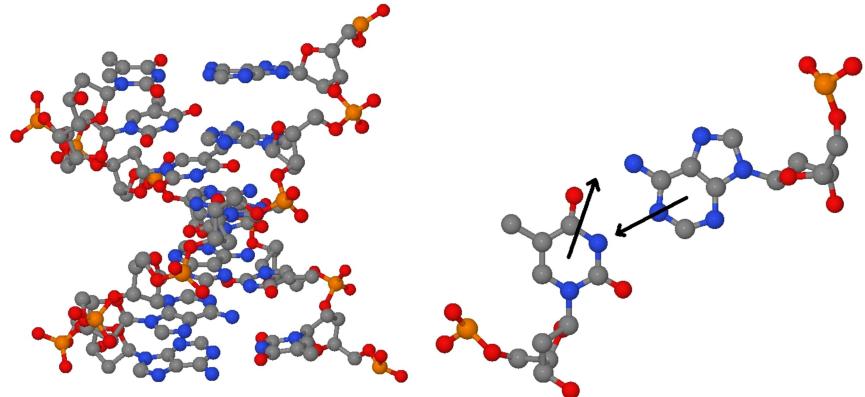
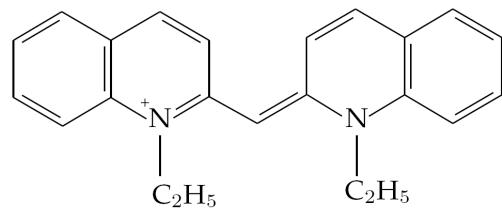
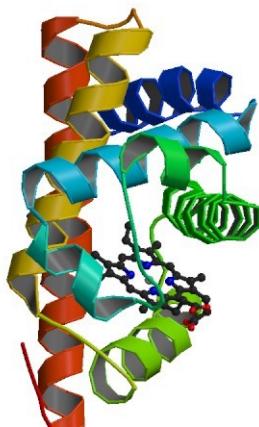


# Nonlinear optical response of excitons in a quantum mechanical bath

Arend G. Dijkstra  
Kyoto University

Prof. Y. Tanimura (Kyoto University)  
Prof. J. Knoester and dr. T.I.C. Jansen (University of Groningen)

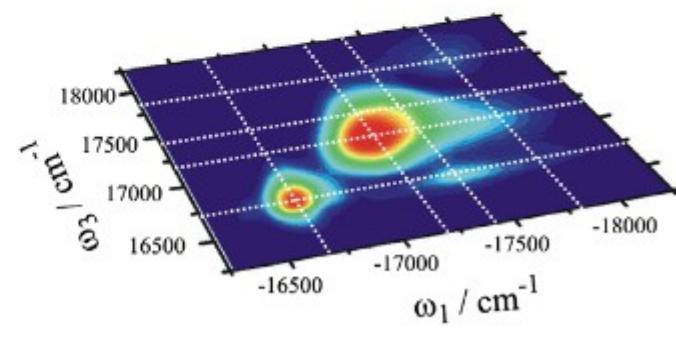
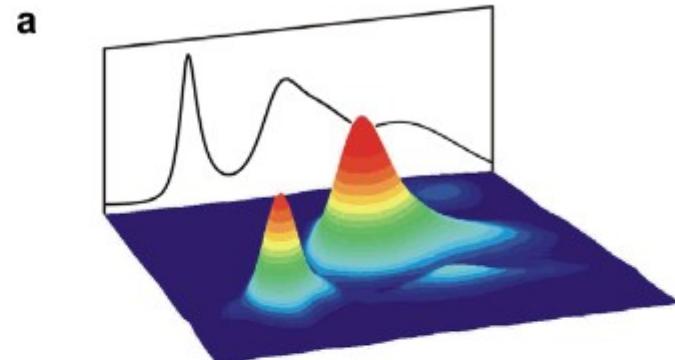
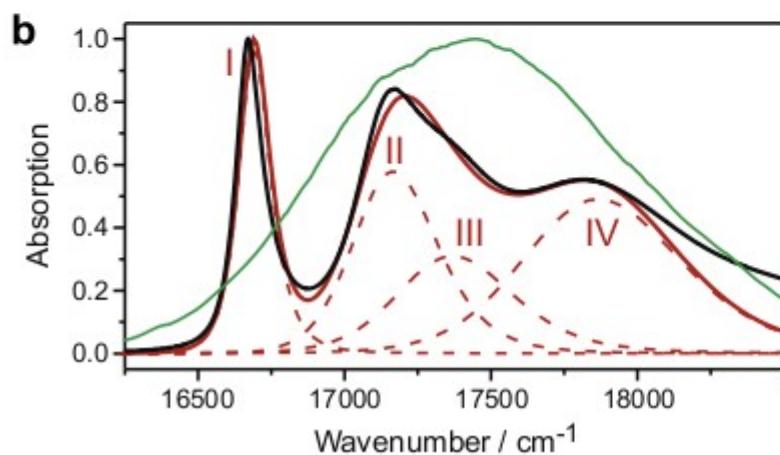
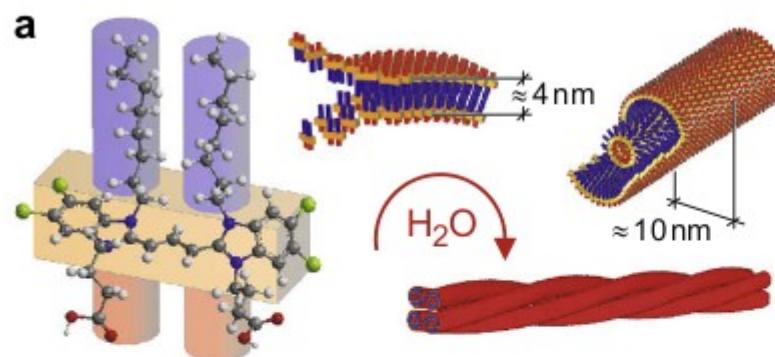


# Nonlinear spectroscopy: excitons

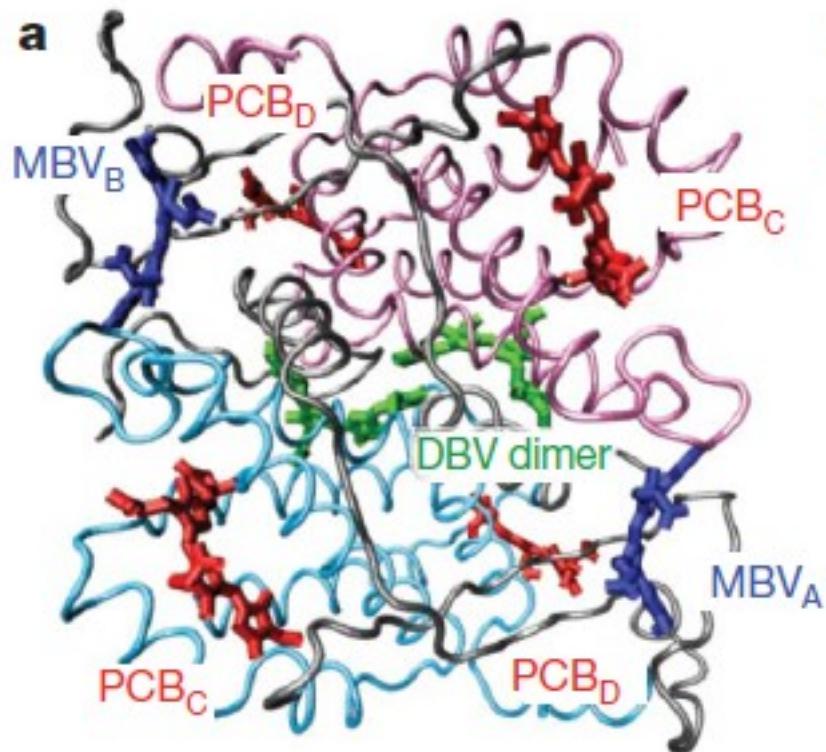
double-wall cylinders (Kauffmann)

2

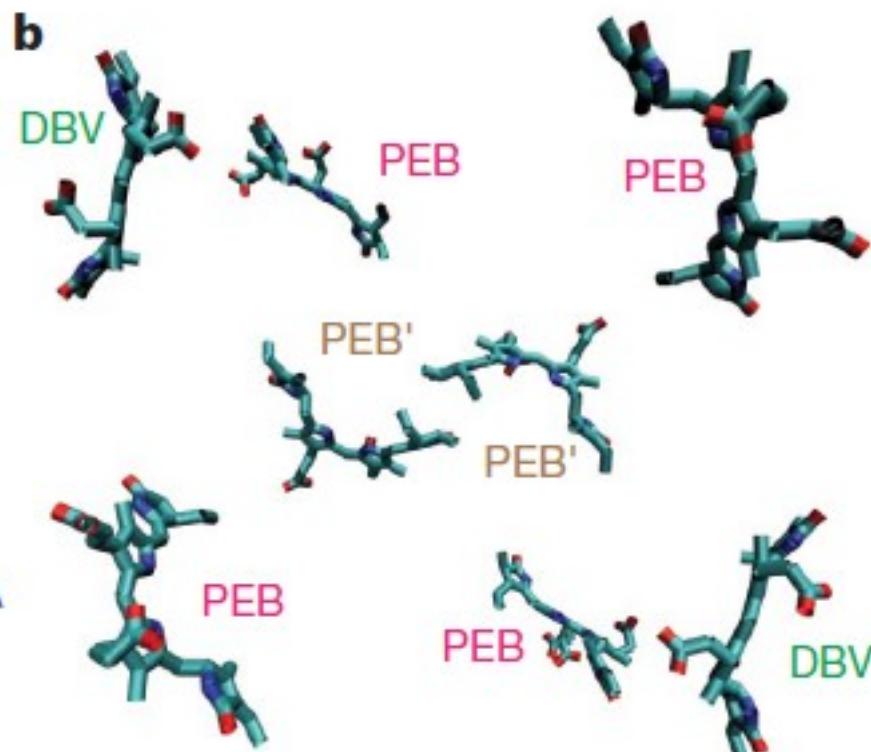
A. Nemeth et al./Chemical Physics Letters xxx (2009) xxx-xxx



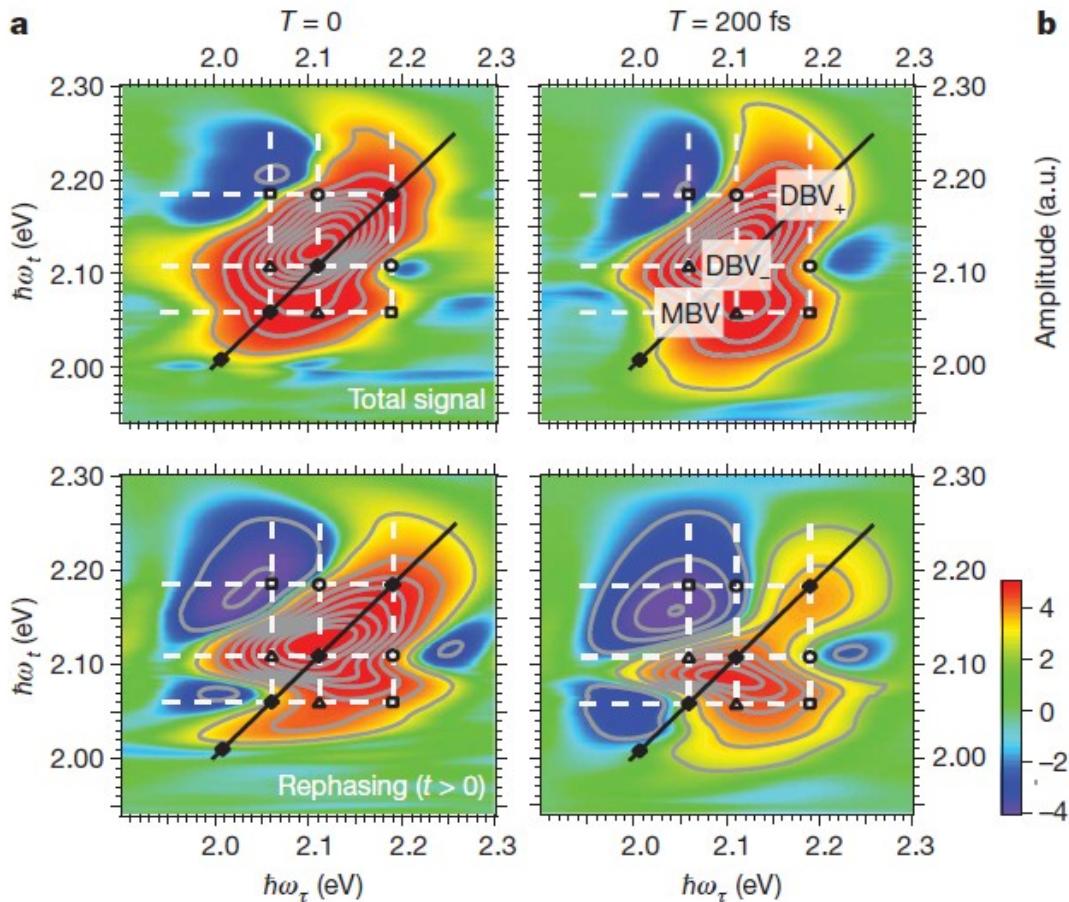
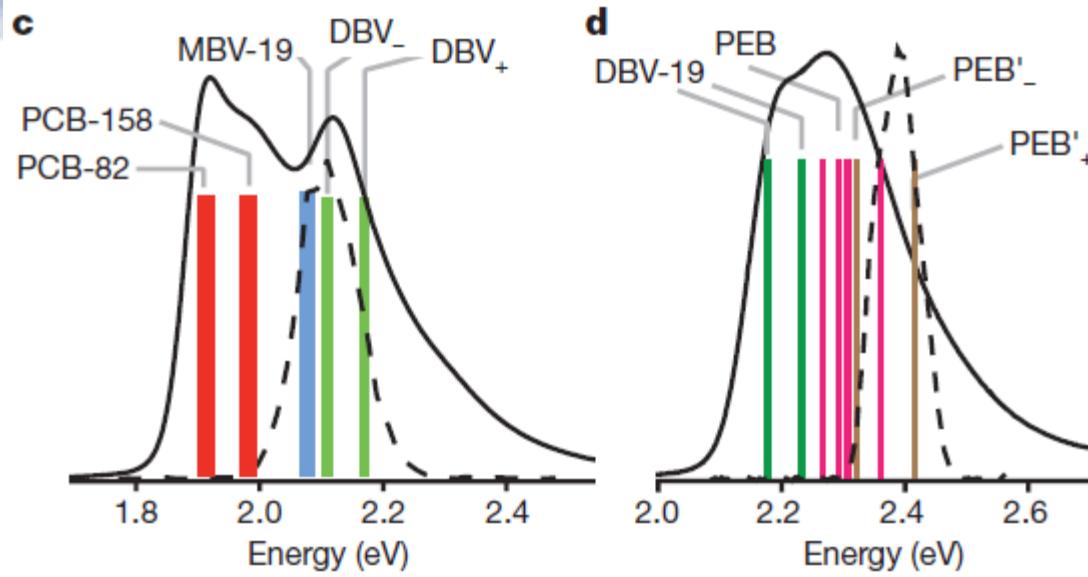
PC645



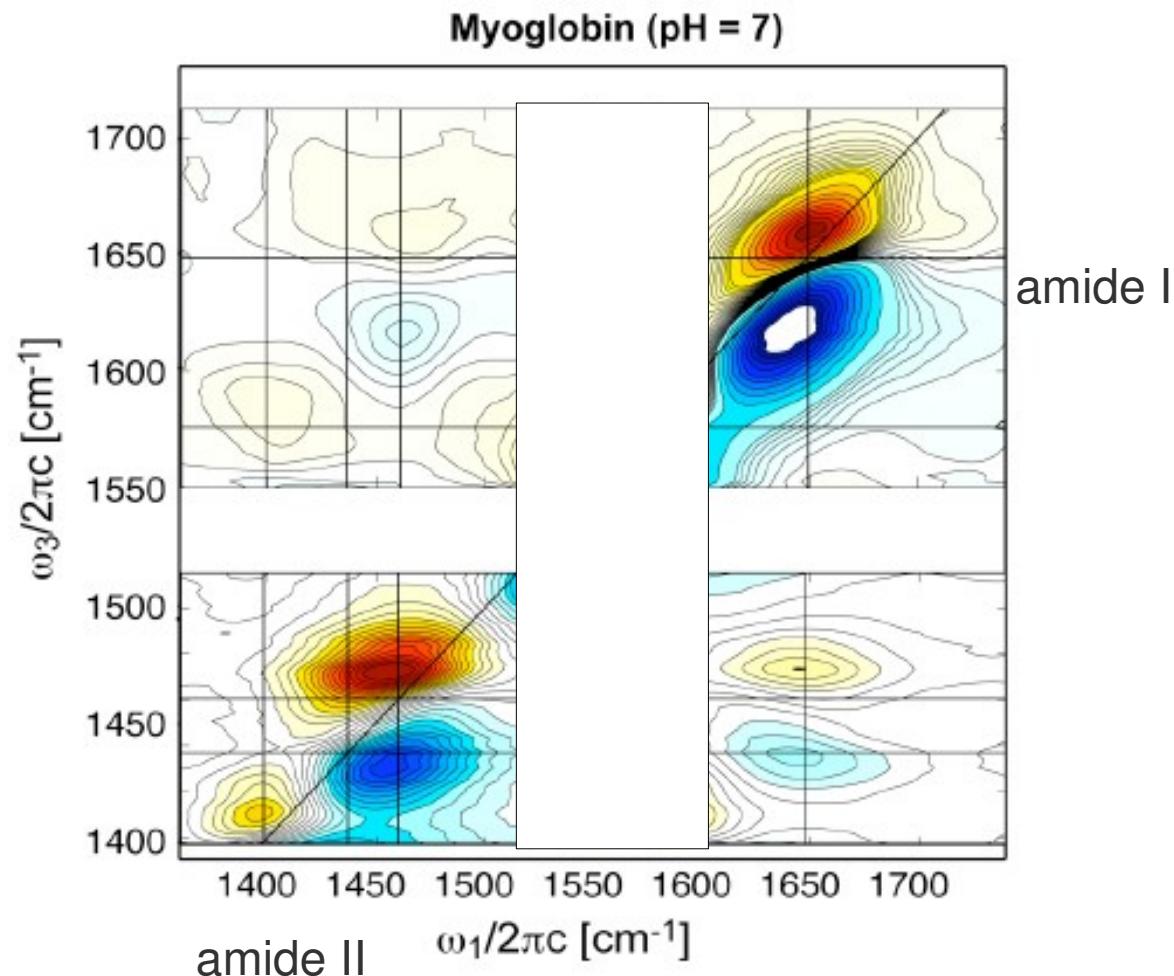
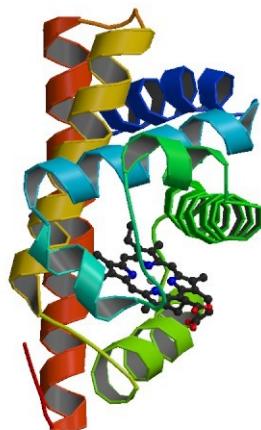
PE545



E. Collini et al. Nature 463, 644 (2010)



# Nonlinear spectroscopy: vibrons



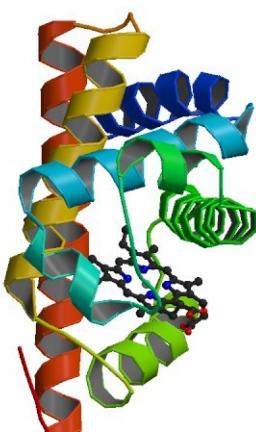
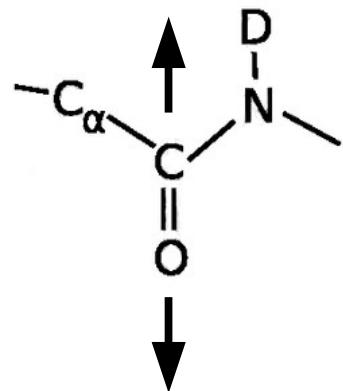
Two vibrational modes in Myoglobin (DeFlores and Tokmakoff)

# Outline

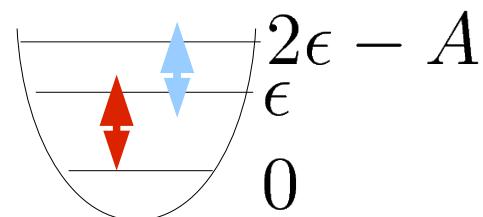
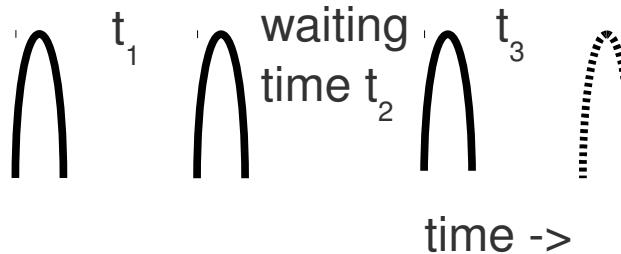
- Classical bath: vibrations in peptides
- Quantum mechanical bath: aggregates
  - Slow bath
  - Fast bath
- Quantum mechanical bath: DNA
  - Time scale of the bath

# 2DIR spectra of proteins

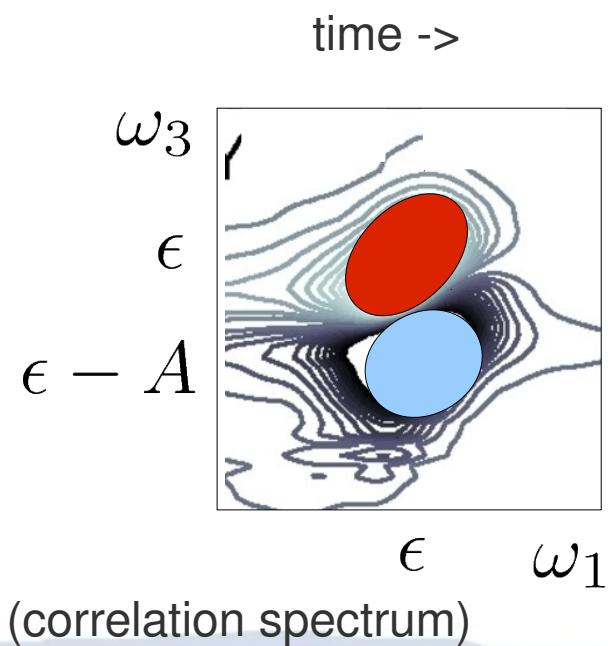
Amide I



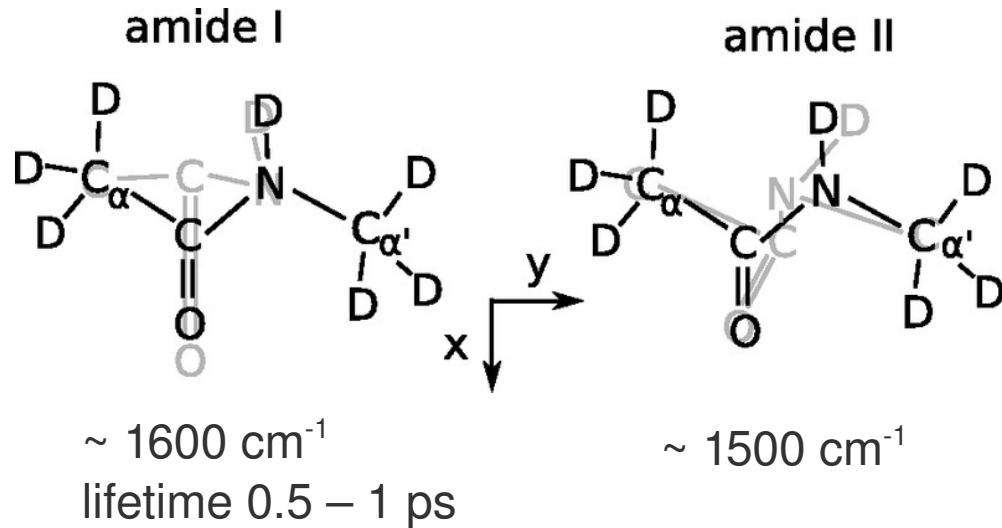
- popular in nonlinear IR experiments (large dipole)
- sensitive to the environment
- sensitive to secondary structure (Miyazawa, 1960)
- lifetime 0.5 – 1 ps



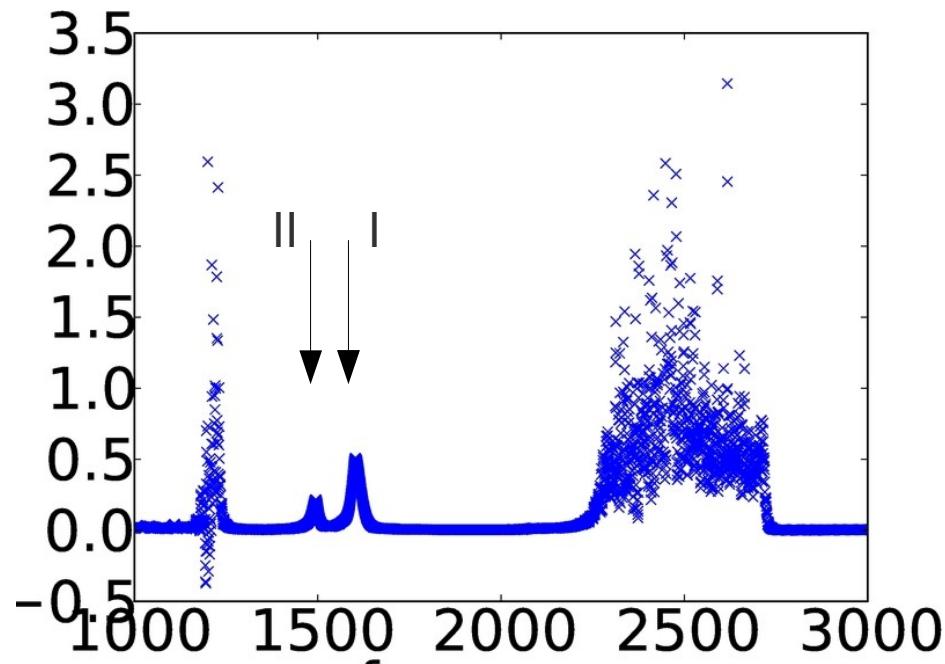
Model system: NMA  
single amide I vibration



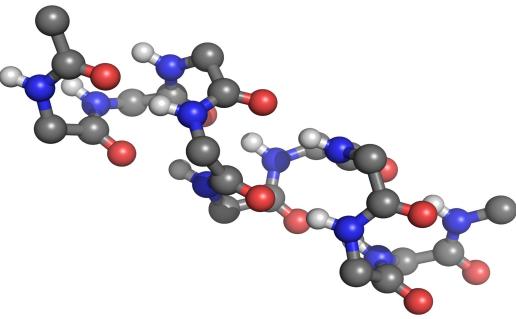
# Amide I and II modes



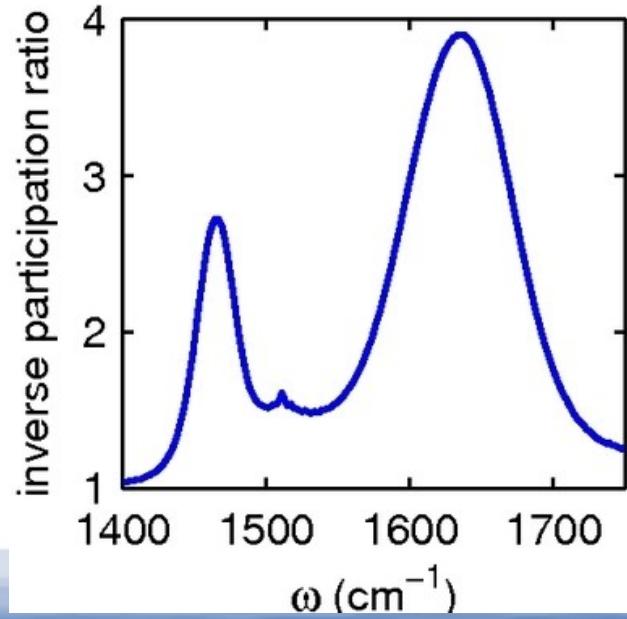
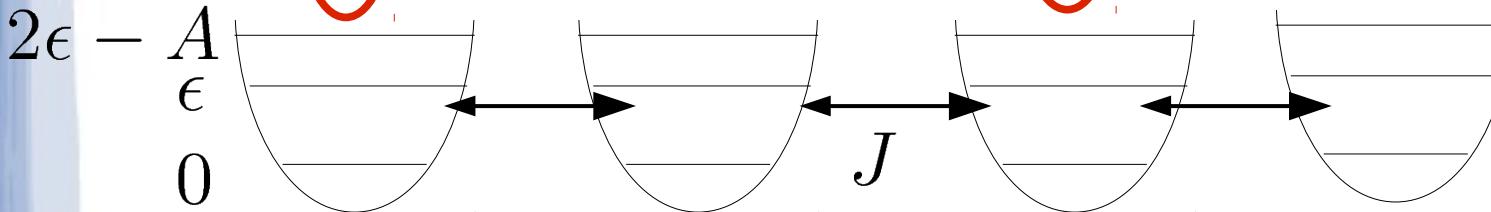
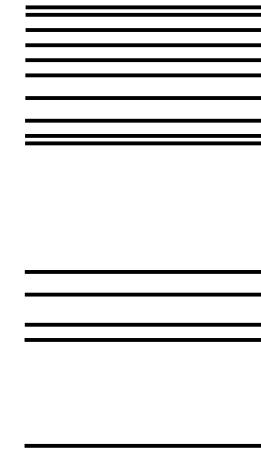
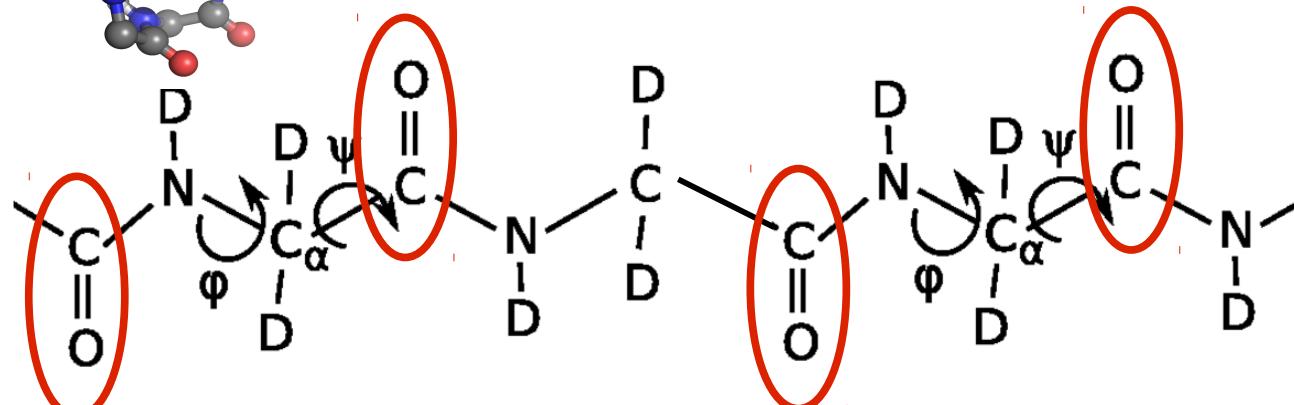
NMA-d7  
Amide I, II: independent?



Lot of detail about a small region of the spectrum



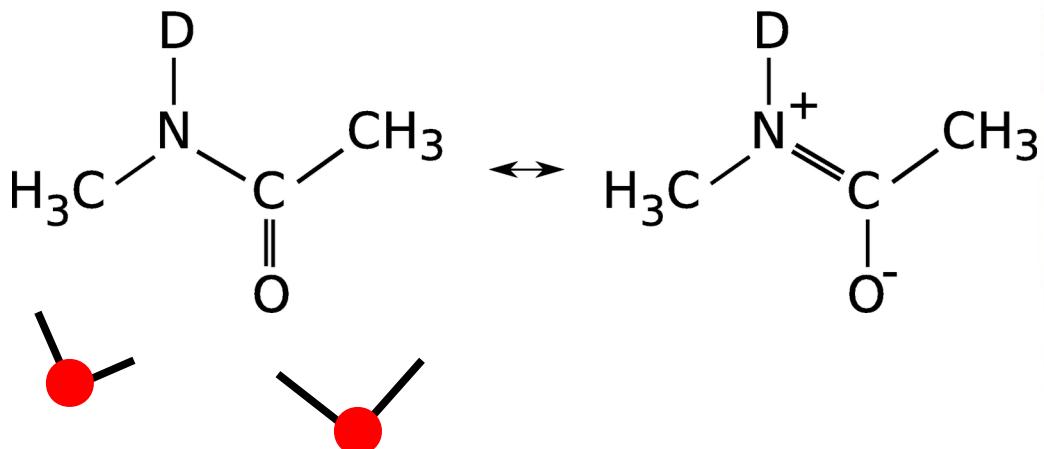
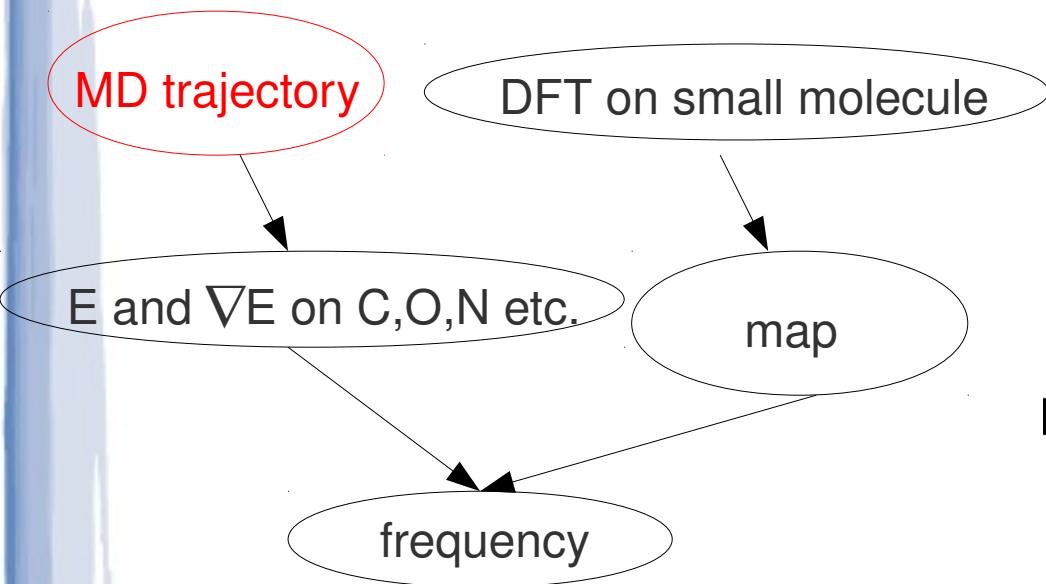
# Collective vibrations



Competition:  
coupling  $\rightarrow$  collective states  
interaction with environment  $\rightarrow$  localized states  
  
 $\rightarrow \sim 4$  amide I local modes vibrate collectively

# Modeling the environment

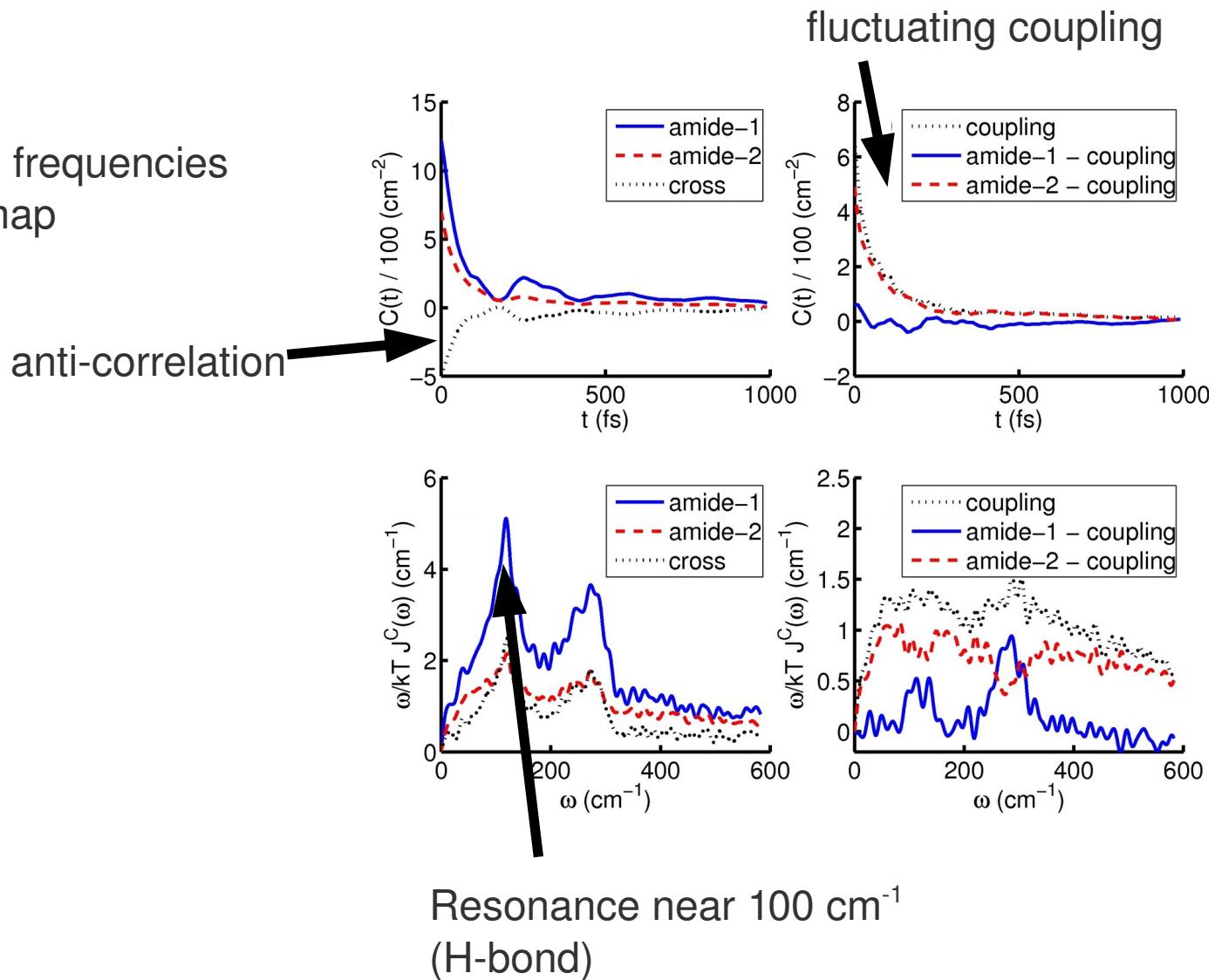
Stark shift: frequency depends on electric fields caused by the environment  
(main effect: hydrogen-bonding)



Time-dependent frequencies and amide I-II coupling

# Correlation functions (NMA-d<sub>7</sub>)

- 2 coupled modes
- water -> fluctuating frequencies
- electrostatic DFT map
- MD trajectories



# NISE

$$\frac{\hbar}{i} \frac{d}{dt} \psi(t) = H(t) \psi(t)$$

Time-dependent Hamiltonian

$$H(t) = \sum_n \epsilon_n(t) b_n^\dagger b_n + \sum_{nm} J_{nm}(t) b_n^\dagger b_m - \sum_n \frac{A_n}{2} b_n^\dagger b_n^\dagger b_n b_n$$

Fluctuations in site energies as well as couplings

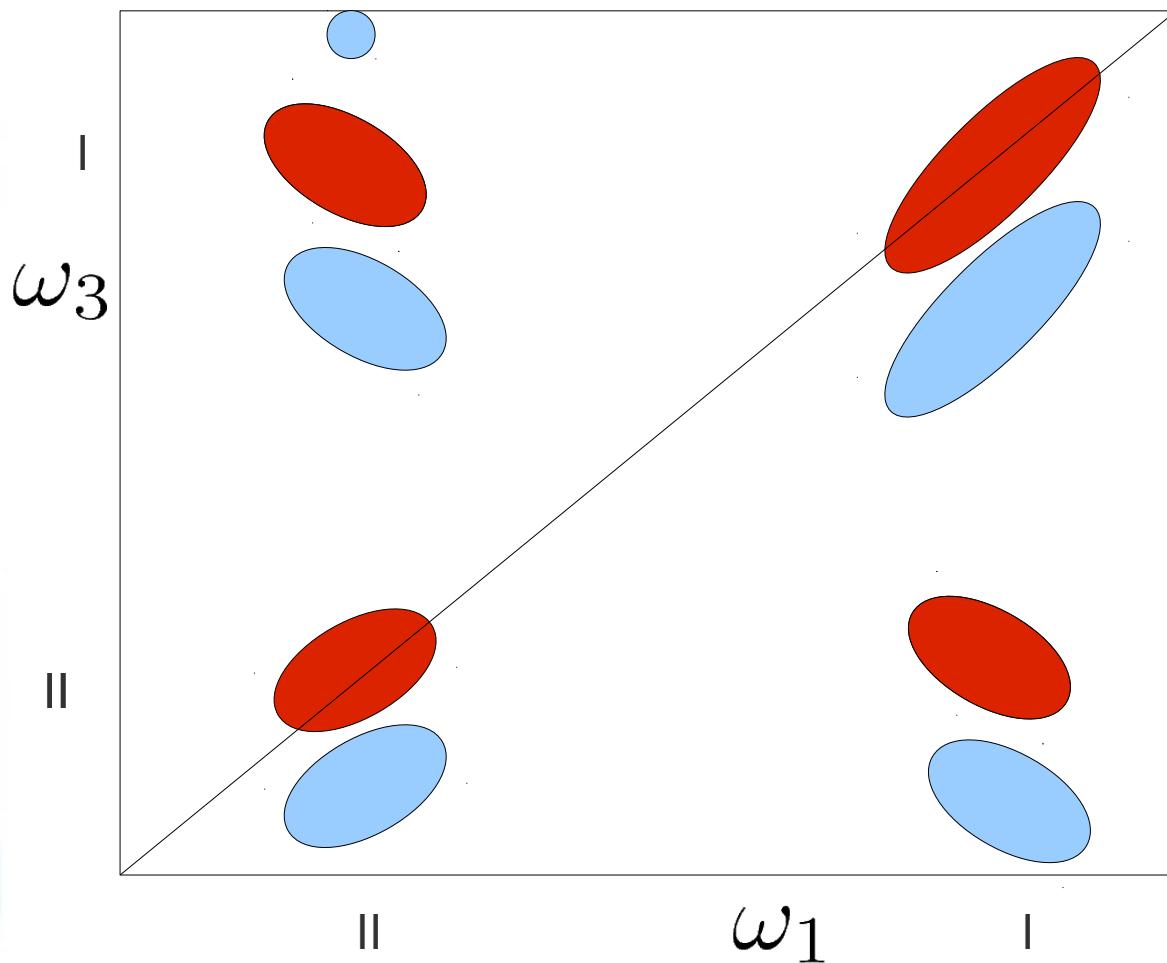
$$\psi(t + \Delta t) = e^{-iH(t)\Delta t} \psi(t) \rightarrow \text{Vibrational dynamics (nonlinear) spectra}$$

Nonadiabatic and non-Markovian dynamics

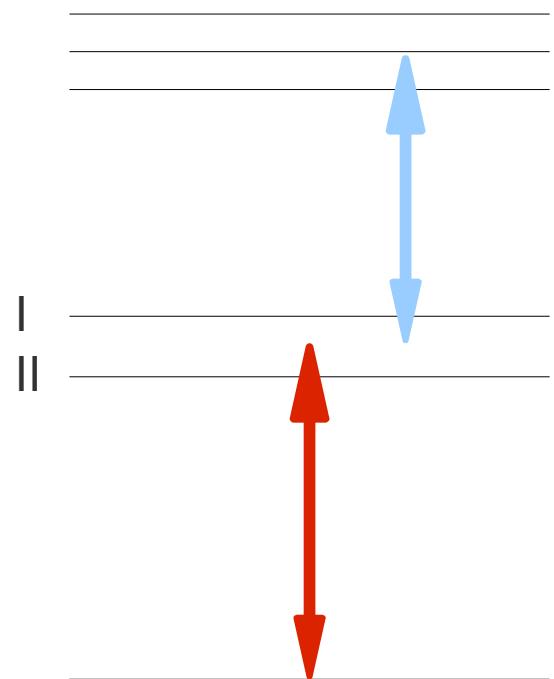
High T; OK for amide I modes, reasonable for amide I + II

# 2DIR of two coupled vibrations

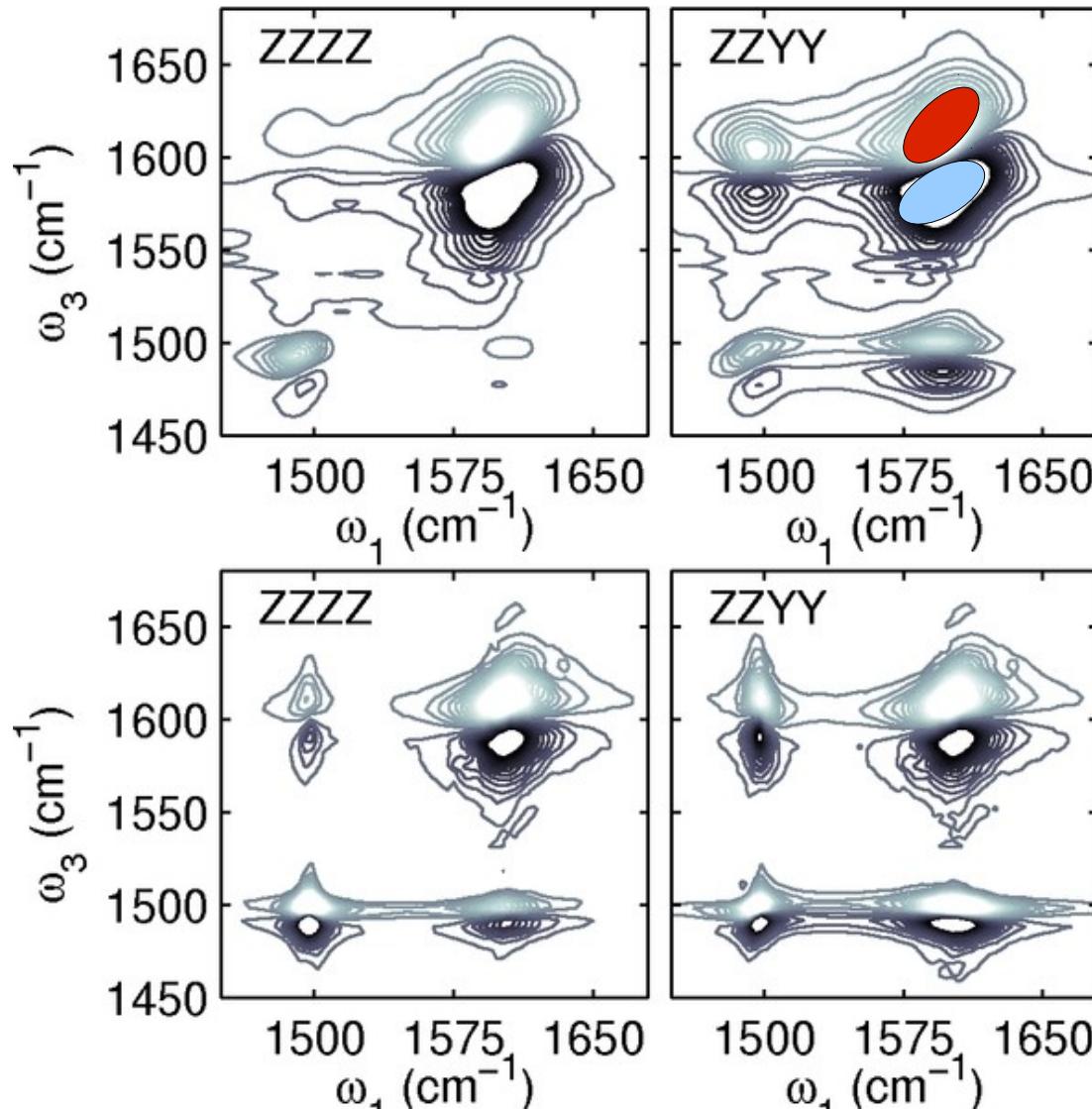
**cross peak: mode coupling**  
(anti-diagonal slant: anti-correlation)



diagonal peak. slant:  
inhomogeneous distribution



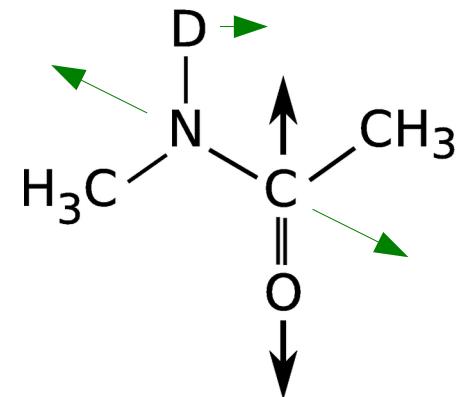
# Amide I – II cross peak in NMA-d<sub>7</sub>



**Strong cross peak in ZZYY polarization:** perpendicular transition dipoles

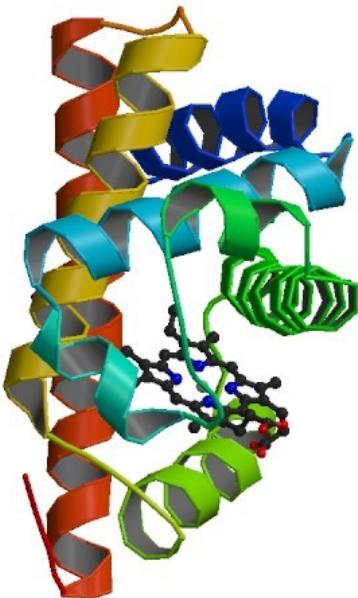
Cross peak means coupling  
=> there must be energy transport

Relaxation rate can be obtained from the cross peak intensity

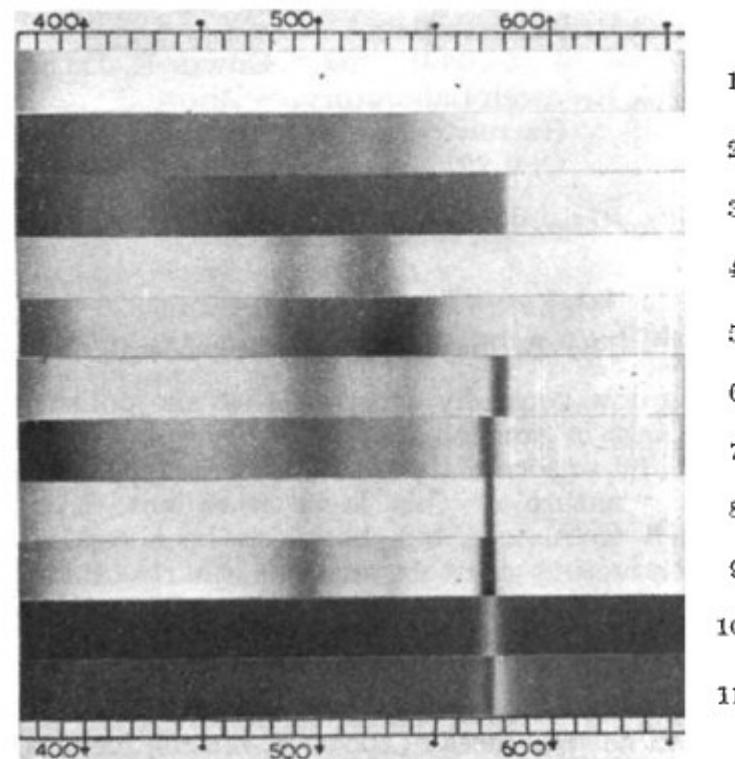
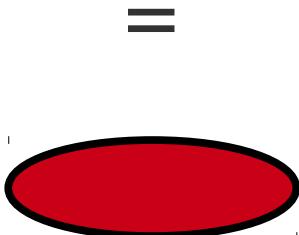
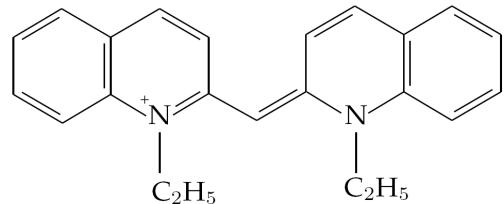


# Vibrations in peptides: classical bath

- amide vibrations in peptides: classical fluctuating environment
- no dissipation, all states equally populated in equilibrium
- only good if the system bandwidth is much smaller than the thermal energy



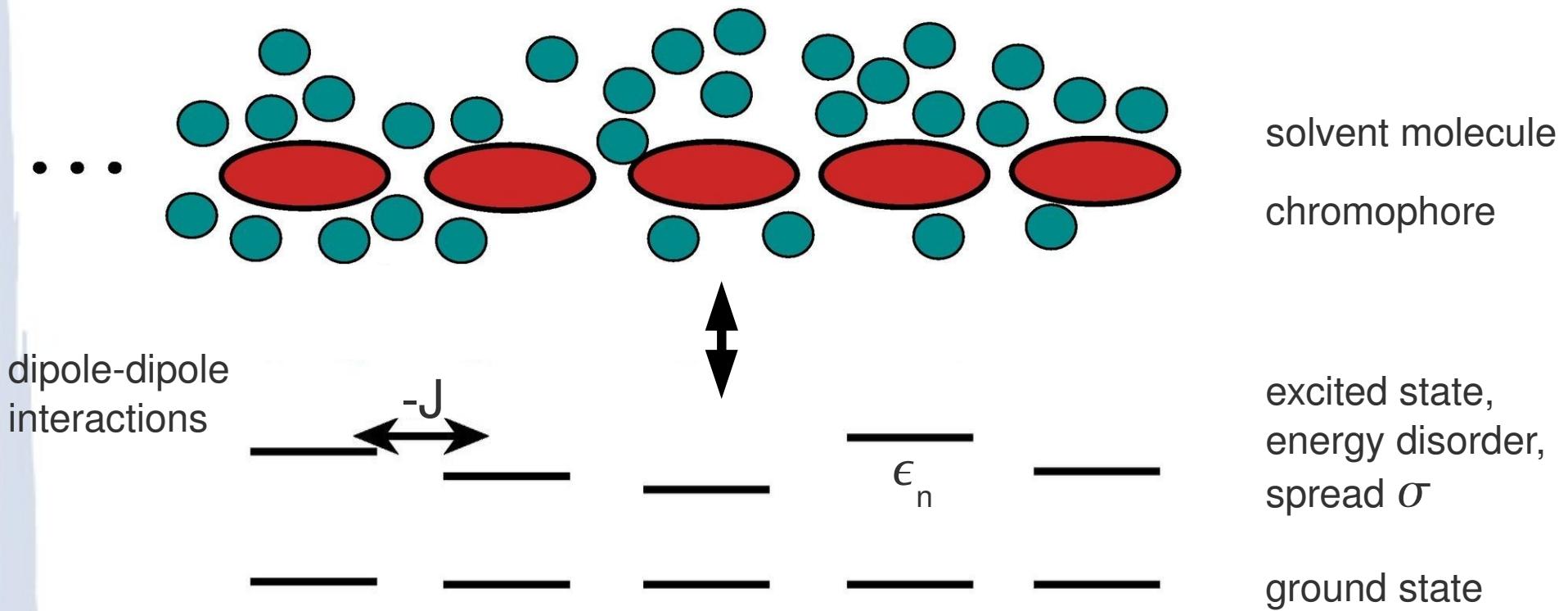
# pseudo-isocyanine (PIC)



Jolley (Nature, 1936)

- (1) Spectrum of tungsten arc lamp (containing neon).
- (2) and (3) The two principal absorptions corresponding to the  $X$  and  $Y$  vibration directions of crystals of 1:1' diethyl- $\varphi$ -cyanine chloride.
- (4) Absorption spectrum of the dye in methyl alcohol.
- (5) Absorption spectrum of the dye in nitrobenzene.
- (6) Absorption spectrum of a molecular suspension of the dye in toluene.
- (7) Absorption spectrum of the dye molecularly dispersed in benzophenone crystals.
- (8) and (9) Absorption spectra of two different concentrations of the dye in sodium chloride solution.
- (10) Fluorescence spectrum of the dye in sodium chloride solution.
- (11) Fluorescence spectrum of the dye in sodium chloride solution. In this case the fluorescent light has been made to traverse a small thickness of unilluminated solution, so that the absorption band is seen superimposed on the fluorescence band. It will be noticed that the two do not exactly correspond.

# Model at a low temperature (< 5 K)



$$H = \sum_n \epsilon_n c_n^\dagger c_n + \sum_{m \neq n} \frac{-J}{|n - m|^3} c_n^\dagger c_m.$$

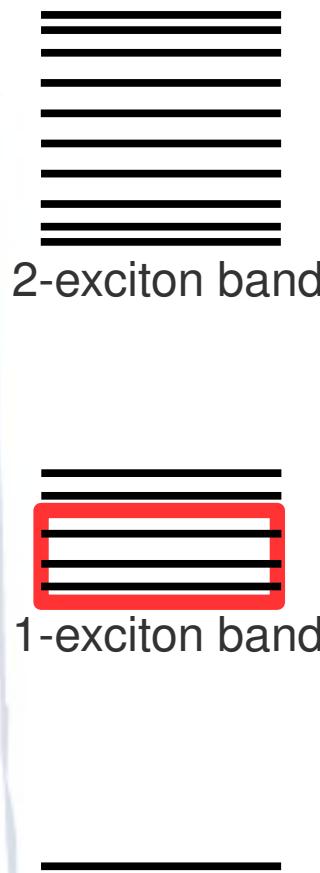
# Excitons: homogeneous aggregates



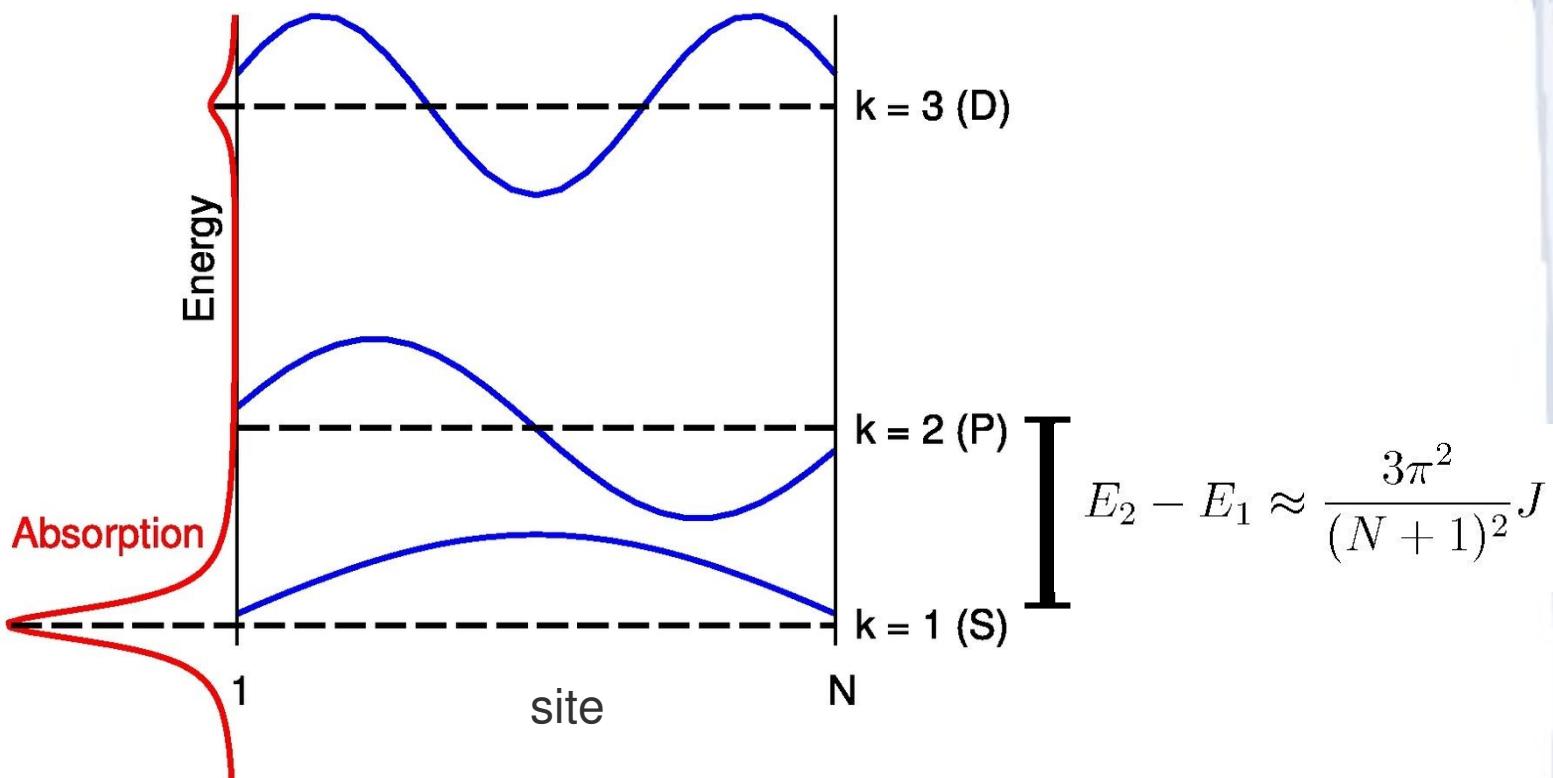
Simplifications:

- no disorder

- only nearest neighbour interactions



Noninteracting fermions, 2-exciton states as anti-symmetric products

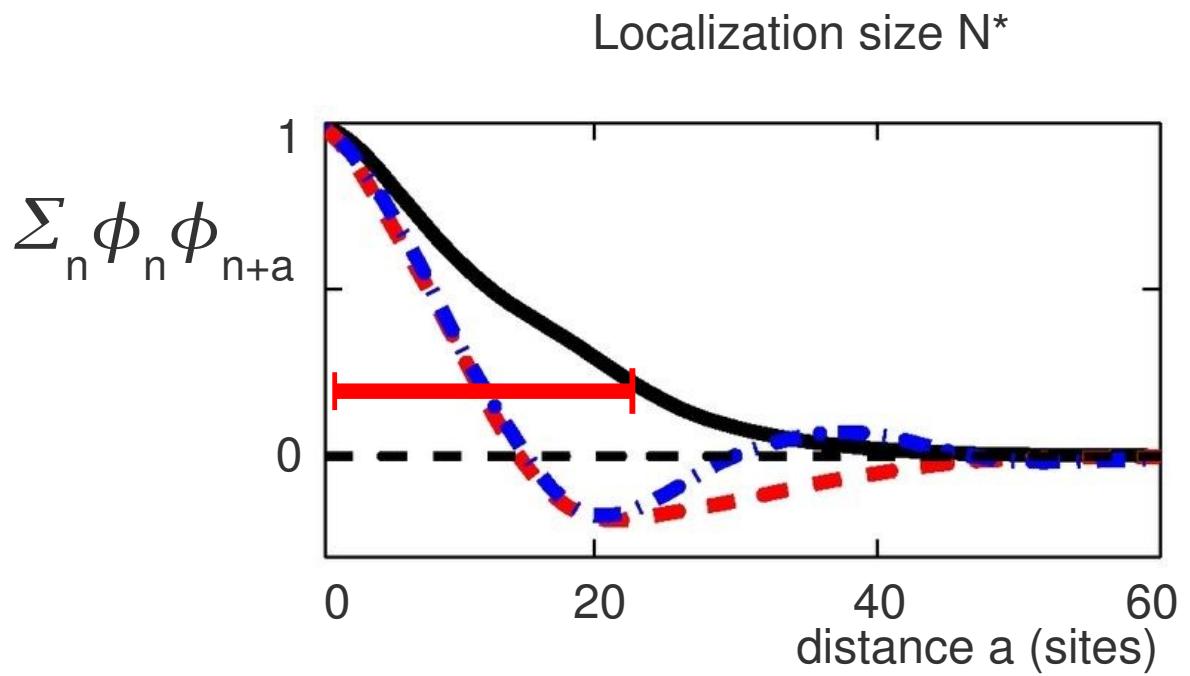
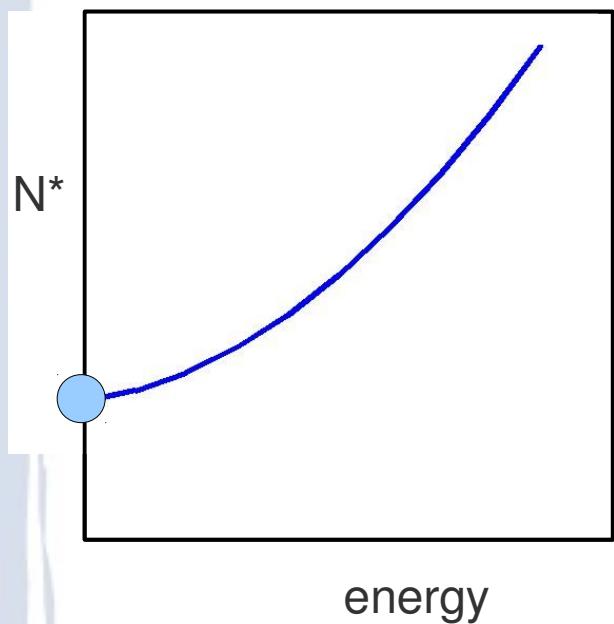
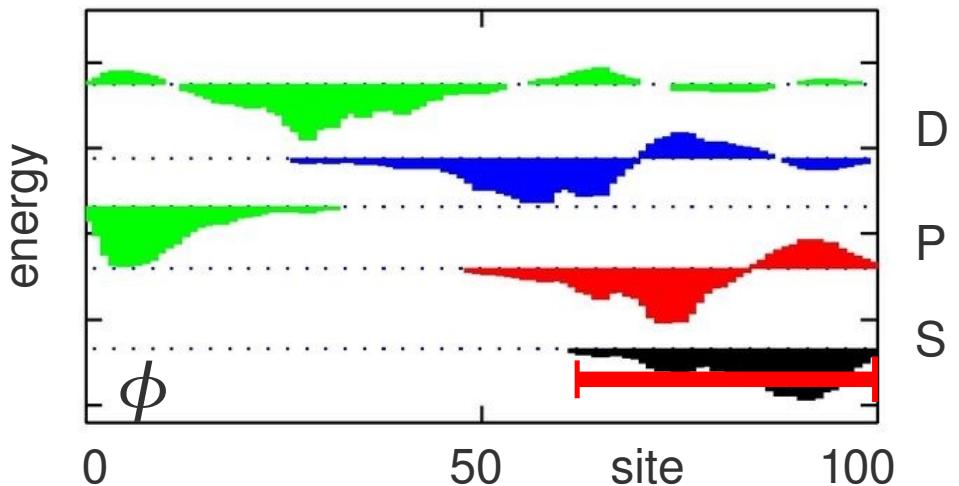


# Excitons: disordered aggregate

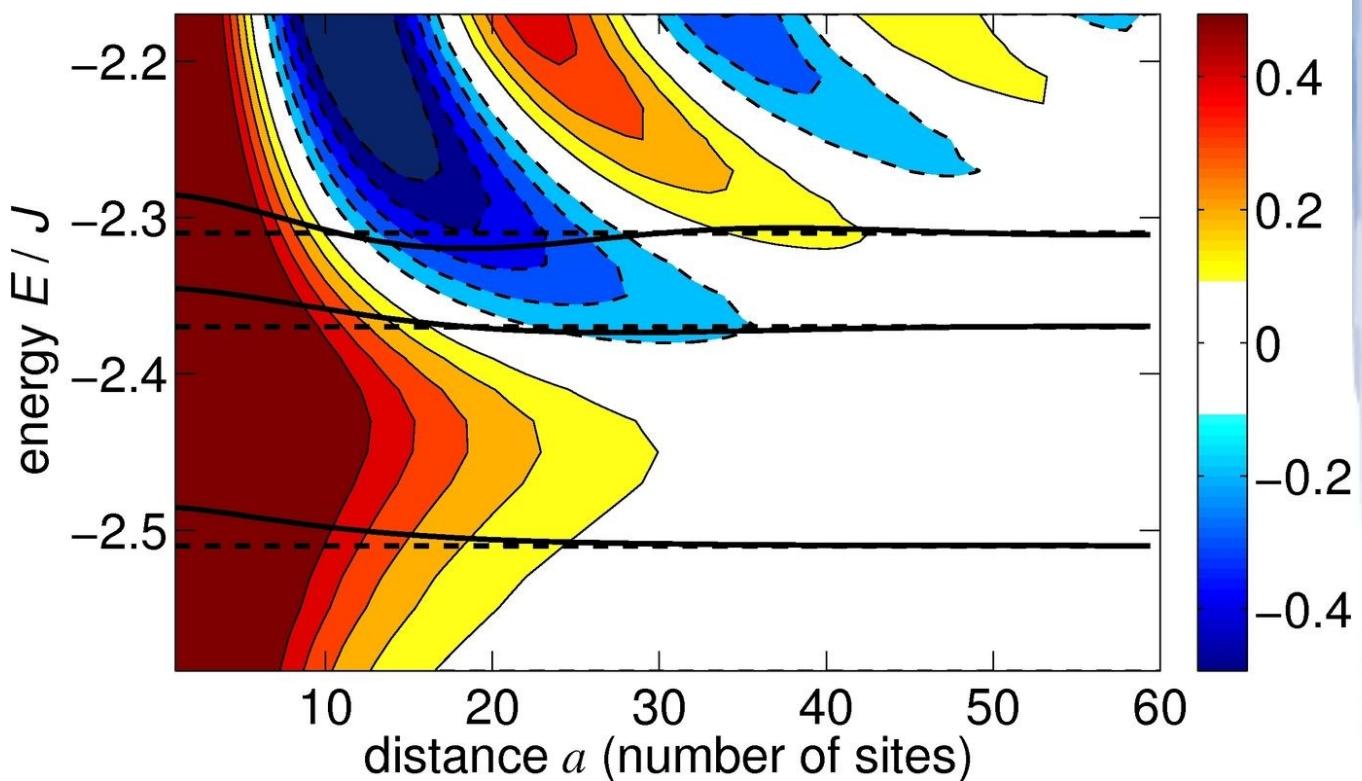
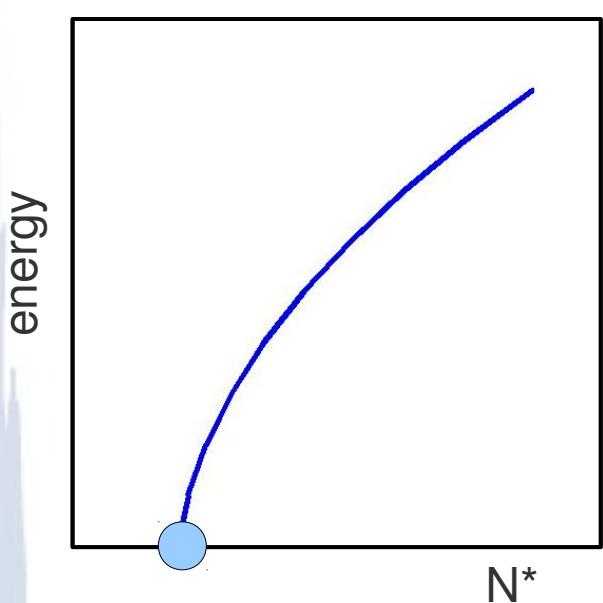
Full model

- site energy disorder
- long range interactions

numerical 1- and 2-exciton states

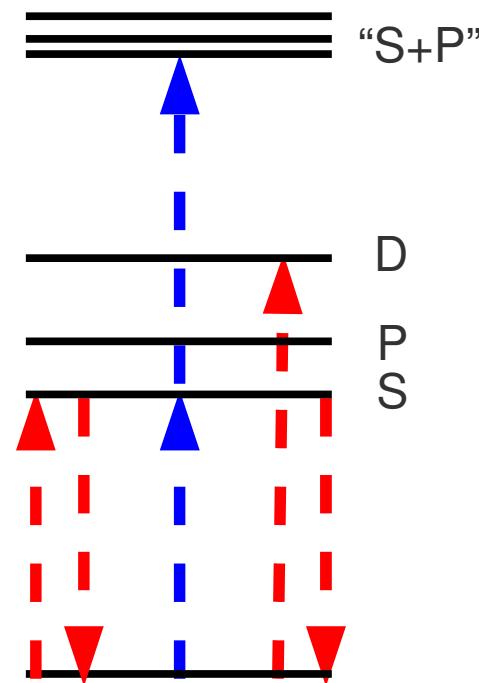
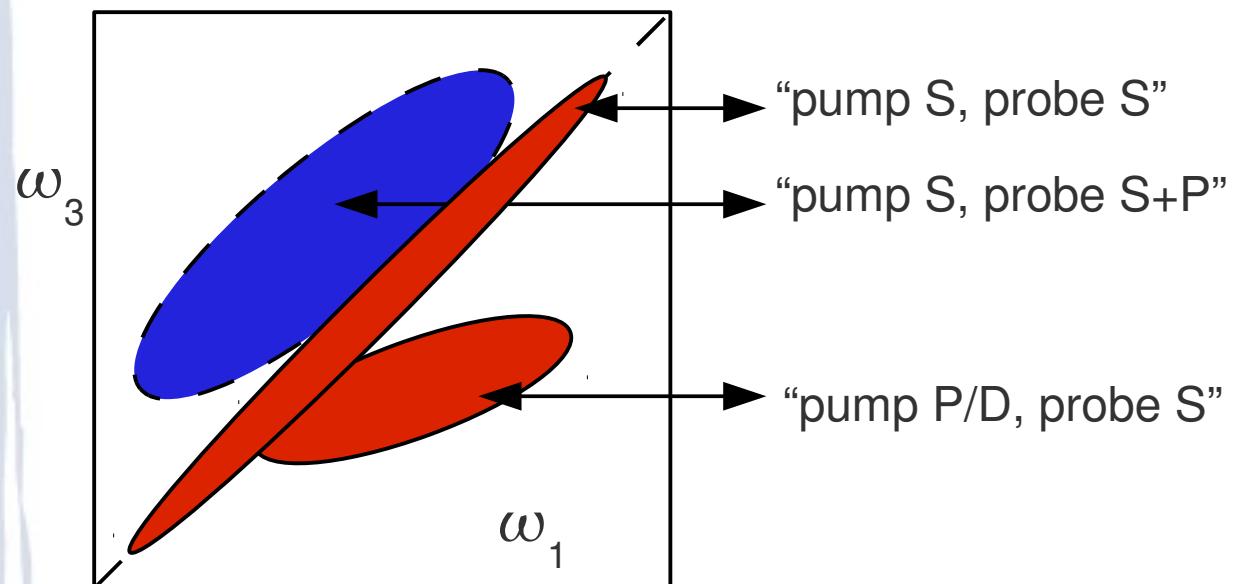
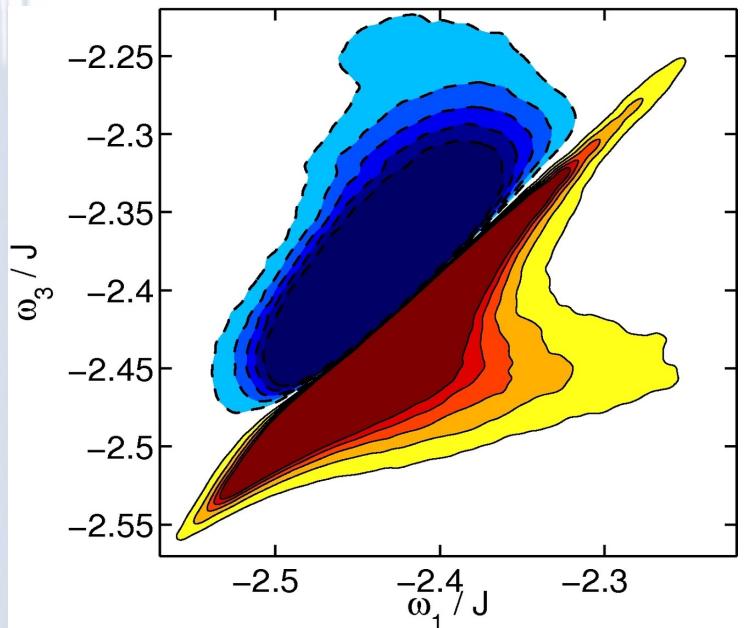


# Localization: position correlation function



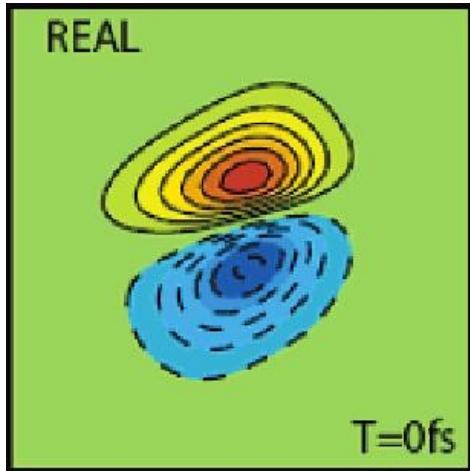
$$C(a; E) = \frac{\langle \sum_k \sum_n \phi_{k,n} \phi_{k,n+a} \delta(E - E_k) \rangle}{\langle \sum_k \delta(E - E_k) \rangle}$$

# 2D spectrum: zero waiting time

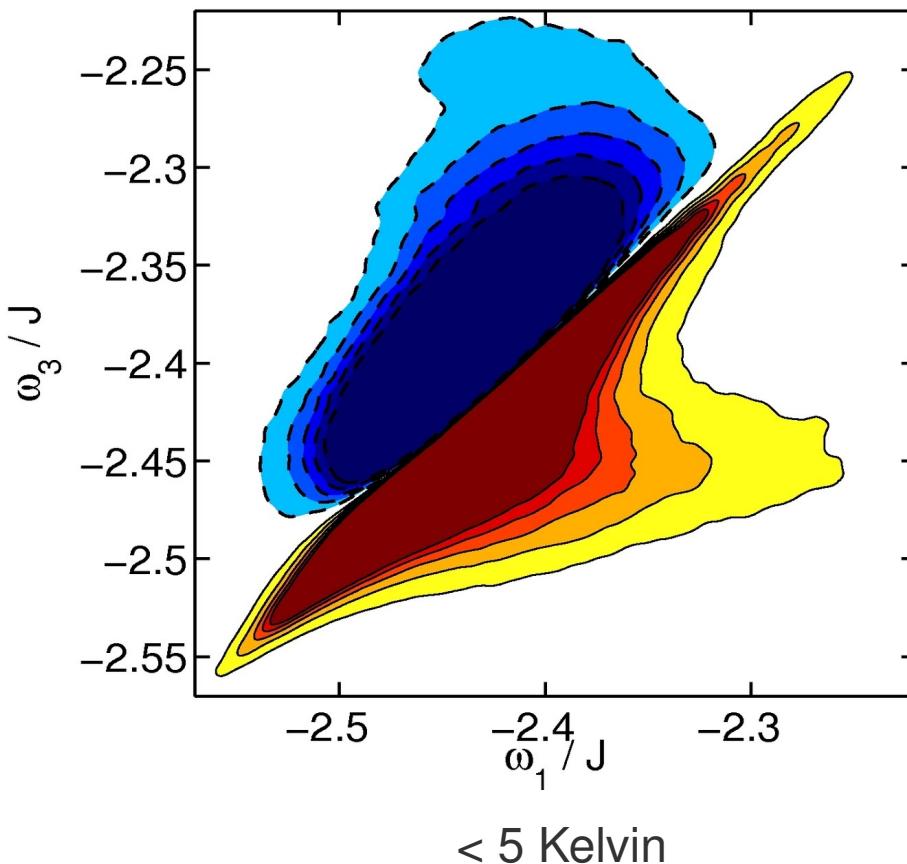


exciton states in  
localization segment

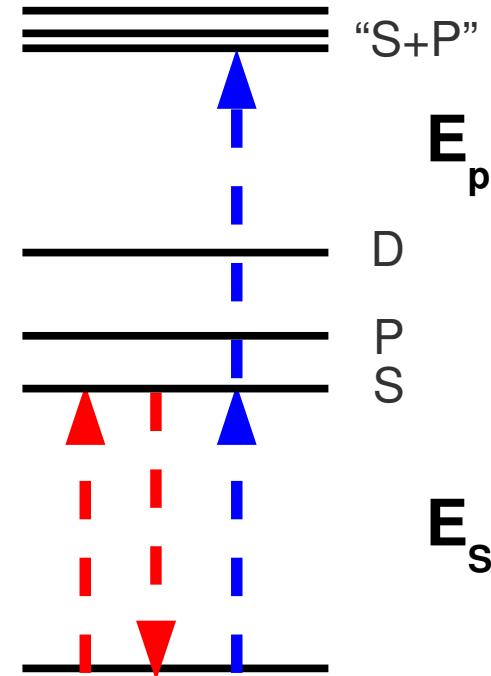
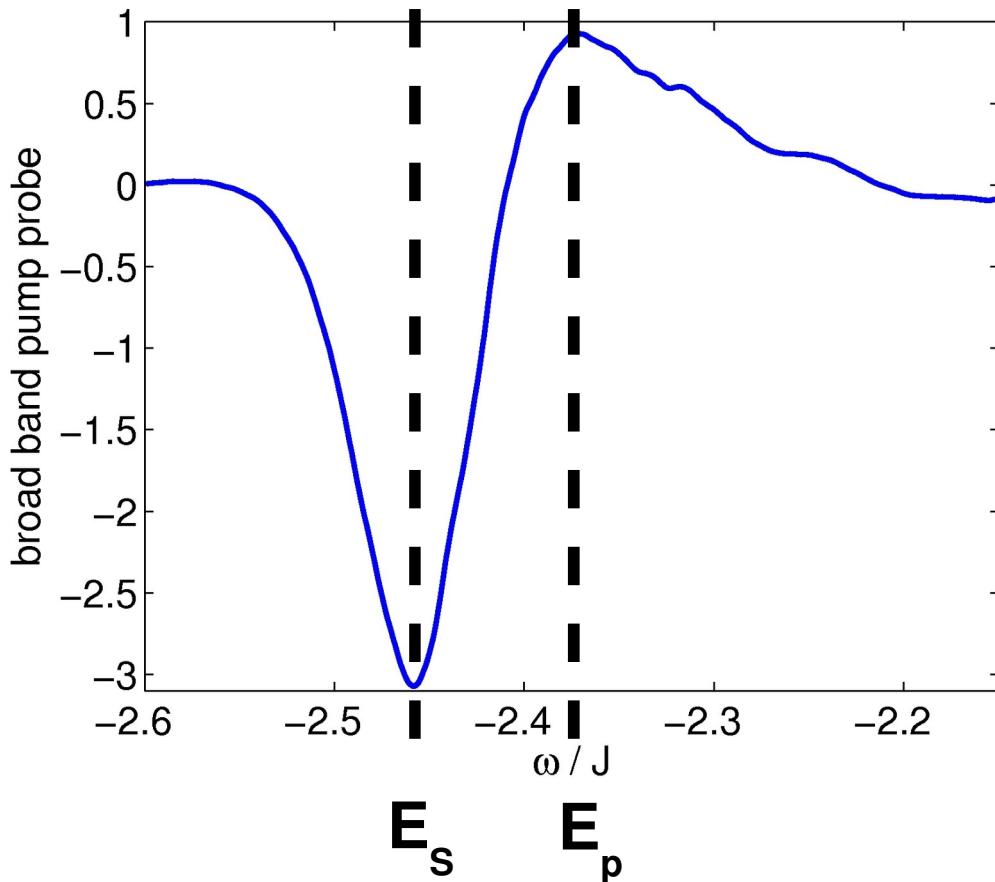
# Low temperature cross peak



room temperature  
(I. Stiopkin et al.  
JPCB **110**, 20032  
(2006))

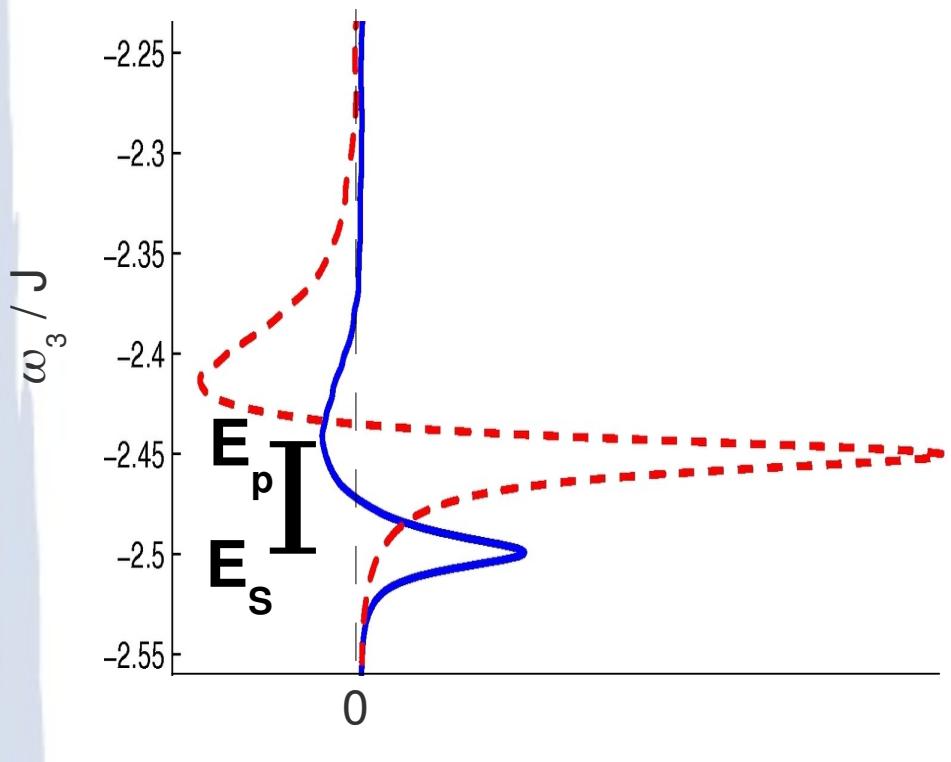


# Average localization length

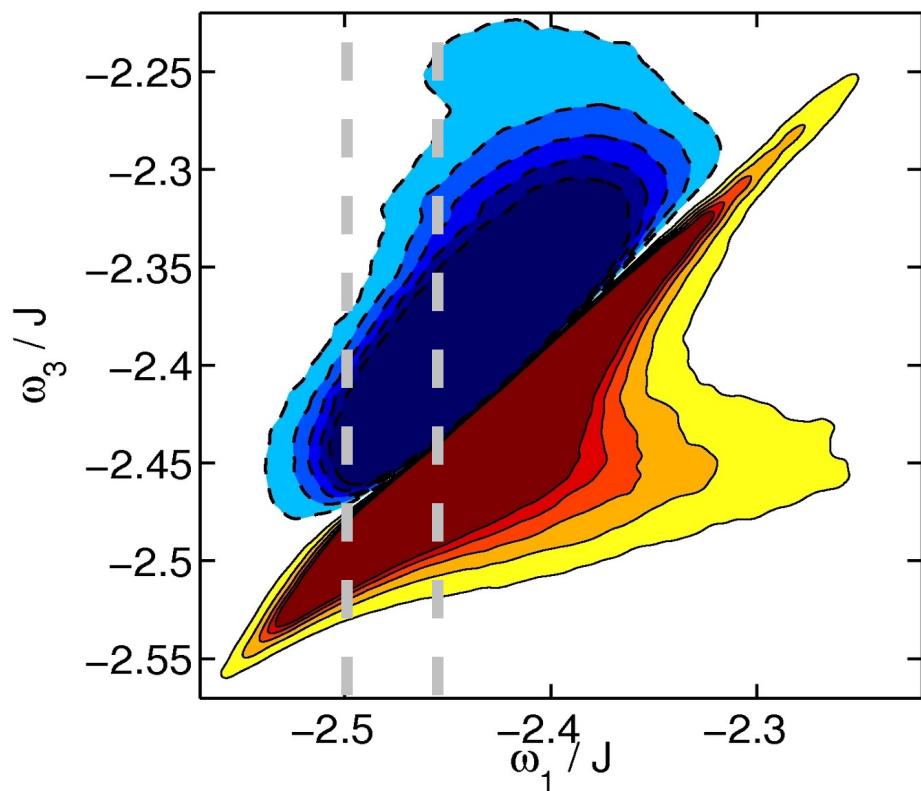


$$E_P - E_S \approx \frac{3\pi^2}{(N^* + 1)^2} J$$

# Extraction of localization length

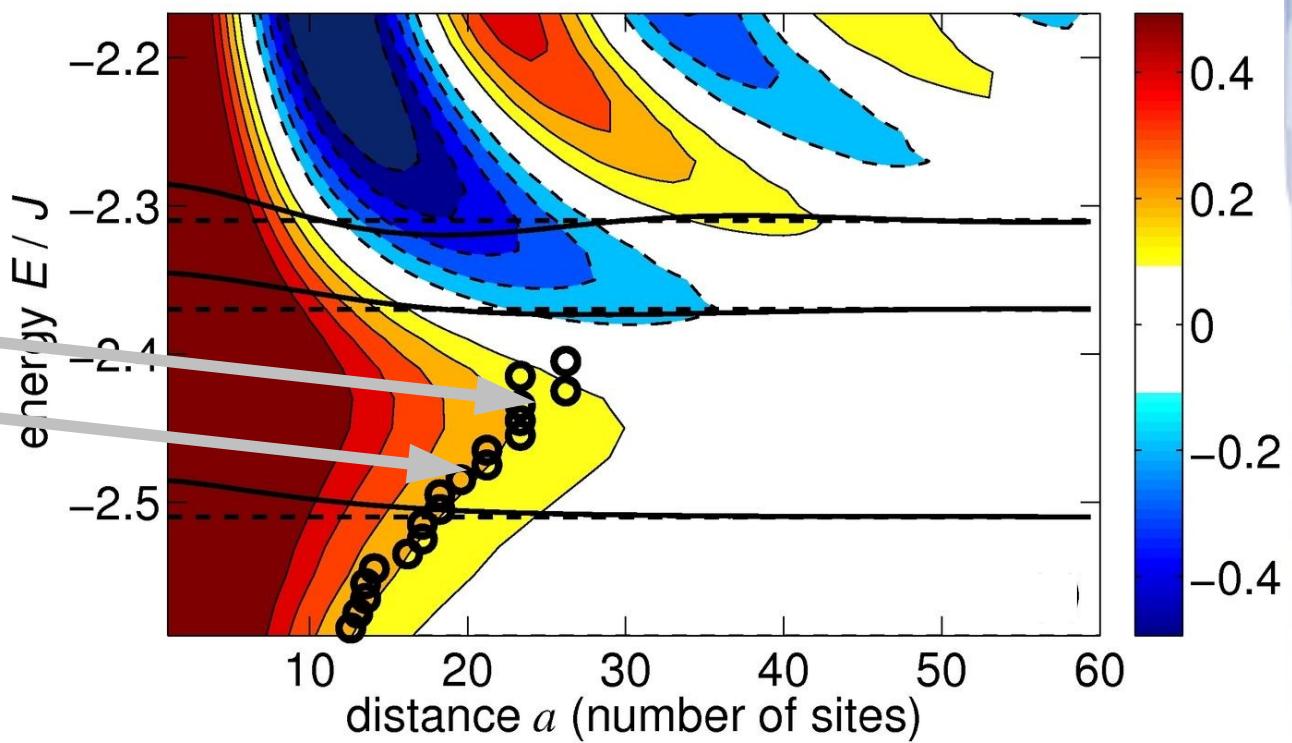
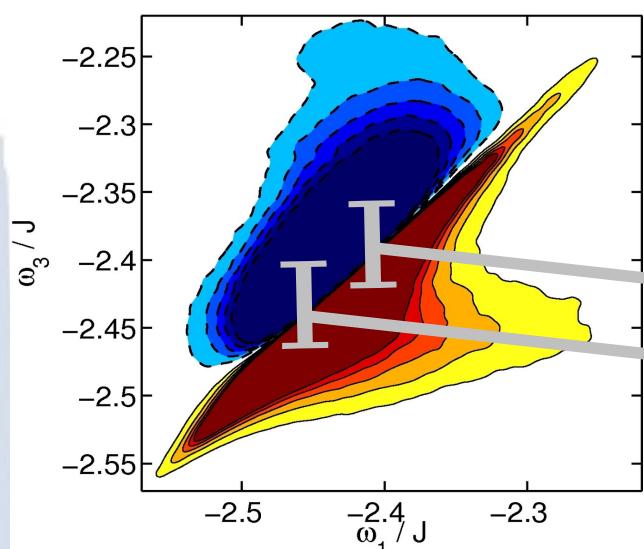


$$E_p - E_s \approx \frac{3\pi^2}{(N^* + 1)^2} J$$

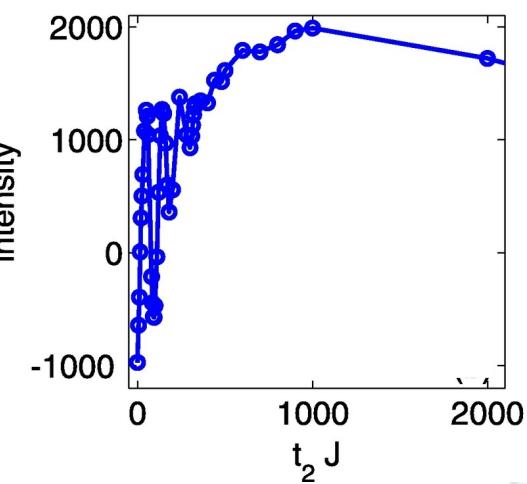
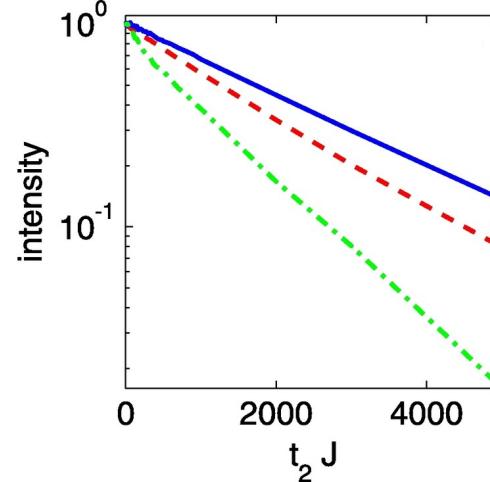
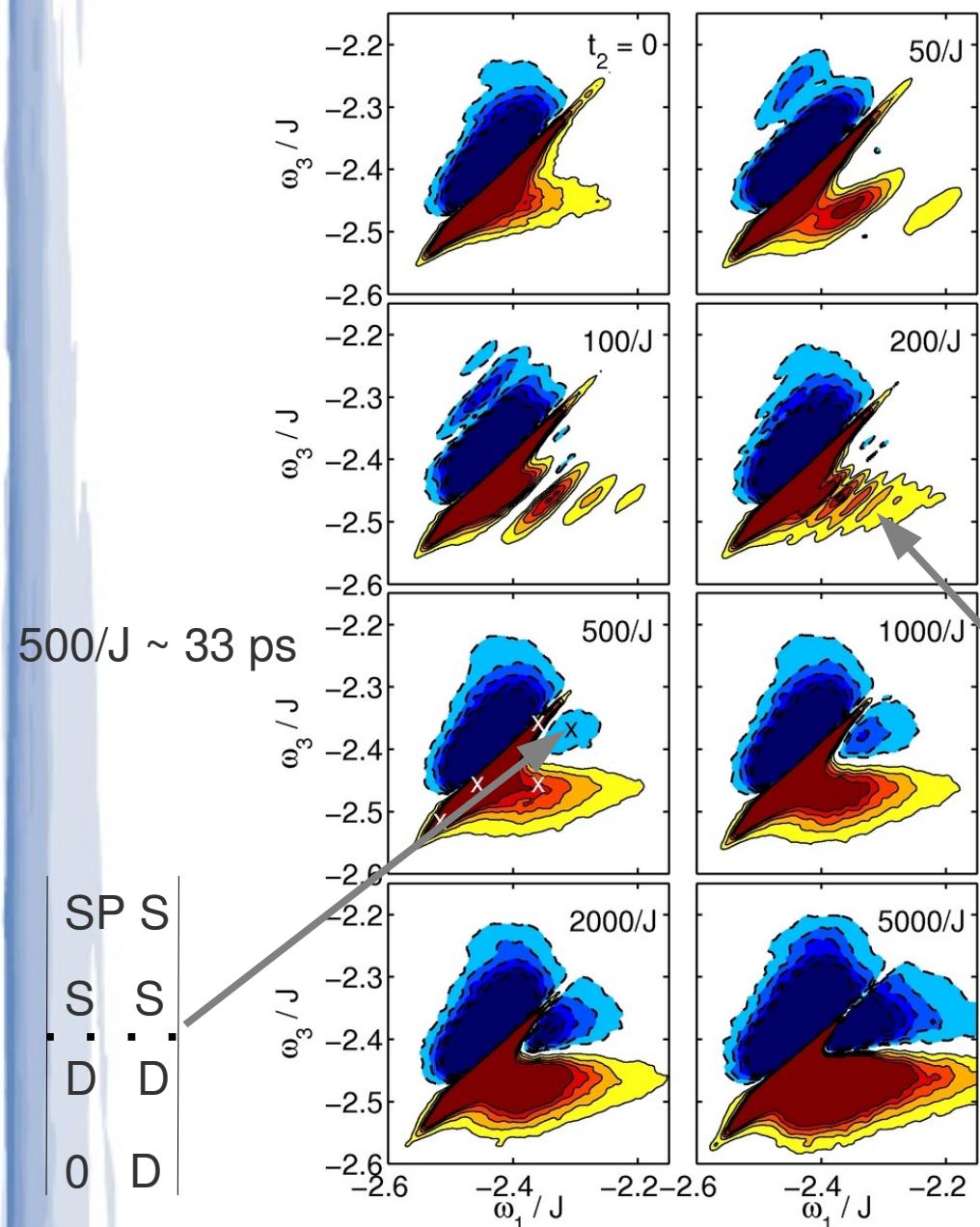


Cross section at fixed  $\omega_1$   
“excites segments with  $E_s = \omega_1$ ”

# Extracted localization length



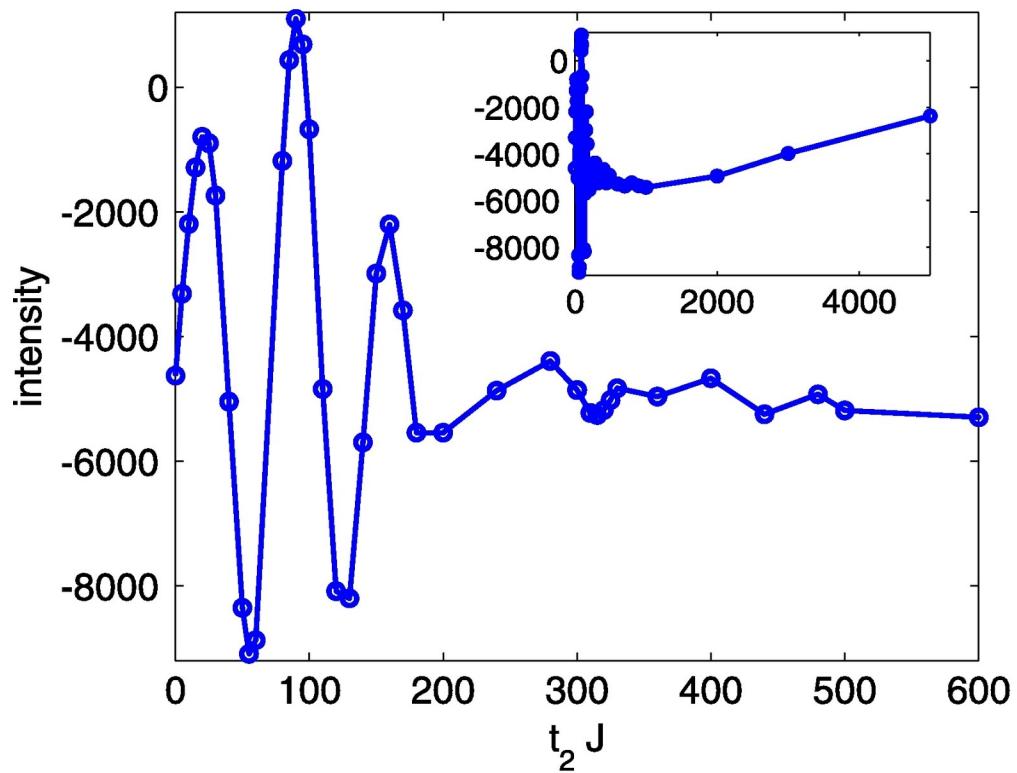
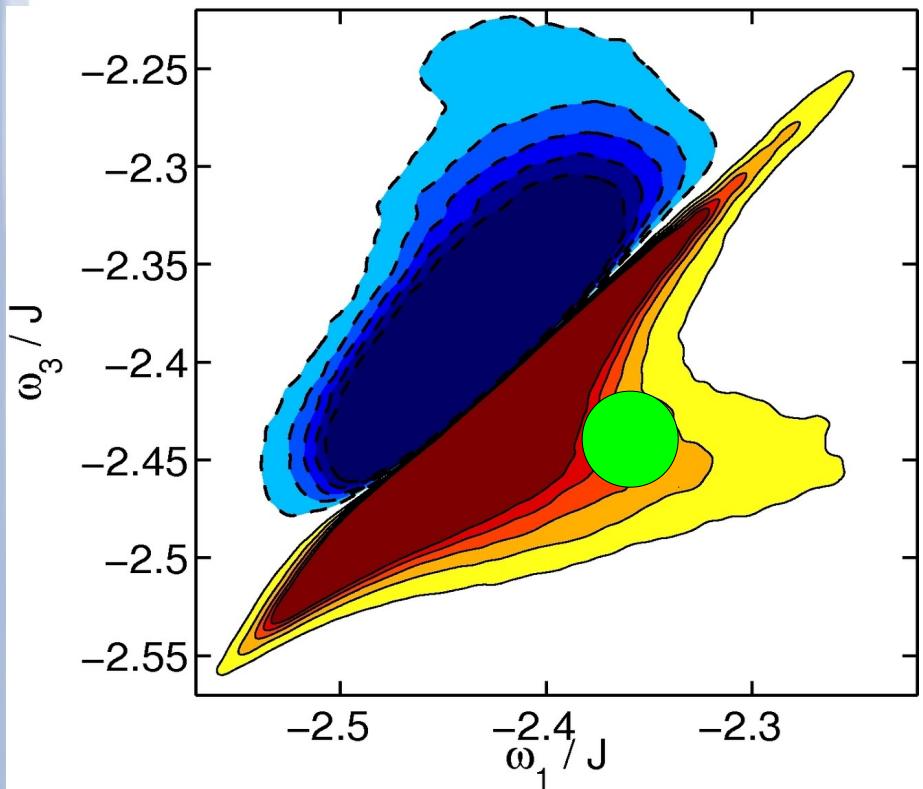
# Dynamics: 2D spectra



S  
P  
0  
P

$$e^{-i(E_S - E_P)t_2} = e^{-i(\omega_3 - \omega_1)t_2}$$

# Coherent oscillations



“long-lived coherent oscillations”

# Aggregates – static bath

- Molecular aggregates at low temperature: static bath (no dissipation)
- Response from sum over states
- Dynamics due to remaining fast degrees of freedom
- Fast bath modes described using Redfield theory

# Quantum and slow bath

$$H = H_S + H_B + H_{SB}$$

$$H_B = \sum_{\alpha} \left( p_{\alpha}^2 / 2m_{\alpha} + m_{\alpha}\omega_{\alpha}^2 x_{\alpha}^2 / 2 \right)$$

$$H_{SB} = - \sum_{n=1}^N \sum_{i \in \{A,T\}} \sum_{\alpha} g_{ni,\alpha} V_{ni} x_{\alpha} \quad V_{ni} = c_{ni}^{\dagger} c_{ni}$$

↙ bath time scale

$$\mathcal{J}_{ni,mj}(\omega) = \sum_{\alpha} \frac{g_{ni,\alpha} g_{mj,\alpha}}{2m_{\alpha}\omega_{\alpha}} \delta(\omega - \omega_{\alpha}) = 2\lambda_{ni,mj} \gamma_{ni,mj} \frac{\omega \gamma_{ni,mj}}{\gamma_{ni,mj}^2 + \omega^2}$$

- Common approximations:
  - Bath dynamics much faster than system dynamics
  - Factorized initial state  $\rho_{\text{system+bath}}(t=0) = \rho_{\text{system}}(t=0) \otimes \rho_{\text{bath}}(t=0)$
- Break down for a slow bath and low temperature
- To treat a slow quantum bath: hierarchy of equations of motion method  
(Y. Tanimura, J. Phys. Soc. Jpn. 2006)

# Hierarchy of equations of motion

- In the high temperature limit

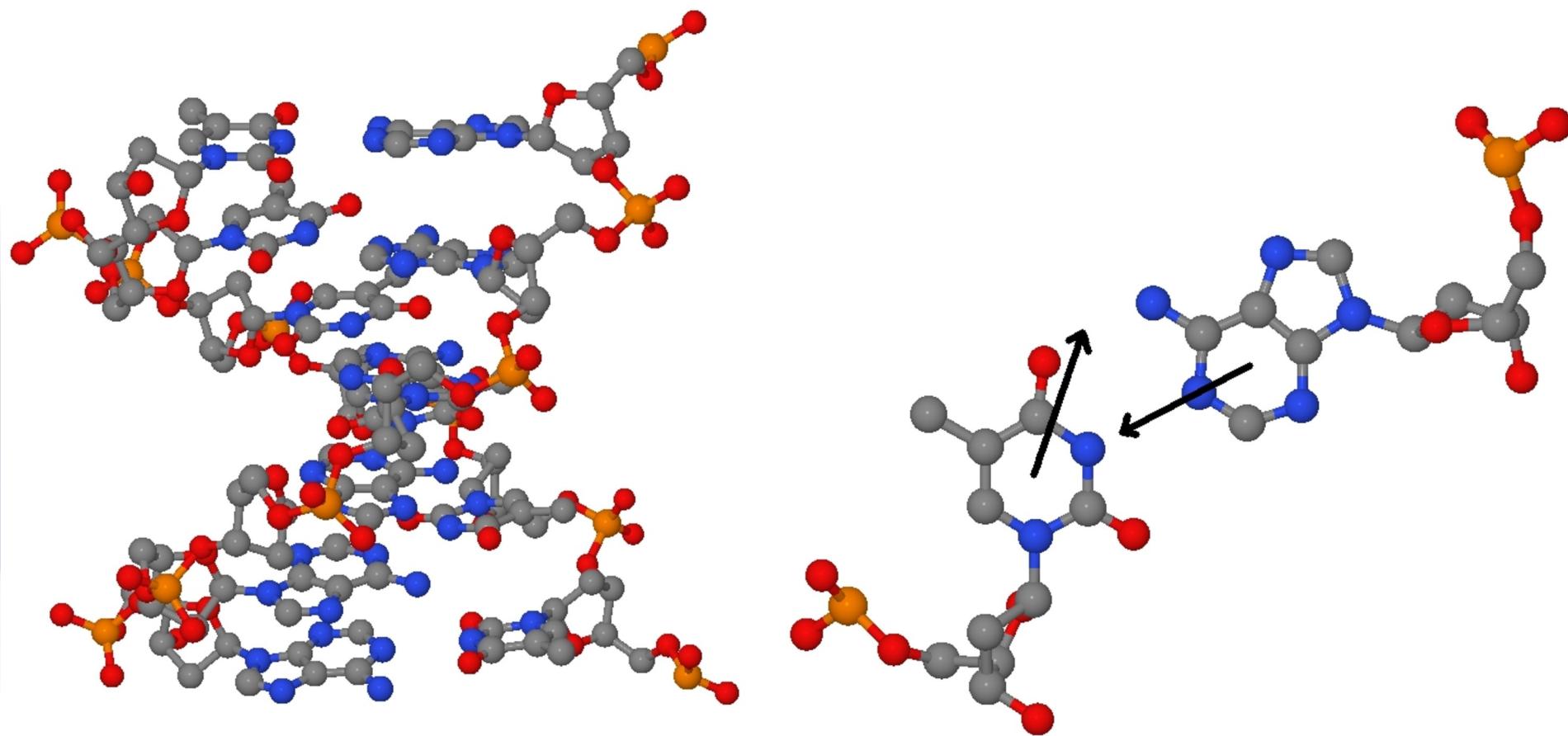
$$L_{ni,mj}(t) = c_{ni,mj} e^{-\gamma_{ni,mj}|t|} \quad c_{ni,mj} = \lambda_{ni,mj} \left( -i\gamma_{ni,mj} + \frac{2}{\beta} \right)$$

- Equations of motion

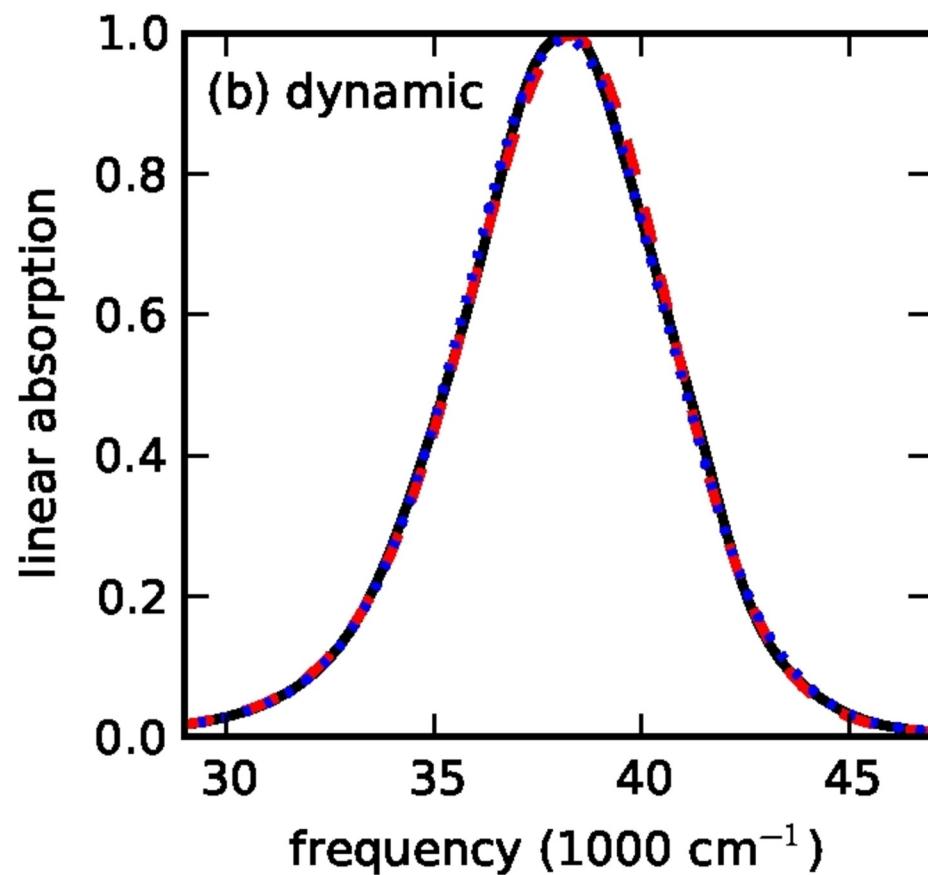
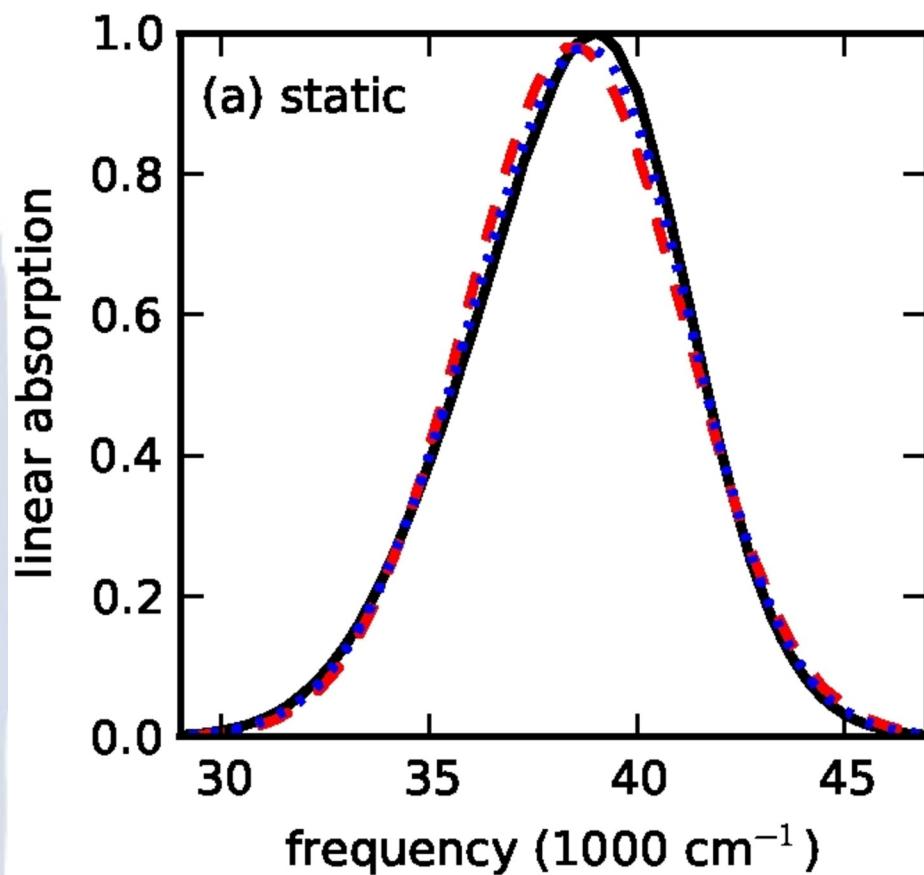
$$\begin{aligned} \dot{\rho}^{\{n\}}(t) = & - \left( iH_S^\times + \sum_s n_s \gamma \right) \rho^{\{n\}}(t) \\ & - i \sum_s n_s \left( c V_s \rho^{n_s^-} - c^* \rho^{n_s^-} V_s \right) - i \sum_s V_s^\times \rho^{n_s^+} \end{aligned}$$

dissipation

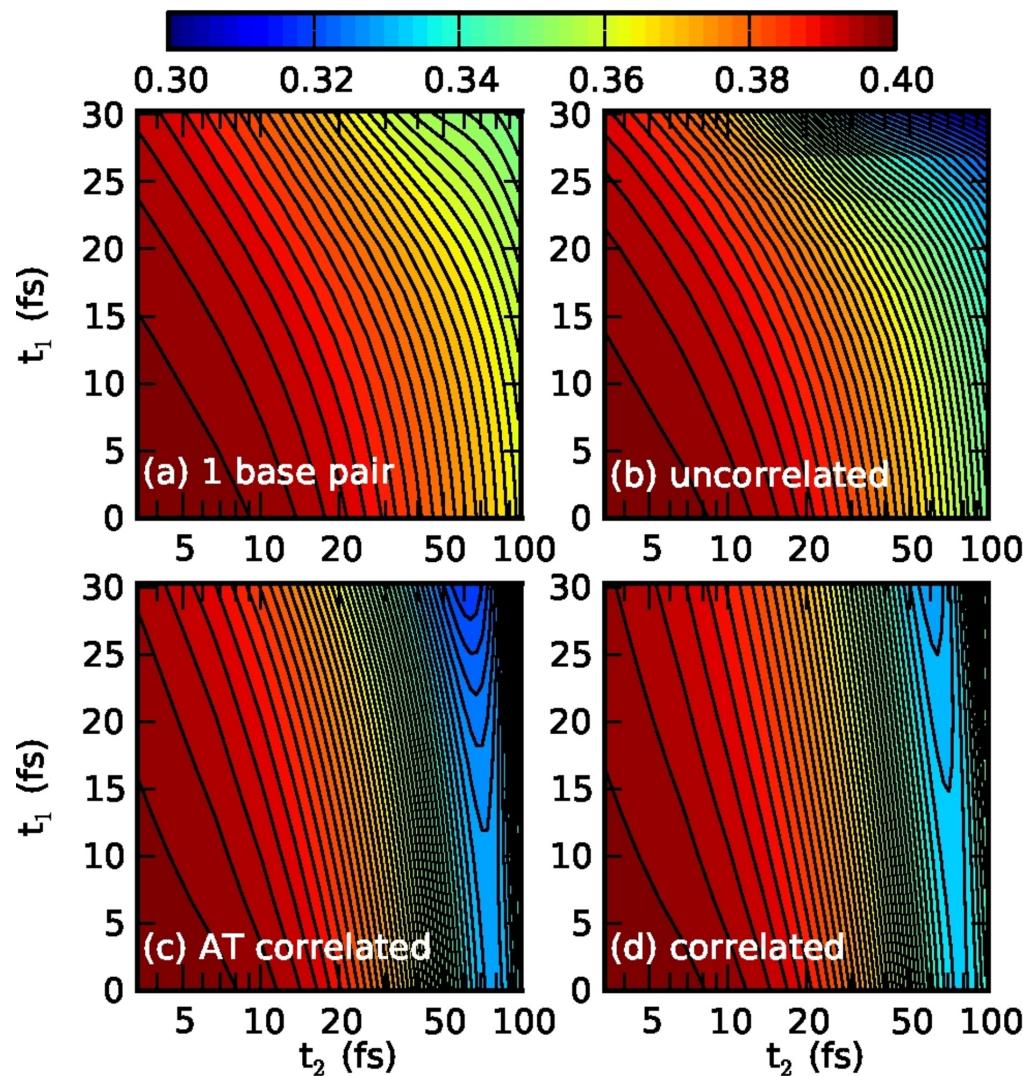
# Application: excitons in B-DNA



# Linear response



# Two-time anisotropy



# Conclusions

- Dynamics of coupled oscillators
- Environment
  - Vibrations in peptides: classical environment
  - Low temperature excitons: static environment
  - Room temperature: hierarchy of EOM
- Calculate nonlinear optical observables